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(54) Title: ORALLY ACTIVE, ANTIMALARIAL, ANTICANCER, ARTEMISININ-DERIVED TRIOXANE DIMERS WITH HIGH SELECTIVITY, STABILITY AND EFFICACY AND METHODS OF MAKING THE SAME

(57) Abstract: In only two steps and in 65% overall yield, natural trioxane artemisinin (I) was converted on gram scale into C-10-carba trioxane dimer (3). This new, very stable dimer was then transformed easily in one additional step into four different dimers (4-7). Alcohol and diol dimers (4 and 5) and ketone dimer (7) are 10 times more antimalarially potent *in vitro* than artemisinin (I), and alcohol and diol dimers (4 and 5) are strongly inhibitory but not cytotoxic toward several human cancer cell lines. Water-soluble carboxylic acid derivatives (8a-10c and 12) were easily prepared from dimers (4-6); they are thermally stable even at 60°C for 24 hours, are more orally efficacious as antimalarials than either artelinic acid or sodium artesunate, and have potent and selective anticancer activities. Further derivitization of the alcohol dimers (4 and 17), diol dimer (5) and ketone (7) has produced a number of analogs also antimalarially active *in vitro* at sub-nanomolar concentrations (most notably: pyridine N-oxides (13, 15, 18, 23, 24 and 25), phosphoric acid triesters (26 and 27), sulfonamide (40) and cyclic carbonate (41)). In addition, dimers (13 and 19) are more efficacious (when administered both orally and i.v.) and less toxic (when administered intraperitoneally to mice as a single dose) than clinically-used sodium artesunate, thereby giving them a better antimalarial therapeutic index than sodium artesunate.

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**ORALLY ACTIVE, ANTIMALARIAL, ANTICANCER,
ARTEMISININ-DERIVED TRIOXANE DIMERS
WITH HIGH SELECTIVITY, STABILITY AND EFFICACY
AND METHODS OF MAKING THE SAME**

5

CROSS REFERENCE TO OTHER APPLICATIONS

This application claims the benefit of U.S. provisional application numbers 60/414,786 filed September 27, 2002 and 60/445,526 filed February 6, 2003 which are both incorporated herein by reference.

10

CONTRACTUAL ORIGIN OF THE INVENTION

This study was supported in-part by National Institutes of Health grant AI 34885 (to G.H.P.) and grant RR-00052 and joint inventors G.H.P., I.P., A.J.M., S.S., K.B., T.L. and T.A.S., have assigned their rights to the Johns Hopkins University.

15

BACKGROUND OF THE INVENTION

1. Field of the Invention:

The present invention relates to the formation of a novel, stable C-10 carba trioxane dimer that may easily be transformed into different trioxane dimers. Some of these dimers are 10 times more antimalarially potent *in vitro* than artemisinin. These anticancer dimers selectively inhibit growth *in vitro* of several human cancer cell lines without being cytotoxic. The present invention relates also to water-soluble carboxylic acid derivatives that are thermally stable, that are more orally efficacious antimalarials than either artelinic acid or clinically used sodium artesunate, and that have potent and selective anticancer activities. The present invention further relates to antimalarial trioxane dimers that are safer and more efficacious than sodium artesunate.

25

2. Description of the State of Art:

Each year approximately 200-300 million people experience a malarial illness and over 1 million individuals die. In patients with severe and complicated disease, the mortality rate is between 20 and 50%.

Plasmodium is the genus of protozoan parasites which is responsible for all cases of human malaria and *Plasmodium falciparum* is the species of

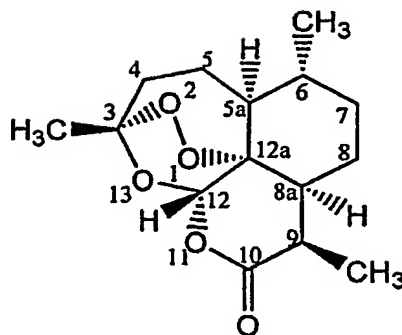
parasite that is responsible for the vast majority of fatal malaria infections. Malaria has traditionally been treated with quinolines such as chloroquine, quinine, mefloquine, and primaquine and with antifolates such as sulfadoxine-pyrimethamine. Unfortunately, most *P. falciparum* strains have now become
5 resistant to chloroquine, and some, such as those in Southeast Asia, have also developed resistance to mefloquine and halofantrine; multidrug resistance is developing in Africa also.

The endoperoxides are a promising class of antimalarial drugs which may meet the dual challenges posed by drug-resistant parasites and the rapid
10 progression of malarial illness. The first generation endoperoxides include natural artemisinin and several synthetic derivatives, discussed in further detail below.

Artemisia annua L., also known as qinghao or sweet wormwood, is a pervasive weed that has been used for many centuries in Chinese traditional
15 medicine as a treatment for fever and malaria. Its earliest mention, for use in hemorrhoids, occurs in the *Recipes for 52 Kinds of Diseases* found in the Mawangdui Han dynasty tomb dating from 168 B.C. Nearly five hundred years later Ge Hong wrote the *Zhou Hou Bei Ji Fang* (Handbook of Prescriptions for Emergency Treatments) in which he advised that a water
20 extract of qinghao was effective at reducing fevers. In 1596, Li Shizhen, the famous herbalist, wrote that chills and fever of malaria can be combated by qinghao preparations. Finally, in 1972, Chinese chemists isolated from the leafy portions of the plant the substance responsible for its reputed medicinal action. This crystalline compound, called qinghaosu, also referred to as QHS
25 or artemisinin, is a sesquiterpene lactone with an internal peroxide linkage.

Artemisinin is a member of the amorphane subgroup of cadinenes and has the following structure (I); "Art" will be used for shorthand in many of the figures.

I



Artemisinin or QHS was the subject of a 1979 study conducted by the Qinghaosu Antimalarial Coordinating Research Group involving the treatment of 2099 cases of malaria (*P. vivax* and *P. falciparum* in a ratio of about 3:1) with different dosage forms of QHS, leading to the clinical cure of all patients. See, Qinghaosu Antimalarial Coordinating Research Group, *Chin. Med. J.*, **92**:811 (1979). Since that time artemisinin has been used successfully in many thousand malaria patients throughout the world including those infected with both chloroquine-sensitive and chloroquine-resistant strains of *P.*

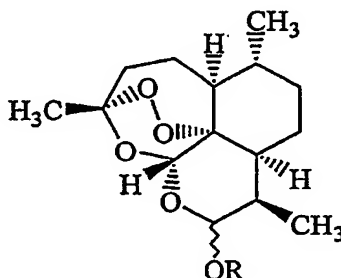
falciparum. Assay of artemisinin against *P. falciparum* *in vitro* revealed that its potency is comparable to that of chloroquine in two Hanian strains (Z. Ye, *et al.*, *J. Trad. Chin. Med.*, **3**:95 (1983)) and of mefloquine in the Camp (chloroquine-susceptible) and Smith (chloroquine-resistant) strains, D.L. Klayman, *et al.*, *J. Nat. Prod.*, **47**:715 (1984).

Although artemisinin is effective at suppressing the parasitemias of *P. vivax* and *P. falciparum*, the problems encountered with recrudescence, and the compound's insolubility in water, led scientists to modify artemisinin chemically, a difficult task because of the chemical reactivity of the peroxide linkage which is believed to be an essential moiety for antimalarial activity.

Reduction of artemisinin in the presence of sodium borohydride results in the production of dihydroartemisinin (II-1) or DHQHS, (illustrated in structure II below), in which the lactone group is converted to a lactol (hemiacetal) function, with properties similar to artemisinin. Artemisinin in methanol is reduced with sodium borohydride to an equilibrium mixture of α -

and β -isomers of dihydroartemisinin. Using dihydroartemisinin as a starting compound a large number of other derivatives, such as,

II



1R=H

5

2R=CH₃

3R=CH₂CH₃

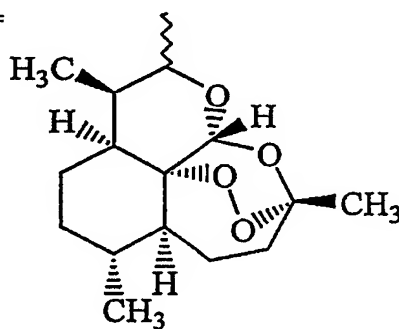
4R=COCH₂CH₂COONa

5R=CH₂C₆H₄COOH

6R=CH₂C₆H₄COONa

7

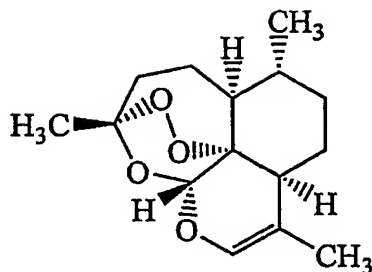
R=



10

artemether (compound II-2), arteether (II-3), sodium artesunate (II-4), artelinic acid (II-5), sodium artelinate (II-6), dihydroartemisinin condensation by-product (II-7) and the olefinic compound, structure III,

III



have been produced.

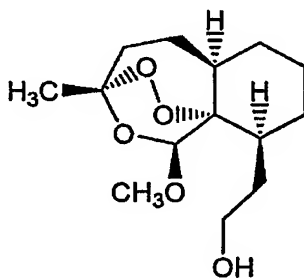
Artemether (II-2) is produced by reacting β -dihydroartemisinin with boron trifluoride (BF₃) etherate or HCl in methanol:benzene (1:2) at room temperature. A mixture of β - and α -artemether (70:30) is obtained, from which the former is isolated by column chromatography and recrystallized from hexane or methanol, R. Haynes, *Transactions of the Royal Society of Tropical Medicine and Hygiene*, **88**(I): S1/23-S1/26 (1994). For arteether (II-3), (Brossi, *et al.*, 1988), the α -isomer is equilibrated (epimerized) to the β -isomer in ethanol:benzene mixture containing BF₃ etherate. Treatment of dihydroartemisinin with an unspecified dehydrating agent yields both the olefinic compound, (III), and the dihydroartemisinin condensation by-product (II-7), formed on addition of dihydroartemisinin to (III), M. Cao, *et al.*, *Chem. Abstr.*, 100:34720k (1984). Until recently, the secondary hydroxy group in dihydroartemisinin (II-1) provided the only site in an active artemisinin-related compound that had been used for derivatization. See B. Venugopalan, "Synthesis of a Novel Ring Contracted Artemisinin Derivative," *Bioorganic & Medicinal Chemistry Letters*, **4**(5):751-752 (1994).

The potency of various artemisinin-derivatives in comparison to artemisinin as a function of the concentration at which the parasitemia is 90 percent suppressed (SD₉₀) was reported by D.L. Klayman, "Qinghaosu (Artemisinin): An Antimalarial Drug from China," *Science* **228**:1049-1055 (1985). Dr. Klayman reported that the olefinic compound III is inactive against *P. berghei*-infected mice, whereas the dihydroartemisinin condensation by-

product (II-7) has an SD_{90} of 10 mg/Kg in *P. berghei*-infected mice. Thus, the dihydroartemisinin ether dimer proved to be less potent than artemisinin, which has an SD_{90} of 6.20 mg/Kg. Following, in order of their overall antimalarial efficacy, are the three types of derivatives of dihydroartemisinin (II-1) that have been produced: (artemisinin) < ethers (II, R = alkyl) < esters [II, R = C(=O)-alkyl or -aryl] < carbonates [II, R = C(=O)O-alkyl or -aryl].

Other rational design of structurally simpler analogs of artemisinin has led to the synthesis of various trioxanes, some of which possess excellent antimalarial activity. Posner, G.H., *et al.*, reported the chemistry and biology of a series of new structurally simple, easily prepared, racemic 1,2,4-trioxanes (disclosed in U.S. Patent No. 5,225,437 and incorporated herein by reference) that are tricyclic (lacking the lactone ring present in tetracyclic artemisinin I) and that are derivatives of trioxane alcohol IV

IV



having the relative stereochemistry shown above. Especially attractive features of trioxane alcohol IV are the following: (1) its straightforward and easy preparation from cheap and readily available starting materials, (2) its availability on gram scale, and (3) its easy one-step conversion, using standard chemical transformations, into alcohol derivatives such as esters and ethers, without destruction of the crucial trioxane framework. See, Posner, G.H., *et al.*, *J Med. Chem.*, **35**:2459-2467 (1992), incorporated herein by reference. The complete chemical synthesis of artemisinin and a variety of other derivatives is reviewed by Sharma, R.P., *et al.*, *Heterocycles*, **32**(8):1593-1638 (1991), and is incorporated herein by reference.

Metabolic studies by Baker, *et al.*, demonstrated that β -arteether (II-3), one of the antimalarial derivatives discussed previously, was rapidly converted by rat liver microsomes into dihydroartemisinin (II-1). See Baker, J.K., *et al.*, *Biol. Mass. Spect.*, **20**:609-628 (1991). This finding and the fact
5 that the most effective artemisinin derivatives against malaria have been ethers or esters of (II-1) suggest that they were prodrugs for (II-1). The controlled slow formation of (II-1), however, is not desirable in view of its short half-life in plasma (less than two hours) and relatively high toxicity.

Unfortunately C-10 acetal derivatives are often unstable (i.e. easily
10 hydrolyzed) in water. Therefore making hydrolytically stable C-10 non-acetal carba-derivatives has become a high priority internationally.

Over the past thirty years only a few drugs isolated from higher plants have yielded clinical agents, the outstanding examples being vinblastine and vincristine from the Madagascan periwinkle, *Catharanthus roseus*, etoposide,
15 the semi-synthetic lignan, from Mayapple *Podophyllum peltatum* and the diterpenoid taxol, commonly referred to as paclitaxel, from the Pacific yew, *Taxus brevifolia*. Of these agents, paclitaxel is the most exciting, recently receiving approval by the Food and Drug Administration for the treatment of refractory ovarian cancer. Since the isolation of artemisinin, there has been a
20 concerted effort by investigators to study other therapeutic applications of artemisinin and its derivatives.

National Institutes of Health reported that artemisinin is inactive against P388 leukemia. See NCI Report on NSC 369397 (tested on 25 October 1983). Later anticancer studies that have been conducted on cell line panels
25 consisting of 60 lines organized into nine, disease-related subpanels including leukemia, non-small-cell lung cancer, colon, CNS, melanoma, ovarian, renal, prostate and breast cancers, further confirm that artemisinin displays very little anticancer activity. A series of artemisinin-related endoperoxides were tested for cytotoxicity to Ehrlich ascites tumor (EAT) cells using the
30 microculture tetrazolum (MTT) assay, H.J. Woerdenbag, *et al.*, "Cytotoxicity of Artemisinin-Related Endoperoxides to Ehrlich Ascites Tumor Cells," *Journal of Natural Products*, **56**(6):849-856 (1993). The MTT assay, used to test the

artemisinin-related endoperoxides for cytotoxicity, is based on the metabolic reduction of soluble tetrazolium salts into insoluble colored formazan products by mitochondrial dehydrogenase activity of the tumor cells. As parameters for cytotoxicity, the IC_{50} and IC_{80} values, the drug concentrations causing
5 respectively 50% and 80% growth inhibition of the tumor cells, were used. Artemisinin (I) had an IC_{50} value of 29.8 μM . Derivatives of dihydroartemisinin (II-1) being developed as antimalarial drugs (artemether (II-2), arteether (III-3), sodium artesunate (II-4), artelinic acid (II-5) and sodium artelinate (II-6)), exhibited a somewhat more potent cytotoxicity. Their IC_{50} values ranged from
10 12.2 μM to 19.9 μM . The dihydroartemisinin condensation by-product dimer (II-7), disclosed previously by M. Cao, *et al.*, 1984, was the most potent cytotoxic agent, its IC_{50} being 1.4 μM . At this drug concentration the condensation by-product (II-7) is approximately twenty-two times more cytotoxic than artemisinin and sixty times more cytotoxic than
15 dihydroartemisinin (II-1), the parent compound.

While artemisinin and its related derivatives (II 1-6) discussed above demonstrated zero to slight antiproliferative and antitumor activity, it has been discovered that a class of artemisinin dimer compounds exhibits antiproliferative and antitumor activities that are, *in vitro*, equivalent to or
20 greater than known antiproliferative and antitumor agents. See, U.S. Patent No. 5,677,468 incorporated herein by reference. Unfortunately, while the *in vitro* results of these artemisinin compounds are encouraging these compounds do not appear to have significant antitumor activity on the treatment of tumor cells in mice.

25 There is still a need, therefore, to develop methods for the formation of hydrolytically stable C-10 non-acetal carba-derivatives and structural analogs thereof having antimalarial, and antiproliferative and antitumor activities that are equivalent to or greater than those of known antimalarial, and antiproliferative and antitumor agents, respectively.

SUMMARY OF THE INVENTION

Accordingly, this invention provides a class of artemisinin related dimers which demonstrate antimalarial and antitumor activities.

More specifically, this invention provides a class of trioxane dimers
5 which demonstrate antimalarial and antitumor activities and that are considerably more stable toward hydrolysis than artemether and related C-10 ethers and esters.

This invention further provides artemisinin dimers to be used clinically as chemotherapeutic antimalarial and anticancer drugs.

10 This invention further provides artemisinin C-10-acetal derivatives for comparison to the corresponding dimers, and to be used clinically as chemotherapeutic antimalarial drugs.

This invention further provides a C-10-carba trioxane dimer that is easily transformed in one additional step into different dimers.

15 This invention further provides water-soluble carboxylic acid dimers that are thermally stable, that are more orally efficacious antimalarials than either artelinic acid or clinically used sodium artesunate, and that have potent and selective anticancer activities.

Additional objects, advantages and novel features of this invention
20 shall be set forth in part in the description and examples that follow, and in part will become apparent to those skilled in the art upon examination of the following specification or may be learned by the practice of the invention. The objects and advantages of the invention may be realized and attained by means of the instrumentalities, combinations, compositions, and methods
25 particularly pointed out in the appended claims.

BRIEF DESCRIPTION OF THE DRAWINGS

The accompanying drawings, which are incorporated in and form a part of the specification, illustrate the preferred embodiments of the present invention, and together with the description serve to explain the principles of
30 the invention.

In the drawings, Figures 7-15, the horizontal axis depicts various dilutions of the test compound, ranging from 10^{-4} to 10^{-9} molar, that were

exposed to the specified cancer cell lines. The vertical axis (percentage growth) depicts the growth of the specified cancer cell line when exposed to a specific concentration of the tested compound as compared to the growth of the same cancer cell line not exposed to any compound.

5 In the Drawings:

Figure 1 schematically depicts the method of converting artemisinin I into C-10 acetate 2 of the present invention.

Figure 2 schematically depicts the method of converting C-10 acetate 2 into C-10 non-acetal trioxane dimer 3 of the present invention.

10 Figure 3 schematically depicts the method of converting C-10 non-acetal trioxane dimer 3 of the present invention into bis-trioxane primary alcohol 4 of the present invention.

Figure 4 schematically depicts the method of converting C-10 non-acetal trioxane dimer 3 of the present invention into bis-trioxane vicinal diol 5
15 of the present invention.

Figure 5 schematically depicts the method of converting C-10 non-acetal trioxane dimer 3 of the present invention into bis-trioxane epoxide 6 of the present invention.

Figure 6 schematically depicts the method of converting C-10 non-acetal trioxane dimer 3 of the present invention into bis-trioxane ketone 7 of
20 the present invention.

Figure 7a depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of paclitaxel.

Figure 7b depicts the dose response curves generated by exposing
25 various leukemia cancer cell lines to various concentrations of artemisinin.

Figure 7c depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane primary alcohol 4 of the present invention.

Figure 7d depicts the dose response curves generated by exposing
30 various leukemia cancer cell lines to various concentrations of the bis-trioxane vicinal diol 5 of the present invention.

Figure 7e depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

5 Figure 7f depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

Figure 7g depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

10 Figure 7h depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 7i depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane
15 O-acetic acid **12b** of the present invention.

Figure 7j depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

20 Figure 7k depicts the dose response curves generated by exposing various leukemia cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

Figure 8a depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of paclitaxel.

25 Figure 8b depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the artemisinin.

Figure 8c depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the
30 bis-trioxane primary alcohol **4** of the present invention.

Figure 8d depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

Figure 8e depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 8f depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

Figure 8g depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 8h depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 8i depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

Figure 8j depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 8k depicts the dose response curves generated by exposing various non-small cell lung cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

Figure 9a depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of paclitaxel.

Figure 9b depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of artemisinin.

Figure 9c depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 9d depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

5 Figure 9e depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 9f depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

10 Figure 9g depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 9h depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane
15 tertiary alcohol benzoic acid **11b** of the present invention.

Figure 9i depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

20 Figure 9j depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 9k depicts the dose response curves generated by exposing various colon cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

25 Figure 10a depicts the dose response curves generated by exposing various CNS cancer cell lines to various concentrations of paclitaxel.

Figure 10b depicts the dose response curves generated by exposing various CNS cancer cell lines to various concentrations of artemisinin.

30 Figure 10c depicts the dose response curves generated by exposing various CNS cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 10d depicts the dose response curves generated by exposing various CNS cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

Figure 10e depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 10f depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

Figure 10g depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 10h depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 10i depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

Figure 10j depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 10k depicts the dose response curves generated by exposing various CNS cancer cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

Figure 11a depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of paclitaxel.

Figure 11b depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of artemisinin.

Figure 11c depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 11d depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

5 Figure 11e depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 11f depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

10 Figure 11g depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 11h depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-
15 trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 11i depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

Figure 11j depicts the dose response curves generated by exposing
20 various melanoma cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 11k depicts the dose response curves generated by exposing various melanoma cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

25 Figure 12a depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of paclitaxel.

Figure 12b depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of artemisinin.

Figure 12c depicts the dose response curves generated by exposing
30 various ovarian cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 12d depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

5 Figure 12e depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 12f depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

10 Figure 12g depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 12h depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

15 Figure 12i depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

Figure 12j depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 12k depicts the dose response curves generated by exposing various ovarian cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

25 Figure 13a depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of paclitaxel.

Figure 13b depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of artemisinin.

30 Figure 13c depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 13d depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

5 Figure 13e depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 13f depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

10 Figure 13g depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 13h depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 13i depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

20 Figure 13j depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 13k depicts the dose response curves generated by exposing various renal cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

25 Figure 14a depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of paclitaxel.

Figure 14b depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

30 Figure 14c depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

Figure 14d depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

5 Figure 14e depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

Figure 14f depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

10 Figure 14g depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 14h depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane
15 O-acetic acid **12b** of the present invention.

Figure 14i depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

20 Figure 14j depicts the dose response curves generated by exposing various prostate cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

Figure 15a depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of paclitaxel.

25 Figure 15b depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane primary alcohol **4** of the present invention.

Figure 15c depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane vicinal diol **5** of the present invention.

30 Figure 15d depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 15e depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the succinate ester **9** of the present invention.

Figure 15f depicts the dose response curves generated by exposing
5 various breast cancer cell lines to various concentrations of the bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 15g depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

10 Figure 15h depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane O-acetic acid **12b** of the present invention.

Figure 15i depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane
15 primary alcohol isonicotinate N-oxide **13** of the present invention.

Figure 15j depicts the dose response curves generated by exposing various breast cancer cell lines to various concentrations of the bis-trioxane diphenyl phosphate dimer **14** of the present invention.

Figure 16 schematically depicts the method of converting bis-trioxane
20 primary alcohol **4** into bis-trioxane primary succinate monoester **8a** of the present invention.

Figure 17 schematically depicts the method of converting bis-trioxane vicinal diol **5** into the tertiary alcohol primary succinate ester **9** of the present invention.

25 Figure 18 schematically depicts the method of converting bis-trioxane epoxide **6** into bis-trioxane β -hydroxysulfide ester **10a**, sulfide to sulfone oxidation gave bis-trioxane β -hydroxysulfone ester **10b** that was saponified into bis-trioxane β -hydroxysulfone benzoic acid **10c** of the present invention.

Figure 19 schematically depicts the method of converting bis-trioxane
30 ketone **7** into bis-trioxane styryl tertiary alcohol **11a** and bis-trioxane tertiary alcohol benzoic acid **11b** of the present invention.

Figure 20 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane O-allyl ether **12a** and bis-trioxane O-acetic acid **12b** of the present invention.

5 Figure 21 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane primary alcohol isonicotinate **8b** of the present invention.

Figure 22 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane primary alcohol isonicotinate N-oxide **13** of the present invention.

10 Figure 23 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane primary alcohol nicotinate N-oxide **15** of the present invention.

Figure 24 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane phosphonic acid monoester **16** of the present invention.

Figure 25 schematically depicts the method of converting primary alcohol **4** to the bis-trioxane diphenyl phosphate **14** of the present invention.

Figure 26 schematically depicts the method of converting bis-trioxane ketone **7** to the bis-trioxane secondary alcohol **17** of the present invention.

20 Figure 27 schematically depicts the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane vicinal diol isonicotinate N-oxide **18** of the present invention.

Figure 28 schematically depicts the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane isobutyric acid **19** of the present invention.

25 Figure 29 schematically depicts the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane β -hydroxy O-allyl ether **20** and then on to the bis-trioxane β -hydroxy O-acetic acid **21** of the present invention.

Figure 30 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol picolinate **22** of the present invention.

Figure 31 schematically depicts the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane vicinal diol nicotinate N-oxide **23** of the present invention.

Figure 32 schematically depicts the method of converting bis-trioxane secondary alcohol **17** to the bis-trioxane secondary alcohol isonicotinate N-oxide **24** of the present invention.

Figure 33 schematically depicts the method of converting bis-trioxane secondary alcohol **17** to the bis-trioxane secondary alcohol nicotinate N-oxide **25** of the present invention.

Figure 34 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol diethyl phosphoric acid triester **26** of the present invention.

Figure 35 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol dimethyl phosphoric acid triester **27** of the present invention.

Figure 36 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol p-trifluoromethylbenzoate **28** of the present invention.

Figure 37 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol 3,5-ditrifluoromethylbenzoate **29** of the present invention.

Figure 38 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol dimethylglycinate **30** of the present invention.

Figure 39 schematically depicts the method of converting artemisinin **1** to the dihydroartemisinin **31** of the present invention.

Figure 40 schematically depicts the method of converting dihydroartemisinin **31** to the α -dihydroartemisinin dimethylglycinate **32** of the present invention.

Figure 41 schematically depicts the method of converting dihydroartemisinin **31** to the α -dihydroartemisinin isonicotinate N-oxide **33** of the present invention.

Figure 42 schematically depicts the method of converting bis-trioxane ketone **7** to the bis-trioxane allyl tertiary alcohol **34** and the method of converting bis-trioxane allyl tertiary alcohol **34** to the bis-trioxane tertiary alcohol carboxylic acid **35** of the present invention.

5 Figure 43 schematically depicts the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol methyl carbonate **36** and the method of converting bis-trioxane primary alcohol **4** to the bis-trioxane primary alcohol ethyl carbonate **37** of the present invention.

10 Figure 44 schematically depicts the method of converting bis-trioxane ketone **7** to the bis-trioxane tertiary alcohol methyl sulfonate **38**, the method of converting bis-trioxane ketone **7** to the bis-trioxane bis-trioxane tertiary alcohol isopropyl sulfonate **39** and the method of converting bis-trioxane ketone **7** to the bis-trioxane tertiary alcohol N,N-dimethylsulfonamide **40** of the present invention.

15 Figure 45 schematically depicts the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane vicinal diol cyclic carbonate **41** and the method of converting bis-trioxane vicinal diol **5** to the bis-trioxane vicinal diol cyclic sulfate **42** of the present invention.

20 Figure 46 schematically depicts the method of converting bis-trioxane epoxide **6** to the bis-trioxane tertiary alcohol pyridine sulfide **43** of the present invention.

25 Figure 47 schematically depicts the method of converting bis-trioxane ketone **7** to the bis-trioxane ketone O-TMS cyanohdrin **44** and the method of converting bis-trioxane ketone **7** to the bis-trioxane tertiary alcohol nitrile **45** of the present invention.

Figure 48 schematically depicts the method of converting bis-trioxane ketone **7** to the bis-trioxane methyl enol ether **46** of the present invention.

Figure 49 schematically depicts the method of converting bis-trioxane methyl enol ether **46** to the bis-trioxane aldehyde **47** of the present invention.

30 Figure 50 schematically depicts the method of converting bis-trioxane aldehyde **47** to the bis-trioxane aniline **48** of the present invention.

Figure 51 schematically depicts the method of converting bis-trioxane primary alcohol **4** and bis-trioxane diol **5** to a number of compounds according to the present invention. In cases where the derivatization of bis-trioxane primary alcohol **4** is shown, conversion of bis-trioxane secondary alcohol **17** and bis-trioxane diol **5** to their corresponding analogs is also in accordance with the present invention.

Figure 52 schematically depicts the method of converting a number of compounds of the present invention to ten other compounds that can be synthesized according to the present invention.

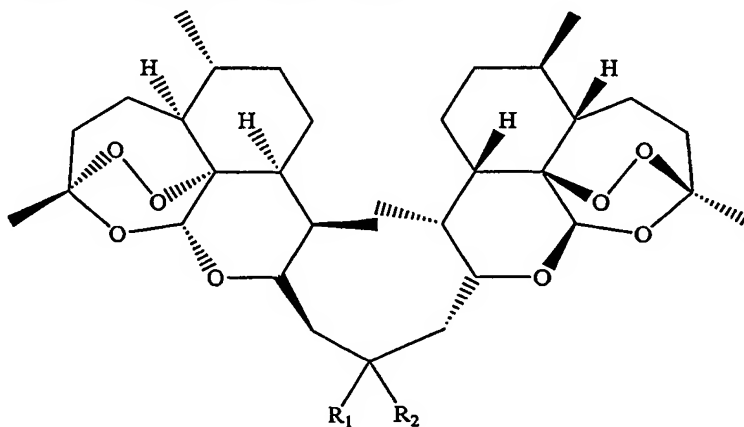
Figure 53 schematically depicts the method of converting bis-trioxane aldehyde **47** to a range of substituted and unsubstituted amine/amide derivatives of the current invention.

Figure 54 represents other compounds that can be synthesized according to the present invention.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

The present invention provides a direct method for converting a novel C-10-carba trioxane dimer in one step into a number of different dimers, all of which are hydrolytically stable. The resulting trioxane dimers may then be further reacted to form 3-carbon atom linked, oxygenated dimers. These C-10 non-acetal derivatives of natural trioxane artemisinin have high *in vitro* antimalarial and antitumor activities.

In general, the compound of the present invention includes resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, and may be illustrated as follows:

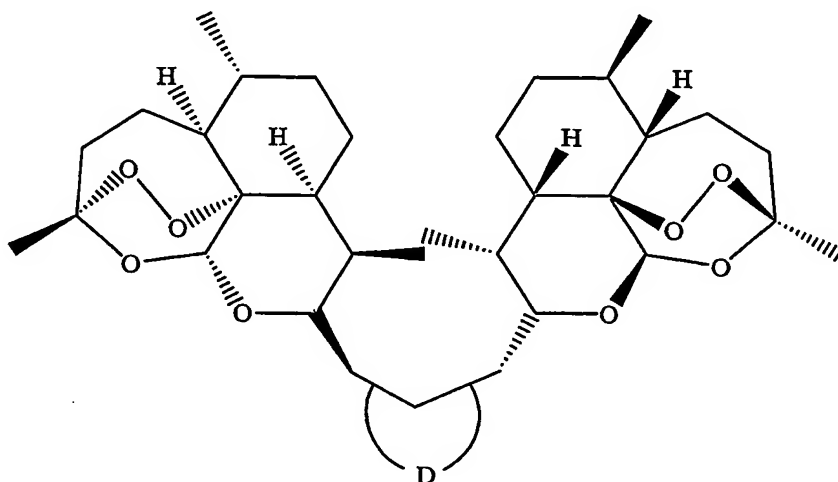


Where if R_1 is hydrogen or OH then R_2 is AX, and if R_2 is hydrogen or OH then R_1 is AX, and A may be absent or A may be any alkyl or aryl group where X is hydrogen, a phosphate group, a phosphonic acid derivative group, an alcohol group, a carboxylic acid group, an ether group, an ester group, a nitrile group, a sulfone group, a sulfide group, an amino acid derivative group, an amine group, and amide group, an aldehyde group, or an aromatic group.

The alcohol group may be represented by $-R^3OH$, where R^3 is a straight chained or branched alkyl group having 1 to 5 carbon atoms. The carboxylic acid group may be represented by $-R^4COOH$, where R^4 is at least one saturated or unsaturated alkyl group, an ester group, an ether group or a combination thereof. When R^4 is an ester group, it may be represented by $-R^5COO-$, where R^5 is bonded to the carboxylic acid group and has 0 to 5 carbon atoms. When R^4 is an ether group, it may be represented by R^6-O-R^7 wherein R^6 and R^7 are, independently, an alkyl or allyl group having 0 to 5 carbon atoms. The aromatic group may be represented by $Ar-(R^8)_m$, where Ar represents a benzene ring, and m is 1 or 2, and R^8 may represent $-CH=CH_2$, or $-COOH$. The ester group may be represented by $-CR^9$, where R^9 is an ester of nicotinic acid, an ester of isonicotinic acid, or the ester group is represented by $-CO(C=O)R^{9a}$, where R^{9a} is $Ph(CY_3)_o$, where o is 1 or 2, and Y may be, independently, H, F, Cl, Br, or I, or where R^{9a} is a substituted heterocyclohexane compound. The phosphonic acid derivative group may be represented by $-CO-P(R^{10})OOH$, where R^{10} is an alkyl group having 0 to 5 carbon atoms. The phosphate group is $-COP(O)(OR^{11})_2$, where R^{11} is an alkyl group having 0 to 5 carbon atoms, or a phenyl group. The nitrile group is $R^{12}CN$, where R^{12} is an alkyl group having 0 to 5 carbon atoms. The sulfone group may be represented by $-CS(=O)_2R^{13}$, wherein R^{13} is $-N(CH_3)_2$, $-OR^{14}$, or $-Ph-COOR^{14}$, where R^{14} is H, CH_3 , or $-CH(CH_3)_2$. The sulfide group may be represented by $-CSR^{15}$, where R^{15} is pyridine or $-Ph-COOR^{16}$, where R^{16} is H or CH_3 . The amino acid derivative group is $-COC(=O)CN(R^{17})_2$, where each R^{17} group is, independently, H or CH_3 . The amine group may be represented by $-CN(R^{18})_2$, where each R^{18} group is, independently, H, an alkyl group, or a phenyl group. The ether group may be represented by $-C-O-CR^{19}$, where R^{19} is a substituted pyridine. The amide

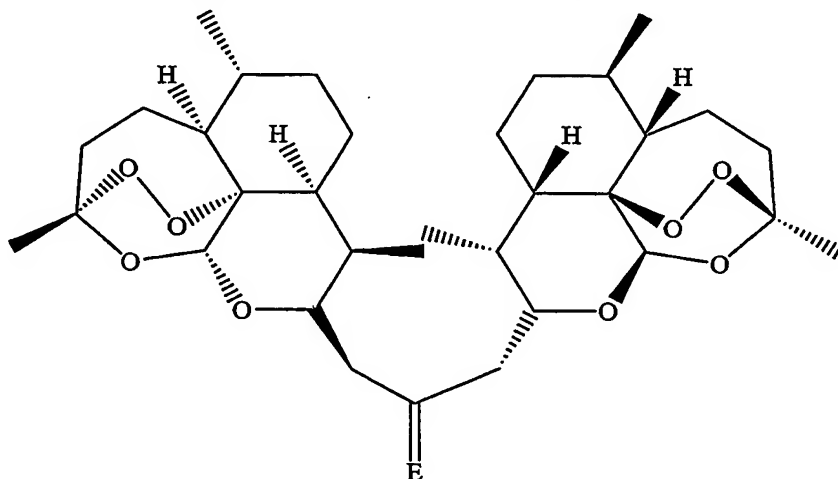
group is $-(C=O)N(R^{20})_2$, or $-CH_2(C=O)N(R^{20})_2$ where each R^{20} is, independently, H or $-CH_2CH_2N(CH_3)_2$.

The compound of the present invention may also be illustrated as:



- 5 where D forms a heterocyclic ring having 3 to 5 atoms. For example, the heterocyclic ring may be a 3-membered ring and one of the atoms in the ring is oxygen. The heterocyclic ring may also be a 5-membered ring and two of the atoms in the ring are oxygen, and the ring may also be substituted with an oxygen atom and another atom in the 5-membered ring may be a sulfur or
- 10 phosphorus atom. In another example, the 5-membered ring may also be substituted with 1 or 2 oxygen atoms bonded to the sulfur atom.

The compound of the present invention may further be illustrated as:



- 15 where where E is H, O, NR, CH₂ or S wherein R may be hydrogen, alkyl, aryl or any other substituent.

In general, the first step of the preferred process of the present invention, shown in Figure 1, conversion of artemisinin (**1**) into α -dihydroartemisinin acetate (**2**), is accomplished by using an effective amount of a hydride nucleophile in combination with an acetylating agent which chemoselectively reacts at the C-10 position of the artemisinin skeleton without disrupting the O-O bond in this trioxane. Effective nucleophilic hydride reagents include, but are not limited to sodium borohydride, and DIBAL. The relative amounts of the various possible nucleophilic hydride reagents depends upon the concentration employed and other conditions of the reaction. Various amounts of the nucleophilic reagents can be employed, but generally it should be present in the range of 1.0 to 1.5 molar equivalents of nucleophilic hydride per molar equivalent of artemisinin for the reaction to proceed to completion. Effective acetylating agents include, but are not limited to acetyl chloride and acetic anhydride. The relative amounts of the various possible acetylating agents depends upon the concentration employed and other conditions of the reaction. Various amounts of the acetylating agents can be employed, but generally it should be present in the range of 1.0 to 1.5 molar equivalents of acetylating agent per molar equivalent of dihydroartemisinin for the reaction to proceed to completion.

The next step, shown in Figure 2, in the production of the novel meso trioxane dimer **3** involves the substitution of the acetoxy group from the C-10 acetate **2** product to form the meso trioxane dimer **3** product. Inspiration for this bis-allylation of C-10 acetate **2** was based on the pioneering trioxane mono-allylations of Ziffer (Pu, Y.M.; Ziffer, H., *Synthesis and Antimalarial Activities of 12 β -Allyldeoxoartemisinin*. *J. Med. Chem.* **1995**, *38*, 613-616) and O'Neill (O'Neill, et al., *J. Med. Chem.* **1999**, *42*, 5487-5493; and O'Neill, P. M., et al., *J. Chem. Soc., Perkin Trans.* **2001**, *1*, 2682-2689.) using allylsilane and based also on allylic bis-silane chemistry, see Rychnovsky, S.D., et. al., *Tetrahedron Lett.* **1999**, *40*, 41-44. The requisite allylic bis-silane is easily prepared in one step from the corresponding commercial allylic dichloride (Shown in Figure 2); in the presence of fresh tin tetrachloride, the allylic bis-silane converts acetate **2** on gram scale into meso dimer **3**,

characterized by ^1H NMR spectroscopy as done before in structurally related trioxanes. This double substitution reaction undoubtedly proceeds sequentially via initial monoallylation, producing an intermediate C-10 allylic silane that then reacts with another molecule of trioxane acetate **2** to form the product dimer **3**. This new dimer **3**, with an unsaturated 3-carbon atom linker between the two trioxane units, is stable in air and light at room temperature for at least six months, and its preparation on much larger industrial scale should be feasible.

In contrast to most simple peroxides that are easily cleaved by reducing agents and by reactive organometallics, the peroxide linkage in artemisinin-like trioxanes is relatively robust. Therefore, several different chemical transformations chemospecifically involving only the linker isobutylene carbon-carbon double bond in dimer **3** may be performed and are discussed in further detail below.

Borane reduction and *in situ* oxidation produced bis-trioxane primary alcohol **4**, shown in Figure 3. Dihydroxylation using osmium tetroxide gave bis-trioxane vicinal diol **5**, shown in Figure 4. Dimethyldioxirane formed bis-trioxane epoxide **6**, shown in Figure 5. Oxidative cleavage using catalytic osmium tetroxide and Oxone led to bis-trioxane ketone **7**, shown in Figure 6. Bis-Trioxanes **4-7** are stable even when heated neat in air for 24 h at 60 °C; ^1H NMR spectroscopy showed less than 5% decomposition under these accelerated aging conditions.

To illustrate further the chemical inertness of the peroxide unit in these trioxane dimers, and especially to generate some water-soluble dimers that can be easily administered *in vivo*, each dimer **4-7** was converted into a carboxylic acid (Figures 16-20). Primary alcohol **4** opened succinic anhydride to form bis-artesunate **8a** in high yield, see Figure 16. Likewise, diol **5** opened succinic anhydride to produce the tertiary alcohol primary succinate ester **9**, see Figure 17. Epoxide **6** reacted chemospecifically with a benzenethiol in the presence of chromatographic alumina to give β hydroxysulfide **10a** in high yield; sulfide to sulfone oxidation gave dimer benzoate ester **10b** that was saponified into benzoic acid **10c**, see Figure 18. Ketone **7** underwent

5 Each of these new carboxylic acid dimers is at least as soluble in water at pH 7.4 as is the antimalarial drug candidate artelinic acid, and tertiary alcohol primary succinate ester **9** is close to 30 times more water-soluble than artelinic acid, and phosphonic acid **16** is even more water soluble than succinate ester **9**. Trioxane dimer carboxylic acids **8a-10c** and **12b** are stable
10 in air even at 60°C for 24 hours.

Primary alcohol **4** may also be transformed into a variety of other compounds as shown in Figures 21 – 25, 28, 30, 34 – 38 and 43. Likewise, diol **5** may be converted into a variety of compounds as shown in Figures 27, 29, 31 and 45. Additionally, secondary alcohol **17** may be transformed into a variety of compounds as shown in Figures 32 and 33. Likewise, ketone **7** may be converted into a variety of compounds as shown in Figures 42, 44, 47 and 48 and epoxide **6** may be reacted further as shown in Figure 46. Finally, dimer **3** may be converted to a variety of compounds as shown in Figures 51, 52 and 53.

To determine the antimalarial effect of various dimers of the present invention, screening assays were performed against chloroquine-sensitive *P. falciparum* (NF54), according to the method described below, and their IC₅₀ values are included in Table 1.

28

(Gibco BRL # 13200-076), supplemented with 25 mM *N*-2-hydroxyethylpiperazine-*N'*-2-ethanesulfonic acid (HEPES; Calbiochem #391338), 27 mM NaHCO₃ (Gibco BRL # 11810-025), and 10% heat-inactivated human type O⁺ serum (Interstate Blood Bank, Inc.), under 3% O₂,
5 4% CO₂, and 93% N₂. Parasitemia was maintained at 0.05-3% and doubling time at approximately 15 hours by twice weekly change of medium and replenishment with fresh erythrocytes.

Stock solutions (approximately 2.5 mg/mL of HPLC-purified or recrystallized test compound) were prepared in dimethyl sulfoxide (DMSO; Sigma-Aldrich #27,043-1). DMSO solutions were diluted 500-fold in medium,
10 serially diluted in 0.2% DMSO in medium (to maintain constant solvent concentration), then 100 μ L aliquots were pipetted into microtiter plate wells (Costar3595). Provisional EC₅₀ values were obtained in a survey of seven 5-fold dilutions yielding final concentrations (in triplicate) of 0.16-2500 ng/mL.
15 Assays were later expanded to include ten concentrations (in quadruplicate) of approximately 1.8-fold dilutions which flank the provisional EC₅₀. Plates included at least 8 wells of no drug controls (4 with and 4 without DMSO) and 4 wells of uninfected erythrocytes. Parasite culture (0.25% parasitemia in 2.4% hematocrit; 100 μ L per well) was added and the plate was incubated for 48
20 hours prior to the addition of 25 μ L [³H]hypoxanthine (14.1 Ci/mmol, 1 mCi/mL in 70% ethanol, New England Nuclear NET -177, diluted to 25 μ Ci/mL with medium) and subsequent 20 hour incubation. Cells were harvested (Brandel MB-48R) onto GF-C glass filters (Brandel). The filters were washed five times with 3 mL water per sample spot, dried under a heat lamp, and counted
25 (Beckman Model LS-6500) in scintillation cocktail (ICN Cytoscint).

Decays per minute (dpm) values were downloaded and analyzed (Power Macintosh 7200/90; Microsoft Excel 5.0), to yield the mean and standard deviation at each drug concentration. Dose-response curves were fit to the experimental data (Delta Point DeltaGraph 3.5.3) by means of the
30 Marquardt algorithm, were solved for the drug concentration that kills 50% of parasites, and were analyzed for goodness of fit (R² value).

Table I
Antimalarial Activities *in vitro*^a

<u>Trioxane</u>	<u>IC₅₀(nM)</u>
3	24.0
4	0.87
5	0.59
6	2.8
7	0.91
8a	2.0
8b	1.7
9	3.0
10c	2.4
11b	2.1
12b	3.2
13	0.53
14	10.0
15	0.58
16	320
17	0.66
18	0.34
19	2.4
21	4.4
23	0.40
24	0.42
25	0.31
26	0.74
27	0.48
32	1.1
33	0.81
35	2.0
36	1.7
37	2.8
38	200
40	0.73
41	0.59
42	19
48	4.8
Artemisinin	9.2 ± 1.7

^a The standard deviation for each set of quadruplicates was an average of 8.6% ($\leq 22\%$) of the mean. R^2 values for the fitted curves were ≥ 0.933 . Artemisinin activity is mean \pm standard deviation of concurrent control (n = 34).

- 5 In sharp contrast to the potency of the natural trioxane artemisinin (IC₅₀ = 7.7 nM) and of the initial olefinic dimer **3** (IC₅₀ = 24 nM), alcohol dimers **4** and **17**, diol dimer **5** and ketone dimer **7** all have substantially enhanced potencies, with IC₅₀ values below one nM. In addition, further derivitization of these potent antimalarials has produced a number of analogs

also antimalarially active in sub-nanomolar concentrations (most notably: pyridine N-oxides **13**, **15**, **18**, **23**, **24** and **25**, phosphoric acid triesters **26** and **27**, sulfonamide **40** and cyclic carbonate **41**).

The *in vivo* antimalarial efficacies of water-soluble dimer carboxylic acids **8a-10c** and **12b** (**11b** not tested), as measured in mice according to a published protocol, are shown in Table II. In all cases, these water-soluble dimeric trioxanes are more efficacious than the drug candidate artelinic acid administered intravenously (IV) and than the clinically used drug sodium artesunate administered orally (PO). At an IV dose of 10 mg/kg of mouse body weight, each dimeric trioxane **8a-10c** and **12b** suppressed *P. berghei* NY malaria parasite growth considerably better (>80%) than did artelinic acid or sodium artesunate. Neither overt toxicity nor behavioral modification was observed due to drug administration in any of these mouse experiments.

Table II. Antimalarial Efficacies *in vivo*

<u>Trioxane Carboxylic Acid</u>	<u>Administration Route</u>	<u>ED₅₀^b</u>	<u>ED₉₀^b</u>	<u>%Parasite Suppression at 10 mg/kg</u>
8a	IV	2.2	6.3	81.0
	PO	9.0	15.0	79.2
12b	IV	1.0	3.5	95.7
	PO	6.5	17.0	46.5
Artelinic Acid Sodium Artesunate	IV	11.0	25.0	54.7
	PO	10.0	39.0	52.8
9	IV	2.4	11.0	83.4
	PO	4.8	34.0	55.4
10c	IV	2.7	10.0	83.4
	PO	7.5	35.0	46.1
Artelinic Acid Sodium Artesunate	IV	5.6	43.0	42.4
	PO	5.5	70.0	50.9

^b mg/kg x 4

Determination of Antitumor Activities

Growth Inhibition. To determine the growth inhibition (GI) and cytotoxicity of dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, **12b**, **13**, and **14** respectively, of

the present invention, screening assays were performed by the National Cancer Institute using a 60 cell line panel; these activities are summarized in Tables III, IV and V (set out below). The screening assay is performed on 96-well microtitre plates. Relatively high initial inoculation densities are used, in order to permit measurement of "time-zero" values and to enhance the screen's ability to detect and provide some differentiation between growth inhibition and cytotoxic response parameters. The specific inoculation densities (which range from 5,000 to 40,000 cells/well) used for each cell line are those which, for the respective line, were determined to give an optical density signal for both the "time-zero" value (at 24 hours) and the "no-drug" control (at 72 hours) above the noise level and within the linear range of the end-point assay (which measures cellular protein). The inoculated microtitre plates are pre-incubated for 24 hours at 37°C prior to drug additions. The five drug dilutions tested routinely range from 10^{-4} to 10^{-8} molar. Higher or lower concentration ranges may be selected on a non-routine basis if appropriate solubility and/or prior biological information or other screening data so dictate. Duplicate wells are prepared for all concentrations, (concentration is often denoted by placing brackets around a number); "time-zero" and "no drug" controls are also provided for each test. The minimum amount of compound required for a one-time evaluation in the routine screen can be calculated from the knowledge that each test requires a total of approximately 40 ml (0.04 liter) of cell culture medium containing the highest desired drug concentration. Thus, the amount (grams) of sample required (assuming an upper test concentration limit of 10^{-4} M) is: molecular weight of compound x 10^{-4} x 0.04. After a 48 hour incubation (37°C) with the test compound, the cells are fixed *in situ* to the bottoms of the microtitre wells by addition of 50 μ l of either 50% trichloroacetic acid (for adherent cell lines) or 80% trichloroacetic acid (for settled cell suspension lines), followed by incubation for 60 minutes at 4°C. The cellular protein in each well is assayed using a sulforhodamine B (SRB) stain procedure. Briefly, after discarding the supernatants, the microtitre plates are washed 5 times with deionized water and air-dried. One hundred microliters of SRB solution (0.4% w/v in 1%

acetic acid) is added to each microtitre well and incubated for 10 minutes at room temperature. Unbound SRB is removed by washing 5 times with 1% acetic acid. The plates are air-dried, the bound stain is solubilized with Tris buffer, and the optical densities read at 515nm. SRB is a bright pink anionic dye which, in dilute acetic acid, binds electrostatically to the basic amino acids of TCA-fixed cells. Cryopreserved master stocks of all the lines are maintained, and cultures used for screening are replaced from the master stock after no more than twenty passages in the screening laboratory. The cell line panel consists of 60 lines, organized into nine, disease-related subpanels including leukemia, non-small-cell lung cancer, colon, CNS, melanoma, ovarian, renal, prostate and breast cancers.

The response parameters GI_{50} and LC_{50} are interpolated values representing the concentrations at which the percentage growth (PG) is +50 and -50, respectively:

GI_{50} is the concentration for which the $PG = +50$. At this value the increase from time t_{zero} , in the number or mass of cells in the test well is only 50% as much as the corresponding increase in the control well during this period of the experiment, see Table III. A drug effect of this intensity is interpreted as primary growth inhibition.

TGI is the concentration for which $PG = 0$. At this value the number or mass of cells in the well at the end of the experiment equals the number or mass of cells in the well at time t_{zero} , see Table IV. A drug effect of this intensity is regarded as cytostasis.

LC_{50} is the concentration for which the $PG = -50$. At this value, the number or mass of cells in the test well at the end of the experiment is half that at time t_{zero} , see Table V. This is interpreted as cytotoxicity.

TABLE III

Panel/Cell Line	Log ₁₀ GI ₅₀										
	Artem- isinin	Trioxane Dimers									Pacil- taxel
		4	5	8a	9	10c	11b	12b	13	14	
Leukemia											
CCRF-CEM	—	—	-4.71	< -8.30	< -8.30	< -8.30	-8.07	< -8.30	5.73	> -4.30	-11.61
HL-60(TB)	-4.26	-7.78	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30		< -8.30	-11.57
K-562	-4.33	-7.91	< -8.00	< -8.30	-8.30	< -8.30	< -8.30	< -8.30		< -8.30	-10.83
MOLT-4	-4.73	-7.92	< -8.00	-8.18	-5.15	—	< -8.30	-7.66		—	-11.07
RPMI-8226	> -4.00	< -8.00	—	< -8.30	< -8.30	< -8.30	—	—	< -8.30	< -8.30	< -13.00
SR	> -4.00	< -8.00	—	—	—	—	—	—	—	—	8.34
Non-Small Cell Lung Cancer											
A549/ATCC	-4.17	-7.29	-7.59	-4.42	-4.86	-5.28	-5.69	> -4.30	-5.98	< -8.30	—
EKVX	> -4.00	-7.12	—	-6.90	—	-4.99	-4.79	> -4.30	-8.19	< -8.30	—
HOP-62	> -4.00	-4.92	-4.66	-6.34	-5.69	< -8.30	-6.96	-6.23	—	-4.90	-9.67
HOP-92	> -4.00	-6.37	-4.93	-6.56	< -8.30	< -8.30	-7.43	-7.00	-8.04	< -8.30	—
NCI-H226	> -4.00	-4.82	-4.87	< -8.30	—	< -8.30	—	—	—	—	—
NCI-H23	> -4.00	-7.84	-4.83	< -8.30	< -8.30	< -8.30	-8.19	< -8.30	< -8.30	< -8.30	—
NCI-H322M	—	-4.97	-4.71	-5.11	-5.07	-6.03	-6.72	-5.03	-6.26	< -8.30	-10.12
NCI-H460	> -4.00	-7.35	-5.02	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-12.16
NCI-H522	—	-7.53	-7.33	-6.16	-4.91	< -8.30	-6.54	> -4.30	—	—	< -13.00
Colon Cancer											
COLO 205	> -4.00	< -8.00	< -8.00	< -8.30	—	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-11.07
HCC-2998	> -4.00	-7.45	-4.65	—	—	—	—	—	—	—	-12.34
HCT-116	-4.18	< -8.00	-4.72	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -13.00
HCT-15	> -4.00	< -8.00	-5.28	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-6.37
HT29	> -4.00	-7.75	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	—	—	< -13.00
KM12	> -4.00	< -8.00	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	—	-11.43
SW-620	> -4.00	< -8.00	-4.97	—	—	—	—	—	< -8.30	< -8.30	-11.60
CNS Cancer											
SF-268	—	-5.40	-4.94	-7.11	-5.48	< -8.30	-7.72	-6.82	-5.29	—	—
SF-295	—	-5.59	-4.94	—	< -8.30	—	—	—	-5.21	< -8.30	—
SF-539	—	-7.05	-4.86	-5.27	-4.90	-5.19	-4.80	-4.53	< -8.30	< -8.30	-11.09
SNB-19	> -4.00	-4.88	-4.82	-5.62	—	-6.72	-6.53	-6.03	-5.08	—	-8.98
SNB-75	> -4.00	-4.99	-4.53	-6.47	—	—	-7.40	—	—	—	—
U251	> -4.00	-7.27	-7.68	-8.20	—	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-11.29
Melanoma											
LOX IMVI	—	-7.26	< -8.00	—	—	—	—	—	< -8.30	< -8.30	-11.80
MALME-3M	—	-7.31	—	< -8.30	< -8.30	< -8.30	< -8.30	—	< -8.30	-7.22	—
M14	—	-5.87	-4.59	-6.94	—	< -8.30	-7.81	-6.68	-5.81	—	-11.73
SK-MEL-2	—	-7.33	-4.94	-5.12	-4.88	—	-6.43	-5.62	-5.12	> -4.30	-9.53
SK-MEL-28	> -4.00	-4.95	-4.79	-6.09	< -8.30	-5.45	-6.40	-4.92	-5.13	—	—
SK-MEL-5	> -4.10	-7.90	-7.69	—	—	—	—	—	< -8.30	< -8.30	—
UACC-257	> -4.00	-7.31	< -8.00	-5.02	-4.89	—	—	> -4.30	-5.15	< -8.30	-10.30
UACC-62	> -4.00	-6.86	—	—	—	—	—	—	-8.00	< -8.30	-10.46
Ovarian Cancer											
IGROVI	-4.31	-5.36	-4.97	-5.07	-4.97	—	-5.98	-5.06	-5.36	-5.81	-8.61
OVCAR-3	—	-7.68	-5.09	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-7.75	< -8.30	-10.40
OVCAR-4	—	-5.44	-4.99	—	—	—	—	—	< -8.30	< -8.30	-5.00
OVCAR-5	> -4.00	-7.79	-4.78	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-9.38
OVCAR-8	> -4.00	-7.33	< -8.00	> -4.30	-5.12	-5.25	-5.26	> -4.30	-7.69	< -8.30	-10.75
SK-OV-3	—	-4.85	-4.62	-4.89	—	-6.82	—	> -4.30	-5.13	> -4.30	—

Panel/Cell Line	Log ₁₀ GI ₅₀										
	Artem- isinin	Trioxane Dimers									Pacil- taxel
		4	5	8a	9	10c	11b	12b	13	14	
Leukemia											
CCRF-CEM	—	—	-4.71	< -8.30	< -8.30	< -8.30	-8.07	< -8.30	5.73	> -4.30	-11.61
HL-60(TB)	-4.26	-7.78	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30		< -8.30	-11.57
K-562	-4.33	-7.91	< -8.00	< -8.30	-8.30	< -8.30	< -8.30	< -8.30		< -8.30	-10.83
MOLT-4	-4.73	-7.92	< -8.00	-8.18	-5.15	—	< -8.30	-7.66		—	-11.07
RPMI-8226	> -4.00	< -8.00	—	< -8.30	< -8.30	< -8.30	—	—	< -8.30	< -8.30	< -13.00
SR	> -4.00	< -8.00	—	—	—	—	—	—	—	—	8.34
Non-Small Cell Lung Cancer											
A549/ATCC	-4.17	-7.29	-7.59	-4.42	-4.86	-5.28	-5.69	> -4.30	-5.98	< -8.30	—
EKVX	> -4.00	-7.12	—	-6.90	—	-4.99	-4.79	> -4.30	-8.19	< -8.30	—
HOP-62	> -4.00	-4.92	-4.66	-6.34	-5.69	< -8.30	-6.96	-6.23	—	-4.90	-9.67
HOP-92	> -4.00	-6.37	-4.93	-6.56	< -8.30	< -8.30	-7.43	-7.00	-8.04	< -8.30	—
NCI-H226	> -4.00	-4.82	-4.87	< -8.30	—	< -8.30	—	—	—	—	—
NCI-H23	> -4.00	-7.84	-4.83	< -8.30	< -8.30	< -8.30	-8.19	< -8.30	< -8.30	< -8.30	—
NCI-H322M	—	-4.97	-4.71	-5.11	-5.07	-6.03	-6.72	-5.03	-6.26	< -8.30	-10.12
NCI-H460	> -4.00	-7.35	-5.02	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-12.16
NCI-H522	—	-7.53	-7.33	-6.16	-4.91	< -8.30	-6.54	> -4.30	—	—	< -13.00
Renal Cancer											
786-0	> -4.00	-7.48	-4.87	-6.45	< -8.30	< -8.30	-7.17	-6.59	-7.81	< -8.30	-8.01
A498	> -4.00	-6.69	-4.81	—	—	—	—	—	-6.97	-7.20	-7.14
ACHN	> -4.00	-7.12	< -8.00	—	-6.67	—	—	—	< -8.30	< -8.30	—
CAKI-1	—	-7.05	-4.94	-6.84	—	< -8.30	-7.74	-6.91	-7.36	< -8.30	—
RXF 393	-4.08	-4.80	-4.89	< -8.30	-7.67	< -8.30	< -8.30	< -8.30	-5.20	—	-8.32
SN12C	-4.21	-7.46	-7.60	—	—	—	—	—	< -8.30	< -8.30	-9.53
TK-10	> -4.00	-6.60	-4.91	-6.92	< -8.30	< -8.30	-7.81	-7.22	< -8.30	< -8.30	-7.89
UO-31	-4.06	-7.46	-7.38	-5.46	-5.05	-5.91	-6.78	-5.84	< -8.30	< -8.30	-6.09
Prostate Cancer											
PC-3	-4.17	< -8.00	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-10.85
DU-145	—	-6.24	-4.83	—	—	—	—	—	-7.48	—	-9.38
Breast Cancer											
MCF7	> -4.00	-7.22	< -8.00	< -8.30	< -8.30	< -8.30	< -8.30	-7.37	< -8.30	< -8.30	-11.69
NCI/ADR-RES	—	-7.45	-4.75	-5.64	-5.61	-5.73	-6.84	-6.45	-7.60	-6.07	-8.48
MDA-MB231/ATCC	-4.20	-5.40	-4.99	-7.19	—	< -8.30	—	—	-7.64	< -8.30	-8.54
HS 578T	> -4.00	-4.82	-4.72	> -4.30	-4.36	-4.82	-5.15	> -4.30	< -8.30	< -8.30	—
MDA-MB-435	—	-7.50	—	—	< -8.30	< -8.30	-8.16	-6.97	-7.62	< -8.30	< -13.00
MDA-N	> -4.00	-7.54	—	-7.29	—	—	—	—	—	—	< -13.00
BT-549	-4.06	-7.11	< -8.00	-7.91	—	-6.67	-5.82	—	< -8.30	< -8.30	-9.31
T-47D	—	< 8.00	-7.23	< -8.30	—	< -8.30	< -8.30	< -8.30	< -8.30	—	-9.81
MG MID	—	-6.87	-5.97	-7.01	-7.02	-7.57	-7.37	-6.73	-7.38	-7.79	—
Delta	-4.07	1.13	2.03	1.29	1.28	-0.73	0.93	1.57	0.92	-5.1	-10.15
Range	0.73	3.20	3.47	4.00	3.94	3.48	3.52	4.00	3.22	4.00	8.00

TABLE IV

Panel/Cell Line	Log ₁₀ TGI ₅₀										
	Artem- isinin	Trioxane Dimers									Pacli- taxel
		4	5	8a	9	10c	11b	12b	13	14	
Leukemia											
CCRF-CEM	—	-4.66	> -4.00	> -4.30	> -4.30	-4.84	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
HL-60(TB)	> -4.00	-4.88	—	-7.61	—	< -8.30	-8.11	< -8.30	—	> -4.30	-4.53
K-562	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	-4.75	> -4.30	> -4.30	—	> -4.30	> -4.00
MOLT-4	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—	—	> -4.00
RPMI-8226	> -4.00	-6.91	> -4.00	-6.19	> -4.30	-5.03	—	—	> -4.30	> -4.30	> -4.00
SR	> -4.00	-4.58	—	—	—	—	—	—	—	> -4.30	> -4.00
Non-Small Cell Lung Cancer											
A549/ATCC	> -4.00	-4.71	-4.82	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.61	> -4.30	—
EKVX	> -4.00	-4.77	—	-5.47	—	-4.63	> -4.30	> -4.30	-5.08	> -4.30	—
HOP-62	> -4.00	-4.56	-4.26	> -4.30	> -4.30	-5.01	> -4.30	> -4.30	-4.91	> -4.30	-4.80
HOP-92	> -4.00	-4.65	-4.49	> -4.30	> -4.30	-5.05	> -4.30	> -4.30	-5.27	> -4.30	—
NCI-H226	> -4.00	-4.40	-4.49	-6.67	—	< -8.30	—	—	—	—	—
NCI-H23	> -4.00	—	-4.26	> -4.30	-4.65	-5.25	-4.63	> -4.30	-5.17	—	—
NCI-H322M	—	-4.63	-4.44	> -4.30	> -4.30	-4.98	> -4.30	> -4.30	-5.06	> -4.30	-4.46
NCI-H460	> -4.00	-4.71	-4.40	> -4.30	> -4.30	-5.11	> -4.30	> -4.30	-4.92	> -4.30	-4.92
NCI-H522	—	-4.88	-4.59	> -4.30	> -4.30	-4.34	> -4.30	> -4.30	-4.60	> -4.30	-11.20
Colon Cancer											
COLO 205	> -4.00	-6.43	-4.87	-6.29	—	< -8.30	-6.99	-5.28	> -4.30	—	—
HCC-2998	> -4.00	-5.39	-4.23	—	—	—	—	—	-4.91	-5.33	-4.77
HCT-116	> -4.00	—	> -4.00	> -4.30	> -4.30	-5.16	-4.40	> -4.30	-5.06	> -4.30	-4.82
HCT-15	> -4.00	-4.78	-4.53	-5.61	-5.01	-5.23	-6.70	-5.00	-8.19	-4.96	> -4.00
HT29	> -4.00	-4.67	> -4.00	> -4.30	-4.91	> -4.30	> -4.30	> -4.30	—	—	—
KM12	> -4.00	-4.79	-4.47	< -8.30	< -8.30	< -8.30	< -8.30	< -8.30	-5.04	> -4.30	-4.36
SW-620	> -4.00	-4.89	> -4.00	—	—	—	—	—	-5.05	> -4.30	> -4.00
CNS Cancer											
SF-268	—	-4.63	-4.53	> -4.30	> -4.30	-5.14	> -4.30	> -4.30	-4.80	> -4.30	—
SF-295	—	-4.80	-4.49	—	< -8.30	—	—	—	-4.77	> -4.30	—
SF-539	—	-4.72	-4.41	> -4.30	-4.50	-4.66	> -4.30	> -4.30	—	—	—
SNB-19	> -4.00	-4.50	-4.34	> -4.30	—	-4.96	> -4.30	> -4.30	> -4.30	—	> -4.00
SNB-75	> -4.00	-4.55	-4.04	> -4.30	> -4.30	-4.80	> -4.30	> -4.30	—	—	—
U251	> -4.00	-4.47	-4.61	> -4.30	-4.75	-5.22	-7.39	> -4.30	-5.08	> -4.30	-4.32
Melanoma											
LOX IMVI	—	-4.45	-5.47	—	—	—	—	—	< -8.30	< -8.30	-4.65
MALME-3M	-4.06	-4.74	-4.50	-6.82	-5.14	< -8.30	-7.46	-6.40	< -8.30	—	-4.46
M-14	> -4.00	-4.68	> -4.00	> -4.30	> -4.30	-4.91	> -4.30	> -4.30	-4.75	> -4.30	-4.62
SK-MEL-2	> -4.00	-4.92	-4.53	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—
SK-MEL-28	> -4.00	-4.54	-4.31	> -4.30	> -4.30	-4.63	> -4.30	> -4.30	-4.73	> -4.30	—
SK-MEL-5	> -4.00	-7.18	-4.64	—	—	—	—	—	-5.28	-6.49	—
UACC-257	> -4.00	-4.71	-4.90	> -4.30	> -4.30	-4.38	> -4.30	> -4.30	-4.65	> -4.30	-4.52
UACC-62	> -4.00	-4.74	-4.67	—	—	—	—	—	-5.07	> -4.30	-4.71
Ovarian Cancer											
IGROVI	> -4.00	-4.68	-4.57	> -4.30	> -4.30	> -4.30	-4.50	> -4.30	> -4.30	> -4.30	-4.19
OVCAR-3	—	-4.92	-4.45	-8.21	< -8.30	< -8.30	-8.07	—	-4.98	> -4.30	-4.55
OVCAR-4	—	-4.62	-4.55	—	—	—	—	—	-4.93	> -4.30	-4.19
OVCAR-5	> -4.00	-6.42	-4.33	-6.62	—	< -8.30	-6.95	-6.39	-5.14	> -4.30	-4.92
OVCAR-8	> -4.00	-4.69	-4.47	> -4.30	> -4.30	-4.48	> -4.30	> -4.30	> -4.30	> -4.30	—
SK-OV-3	—	-4.53	-4.15	> -4.30	—	-4.92	> -4.30	> -4.30	-4.72	> -4.30	—

Panel/Cell Line	Artem- isinin	Log ₁₀ TGI ₅₀ Trioxane Dimers									Pacli- taxel
		4	5	8a	9	10c	11b	12b	13	14	
Renal Cancer											
786-0	> -4.00	-4.83	-4.36	> -4.30	> -4.30	-5.08	> -4.30	> -4.30	-4.94	> -4.30	> -4.00
A498	> -4.00	-5.00	-4.51	---	---	---	---	---	-5.31	> -4.30	---
ACHN	> -4.00	-4.76	< -8.00	---	-4.74	---	---	---	-6.57	-6.61	-4.90
CAKI-1	---	-4.84	-4.54	> -5.99	---	-6.71	-6.91	-6.40	-5.11	-6.30	-4.04
RXF 393	> -4.00	-4.41	-4.53	-6.77	-4.45	-5.99	-7.60	< -8.30	-4.69	> -4.30	> -4.00
SN12C	> -4.00	-4.84	-4.72	---	---	---	---	---	-5.29	-6.02	-4.29
TK-10	> -4.00	-4.81	-4.49	-4.52	-4.60	-5.40	-5.86	-5.22	-5.34	---	---
UO-31	> -4.00	-5.77	-4.68	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-5.10	-4.49	---
Prostate Cancer											
PC-3	-4.00	-4.89	-4.49	> -4.30	> -4.30	-5.23	-5.75	-4.92	-5.11	> -4.30	> -4.00
DU-145	---	-4.72	-4.48	---	---	---	---	---	-5.00	> -4.30	> -4.00
Breast Cancer											
MCF7	> -4.00	-4.67	-4.88	> -4.30	> -4.30	-5.13	> -4.30	> -4.30	-5.03	> -4.30	-4.05
NCI/ADR-RES	---	-6.41	> 4.00	> -4.30	> -4.30	-5.09	> -4.30	> -4.30	-5.18	> -4.30	> -4.00
MDA-MB231/ATCC	> -4.00	-4.33	-4.57	-6.35	---	< -8.30	---	---	-6.32	-7.94	-4.84
HS 578T	> -4.00	> -4.00	-4.08	> -4.30	> -4.30	-4.53	> -4.31	> -4.30	-5.00	> -4.30	---
MDA-MB-435	---	-4.78	-4.54	> -4.30	> -4.30	-5.17	> -4.30	> -4.30	-4.94	> -4.30	---
MDA-N	> -4.00	-4.90	---	---	---	---	---	---	---	---	---
BT-549	> -4.00	-4.73	-4.81	-7.39	---	-5.54	-5.32	> -4.30	-5.07	> -4.30	-6.32
T-47D	> -4.00	-4.73	-4.48	-4.77	---	---	> -4.30	> -4.30	-5.36	> -4.30	-4.05
MG MID	---	-4.87	-4.50	-5.04	-4.72	-5.51	-5.12	-4.81	-5.15	-4.66	---
Delta	-4.00	2.31	3.50	3.26	3.58	2.79	3.18	3.49	3.15	3.64	-4.54
Range	0.06	3.18	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	7.20

TABLE V

Panel/Cell Line	Log ₁₀ LC ₅₀										
	Artemisinin	Troxane Dimers									Paclitaxel
		4	5	8a	9	10c	11b	12b	13	14	
Leukemia											
CCRF-CEM	—	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
HL-60(TB)	> -4.00	-4.13	> -4.00	> -4.30	> -4.30	> -4.30	-4.55	> -4.30	—	> -4.30	> -4.53
K-562	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—	> -4.30	> -4.00
MOLT-4	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—	—	> -4.00
RPMI-8226	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	—	—	> -4.30	> -4.30	> -4.00
SR	> -4.00	> -4.00	—	—	—	—	—	—	—	> -4.30	> -4.00
Non-Small Cell Lung Cancer											
A549/ATCC	> -4.00	-4.11	-4.11	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—
EKVX	> -4.00	-4.36	—	> -4.30	—	> -4.30	> -4.30	> -4.30	-4.59	> -4.30	—
HOP-62	> -4.00	-4.20	> -4.00	> -4.30	> -4.30	-4.62	> -4.30	> -4.30	-4.54	> -4.30	-4.10
HOP-92	> -4.00	-4.22	-4.05	> -4.30	> -4.30	-4.52	> -4.30	> -4.30	-4.74	> -4.30	—
NCI-H226	> -4.00	> -4.00	-4.11	> -4.30	—	-4.82	—	—	—	—	—
NCI-H23	> -4.00	-4.49	> -4.00	> -4.30	> -4.30	-4.77	> -4.30	> -4.30	-4.62	> -4.30	—
NCI-H322M	—	-4.29	-4.16	> -4.30	> -4.30	-4.57	> -4.30	> -4.30	-4.59	> -4.30	> -4.00
NCI-H460	> -4.00	-4.19	> -4.00	> -4.30	> -4.30	-4.62	> -4.30	> -4.30	-4.36	> -4.30	> -4.00
NCI-H-522	—	-4.40	-4.17	> -4.30	> -4.30	> -4.30	> -4.30	< -4.30	> -4.30	> -4.30	> -4.00
Colon Cancer											
COLO 205	> -4.00	-4.50	-4.15	-4.78	—	-5.55	-5.28	> -4.30	> -4.30	> -4.30	> -4.41
HCC-2998	> -4.00	-4.59	> -4.00	—	—	—	—	—	-4.55	> -4.30	-4.26
HCT-116	> -4.00	-4.57	> -4.00	> -4.30	> -4.30	-4.72	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
HCT-15	> -4.00	-4.06	-4.03	> -4.30	> -4.30	-4.70	> -4.30	> -4.30	-4.81	> -4.30	> -4.00
HT29	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—	—	-4.39
KM12	> -4.00	-4.29	> -4.00	-4.60	-4.36	-5.23	-5.95	-5.02	-4.37	> -4.30	> -4.00
SW-620	> -4.00	-4.44	> -4.00	—	—	—	—	—	-4.53	> -4.30	> -4.00
CNS Cancer											
SF-268	—	-4.23	-4.12	> -4.30	> -4.30	-4.59	> -4.30	< -4.30	-4.31	> -4.30	—
SF-295	—	-4.39	-4.04	—	-4.60	—	—	—	-4.33	> -4.30	—
SF-539	—	-4.34	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.56	> -4.30	> -4.00
SNB-19	> -4.00	-4.13	> -4.00	> -4.30	—	-4.44	> -4.30	> -4.30	> -4.30	—	> -4.00
SNB-75	> -4.00	-4.12	> -4.00	> -4.30	> -4.30	-4.32	> -4.30	> -4.30	—	—	> -4.00
U251	> -4.00	> -4.00	-4.08	> -4.30	> -4.30	-4.72	> -4.30	—	-4.69	> -4.30	-4.15
Melanoma											
LOX IMVI	—	> -4.00	-4.44	—	—	—	—	—	-5.78	-4.39	> -4.15
MALME-3M	> -4.00	-4.28	-4.11	> -4.30	> -4.30	-4.80	> -4.30	> -4.30	-4.66	> -4.30	-4.11
M14	> -4.00	-4.14	> -4.00	> -4.30	> -4.30	-4.42	> -4.30	> -4.30	> -4.30	> -4.30	-4.13
SK-MEL2	> -4.00	-4.46	-4.12	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
SK-MEL-28	> -4.00	-4.12	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.34	> -4.30	—
SK-MEL-5	> -4.00	-4.53	-4.21	—	—	—	—	—	-4.73	> -4.30	—
UACC-257	> -4.00	-4.30	-4.36	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.03
UACC-62	> -4.00	-4.29	-4.24	—	—	—	—	—	-4.66	> -4.30	-4.19
Ovarian Cancer											
IGROVI	> -4.00	-4.32	-4.17	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
OVCAR-3	—	-4.42	> -4.00	> -4.30	> -4.30	-5.01	> -4.30	> -4.30	-4.51	> -4.30	> -4.00
OVCAR-4	—	-4.16	-4.12	—	—	—	—	—	-4.42	> -4.30	> -4.00
OVCAR-5	> -4.00	-4.54	> -4.00	> -4.30	> -4.30	-4.88	> -4.30	> -4.30	-4.66	> -4.30	> -4.00
OVCAR-8	> -4.00	-4.03	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
SK-OV-3	—	-4.21	> -4.00	> -4.30	—	-4.47	> -4.30	> -4.30	-4.31	> -4.30	—

Panel/Cell Line	Log ₁₀ LC ₅₀										
	Artemisinin	Trioxane Dimers									Paclitaxel
Renal Cancer											
786-0	> -4.00	-4.31	> -4.00	> -4.30	> -4.30	-4.64	> -4.30	> -4.30	-4.43	> -4.30	> -4.00
A498	> -4.00	-4.50	-4.21	—	—	—	—	—	-4.74	> -4.30	-4.13
ACHN	> -4.00	-4.30	4.47	—	> -4.30	—	—	—	-4.79	> -4.30	-4.45
CAKI-1	—	-4.38	-4.13	> -4.30	> -4.30	-4.93	-4.44	-4.47	-4.66	> -4.30	> -4.00
RXF 393	> -4.00	-4.01	-4.18	> -4.30	> -4.30	-4.73	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
SN12C	> -4.00	-4.35	-4.26	—	—	—	—	—	-4.72	> -4.30	> -4.00
TK-10	> -4.00	-4.38	-4.06	> -4.30	> -4.30	-4.62	> -4.30	> -4.30	-4.53	> -4.30	—
UO-31	> -4.00	-5.14	-4.26	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	—
Prostate Cancer											
PC-3	> -4.00	-4.34	> -4.00	> -4.30	> -4.30	-4.62	> -4.30	> -4.30	-4.63	> -4.30	> -4.00
DU-145	—	-4.36	-4.13	—	—	—	—	—	-4.6	> -4.30	> -4.00
Breast Cancer											
MCF7	> -4.00	-4.03	-4.22	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.44	> -4.30	> -4.00
NCI/ADR-RES	—	-4.52	> -4.00	> -4.30	> -4.30	-4.63	> -4.30	> -4.30	-4.65	> -4.30	> -4.00
MDA-MB231/ATCC	> -4.00	> -4.00	-4.15	> -4.30	—	-5.02	—	—	-4.77	> -4.30	-4.29
HS 578T	> -4.00	> -4.00	> -4.00	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	-4.35	> -4.30	—
MDA-MB-435	—	-4.30	-4.09	> -4.30	> -4.30	-4.60	> -4.30	> -4.30	-4.46	> -4.30	—
MDA-N	> -4.00	-4.45	—	—	—	—	—	—	—	—	—
BT-549	> -4.00	-4.31	-4.30	> -4.30	—	-4.61	> -4.30	> -4.30	-4.61	> -4.30	> -4.00
T-47D	> -4.00	-4.17	> -4.00	> -4.30	—	> -4.30	> -4.30	> -4.30	> -4.30	> -4.30	> -4.00
MG MID	—	-4.26	-4.09	-4.32	-4.31	-4.55	-4.37	-4.32	-4.51	-4.30	—
Delta	-4.00	0.88	0.37	0.46	0.29	1.00	1.58	0.70	1.27	0.08	-4.06
Range	0.00	1.14	0.47	0.48	0.30	1.25	1.65	0.72	1.49	0.09	.045

The data indicate that alcohol and diol dimers **4** and **5** are selectively and strongly inhibitory to leukemia cell line RPM I-8226, to melanoma cell line SK-MEL-5, and to renal cancer cell line ACHN. The epoxide dimer **6** and the ketone dimer **7** are less inhibitory. The data further indicate that the carboxylic acid derivatives **8a**, **9**, **10c**, **11b**, and **12b** are also selectively and strong inhibitory to colon cancer KM12, ovarian cancer OVCAR 3, and to breast cancer MDA-MB-435. The data lastly indicate that the acetic acid derivative **12b** is highly selective and especially inhibitory.

The dimers **4**, **5**, **8a**, **9**, **10c**, **11b** and **12b** of the present invention in most instances are as potent and in some instances more potent than paclitaxel. The data in Tables III, IV and V are graphically represented in Figures 7a, 7b, 7c, 7d, 7e, 7f, 7g, 7h, 7i, 7j and 7k through Figures 15j. Dose response curves, shown in the above mentioned Figures, are obtained by exposing various cancer cell lines to compounds having a known concentration ([log₁₀M]), as discussed in detail above, and then plotting the

percentage growth of each cell line for each concentration. The drug concentration limits that are tested are between 10^{-4} or -4.00M and 10^{-9} or -9.00M. The -4.00M value being the high concentration and the -9.00M value being the low concentration. Percentage growth is determined by dividing the number or mass of cells in the test well by the number or mass of cells in a control well. Referring to the leukemia cell line MOLT-4 in Figures 7a, 7b, 7c, 7d, 7e, 7f, 7g, 7h, 7i, 7j and 7k the first comparison that is made between artemisinin, paclitaxel, and the dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, and **12b** of the present invention are the drug concentrations which are necessary to inhibit growth, graphically represented in Figures 7a, 7b, 7c, 7d, 7e, 7f, 7g, 7h, 7i, 7j and 7k as the concentration necessary to achieve the percentage growth value of +50. As discussed previously, the five drug dilutions routinely tested range from 10^{-4} to 10^{-9} molar. Therefore, concentrations less than or greater than 10^{-9} and 10^{-4} molar, respectively, that are required to achieve a desired result are not determined. Referring now to Figure 7a, some concentration of paclitaxel that is less than 10^{-8} M is necessary to achieve primary growth inhibition; in fact the lower concentrations have been determined for this drug and the concentration at which primary growth inhibition occurs using paclitaxel is at 10^{-11} molar. Figure 7b indicates that some concentration of artemisinin that is greater than 10^{-4} molar is necessary to achieve primary growth inhibition. Referring to the alcohol and diol trioxane dimers **4** and **5**, dose response curves in Figures 7c and 7d, respectively, the leukemia cell line MOLT-4 displays primary growth inhibition at drug concentrations that are less than 10^{-7} and less than 10^{-8} molar, respectively. Referring to the carboxylic acid derivatives **8a**, **9**, **10c**, **11b** and **12b**, dose response curves in Figures 7e, 7f, 7g, 7h and 7i, respectively, the leukemia cell line MOLT-4 displays primary growth inhibition at drug concentrations that are less than 10^{-8} , 10^{-5} , 10^{-8} and 10^{-7} molar respectively. The drug concentration at which artemisinin is considered cytostatic, i.e., percentage growth is equal to 0, is at a concentration greater than 10^{-4} molar. The dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, and **12b** reach cytostasis at some concentration greater than 10^{-4} M, while the paclitaxel concentration necessary to achieve cytostasis is also some value

greater than 10^{-4} M. Cytotoxicity, i.e., the concentration for which the percentage growth is equal to -50, occurs at a concentration greater than 10^{-4} M for paclitaxel, artemisinin, for both alcohol and diol trioxane dimers **4** and **5**, and for carboxylic acid dimers **8a**, **9**, **10c**, **11** and **12b**, respectively.

5 The potency of the dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, and **12b**, respectively, of the present invention as compared to artemisinin and paclitaxel varies from cell line to cell line. The mean values for each drug are presented at the end Tables III, IV and V and the dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, and **12b** of the present invention are more potent than artemisinin and equivalent to and in
10 many instances higher in potency than paclitaxel.

 The dihydroartemisinin dimer disclosed by M. Cao, *et al.*, and tested by D.L. Klayman and H.J. Woerdenbag, discussed previously, was approximately twenty-two times more potent at causing 50% growth inhibition in one cancer cell line than artemisinin. With respect to the drug
15 concentrations causing 50% growth inhibition, the dimers **4**, **5**, **8a**, **9**, **10c**, **11b**, and **12b** were at least 100 times more potent than artemisinin. When interpreting the mean values, it is important to take into consideration that drug concentrations less than 10^{-9} M and greater than 10M were not collected, and this factor is reflected in the range.

20 For a further comparison on the effects of the trioxane dimers of the present invention on various cancer cell lines versus the effects of artemisinin and paclitaxel on the same cell lines see Figures 8a, b, c, d, e, f, g, h, i, j and k for non-small cell lung cancer cell lines, Figures 9a, b, c, d, e, f, g, h, i, j and k for colon cancer cell lines, Figures 10 a, b, c, d, e, f, g, h, i, j and k for CNS
25 cancer cell lines, Figures 11a, b, c, d, e, f, g, h, i, j and k for melanoma cancer cell lines, Figures 12a, b, c, d, e, f, g, h, i, j and k for ovarian cancer cell lines, Figures 13a, b, c, d, e, f, g, h, i, j and k for renal cancer cell lines, Figures 14a, b, c, d, e, f, g, h, i, and j for prostate cancer cell lines and Figures 15a, b, c, d, e, f, g, h, i, and j for breast cancer cell lines.

30

Acute Toxicity Study of Three Anti-Malarial Compounds in Male CD-1 Mice

The purpose of this study was to determine the relative toxicity of four structurally similar anti-malarial compounds (**13**, **16**, **19**, and sodium artesunate) following a single intraperitoneal (ip) injection. This study was not performed in compliance with the U.S. FDA "Good Laboratory Practice for Nonclinical Laboratory Studies" (GLP) as described in 21 CFR Part 58; however, documentation of all procedures and quality control checking of data was performed as for GLP studies.

MATERIALS AND METHODS

A. Test Article and Dose Preparation

Compounds **13**, **16**, **19** and sodium artesunate (Mepha Ltd., Lot No. 1), were provided by NIAID via McKesson BioServices HBOC (Rockville, MD). Each compound was dissolved in DMSO (Mallinkrodt Lot No. V18H15) to achieve a concentration of 200 mg/ml, and then 3 parts sesame oil (Spectrum Lot No. MI0656) was added to make a working concentration of 50 mg/ml for dose administration. Stability, strength, and uniformity of the test articles in the dose formulations were not determined for this study.

B. Test System

Forty-two CD1 male mice purchased from Charles River Laboratories (Wilmington, MA) were used in the study. Mice were quarantined for 3 days prior to initiation of the study. General procedures for animal care and housing were in accordance with the National Research Council (NRC) *Guide for the Care and Use of Laboratory Animals* (1996) and the animal welfare standards incorporated in 9 CFR Part 3, 1991. Mice were approximately 6 weeks old and weighed 25.2-31.2 g at study initiation. They were individually housed under a 12 hr light-dark cycle, with a temperature range of 68-72°F and 33-67% humidity. Purina Certified Rodent Chow #5002 and purified tap water were available *ad libitum*.

C. Experimental Design and Data Collection

Mice were weighed and randomized into treatment groups on the day prior to the first dose administration. Due to limitations in the amount of

available test materials, fewer mice in 1000 mg/kg groups were treated than specified in the protocol. The numbers of animals intended to be used and actually used with each compound at each dose level are shown in the table below; the only differences are in the high-dose groups.

5 Mice were administered either compounds **13**, **16**, **19** or sodium artesunate once ip at does levels at 125, 250, 500, or 1000 mg/kg. Control animals were administered a vehicle solution (25% DMSO and 75% sesame oil) at a volume of 20 ml/kg. Surviving animals were sacrificed on Day 8 and blood was collected for clinical pathology evaluations.

10 Clinical signs were observed daily, including evaluation of the injection site 1 to 2 hr. after treatment on Day 1 and once daily on Days 2-8. Mortality and morbidity were checked twice daily on weekdays and once daily on the weekend, see Table VI. Animals were weighed daily and prior to necropsy.

Table VI

Mortality Data

	750 mg/kg	500 mg/kg	250 mg/kg	125 mg/kg
Control		No death (12.5 and 18.75 ml/kg)		
Na Artesunate	-----	3/3 dead	1/3 dead	No death
13	2/3 dead	3/3 dead	No death	No death
16	3/3 dead	3/3 dead	No death	No death
19	3/3 dead	1/3 dead	No death	No death

Table VII

**Efficacy Study
Blood Schizontocidal Activity**

Compound	Route : IV		Route : PO	
	ED ₅₀	ED ₉₀	ED ₅₀	ED ₉₀
Na Artesunate	5.0 (2.9-7.5)	40.0 (23.0-60.0)	6.0 (4.0-10.0)	60.0 (40.0-100.0)
13	0.55 (0.1-0.8)	2.0 (0.8-3.2)	4.0 (2.0-7.0)	9.0 (4.3-17.0)
16	60.0	1000.0		
12b	5.2 (4.9-6.0)	10.0 (6.9-15.0)	6.0 (2.7-13.0)	19.0 (8.0-39.0)
19	1.7 (1.1-4.9)	4.5 (3.0-11.0)	5.5 (2.9-13.0)	18.0 (8.0-38.0)

In conclusion, new C-10 non-acetal trioxane dimer **3**, easily prepared on gram scale and thermally stable, can be used to make a diverse series of 3-carbon atom linked, oxygenated dimers **4-7** without destroying the critical pharmacophore peroxide bond. Each of the new trioxane dimers **4**, **5**, and **7** is 10 times more antimalarially potent *in vitro* than the natural trioxane artemisinin (**1**), and alcohol and diol dimers **4** and **5** are strongly inhibitory but not cytotoxic toward several human cancer cell lines. Moreover, water-soluble trioxane dimers **8a-10c** and **12b** are orally active new antimalarials that are more efficacious than artelinic acid and than sodium artesunate in mice. These semi-synthetic new chemical entities **4** and **5** and especially **8a-10c** and **12b**, therefore, deserve further preclinical evaluation as potential drug candidates for chemotherapy of malaria and cancer. Dimers **13** and **19** are more efficacious (when administered both orally and i.v.) and less toxic (when administered intraperitoneally to mice as a single dose) than clinically-used sodium artesunate, thereby giving them a better antimalarial therapeutic index than sodium artesunate.

The invention is further illustrated by the following non-limited examples. All scientific and technical terms have the meanings as understood by one with ordinary skill in the art. The specific examples which follow illustrate the synthesis of representative compounds of the instant invention and are not to be construed as limiting the invention in sphere or scope. The methods may be adapted to variation in order to produce compounds embraced by this invention but not specifically disclosed. Further, variations of the methods to produce the same compounds in somewhat different fashion will be evident to one skilled in the art. The synthetic descriptions and specific examples that follow are only intended for the purposes of illustration, and are not to be construed as limiting in any manner to make compounds of the present invention by other methods.

EXAMPLES

Unless otherwise noted, reactions were run in oven-dried glassware under an atmosphere of argon. Diethyl ether (ether) and tetrahydrofuran (THF) were distilled from sodium benzophenone ketyl prior to use. Methylene

chloride (CH_2Cl_2) was distilled from calcium hydride prior to use. All other compounds were purchased from Aldrich Chemical Company and used without further purification. Analytical thin-layer chromatography (TLC) was conducted with Silica Gel 60 F254 plates (250 micrometer thickness, Merck).

5 Column chromatography was performed using flash silica gel (particle size 400-230 mesh). Yields are not optimized. Purity of final products was judged to be >95% based on their chromatographic homogeneity. High performance liquid chromatography (HPLC) was carried out with a Rainin HPLX system equipped with two 25 mL/min preparative pump heads using a Rainin

10 Dynamax 10 mm x 250 mm (semi-preparative) column packed with 60 Å silica gel (8 μm pore size) as bare silica. Melting points were measured using a Mel-Temp metal-block apparatus and are uncorrected. Nuclear magnetic resonance (NMR) spectra were obtained on a Varian XL-400 spectrometer, operating at 400 MHz for ^1H , 162 MHz for ^{31}P 100 MHz for ^{13}C . ^1H and ^{13}C

15 chemical shifts are reported in parts per million (ppm) downfield from tetramethylsilane, ^{31}P chemical shifts are measured using 85% phosphoric acid as an external reference. Splitting patterns are described as singlet(s), doublet(d), triplet(t), quartet(q), quintet (qt), sextet (st), multiplet (m) and broad (br). Infrared (IR) spectra were obtained using a Perkin-Elmer 1600 FT-IR

20 spectrometer. Resonances are reported in wavenumbers (cm^{-1}). Low and high resolution mass spectra (LRMS and HRMS) were obtained with electronic or chemical ionization (EI or CI) either (1) at Johns Hopkins University on a VG Instruments 70-S Spectrometer run at 70 eV for EI and run with ammonia (NH_3), butane (C_4H_{10}) or methane (CH_4) as carrier gas for

25 CI or (2) at Ohio State University with a 3-Tesla Finnigan FTMS-2000 Fourier Transform mass spectrometer. Samples were sprayed from a commercial Analytica electrospray ionization source, and then focused into the FTMS cell using a home-built set of ion optics. For ESI analysis, most compounds were sprayed from a micromolar concentration of the analyte in various solvent

30 mixtures, such as tetrahydrofuran/ CH_3OH , with added NaCl. This process generated the sodiated molecular ion (usually as the singly-charged species), denoted as $(\text{M}+\text{Na})^+$. But in some cases, acetic acid or trifluoroacetic acid was used to generate the protonated molecular ion $(\text{M}+\text{H})^+$ instead. Electron

impact (EI) ionization was performed with a Kratos MS-25, using 70 eV ionization conditions. Combustion analyses were conducted by Atlantic Microlab (Norcross, GA). Various methods of purifying the products of the present invention are known and understood by those skilled in the art and the purification methods presented in the Examples is solely listed by way of example and is not intended to limit the invention.

Example 1

Synthesis of bis-trioxane O-allyl ether 12a and bis-trioxane O-acetic acid 12b

10 Synthesis of bis-trioxane O-allyl ether 12a.

To a solution of bis-trioxane primary alcohol 4 (200 mg, 0.33 mmol) in anhydrous tetrahydrofuran (20mL) at -78° C was added KHMDS (0.5 M sol. in toluene, 2 mL, 1 mmol) in a dropwise manner. The reaction was stirred at -78° C for 30 mins, at which time allyl bromide (0.4 mL, 1 mmol) and 18-crown-6 (catalytic) were added. The reaction was then stirred at -78° C for a further 1 hour, at which time TLC analysis showed complete consumption of starting material. The reaction was quenched with water (5 mL) and organics were extracted with methylene chloride (3 x 20 mL), dried (MgSO₄) and concentrated *in vacuo*. Column chromatography on silica eluting with 10-12% ethyl acetate/hexanes isolated bis-trioxane O-allyl ether 12a as a viscous oil (204 mg, 96%). ¹H NMR (400 MHz, CDCl₃) δ 5.98 - 5.88 (m, 1H), 5.32 (s, 2H), 5.28 - 5.23 (m, 1H), 5.14 - 5.10 (m, 1H), 4.34 - 4.28 (m, 1H), 4.24 - 4.17 (m, 1H), 4.04 - 3.90 (m, 2H), 3.61 - 3.54 (m, 2H), 2.76 - 2.60 (m, 2H), 2.38 - 3.26 (m, 2H), 2.10 - 1.20 (m), 0.99 - 0.81 (m, 14H); ¹³C NMR (100 MHz, CDCl₃) δ 135.4, 116.1, 103.1, 102.9, 88.9, 88.4, 81.1, 74.7, 72.7, 71.7, 71.5, 52.4, 52.2, 44.6, 44.4, 37.3, 36.5, 35.6, 34.5, 34.4, 30.5, 30.4, 29.9, 29.6, 26.1, 26.0, 24.9, 24.6, 24.5, 20.2, 20.1, 13.3, 13.0.

25 Synthesis of bis-trioxane O-acetic acid 12b.

To a solution of bis-trioxane O-allyl ether 12a (115 mg, 0.178 mmol) in ethyl acetate (3.0 mL), acetonitrile (3.0 mL) and water (1.0 mL) was added ruthenium (III) chloride hydrate (7.4 mg, 0.036 mmol) and sodium periodate (266 mg, 1.25 mmol) (on addition of ruthenium chloride the solution turned

black). The reaction was stirred at room temperature for 30 mins (the color of the solution turned to pale orange) before the reaction mixture was poured into a mixture of ethyl acetate (30 mL) and saturated aqueous NH_4Cl solution (20 mL). Organics were extracted with ethyl acetate (2 x 30 mL), filtered through a pad of celite, dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography (1% acetic acid in 40% ethyl acetate/hexane) isolated bis-trioxane O-acetic acid **12b** as a white solid (52.0 mg, 0.078 mmol, 44 %). Mp = 86 - 90 °C; ^1H NMR (400 MHz, CDCl_3) δ 5.33 (s, 2H), 4.46 - 4.39 (m, 1H), 4.38 - 4.33 (m, 1H), 4.15 (d, 1H, J = 16.4), 4.03 (d, 1H, J = 16.4), 3.86 (dd, 1H, J = 8.8, J = 5.2), 3.57 (dd, 1H, J = 8.8, J = 4.4), 2.71 - 2.52 (m, 2H), 2.37 - 2.29 (m, 2H), 2.08 - 1.18 (m, 30H, including two singlets at 1.39 and 1.38), 0.98 - 0.85 (m, 2H), 0.94 (apparent doublet, 6H, J = 6.0), 0.86 (d, 3H, J = 7.2), 0.85 (d, 3H, J = 7.2); ^{13}C NMR (100 MHz, CDCl_3) δ 171.7, 103.4, 103.1, 89.4, 89.2, 81.1, 81.0, 74.3, 73.6, 71.5, 69.1, 52.2, 52.1, 44.3, 44.1, 37.5, 36.5, 36.0, 34.4, 30.7, 30.7, 30.6, 26.0, 25.9, 24.8, 24.7, 20.2, 20.1, 12.9, 12.8; IR (film, cm^{-1}) 3506 (br), 2939, 2875, 1759, 1739, 1451, 1377, 1279, 1251, 1206, 1187, 1125, 1052, 1011, 929, 910, 826, 731; HRMS(ES) m/z calcd for $\text{C}_{36}\text{H}_{56}\text{O}_{11}\text{Na}$ ($M+\text{Na}$) 687.3715, found 687.3695.

Example 2

Synthesis of α -dihydroartemisinin acetate 2

To a solution of artemisinin **1** (565 mg, 2.00 mmol) in anhydrous methylene chloride (15.0 mL) at -78°C was added DIBAL (1.0 M in toluene, 2.4 mL, 2.4 mmol) in a dropwise manner. The reaction was stirred at -78°C for one hour, at which time TLC analysis confirmed complete consumption of starting material. Pyridine (0.50 mL, 6.18 mmol), 4-(dimethylamino)-pyridine (292 mg, 2.4 mmol) and finally acetic anhydride (0.760 mL, 8.05 mmol) were added and the reaction was stirred vigorously at -78°C for 3 hours before being allowed to warm to room temperature and stir overnight. The reaction was then quenched with saturated NH_4Cl solution (20 mL) and organics were extracted with methylene chloride (3 x 20 mL), washed with brine, dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 14% ethyl acetate/hexanes isolated α -dihydroartemisinin acetate

2 as a white solid (600 mg, 1.84 mmol, 92%) with the following characteristic peaks: ^1H NMR (400 MHz, CDCl_3) δ 5.78 (d, 1H, $J = 9.6$), 5.44 (s, 1H), 2.60 - 2.51 (m, 1H), 2.41 - 2.33 (m, 1H), 2.12 (s, 3H), 1.43 (s, 3H), 0.96 (d, 3H, $J = 6.0$), 0.84 (d, 3H, $J = 7.2$).

5

Example 3

Synthesis of trioxane isobutene dimer 3

A solution of dihydroartemisinin acetate 2 (DHA acetate) (872 mg, 2.67 mmol) and the allylic bis-silane linker (267 mg, 1.34 mmol) in methylene chloride (50 mL) was cooled to -78°C . Tin tetrachloride (1M solution in CH_2Cl_2 , 2.67 mL, 2.67 mmol) was further diluted with CH_2Cl_2 (3 mL) and was added to the reaction mixture dropwise using a syringe pump at the rate of 6 ml/hour. The reaction was stirred at -78°C for a further 45 minutes at which time TLC analysis confirmed complete consumption of starting material. Saturated ammonium chloride solution (10 mL) was then added and the reaction was allowed to warm to room temperature naturally. Organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo*. Gradient column chromatography on silica eluting with 5%, 7%, 8% and finally 10% ethyl acetate/hexanes isolated trioxane isobutene dimer 3 as a white solid (564 mg, 0.96 mmol, 71%). $\text{Mp} = 132$ - 133°C ; ^1H NMR (400 MHz, CDCl_3) δ 5.41 (s, 2H), 4.87 (s, 2H), 4.30 (ddd, 2H, $J=10.4$, $J = 5.6$, $J = 4.0$), 2.75 - 2.67 (m, 2H), 2.61 - 2.54 (m, 2H), 2.37 - 2.23 (m, 4H), 2.04 - 1.98 (m, 2H), 1.91 - 1.84 (m, 2H), 1.81 - 1.75 (m, 2H), 1.67 - 1.56 (m, 4H), 1.52-1.32 (m, 6H), 1.39 (s, 6H), 1.23 - 1.20 (m, 2H), 0.94 (d, 6H, $J = 6.0$), 0.90 (d, 6H, $J = 7.6$), 0.98 - 0.86 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 147.4, 113.1, 103.1, 88.6, 81.2, 75.5, 52.4, 44.5, 37.1, 36.7, 35.2, 34.5, 30.6, 26.2, 24.7, 24.6, 20.2, 13.3; IR (film, cm^{-1}) 3060, 2938, 2875, 1641, 1451, 1375, 1121, 1092, 1054, 1007, 878, 732; HRMS(ES) m/z calcd for $\text{C}_{34}\text{H}_{52}\text{O}_8\text{Na}$ ($\text{M}+\text{Na}$) 611.3560, found 611.3555; Anal ($\text{C}_{34}\text{H}_{52}\text{O}_8$) C, H.

30

Example 4

Synthesis of bis-trioxane primary alcohol 4

A solution of trioxane isobutene dimer 3 (0.89 g, 1.51 mmol) in anhydrous tetrahydrofuran (25 mL) was cooled to 0°C . Borane-dimethyl

sulfide complex ($\text{BH}_3\cdot\text{DMS}$) (2.0 M solution in diethyl ether, 0.9 mL, 1.80 mmol) was carefully added and the reaction was allowed to warm to room temperature and stir for 3 hours. At this time TLC analysis confirmed that no starting material remained. A suspension of sodium perborate $\cdot 4\text{H}_2\text{O}$ (NaBO₃·4H₂O) (1.17 g, 7.60 mmol) in water (12 mL) was slowly added and the resulting suspension was allowed to stir for 17 hours. Water (10 mL) and methylene chloride (50 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a white solid. Gradient column chromatography on silica eluting with 20%, 30% and finally 40% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol **4** as a white solid (0.85 g, 1.40 mmol, 93%). ¹H NMR (CDCl₃, 400 MHz) δ 5.35 (s, 1H), 5.34 (s, 1H), 4.47 – 4.40 (m, 1H), 4.33 (qt, 1H, *J* = 6.0), 3.83 – 3.75 (m, 1H), 3.67 – 3.60 (m, 1H), 3.17 (dd, 1H, *J* = 7.6, *J* = 6.1), 2.64 (qt, 1H, *J* = 6.9), 2.59 (qt, 1H, *J* = 6.9), 2.32 (t, br, 2H, *J* = 14.1), 2.06 – 1.97 (m, 3H), 1.95 – 1.87 (m, 2H), 1.86 – 1.74 (m, 2H), 1.70 – 1.55 (m, 8H), 1.50 – 1.30 (m, 14H, including singlet at 1.40), 0.99 – 0.90 (m, 2H), 0.95 (d, 6H, *J* = 5.8), 0.87 (apparent t, 6H, *J* = 6.9); ¹³C NMR (CDCl₃, 100 MHz) δ 103.12, 102.97, 89.47, 89.22, 81.11 (2), 73.86, 71.27, 65.12, 52.20, 52.07, 44.18, 44.01, 37.68, 37.46, 37.43, 36.52, 36.51, 34.40, 34.37, 31.26, 30.74 (2), 30.65, 25.95, 25.89, 24.83 (2), 24.73, 24.70, 20.15, 20.10, 12.89, 12.61; HRMS (EI, *m/z*) for C₃₄H₅₄O₉Na requires 629.3660, found 629.3697; IR (film, cm⁻¹) 3490; Mp. 81 – 82 °C; Anal (C₃₄H₅₆O₁₀) C, H.

Example 5

Synthesis of bis-trioxane vicinal diol **5**

To a solution of trioxane isobutene dimer **3** (21.0 mg, 0.036 mmol) and 4-methylmorpholine *N*-oxide (5.0 mg, 0.043 mmol) in acetone (2.0 mL) was added osmium tetroxide (25 mg/2 mL aqueous solution, 0.016 mL, catalytic) and the reaction was stirred vigorously at room temperature for 24 hrs. The reaction mixture was then quenched with saturated aqueous NaHSO₃ solution (2.0 mL) and stirred for an additional 30 mins (during which time the reaction turned a pale orange color). The reaction mixture was poured into a mixture of diethyl ether (20 mL) and saturated aqueous NH₄Cl solution (20 mL) and

organics were extracted with ethyl acetate (2 x 30 mL), washed with brine, dried (MgSO₄) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 50% ethyl acetate/hexanes) isolated bis-trioxane vicinal diol **5** as a white solid (20.3 mg, 0.033 mmol, 92%). Mp = 159 – 160 °C; ¹H NMR (400 MHz, CDCl₃) δ 5.36 (s, 1H), 5.35 (s, 1H), 4.74 - 4.70 (m, 1H), 4.56 - 4.51 (m, 1H), 4.09 (s, 1H), 3.71 - 3.62 (m, 2H), 3.12 (t, 1H, *J* = 7.2), 2.64 - 2.52 (m, 2H), 2.36 - 2.26 (m, 2H), 2.04 - 1.99 (m, 2H), 1.96 - 1.63 (m, 12H), 1.46 - 1.20 (m, 14H, including two singlets at 1.40 and 1.39), 0.96 (d, 3H, *J* = 6.0), 0.95 (d, 3H, *J* = 6.0), 0.89 (d, 3H, *J* = 7.6), 0.88 (d, 3H, *J* = 7.6), 0.98 - 0.86 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 103.0, 102.9, 89.7, 89.5, 81.1, 81.0, 74.7, 70.6, 70.5, 68.4, 52.0, 43.8, 43.7, 37.8, 37.5, 36.5, 34.9, 34.3, 31.0, 30.9, 25.9, 25.9, 24.9, 24.8, 24.8, 20.1, 12.5, 12.4; IR (film, cm⁻¹) 3499, 2951, 2876, 1453, 1378, 1207, 1108, 1054, 1009, 912, 878, 844, 732; HRMS(ES) *m/z* calcd for C₃₄H₅₄O₁₀Na (M+Na) 645.3609, found 645.3559.

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Example 6

Synthesis of bis-trioxane epoxide **6**

To a solution of trioxane isobutene dimer **3** (34.1 mg, 0.058 mmol) in anhydrous methylene chloride (10.0 mL) at -78 °C was rapidly added dimethyl dioxirane (DMDO) (0.08 M solution in acetone, 3.8 mL, 0.29 mmol). The reaction was stirred at -78 °C for 30 minutes before being allowed to warm to room temperature. The reaction was then concentrated *in vacuo*, giving a yellow oil. Flash column chromatography on silica eluting with 20% ethyl acetate/hexanes isolated bis-trioxane epoxide **6** as a white solid (28.1 mg, 0.047 mmol, 80%). Mp = 147-148 °C; ¹H NMR (400 MHz, CDCl₃) δ 5.48 (s, H), 5.43 (s, H), 4.45 (dd, 1H, *J* = 10.0, *J* = 6.0), 4.22 (dd, 1H, *J* = 10.0, *J* = 6.0), 2.82 - 2.72 (m, 2H), 2.68 - 2.62 (m, 3H), 2.43 - 2.33 (m, 3H), 2.02 - 1.97 (m, 2H), 1.88 - 1.82 (m, 2H), 1.75 - 1.70 (m, 2H), 1.62 - 1.58 (m, 2H), 1.54 - 1.26 (m, 11H), 1.39 (s, 6H), 0.98 - 0.88 (m, 3H), 0.93 (d, 6H, *J* = 6.4), 0.92 (d, 6H, *J* = 6.4), 0.80 (d, 6H, *J* = 7.6); ¹³C NMR (100 MHz, CDCl₃) δ 103.43, 103.40, 87.94, 87.93, 81.10, 81.07, 75.44, 74.37, 60.43, 54.46, 52.54, 52.50, 44.58, 36.91, 36.70, 34.40, 33.50, 33.01, 30.40, 30.36, 26.18, 24.45, 24.35, 24.32, 20.22, 20.19, 13.69, 13.51; IR (film, cm⁻¹) 2939, 2875, 1452, 1376,

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1280, 1208, 1188, 1123, 1091, 1057, 1006, 941, 878, 754; HRMS(ES) m/z calcd for $C_{34}H_{52}O_9Na$ ($M+Na$) 627.3509, found 627.3478.

Example 7

Synthesis of bis-trioxane ketone 7

5 To a solution of trioxane isobutene dimer **3** (30 mg, 0.051 mmol) in anhydrous N,N-dimethylformamide (25 μ L) was added OsO_4 (2.5% weight % in t-BuOH, 0.1 mol%). The reaction mixture was stirred for 5 minutes before Oxone® (63 mg, 0.2 mmol) was added in one portion. The reaction was then stirred for a further 2 hours, at which time TLC analysis confirmed full
10 consumption of starting material. The reaction was quenched with saturated Na_2SO_3 solution (10 mL) and water (10 mL) and stirred for 1 hour before organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 25% ethyl acetate/hexanes) isolated bis-trioxane ketone **7** as a white
15 solid (21 mg, 0.035 mmol, 70%). Mp = 120-121 °C; 1H NMR (400 MHz, $CDCl_3$) δ 5.34 (s, 2H), 4.75 (m, 2H), 2.90 (m, 2H), 2.75 (m, 2H), 2.59 (dd, 2H, $J = 16$, $J = 4$), 2.33 (dt, 2H, $J = 16$, $J = 9$), 2.04 - 1.96 (m, 2H), 1.94 (m, 2H), 1.79 (m, 2H), 1.68 - 1.57 (m, 6H), 1.48 - 1.22 (m, 14H), 0.99 (d, 6H, $J = 6.4$), 0.88 (d, 6H, $J = 7.2$); ^{13}C NMR (100 MHz, $CDCl_3$) δ 196.8, 103.8, 89.3, 71.6,
20 52.5, 44.5, 44.3, 37.7, 36.8, 34.7, 30.3, 26.3, 25.0, 24.8, 20.4, 13.8; IR (film, cm^{-1}) 1722; HRMS(ES) m/z calcd for $C_{33}H_{50}O_9Na$ ($M+Na$) 613.3347, found 613.3303.

Example 8

Synthesis of bis-trioxane primary succinate monoester 8a

25 To a solution of bis-trioxane primary alcohol **4** (50mg, 0.082 mmol) and succinic anhydride (24 mg, 0.24 mmol) in methylene chloride (10 mL) at 0 °C was added 4-(dimethylamino)-pyridine (10 mg, 0.082 mmol). The reaction was allowed to warm to room temperature and was then stirred for 12 hours, at which TLC analysis showed complete consumption of starting material.
30 Organics were extracted with ethyl acetate (3 x 30 mL), washed with brine (10 mL), dried ($MgSO_4$) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 50% ethyl acetate/hexanes to isolated

bis-trioxane primary succinate monoester **8a** as a white solid (40 mg, 0.068 mmol, 84%). ^1H NMR (400 MHz, CDCl_3) δ 5.39 (s, 1H), 5.31 (s, 1H), 4.39 - 4.32 (m, 2H), 4.28 - 4.18 (m, 3H), 2.74 - 2.54 (m, 6H), 2.38 - 2.24 (m, 2H), 2.22 - 2.12 (m, 1H), 2.06 - 1.96 (m, 2H), 1.95 - 1.85 (m, 2H), 1.84 - 1.60 (m, 11H), 1.48 - 1.20 (m, 16H), 1.00 - 0.80 (m, 12H); ^{13}C NMR (100 MHz, CDCl_3) δ 175.4, 172.0, 102.9, 103.6, 89.4, 88.5, 81.1, 74.1, 71.1, 67.1, 52.4, 52.1, 44.6, 44.1, 37.4, 37.3, 36.6, 36.5, 34.4, 34.3, 30.5, 30.4, 30.0, 29.7, 29.4, 29.0, 26.0, 25.9, 24.7, 20.3, 20.1, 13.3, 12.7; HRMS(ES) m/z calcd for $\text{C}_{38}\text{H}_{58}\text{O}_{12}\text{Na}$ (M+Na) 729.3820, found 729.3795.

Example 9

Synthesis of bis-trioxane primary alcohol isonicotinate **8b**

To a stirring suspension of bis-trioxane primary alcohol **4** (30.4 mg, 0.050 mmol) and isonicotinic acid (20.1 mg, 0.163 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (23.5 mg, 0.192 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide (EDC) hydrochloride (39.2 mg, 0.204 mmol). A further 1.5 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), saturated NaHCO_3 solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a white solid. Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 25% ethyl acetate/petroleum ether and then 30% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol isonicotinate **8b** as a white solid (32.6 mg, 0.046 mmol, 91%). $\text{Mp} = 74\text{-}78^\circ\text{C}$; $[\alpha]_{\text{D}}^{24.5} 77.8$ (CHCl_3 , $c = 0.06$); ^1H NMR (400 MHz, CDCl_3) δ 8.78 (s, br, 2H), 7.86 (s, br, 2H), 5.33 (s, 1H), 5.29 (s, 1H), 4.55 (s, 1H), 4.54 (s, 1H), 4.51-4.44 (m, 1H), 4.37-4.30 (m, 1H), 2.70 (st, 1H, $J = 7.0$), 2.59 (st, 1H, $J = 7.0$), 2.45-2.36 (m, br, 1H), 2.31 (td, 2H, $J = 14.0$, $J = 3.7$), 2.05-1.96 (m, 2H), 1.95-1.73 (m, 6H), 1.69-1.50 (m, 6H), 1.41-1.15 (m, 14H, including two singlets at 1.40 and 1.39), 0.98-0.92 (m, 2H), 0.96 (d, 3H, $J = 5.9$), 0.94 (d, 3H, $J = 5.9$), 0.88 (d, 3H, $J = 7.4$), 0.87 (d, 3H, $J = 7.4$); ^{13}C

NMR (100 MHz, CDCl_3) δ 164.99, 150.49 (2), 137.93, 122.91 (2), 103.16, 102.87, 89.52, 88.83, 81.12, 81.11, 73.23, 70.83, 68.10, 52.30, 52.07, 44.32, 44.06, 37.50, 37.47, 36.64, 36.54, 34.42, 34.39, 33.95, 30.61, 30.58, 30.49, 30.15, 26.02, 26.01, 24.97, 24.89, 24.76, 24.67, 20.17, 20.10, 13.08, 12.72;
5 IR (film, cm^{-1}) 2942, 1866, 1726, 1452, 1407, 1372, 1321, 1276, 1210, 1123, 1106, 1057, 1046, 1006, 931, 876, 756, 707; HRMS (EI, m/z) calcd for $\text{C}_{40}\text{H}_{57}\text{NO}_{10}\text{Na}$ ($M+\text{Na}$) 734.3875, found 734.3855; Anal ($\text{C}_{40}\text{H}_{57}\text{NO}_{10}$) C, H.

Example 10

Synthesis of tertiary alcohol primary succinate ester 9

10 To a solution of bis-trioxane vicinal diol 5 (52.9 mg, 0.085 mmol) in anhydrous methylene chloride (3.0 mL) was added succinic anhydride (25.5 mg, 0.255 mmol) and 4-(dimethylamino)-pyridine (10.4 mg, 0.085 mmol). The reaction was stirred for 24 hours at room temperature, at which time TLC analysis confirmed full consumption of starting material. The reaction mixture
15 was then poured into a mixture of methylene chloride (30 mL) and saturated NH_4Cl solution (30 mL). Organics were extracted with methylene chloride (3 x 20 mL), washed with brine, dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 2-propanol/methylene chloride/ethyl acetate (1:3:18) isolated tertiary alcohol primary succinate ester
20 9 as a white solid (52.0 mg, 0.071 mmol, 85%). $M_p = 95 - 97^\circ\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ 5.38 (s, 1H), 5.34 (s, 1H), 4.61 - 4.52 (m, 2H), 4.29 (Abq, 2H, $J_{AB} = 18.8$, $\Delta\nu_{AB} = 11.6$), 2.72 - 2.66 (m, 4H), 2.58 - 2.50 (m, 2H), 2.36 - 2.24 (m, 2H), 2.04 - 1.58 (m, 13H), 1.46 - 1.18 (m, 11H), 1.40 (s, 3H), 1.39 (s, 3H), 0.95 (d, 6H, $J = 6.0$), 0.88 (d, 3H, $J = 7.6$), 0.87 (d, 3H, $J = 7.6$), 0.98 -
25 0.86 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.4, 171.7, 103.2, 102.9, 89.6, 89.0, 81.1, 81.0, 73.3, 70.6, 70.3, 68.6, 52.1, 51.9, 44.1, 43.7, 37.5, 36.6, 36.5, 36.3, 35.1, 34.4, 34.3, 30.8, 30.7, 29.2, 29.1, 25.9, 25.8, 24.8, 24.7, 20.2, 20.1, 12.8, 12.5; IR (film, cm^{-1}) 3502, 2950, 2872, 1737, 1713, 1453, 1378, 1208, 1168, 1106, 1054, 1009, 942, 878, 844, 735; HRMS(ES) m/z
30 calcd for $\text{C}_{38}\text{H}_{58}\text{O}_{13}\text{Na}$ ($M+\text{Na}$) 745.3769, found 745.3726; $[\alpha]_D^{23.5} 56.6$ (CHCl_3 , $c = 0.16$); Anal ($\text{C}_{38}\text{H}_{58}\text{O}_{13}$) C, H.

Example 11

Synthesis of bis-trioxane β -hydroxysulfide ester 10a

To a solution of bis-trioxane epoxide **6** (80.0 mg, 0.132 mmol) in anhydrous diethyl ether (10.0 mL) was added methyl 4-mercaptobenzoate (44.4 mg, 0.264 mmol) and neutral aluminum oxide (1.0 g, type W 200 super I, Woelm Pharma, Germany). The resulting slurry was stirred at room temperature for 3 hours, at which time TLC analysis confirmed full consumption of starting material. The reaction mixture was then filtered through a pad of celite and the remaining solid was washed with ethyl acetate (2 x 30 mL) before concentration of the organics *in vacuo*. Flash column chromatography on silica eluting with 25% ethyl acetate/hexanes isolated bis-trioxane β -hydroxysulfide ester **10a** as a sticky solid (81.2 mg, 0.105 mmol, 80 %). ^1H NMR (400 MHz, CDCl_3) δ 7.67 (ABq, 4H, $J_{\text{AB}} = 8.4$, $\Delta\nu_{\text{AB}} = 154.4$), 5.36 (s, 1H), 5.35 (s, 1H), 4.69 (dd, 1H, $J = 10.4$, $J = 6.0$), 4.69 (dd, 1H, $J = 10.0$, $J = 6.0$), 3.88 (s, 3H), 3.52 (ABq, 2H, $J_{\text{AB}} = 12.4$, $\Delta\nu_{\text{AB}} = 44.4$), 2.65 - 2.55 (m, 1H), 2.52 - 2.42 (m, 1H), 2.34 - 2.24 (m, 2H), 2.08 - 1.58 (m, 12H), 1.46 - 1.18 (m, 10H), 1.36 (s, 3H), 1.31 (s, 3H), 0.94 (d, 3H, $J = 6.0$), 0.85 (d, 3H, $J = 6.8$), 0.84 (d, 3H, $J = 7.2$), 0.98 - 0.86 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ 166.9, 145.0, 129.6, 127.7, 126.4, 102.9, 102.8, 89.7, 89.3, 81.1, 81.0, 74.9, 70.8, 70.5, 52.1, 51.9, 43.9, 43.6, 42.4, 38.3, 37.4, 36.5, 36.4, 36.1, 34.4, 34.3, 30.8, 30.7, 26.0, 25.9, 24.8, 24.7, 20.1, 20.0, 12.7, 12.4; IR (film, cm^{-1}) 3498, 2951, 2876, 1721, 1594, 1430, 1377, 1276, 1182, 1110, 1054, 1008, 942, 880, 845, 762, 736.

Example 12

Synthesis of bis-trioxane β -hydroxysulfone ester 10b

To a solution of bis-trioxane β -hydroxysulfide ester **10a** (40.0 mg, 0.052 mmol) in tetrachloromethane (1.5 mL), acetonitrile (1.5 mL) and H_2O (2.3 mL) was added ruthenium (III) chloride hydrate (catalytic) and periodic acid (23.7 mg, 0.104 mmol, 2.0) (on addition of ruthenium chloride the solution turned black). On stirring for 30 minutes at room temperature, diethyl ether (30 mL) was added and the reaction was stirred for a further 10 minutes. The reaction mixture was then dried (MgSO_4), filtered through a pad

of celite and the remaining solid was washed with ethyl acetate (30 mL). The organics were then concentrated *in vacuo* and flash column chromatography on silica eluting with 30% ethyl acetate/hexanes isolated bis-trioxane β -hydroxysulfone ester **10b** as a white solid (32.0mg, 0.040 mmol, 76 %). Mp = 115-117 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.21 (ABq, 4H, $J_{\text{AB}} = 8.4$, $\Delta\nu_{\text{AB}} = 73.6$), 5.70 (s, 1H), 5.36 (s, 1H), 5.19 (dd, 1H, $J = 10.8$, $J = 6.0$), 4.88 (s, 1H), 4.58 (dd, 1H, $J = 9.6$, $J = 6.4$), 3.95 (s, 3H), 3.81 (ABq, 2H, $J_{\text{AB}} = 13.6$, $\Delta\nu_{\text{AB}} = 278.4$), 2.90 - 2.82 (m, 1H), 2.70 - 2.56 (m, 2H), 2.43 - 2.30 (m, 2H), 2.20 - 1.20 (m, 16H), 1.54 (s, 3H), 1.24 (s, 3H), 0.96 (d, 6H, $J = 6.0$), 0.93 (d, 3H, $J = 8.0$), 0.88 (d, 3H, $J = 7.6$), 0.98 - 0.86 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 165.8, 145.3, 134.3, 129.9, 128.5, 103.6, 102.8, 90.1, 88.5, 81.2, 73.6, 71.6, 71.1, 60.1, 52.6, 52.5, 51.8, 45.0, 43.3, 39.5, 37.5, 37.3, 36.5, 34.5, 34.3, 30.5, 30.4, 26.4, 26.0, 25.0, 24.8, 24.6, 24.3, 20.3, 20.0, 13.7, 11.9; IR (film, cm^{-1}) 3458, 2951, 2875, 1731, 1435, 1377, 1320, 1280, 1153, 1105, 1051, 1015, 911, 880, 733; HRMS(ES) m/z calcd for $\text{C}_{42}\text{H}_{60}\text{O}_{13}\text{SNa}$ ($\text{M}+\text{Na}$) 827.3646, found 827.3661.

Example 13

Synthesis of bis-trioxane β -hydroxysulfone benzoic acid **10c**

A solution of bis-trioxane β -hydroxysulfone ester **10b** (19.7 mg, 0.024 mmol) in 2.5 % KOH / MeOH (1.0 mL) was stirred vigorously at room temperature for three hours, at which time TLC analysis confirmed complete consumption of starting material. The solution was then evaporated to dryness *in vacuo*, before the addition of water (10 mL). This solution was acidified with acetic acid until pH = 4.0 was attained and organics were then extracted with ethyl acetate (2 x 30mL), dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 2-propanol/methylene chloride/ethyl acetate (1:3:18) isolated bis-trioxane β -hydroxysulfone benzoic acid **10c** as a white solid (14.7mg, 0.019 mmol, 79 %). Mp = 143-145 °C; $[\alpha]_{\text{D}}^{23.9}$ 105.2 (CHCl_3 , $c = 0.07$); ^1H NMR (400 MHz, CDCl_3) δ 8.21 (ABq, 4H, $J_{\text{AB}} = 8.4$, $\Delta\nu_{\text{AB}} = 76.4$), 5.71 (s, 1H), 5.38 (s, 1H), 5.19 (dd, 1H, $J = 10.8$, $J = 6.4$), 4.63 (dd, 1H, $J = 9.2$, $J = 6.0$ Hz), 3.82 (ABq, 2H, $J_{\text{AB}} = 13.6$, $\Delta\nu_{\text{AB}} = 266.4$ Hz), 2.85 - 2.82 (m, 1H), 2.70 - 2.56 (m, 2H),

2.43 - 2.30 (m, 2H), 2.20 - 1.20 (m, 22H, including two singlets at 1.56 and 1.25), 0.96 (d, 6H, $J = 6.0$), 0.93 (d, 3H, $J = 8.0$), 0.88 (d, 3H, $J = 7.6$), 0.98 - 0.86 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 168.0, 145.7, 133.5, 130.4, 128.5, 103.8, 102.9, 90.1, 88.7, 81.3, 73.7, 71.8, 71.2, 60.2, 52.6, 51.9, 44.9, 43.3, 39.4, 37.5, 37.4, 37.3, 36.5, 34.5, 34.3, 30.6, 30.5, 26.1, 25.9, 25.0, 24.8, 24.5, 24.4, 20.2, 20.0, 13.6, 11.9; IR (film, cm^{-1}) 3459, 2939, 2875, 1723, 1432, 1403, 1378, 1319, 1300, 1228, 1154, 1099, 1051, 1015, 911, 877, 732, 616; HRMS(ES) m/z calcd for $\text{C}_{41}\text{H}_{58}\text{O}_{13}\text{SNa}$ ($\text{M}+\text{Na}$) 813.3490, found 813.3500; Anal ($\text{C}_{41}\text{H}_{58}\text{O}_{13}$) C, H.

10

Example 14

Synthesis of bis-trioxane styryl tertiary alcohol 11a

A solution of styryllithium was prepared by adding t-BuLi (1.7 M in hexanes, 1.2ml, 2.04 mmol) to a solution of 4-bromostyrene (0.13 mL, 1 mmol) in anhydrous diethyl ether (5 mL). The resulting deep red solution was stirred for 30 minutes to ensure complete formation of styryllithium. To a solution of bis-trioxane ketone **7** (30 mg, 0.051 mmol) in anhydrous tetrahydrofuran was added the solution of styryllithium (0.35 mL, ~ 0.07 mmol) at -78°C and the reaction was stirred for 30 minutes, at which time TLC analysis confirmed full consumption of starting material. Saturated ammonium chloride solution (5 mL) was added and organics were extracted with methylene chloride (3 x 20 mL), dried (MgSO_4) and concentrated *in vacuo*. Gradient column chromatography on silica eluting with 10 and then 15% ethyl acetate/petroleum ether isolated bis-trioxane styryl tertiary alcohol **11a** as a viscous oil (26 mg, 0.037 mmol, 74%). ^1H NMR (400 MHz, CDCl_3) δ 7.45 (d, $J = 8.4$, 2H), 7.35 (d, $J = 8.4$, 2H), 6.68 (dd, $J = 10.9$, $J = 17.6$, 1H), 5.70 (d, $J = 17.6$, 1H), 5.35 (s, 2H), 5.2 (d, $J = 10.9$, 2H), 4.50 - 4.40 (m, 1H), 4.25 - 4.12 (m, 1H), 2.78 - 2.42 (m, 4H), 2.38 - 1.60 (m, 14H), 1.48 - 1.20 (m, 14H), 1.00 - 0.80 (m, 14H).

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Example 15

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Synthesis of bis-trioxane tertiary alcohol benzoic acid 11b

To a solution of bis-trioxane styryl tertiary alcohol **11a** (20 mg, 0.028 mmol) in acetone (3 mL) was added KMnO_4 (large excess) as a solid. The

reaction was stirred at room temperature for 6 hours, at which time TLC analysis confirmed full consumption of starting material. 2-propanol was added to quench any excess KMnO_4 and the reaction mixture was concentrated *in vacuo*. Ethyl acetate (10 mL) and water (10 mL) were added and organics were extracted with ethyl acetate (3 x 10 mL), dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 90% ethyl acetate/petroleum ether isolated bis-trioxane tertiary alcohol benzoic acid **11b** as a white solid (19 mg, 0.014 mmol, 50%). ^1H NMR (400 MHz, CDCl_3) δ 7.80 (d, 2H, $J = 8.4$), 7.39 (d, 2H, $J = 8.4$), 5.36 (s, 2H), 4.40 - 4.32 (m, 1H), 4.12 - 4.06 (m, 1H), 2.80 - 2.42 (m, 4H), 2.40 - 2.10 (m, 3H), 2.10 - 1.60 (m, 15H), 1.48 - 1.20 (m, 14H), 1.00 - 0.80 (m, 14H); ^{13}C NMR (100 MHz, CDCl_3) δ 167.4, 132.5, 130.6, 130.0, 127.7, 104.7, 104.6, 90.5, 89.9, 82.3, 82.1, 78.4, 74.1, 73.6, 54.1, 53.8, 46.4, 45.9, 43.4, 39.8, 38.6, 38.5, 37.8, 37.7, 35.9, 35.8, 32.1, 31.9, 26.4, 26.1, 26.0, 25.9, 25.8, 25.7, 20.8, 20.7, 13.9, 13.6; HRMS(ES) m/z calcd for $\text{C}_{40}\text{H}_{56}\text{O}_{11}\text{Na}$ ($\text{M}+\text{Na}$) 735.3715, found 735.3717.

Example 16

Synthesis of bis-trioxane primary alcohol isonicotinate N-oxide **13**

To a stirring suspension of bis-trioxane primary alcohol **4** (0.14 g, 0.24 mmol) and commercially available (Aldrich) isonicotinic acid *N*-oxide (0.11 g, 0.76 mmol) in anhydrous methylene chloride (10 mL) was added 4-(dimethylamino)-pyridine (0.11 g, 0.93 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (0.17 g, 0.90 mmol). A further 5 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 4 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (10 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 30 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a colorless oil (0.23 g). Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 70% ethyl acetate/petroleum ether and then with 80% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol isonicotinate N-

oxide **13** as a white solid (0.17 g, 0.23 mmol, 98%). ^1H NMR (CDCl_3 , 400 MHz) δ 8.22 (d, 2H, $J = 7.3$), 7.89 (d, 2H, $J = 7.3$), 5.32 (s, 1H), 5.29 (s, 1H), 4.52 (s, 1H), 4.51 (s, 1H), 4.51 – 4.54 (m, 1H), 4.39 – 4.32 (m, 1H), 2.67 (st, 1H, $J = 7.0$), 2.57 (st, 1H, $J = 7.0$), 2.42 – 2.26 (m, 3H), 2.07 – 1.97 (m, 2H), 1.97 – 1.85 (m, 2H), 1.84 – 1.74 (m, 4H), 1.70 – 1.47 (m, 6H), 1.41 – 1.19 (m, 14H, including two singlets at 1.40 and 1.39), 1.00 – 0.91 (m, 2H), 0.96 (d, 3H, $J = 5.6$), 0.95 (d, 3H, $J = 5.6$), 0.88 (d, 3H, $J = 6.9$), 0.86 (d, 3H, $J = 6.9$); ^{13}C NMR (CDCl_3 , 100 MHz) δ 163.27, 139.35 (2), 127.26, 126.39 (2), 103.10, 102.80, 89.64, 89.01, 81.12, 81.09, 72.99, 70.52, 68.37, 52.22, 52.01, 44.21, 43.97, 37.56, 37.49, 36.63, 36.53, 34.40, 34.36, 33.98, 30.95, 30.58, 30.52, 29.68, 26.03, 25.98, 24.92, 24.88, 24.77, 24.75, 20.16, 20.07, 12.96, 12.62; HRMS (EI, m/z) for $\text{C}_{40}\text{H}_{57}\text{NO}_{11}\text{Na}$ requires 750.3824, found 750.3845; IR (film, cm^{-1}) 2931, 2871, 1719, 1609, 1448, 1373, 1261, 1156, 1108, 1049, 1037, 1008, 925, 732; Mp. 114 – 122 °C; $[\alpha]_{\text{D}}^{23.1}$ 51.7 (CHCl_3 , $c = 0.67$); Anal ($\text{C}_{40}\text{H}_{59}\text{NO}_{12}$) C, H, N.

Example 17

Synthesis of bis-trioxane diphenyl phosphate **14**

To a stirring solution of bis-trioxane primary alcohol **4** (8.0 mg, 0.013 mmol) in anhydrous methylene chloride (0.5 mL) was added diphenyl chlorophosphate (10 μL , 0.048 mmol) and pyridine (20 μL , 0.247 mmol). A further 1 mL of anhydrous methylene chloride was used to wash down the inside walls of the flask and the reaction was stirred at room temperature. TLC analysis after 18 hours showed a substantial amount of starting material still present. A second portion of diphenyl chlorophosphate (50 μL , 0.241 mmol) and 4-(dimethylamino)-pyridine (15.0 mg, 0.123 mmol) were added and the reaction stirred at room temperature for a further 2 hours. At this time, TLC analysis confirmed that no starting material remained. Water (10 mL), 3M HCl (2 mL) and methylene chloride (10 mL) were added and organics were extracted with methylene chloride (3 x 15 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a yellow oil. Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 20% and then 25 % ethyl acetate/petroleum ether isolated bis-trioxane diphenyl phosphate **14** as a

cloudy white oil (10.3 mg, 0.012 mmol, 93%). ^1H NMR (CDCl_3 , 400 MHz) δ 7.33 (t, 4H, $J = 7.7$), 7.28 – 7.22 (m, 4H), 7.17 (t, 2H, $J = 7.7$), 5.30 (s, 1H), 5.29 (s, 1H), 4.58 – 4.51 (m, 1H), 4.50 – 4.37 (m, 2H), 4.25 (dd, 1H, $J = 10.1$, $J = 5.9$), 2.68 (q, 1H, $J = 6.6$), 2.52 (q, 1H, $J = 6.6$), 2.37 – 2.24 (m, 3H), 2.05 – 1.97 (m, 2H), 1.95 – 1.79 (m, 4H), 1.79 – 1.33 (m, 16H, including two singlets at 1.40 and 1.37), 1.32 – 1.17 (m, 6H), 0.98 – 0.78 (m, 14H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 150.68 (d, 2, $J = 7.2$), 129.67 (4), 125.11 (2), 120.17 (d, 2, $J = 3.9$), 120.12 (d, 2, $J = 3.9$), 103.19, 102.78, 89.45, 88.73, 81.14, 81.10, 73.80, 71.92 (d, $J = 7.0$), 70.80, 52.39, 52.11, 44.47, 44.09, 37.40, 37.28, 36.64, 36.58, 35.33 (d, $J = 7.7$), 34.46, 34.42, 30.49, 30.42, 30.15, 29.38, 26.11, 26.05, 24.86, 24.79, 24.74, 24.61, 20.19, 20.09, 13.17, 12.61; ^{31}P NMR (162 MHz) δ -11.77; HRMS (EI, m/z) for $\text{C}_{46}\text{H}_{63}\text{O}_{12}\text{PNa}$ requires 861.3949, found 861.3879; IR (film, cm^{-1}) 2919, 2861, 1589, 1485, 1455, 1376, 1290, 1220, 1190, 1108, 1008, 944, 767, 686; $[\alpha]_{\text{D}}^{24.1}$ 67.6 (CHCl_3 , $c = 0.26$).

Example 18

Synthesis of bis-trioxane primary alcohol nicotinate N-oxide 15

To a stirring suspension of bis-trioxane primary alcohol 4 (19.7 mg, 0.033 mmol) and nicotinic acid *N*-oxide (13.1 mg, 0.094 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (14.1 mg, 0.115 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (21.4 mg, 0.112 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 2 hours 30 minutes, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a white cloudy oil (33.2 mg). Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with ethyl acetate and then with 3% methanol/ethyl acetate isolated bis-trioxane primary alcohol nicotinate N-oxide 15 as a glassy solid. Treatment with hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid

(23.6 mg, 0.032 mmol, 99%). ^1H NMR (CDCl_3 , 400 MHz) δ 8.76 (s, br, 1H), 8.33 (d, 1H, $J = 6.3$), 7.87 (d, 1H, $J = 7.9$), 7.36 (t, br, 1H, $J = 7.1$), 5.31 (s, 1H), 5.27 (s, 1H), 4.61 – 4.51 (m, 2H), 4.46 (dd, 1H, $J = 10.0$, $J = 6.2$), 4.35 (dd, 1H, $J = 10.0$, $J = 6.2$), 2.65 (q, 1H, $J = 6.8$), 2.55 (q, 1H, $J = 6.8$), 2.40 – 2.24 (m, 3H), 2.06 – 1.71 (m, 8H), 1.70 – 1.47 (m, 6H), 1.42 – 1.18 (m, 14H, including two singlets at 1.40 and 1.38), 0.99 – 0.89 (m, 8H), 0.86 (apparent triplet, 6H, $J = 7.0$); ^{13}C NMR (CDCl_3 , 100 MHz) δ 162.65, 142.15, 140.12, 130.57, 126.41, 125.68, 103.01, 102.77, 89.63, 89.02, 81.12, 81.10, 73.00, 70.36, 68.34, 52.18, 51.98, 44.16, 43.93, 37.47, 37.45, 36.62, 36.51, 34.36, 34.33, 33.95, 30.96, 30.59, 30.56, 30.30, 25.99, 25.96, 24.90, 24.86, 24.75, 24.66, 20.16, 20.07, 12.92, 12.60; HRMS (EI, m/z) for $\text{C}_{40}\text{H}_{57}\text{NO}_{11}\text{Na}$ requires 750.3824, found 750.3858; IR (film, cm^{-1}) 2931, 2870, 1727, 1435, 1375, 1296, 1223, 1106, 1043, 1008, 879, 749; Mp. 104 – 108 °C (morphological change observed beginning at ~ 80 °C); $[\alpha]_D^{24.4}$ 49.7 (CHCl_3 , $c = 0.205$); Anal ($\text{C}_{40}\text{H}_{57}\text{NO}_{11} + 0.5 \text{H}_2\text{O}$) C, H, N.

Example 19

Synthesis of bis-trioxane phosphonic acid monoester 16

To a solution of methylphosphonic dichloride (32.2 mg, 0.242 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (5.2 mg, 0.043 mmol). The mixture was stirred at room temperature for five minutes before addition of bis-trioxane primary alcohol 4 (25.7 mg, 0.042 mmol) in anhydrous methylene chloride (1.25 mL) by canula. On stirring at room temperature for 1 hour 45 minutes, TLC analysis showed a large amount of starting material remaining and so a second portion of 4-(dimethylamino)-pyridine (5.2 mg, 0.043 mmol) was added. Continued stirring for a further 2 hours followed by TLC analysis showed that starting material still remained. A third portion of 4-(dimethylamino)-pyridine (10.0 mg, 0.082 mmol) was added and the reaction was stirred at room temperature. After 15 hours, TLC analysis confirmed full consumption of starting material and pyridine (0.2 mL, 2.47 mmol) was added and the mixture stirred vigorously for 1 hour before careful addition of 3M HCl (3 mL, ~9 mmol) and continued stirring for a further 1 hour. Methylene chloride (5 mL) and water (5 mL) were added and organics

were extracted with methylene chloride (3 x 10 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a white cloudy oil (34.7 mg). Gradient column chromatography on silica (crude was dry-loaded) eluting with 10% then 30% methanol/ethyl acetate isolated bis-trioxane phosphonic acid monoester **16** as a glassy solid. Treatment with hexanes (2 x 10 mL) and then acetonitrile (3 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid (22.9 mg, 0.033 mmol, 79%). ^1H NMR (CD_3OD , 400 MHz) δ 5.39 (s, br, 2H), 4.30 (dd, 1H, $J = 10.4$, $J = 6.1$), 4.20 (dd, 1H, $J = 10.4$, $J = 6.1$), 3.99 – 3.87 (m, 2H), 2.70 – 2.54 (m, 2H), 2.29 (t, br, 2H, $J = 13.9$), 2.10 – 1.99 (m, 3H), 1.97 – 1.87 (m, 3H), 1.86 – 1.75 (m, 3H), 1.68 (d, br, 2H, $J = 13.2$), 1.61 – 1.33 (m, 16H, including singlet at 1.35), 1.30 – 1.17 (m, 5H, including doublet ($J = 16.6$) at 1.27), 1.02 – 0.94 (m, 2H), 0.97 (d, br, 6H, $J = 6.1$), 0.90 (dd, 6H, $J = 7.5$, $J = 2.8$); ^{13}C NMR (CD_3OD , 100 MHz) δ 104.75, 104.53, 90.54, 90.22, 82.52, 82.49, 75.60, 74.84, 67.77 (d, $J = 5.4$), 54.14, 54.00, 46.28, 46.08, 38.62, 38.60, 37.81, 37.77, 37.52 (d, $J = 7.6$), 35.91, 35.87, 32.12, 32.10, 31.48 (2), 26.46 (2), 26.14, 26.08, 26.01, 25.99, 20.81, 20.74, 13.90, 13.66, 12.56 (d, $J = 139.1$); ^{31}P NMR (CD_3OD , 162 MHz) δ 22.07; HRMS (EI, m/z) for $\text{C}_{35}\text{H}_{56}\text{O}_{11}\text{PNa}_2$ requires 729.3350, found 729.3370; IR (film, cm^{-1}) 2940, 2862, 1449, 1372, 1196, 1188, 1090, 1037, 1008, 937, 896, 878, 826, 732; Mp. 142 – 147 °C; $[\alpha]_{\text{D}}^{23.4}$ 8.9 (CHCl_3 , $c = 3.09$).

Example 20

Synthesis of bis-trioxane secondary alcohol 17

Bis-trioxane ketone **7** (10mg, 0.016 mmol) was dissolved in MeOH (1 mL) and the solution was cooled to 0° C. To this cooled solution was added NaBH_4 (4.2 mg, 0.112 mmol, 7.7eq.) in 4 portions. The reaction mixture was then stirred for 1 hour and quenched with dropwise addition of 1% AcOH-MeOH until neutral pH was attained. The reaction mixture was then concentrated *in vacuo* and the crude product charged onto a silica column. Column chromatography gave bis-trioxane secondary alcohol **17** as a white solid (8 mg, 84 %). ^1H NMR (400 MHz, CDCl_3) δ 5.37 (s, 1H), 5.31 (s, 1H), 4.54 - 4.40 (m, 2H), 4.10 - 3.99 (bm, 1H), 3.75 (s, 1H), 2.76 - 2.60 (m, 2H), 2.38 - 2.24 (m, 2H), 2.08 - 1.20 (m, 28H), 0.96 - 0.84 (m, 14H); ^{13}C NMR (100

MHz, CDCl₃) δ 133.9, 133.7, 128.8, 128.6, 128.5, 103.3, 89.0, 88.9, 81.1, 76.2, 72.4, 71.3, 52.2, 52.4, 52.3, 44.5, 44.2, 37.6, 37.5, 36.6, 36.5, 36.3, 36.2, 34.6, 34.5, 30.6, 30.4, 26.2, 26.1, 24.9, 24.8, 24.7, 20.3, 13.2, 13.1; HRMS (EI, m/z) for C₃₃H₅₂O₉Na⁺ required 615.3503, found 615.3466.

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Example 21

Synthesis of bis-trioxane vicinal diol isonicotinate N-oxide 18

To a stirring suspension of bis-trioxane vicinal diol 5 (19.4 mg, 0.031 mmol) and commercially available (Aldrich) isonicotinic acid N-oxide (14.5 mg, 0.104 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (14.5 mg, 0.119 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (23.0 mg, 0.120 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 1 hour 30 minutes, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a colorless oil. Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 100% ethyl acetate and then with 3% methanol/ethyl acetate isolated bis-trioxane vicinal diol isonicotinate N-oxide 18 as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid (22.7 mg, 0.031 mmol, 99%). ¹H NMR (400 MHz, CDCl₃) δ 8.21 (d, 2H, J = 7.3), 7.92 (d, 2H, J = 7.3), 5.33 (s, 1H), 5.31 (s, 1H), 4.75 (dd, 1H, J = 10.3, J = 6.2), 4.65 – 4.51 (m, 3H), 4.34 (s, br, 1H), 2.58 (qt, 1H, J = 7.0), 2.56 (qt, 1H, J = 7.0), 2.30 (t, br, 2H, J = 13.3), 2.08 – 1.85 (m, 7H), 1.83 – 1.60 (m, 7H), 1.42 – 1.17 (m, 14H, including two singlets at 1.40 and 1.37), 0.99 – 0.93 (m, 8H), 0.89 (apparent triplet, br, 6H, J = 7.7); ¹³C NMR (100 MHz, CDCl₃) δ 163.04, 139.27 (2), 127.39, 126.45 (2), 102.86, 102.84, 89.74, 89.41, 81.02 (2), 73.62, 70.04, 69.85, 51.91, 51.83, 43.64, 43.52, 37.58, 37.55, 36.75, 36.53, 36.46, 35.38 (2), 34.27, 34.25, 30.86, 30.82, 25.92, 25.87, 24.85 (2), 24.77 (2), 20.08, 20.02, 12.50, 12.39; HRMS (EI, m/z) for C₄₀H₅₇NO₁₂Na⁺ requires 766.3773,

found 766.3770; IR (film, cm^{-1}) 3483, 2931, 2861, 1719, 1608, 1443, 1377, 1257, 1165, 1102, 1008, 908, 732; Mp. 114 – 118 °C.

Example 22

Synthesis of bis-trioxane isobutyric acid 19

5 To a solution of bis-trioxane primary alcohol 4 (132 mg, 0.218 mmol) in ethyl acetate (4 mL), acetonitrile (4.0 mL) and H₂O (1.3 mL) was added ruthenium (III) chloride hydrate (4.5 mg, 0.022 mmol) and sodium periodate (326 mg, 1.53 mmol, 7.0) (on addition of ruthenium chloride the solution turned black). After stirring for 30 mins at room temperature (the color of
10 solution turned to pale orange), the reaction mixture was poured into a mixture of ethyl acetate (30 mL) and saturated aqueous NH₄Cl solution (30 mL). Organics were extracted with ethyl acetate (2 x 30 mL), dried (MgSO₄), filtered through a pad of celite and concentrated *in vacuo*. Flash column chromatography on silica eluting with 1% acetic acid in 30% ethyl
15 acetate/hexane isolated bis-trioxane isobutyric acid 19 as a white solid (126 mg, 0.204 mmol, 94 %). Mp = 105-110 °C. Further purification by medium pressure liquid chromatography (MPLC: LiChroprep Si60 (40-63 μ m) - EM Science) with the same solvent system (1% acetic acid in 30% ethyl acetate/hexane) removed unknown impurities and gave a white solid (118
20 mg, 0.190 mmol, 87%); Mp = 110-113 °C ; ¹H NMR (400 MHz, CDCl₃) δ 5.30 (s, 1H), 5.29 (s, 1H), 4.30 - 4.20 (m, 2H), 2.92 – 2.84 (m, 1H), 2.75 - 2.55 (m, 2H), 2.35 - 2.23 (m, 2H), 2.15 - 1.95 (m, 3H), 1.94 - 1.50 (m, 11H), 1.46 - 1.20 (m, 15H, including two singlets at 1.39 and 1.38), 0.94 (d, 6H, *J* = 6.0), 0.86 (d, 6H, *J* = 7.6), 0.98 - 0.86 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 179.9,
25 103.3, 103.1, 89.0, 88.6, 81.1, 81.0, 74.3, 72.8, 52.3, 52.2, 44.4, 44.2, 41.8, 37.4, 37.3, 36.5, 36.5, 34.4, 31.5, 31.4, 30.3, 30.2, 25.9, 25.8, 24.7, 24.6, 20.2, 20.1, 13.1, 12.7; IR (film, cm^{-1}) 3500 (br), 2940, 2875, 1705, 1450, 1377, 1279, 1187, 1094, 1053, 1011, 941, 877, 826, 734; HRMS(ES) *m/z* calcd for C₃₄H₅₂O₁₀Na (M+Na) 643.3457, found 643.3470.

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Example 23

Synthesis of bis-trioxane β -hydroxy O-allyl ether 20

To a suspension of KH (~35% in mineral oil, 16 mg (~ 5.6 mg actual), 0.14 mmol) in anhydrous diethyl ether (1.0 mL) at 0 °C was added a solution of bis-trioxane vicinal diol **5** (28.9 mg, 0.046 mmol) in diethyl ether (0.7 mL). The reaction was stirred for 30 mins (to ensure complete formation of the alkoxide) before addition of allyl bromide (0.012 mL, 0.14 mmol) and 18-crown-6 (2.4 mg, 0.009 mmol). The reaction was then allowed to warm to room temperature and was stirred until TLC analysis showed no starting material remaining. The reaction mixture was poured into a mixture of diethyl ether (20 mL) and saturated aqueous NH₄Cl solution (20 mL) and organics were extracted with diethyl ether (3 x 10 mL), washed with brine, dried (MgSO₄) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 20% ethyl acetate/hexanes isolated bis-trioxane β -hydroxy O-allyl ether **20** as a sticky solid (22.3 mg, 0.034 mmol, 74%); ¹H NMR (400 MHz, CDCl₃) δ 5.98 - 5.88 (m, 1H), 5.32 (s, 2H), 5.25 (ddd, 1H, *J* = 17.2, *J* = 3.6, *J* = 1.6), 5.12 (apparent doublet, br, 1H, *J* = 10.4), 4.33 - 4.28 (m, 1H), 4.23 - 4.17 (m, 1H), 4.02 - 3.90 (m, 2H), 3.60 - 3.53 (m, 2H), 2.75 - 2.60 (m, 2H), 2.35 - 2.25 (m, 2H), 2.10 - 1.18 (m, 28H, including two singlets at 1.40 and 1.39); 0.94 (d, 6H, *J* = 6.0), 0.86 (d, 3H, *J* = 7.6), 0.84 (d, 3H, *J* = 7.6), 0.98 - 0.86 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 135.6, 116.4, 103.1, 103.0, 89.2, 88.6, 81.1, 81.0, 73.9, 73.8, 72.1, 71.5, 71.3, 52.4, 52.2, 44.5, 44.1, 37.4, 36.6, 35.8, 35.0, 34.5, 34.4, 30.7, 30.7, 26.1, 26.0, 24.8, 24.7, 20.2, 20.1, 13.2, 12.9; IR (film, cm⁻¹) 3506, 2953, 2925, 2874, 1452, 1376, 1222, 1190, 1127, 1093, 1053, 1011, 942, 878, 843, 731; HRMS(ES) *m/z* calcd for C₃₇H₅₈NaO₁₀ (M+Na) 685.3922, found 685.3915.

Example 24

Synthesis of bis-trioxane β -hydroxy O-acetic acid 21

To a solution of bis-trioxane β -hydroxy O-allyl ether **20** (40.3 mg, 0.061 mmol) in ethyl acetate (1.2 mL), acetonitrile (1.2 mL) and water (0.4 mL) was added ruthenium (III) chloride hydrate (catalytic) and sodium periodate (65.0 mg, 0.304 mmol) (on addition of ruthenium chloride the solution turned black).

The reaction was stirred at room temperature for 30 mins (the color of solution turned to pale orange), at which time TLC analysis confirmed complete consumption of starting material. The reaction mixture was poured into a mixture of ethyl acetate (20 mL) and saturated aqueous NH_4Cl solution (20 mL) and organics were extracted with ethyl acetate (2 x 20 mL), dried (MgSO_4), filtered through a pad of celite and concentrated *in vacuo*. Flash column chromatography on silica eluting with 1% acetic acid in 4% methanol/methylene chloride isolated bis-trioxane β -hydroxy O-acetic acid **21** as a white solid (5.3 mg, 0.008 mmol, 13 %). Mp = 85-89 °C; ^1H NMR (400 MHz, CDCl_3) δ 5.35 (s, 1H), 5.34 (s, 1H), 4.82 – 4.76 (m, 1H), 4.52 – 4.46 (m, 1H), 4.25 (d, 1H, $J = 17.0$), 4.09 (d, 1H, $J = 17.0$), 3.80 (d, 1H, $J = 9.4$), 3.67 (d, 1H, $J = 9.4$), 1.40 (s, 3H), 1.38 (s, 3H), 0.94 (d, 6H, $J = 6.0$), 0.86 (d, 3H, $J = 7.6$), 0.84 (d, 3H, $J = 7.6$), 0.98 - 0.86 (m, 2H); HRMS(ES) m/z calcd for $\text{C}_{36}\text{H}_{56}\text{NaO}_{12}(\text{M}+\text{Na})$ 703.3664, found 703.3681.

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Example 25

Synthesis of bis-trioxane primary alcohol picolinate **22**

To a solution of bis-trioxane primary alcohol **4** (29.9 mg, 0.049 mmol) in anhydrous methylene chloride (0.5 mL) was added picolinyl chloride hydrochloride (22.1 mg, 0.124 mmol) and 4-(dimethylamino)-pyridine (34.2 mg, 0.280 mmol). A further portion of anhydrous methylene chloride (1 mL) was added to wash down the flask walls and the reaction was stirred at room temperature for 18 hours, at which time TLC analysis confirmed full consumption of starting material. Water (5 mL), saturated NaHCO_3 solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give an off-white solid. Gradient column chromatography on silica (crude was dry-loaded) eluting with 30%, 40% and then 50% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol picolinate **22** as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white bubbly oil (33.1 mg, 0.047 mmol, 94%). ^1H NMR (CDCl_3 , 400 MHz) δ 8.74 (s, br, 1H, $J = 4.8$), 8.12 (d, 1H, $J = 7.8$), 7.83 (dt, 1H, $J = 7.8$, $J = 1.7$),

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7.46 (dd, 1H, $J = 7.8$, $J = 4.8$), 5.35 (s, 1H), 5.33 (s, 1H), 4.62 – 4.54 (m, 2H), 4.44 (dd, 1H, $J = 10.6$, $J = 6.0$), 4.29 (dd, 1H, $J = 10.6$, $J = 6.0$), 2.71 (st, 1H, $J = 6.9$), 2.61 (st, 1H, $J = 6.9$), 2.49 – 2.38 (m, br, 1H), 2.29 (dt, 2H, $J = 13.9$, $J = 3.7$), 2.02 – 1.71 (m, 7H), 1.69 – 1.52 (m, 6H), 1.43 – 1.17 (m, 15H),
5 including two singlets at 1.39 and 1.37), 1.00 – 0.81 (m, 14H); ^{13}C NMR (CDCl₃, 100 MHz) δ 164.85, 149.59, 148.55, 136.93, 126.54, 125.03, 103.13, 102.90, 89.24, 88.58, 81.14, 81.10, 73.59, 71.46, 68.25, 52.39, 52.16, 44.49, 44.21, 37.33, 37.26, 36.61, 36.54, 34.47, 34.43, 34.16, 30.57, 30.46, 29.93, 29.63, 26.08, 26.04, 24.91, 24.83, 24.69, 24.60, 20.20, 20.11, 13.24, 12.87;
10 HRMS (EI, m/z) for C₄₀H₅₇NO₁₀Na requires 734.3875, found 734.3897; IR (film, cm⁻¹) 2941, 2874, 1717, 1458, 1437, 1376, 1303, 1245, 1128, 1104, 1044, 1006, 931, 876, 743; $[\alpha]_{\text{D}}^{23.9}$ 64.5 (CHCl₃, $c = 0.270$).

Example 26

Synthesis of bis-trioxane vicinal diol nicotinate N-oxide 23

15 To a stirring suspension of bis-trioxane vicinal diol 5 (26.0 mg, 0.042 mmol) and commercially available (Aldrich) nicotinic acid N-oxide (19.9 mg, 0.143 mmol) in anhydrous methylene chloride (2 mL) was added 4-(dimethylamino)-pyridine (19.8 mg, 0.162 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (32.0 mg, 0.167
20 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20
25 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a white solid. The crude material was pre-absorbed on silica and gradient column chromatography on silica eluting with 100% ethyl acetate, 5% methanol/ethyl acetate and then 10% methanol/ethyl acetate isolated bis-trioxane vicinal diol nicotinate N-oxide 23 as a colorless oil. Treatment with methylene chloride (2
30 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid (29.1 mg, 0.039 mmol, 94%). ^1H NMR (CDCl₃, 400 MHz) δ 8.80 (s, br, 1H), 8.33 (d, 1H, $J = 6.5$), 7.89 (d, 1H, $J =$

7.9), 7.35 (dd, 1h, $J = 6.5$, $J = 7.9$), 5.32 (s, 1H), 5.28 (s, 1H), 4.74 (dd, 1H, $J = 6.2$, $J = 10.4$), 4.67 – 4.57 (m, 2H), 4.50 (d, 1H, $J = 11.2$), 4.32 (s, br, 1H), 2.58 (st, 1H, $J = 7.0$), 2.52 (st, 1H, $J = 7.0$), 2.34 – 2.23 (m, 2H), 2.06 – 1.61 (m, 14H), 1.43 – 1.18 (m, 14H, including two singlets at 1.40 and 1.36), 1.00 – 0.83 (m, 14H, including doublet at 0.94 ($J = 5.4$) and apparent triplet at 0.88 ($J = 7.4$)); ^{13}C NMR (CDCl_3 , 100 MHz) δ 162.38, 142.07, 140.18, 130.63, 126.53, 126.60, 102.82, 102.78, 89.79, 89.27, 81.02 (2), 73.32, 69.88, 69.82, 69.79, 51.92, 51.79, 43.65, 43.42, 37.48 (2), 36.87, 36.52, 36.43, 35.24, 34.25, 34.21, 30.85, 30.79, 25.87, 25.82, 24.83, 24.81, 24.72, 24.69, 20.08, 20.00, 12.51, 12.31; HRMS (EI, m/z) for $\text{C}_{40}\text{H}_{57}\text{NO}_{12}\text{Na}$ requires 766.3773, found 766.3817; IR (film, cm^{-1}) 2938, 2875, 1732, 1435, 1376, 1298, 1226, 1108, 1054, 1013, 941, 842, 754, 733; Mp. 116 – 118 °C (morphological change observed beginning at 102 °C); $[\alpha]_{\text{D}}^{24.5}$ 33.5 (CHCl_3 , $c = 0.10$); Anal ($\text{C}_{40}\text{H}_{57}\text{NO}_{12}$) C, H, N.

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Example 27

Synthesis of bis-trioxane secondary alcohol isonicotinate N-oxide 24

To a stirring suspension of bis-trioxane secondary alcohol 17 (21.6 mg, 0.036 mmol) and commercially available (Aldrich) isonicotinic acid *N*-oxide (16.2 mg, 0.116 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (16.8 mg, 0.138 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (25.9 mg, 0.135 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a colorless oil (29.5 mg). Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 80% ethyl acetate/petroleum ether and then with 100% ethyl acetate isolated bis-trioxane secondary alcohol isonicotinate N-oxide 24 as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a

white solid (24.4 mg, 0.034 mmol, 94%). ¹H NMR (CDCl₃, 400 MHz) δ 8.18 (d, 2H, *J* = 6.9), 7.92 (d, 2H, *J* = 6.9), 5.48 – 5.39 (m, 1H), 5.35 (s, 1H), 5.31 (s, 1H), 4.46 – 4.37 (m, 2H), 2.65 (st, 2H, *J* = 6.9), 2.35 – 2.22 (m, 2H), 2.16 – 1.54 (m, 13H), 1.49 – 1.17 (m, 15H, including two singlets at 1.31 and 1.26),
5 1.00 – 0.80 (m, 14H, including apparent triplet at 0.95 (*J* = 5.6) and two doublets at 0.88 (*J* = 7.5) and 0.83 (*J* = 7.5)); ¹³C NMR (CDCl₃, 100 MHz) δ 162.93, 139.17 (2), 127.70, 126.61 (2), 103.06, 102.99, 88.95, 88.86, 81.06 (2), 74.27, 71.73, 71.11, 52.23, 52.10, 44.21, 44.08, 37.39, 37.30, 36.51, 35.45, 34.38, 34.34, 32.72, 32.69, 30.39 (2), 25.93, 25.83, 24.72, 24.64,
10 24.62 (2), 20.14, 20.11, 12.99, 12.97; HRMS (EI, *m/z*) for C₃₉H₅₅NO₁₁Na requires 736.3667, found 736.3678; IR (film, cm⁻¹) 2937, 2873, 1717, 1611, 1445, 1376, 1265, 1163, 1094, 1054, 1011; Mp. 114 – 118 °C (morphological change observed beginning at 103 °C); [α]_D^{24.3} 75.5 (CHCl₃, *c* = 0.13); Anal (C₃₉H₅₅NO₁₁) C, H, N.

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Example 28

Synthesis of bis-trioxane secondary alcohol nicotinate N-oxide 25

To a stirring suspension of bis-trioxane secondary alcohol 17 (30.7 mg, 0.052 mmol) and commercially available (Aldrich) nicotinic acid *N*-oxide (21.5 mg, 0.155 mmol) in anhydrous methylene chloride (1 mL) was added 4-
20 (dimethylamino)-pyridine (22.0 mg, 0.180 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (34.5 mg, 0.180 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting
25 material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a colorless oil. Gradient column chromatography on silica (crude was dry-loaded) eluting firstly with 100% ethyl acetate and then with 5% methanol/ethyl acetate
30 isolated bis-trioxane secondary alcohol nicotinate N-oxide 25 as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid

(37.5 mg, 0.052 mmol, 100%). ^1H NMR (CDCl_3 , 400 MHz) δ 8.78 (s, 1H), 8.29 (d, 1H, $J = 6.5$), 7.88 (d, 1H, $J = 8.0$), 7.31 (dd, 1H, $J = 8.0$, $J = 6.5$), 5.52 – 5.42 (m, 1H), 5.32 (s, 1H), 5.30 (s, 1H), 4.45 – 4.34 (m, 2H), 2.64 (st, 2H, $J = 6.9$), 2.34 – 2.21 (m, 2H), 2.16 – 1.53 (m, 13H), 1.49 – 1.15 (m, 15H, including two singlets at 1.28 and 1.27), 1.00 – 0.80 (m, 14H, including four doublets at 0.94 ($J = 6.1$), 0.93 ($J = 6.1$), 0.87 ($J = 7.5$) and 0.83 ($J = 7.5$)); ^{13}C NMR (CDCl_3 , 100 MHz) δ 162.38, 141.93, 140.47, 130.87, 126.66, 125.45, 103.07, 102.98, 88.99, 88.77, 81.03, 81.01, 74.60, 71.93, 70.87, 52.25, 52.09, 44.22, 44.03, 37.35, 37.30, 36.52, 36.43, 34.38, 34.34, 32.93, 32.88, 30.36 (2), 25.85, 25.77, 24.73, 24.66, 24.63, 24.58, 20.13, 20.10, 13.01, 12.88; HRMS (EI, m/z) for $\text{C}_{39}\text{H}_{55}\text{NO}_{11}\text{Na}$ requires 736.3667, found 736.3715; IR (film, cm^{-1}) 2951, 2874, 1726, 1432, 1377, 1299, 1225, 1114, 1094, 1056, 1013, 950, 876, 822, 753; Mp. 106 – 110 °C (morphological change observed beginning at 98 °C); $[\alpha]_{\text{D}}^{24.3}$ 63.1 (CHCl_3 , $c = 0.225$); Anal ($\text{C}_{39}\text{H}_{55}\text{NO}_{11} + 0.5 \text{H}_2\text{O}$) C, H, N.

Example 29

Synthesis of bis-trioxane primary alcohol diethyl phosphoric acid triester 26

To a stirring solution of bis-trioxane primary alcohol 4 (26.1 mg, 0.043 mmol) in anhydrous methylene chloride (2.5 mL) was added diethylchlorophosphate (0.12 mL, 0.860 mmol) and anhydrous pyridine (0.07 mL, 0.860 mmol). On stirring at room temperature for 3 hrs, TLC analysis showed full consumption of starting material. Water (2 mL), 3M HCl (3 mL) and methylene chloride (3 mL) were added and organics were extracted with methylene chloride (3 x 15 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a yellow oil. Gradient column chromatography on silica eluting with 40% and then 50% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol diethyl phosphoric acid triester 26 as a colorless oil (28.1 mg, 0.038 mmol, 88%). ^1H NMR (400 MHz, CDCl_3) δ 5.30 (s, 1H), 5.27 (s, 1H), 4.40–4.35 (m, 1H), 4.24–4.06 (m, 7H), 2.68 (st, 1H, $J = 7.6$), 2.54 (st, 1H, $J = 6.8$), 2.34–1.92 (m, 6H), 1.91–1.14 (m, 31H, including a singlet at 1.38, a singlet at 1.37, and a doublet of triplets at 1.31 ($J = 6.8$, 0.8)), 1.02–0.68 (m, 14H, including doublets at 0.93 ($J = 6.0$), 0.84 ($J = 7.6$) and 0.83 ($J = 7.6$)); ^{13}C NMR

(100 MHz, CDCl_3) δ 103.16, 102.74, 89.35, 89.64, 81.12, 81.08, 74.00, 71.03, 69.72 (d, $J = 5.8$), 63.54 (d, $J = 6.0$), 63.52 (d, $J = 6.0$), 52.40, 52.10, 44.50, 44.10, 37.36, 37.33, 36.61, 36.54, 35.36 (d, $J = 7.5$), 34.46, 34.40, 30.51, 30.43, 30.10, 26.05, 25.99, 24.81, 24.74, 24.70, 24.62, 20.19, 20.07, 16.14 (2, d, $J = 6.8$), 13.20, 12.67; ^{31}P NMR (162 MHz, CDCl_3) δ 0.058; HRMS (EI, m/z) for $\text{C}_{38}\text{H}_{63}\text{O}_{12}\text{PNa}$ requires 765.3949, found 765.3981; IR (film, cm^{-1}) 2941 (m), 2880 (w), 1454 (w), 1373 (w), 1271 (m), 1103 (m), 1037 (s), 1011 (s), 878 (w), 843 (w), 665 (w); $[\alpha]_{\text{D}}^{23.6}$ 13.4 (CHCl_3 , $c = 0.30$).

Example 30

10 **Synthesis of bis-trioxane primary alcohol dimethyl phosphoric acid triester 27**

To a stirring solution of bis-trioxane primary alcohol 4 (21.0 mg, 0.035 mmol) in anhydrous methylene chloride (2.0 mL) was added dimethylchlorophosphate (0.07 mL, 0.70 mmol) and anhydrous pyridine (0.06 mL, 0.70 mmol). On stirring at room temperature for 3 hrs, TLC analysis showed full consumption of starting material. Water (2 mL), 3M HCl (3 mL) and methylene chloride (3 mL) were added and organics were extracted with methylene chloride (3 x 15 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a yellow oil. Gradient column chromatography on silica eluting with 40% and then 50% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol dimethyl phosphoric acid triester 27 as a colorless oil (13.6 mg, 0.020 mmol, 55%). ^1H NMR (400 MHz, CDCl_3) δ 5.32 (s, 1H), 5.29 (s, 1H), 4.45-4.35 (m, 1H), 4.28-4.17 (m, 3H), 3.78 (d, 3H, $J = 10.8$), 3.77 (d, 3H, $J = 11.2$), 2.69 (st, 1H, $J = 7.6$), 2.55 (st, 1H, $J = 6.4$), 2.34-2.25 (m, 2H), 2.17 (s, br, 1H), 2.02-1.18 (m, 28H, including singlet at 1.37), 1.02-0.80 (m, 14H, including doublets at 0.95 ($J = 6.0$), 0.86 ($J = 7.6$) and 0.85 ($J = 7.2$)); ^{13}C NMR (100 MHz, CDCl_3) δ 103.17, 102.77, 89.52, 88.77, 81.16, 81.14, 73.97, 70.72, 69.91 (d, $J = 5.8$), 54.20 (2, d, $J = 6.1$), 52.41, 52.10, 44.50, 44.09, 37.43, 37.38, 36.66, 36.58, 35.36 (d, $J = 7.9$), 34.48, 34.40, 30.57, 30.50, 30.21, 29.57, 26.05, 25.98, 24.85, 24.78, 24.74, 24.67, 20.23, 20.09, 13.20, 12.64; ^{31}P NMR (162 MHz, CDCl_3) δ 2.10; HRMS (EI, m/z) for $\text{C}_{36}\text{H}_{59}\text{O}_{12}\text{PNa}$ requires 737.3636, found 737.3646; IR (film, cm^{-1}) 2955 (m), 2882 (w), 1722

(w), 1452 (w), 1380 (w), 1281 (w), 1189 (w), 1109 (m), 1037 (s), 1004 (s), 879 (w), 846 (w); $[\alpha]_D^{23.7}$ 71.0 (CHCl₃, c = 0.25).

Example 31

Synthesis of bis-trioxane primary alcohol *p*-trifluoromethylbenzoate **28**

5 To a stirring solution of bis-trioxane primary alcohol **4** (47.1 mg, 0.078 mmol) and α,α,α -trifluoro-*p*-toluic acid (46.0 mg, 0.242 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (33.5 mg, 0.274 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (52.8 mg, 0.275 mmol). A further 2 mL of anhydrous methylene
10 chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to
15 give a white solid (101.5 mg). Gradient column chromatography on silica (crude was dry-loaded) eluting with 10% and then 20% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol *p*-trifluoromethylbenzoate **28** as a colorless oil (43.7 mg, 0.056 mmol, 72%). ¹H NMR (CDCl₃, 400 MHz) δ 8.16 (d, 2H, *J* = 7.9), 7.70 (d, 2H, *J* = 7.9), 5.32 (s, 1H), 5.27 (s, 1H), 4.60 – 4.42 (m, 3H), 4.37 – 4.29 (m, 1H), 2.70 (st, 1H, *J* = 7.0), 2.59 (st, 1H, *J* = 7.0), 2.47 – 2.36 (m, 1H), 2.30 (td, 2H, *J* = 14.0, *J* = 3.5), 2.06 – 1.72 (m, 8H), 1.69 – 1.49 (m, 6H), 1.43 – 1.13 (m, 14H, including two singlets at 1.40 and 1.39), 0.98 – 0.84 (m, 14H, including two doublets at 0.95 and 0.91 (*J* = 5.8) and two doublets at 0.87 and 0.86 (*J* = 7.7)); ¹³C NMR (CDCl₃, 100 MHz) δ 165.22, 134.15 (d, *J* = 34.5), 133.99, 129.89 (2), 125.29 (2, d, *J* = 3.7), 123.64 (d, *J* = 271.6), 103.14, 102.85, 89.46, 88.76, 81.09, 81.07, 73.23, 70.98, 67.87, 52.30, 52.08, 44.34, 44.09, 37.43 (2), 36.63, 36.53, 34.41, 34.39, 33.93, 30.56, 30.45, 30.09 (2), 26.05, 25.99, 24.95, 24.87, 24.73, 24.61, 20.08, 20.06, 14.09, 12.73; HRMS (EI, *m/z*) for C₄₂H₅₇F₃O₁₀Na requires 801.3796, found 801.3847; IR (film, cm⁻¹) 2941, 2878, 1725, 1375, 1323, 1275, 1167, 1133, 1098, 1058, 1014, 661; $[\alpha]_D^{24.3}$ 58.2 (CHCl₃, c = 0.265).

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Example 32

Synthesis of bis-trioxane primary alcohol 3,5-ditrifluoromethyl benzoate 29

To a stirring solution of bis-trioxane primary alcohol 4 (48.9 mg, 0.081 mmol) and 3,5-bis(trifluoromethyl)benzoic acid (65.5 mg, 0.254 mmol) in anhydrous methylene chloride (2 mL) was added 4-(dimethylamino)-pyridine (36.0 mg, 0.295 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (55.1 mg, 0.287 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 15 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a bubbly white solid. Gradient column chromatography on silica (crude was dry-loaded) eluting with 5%, 10% and then 20% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol 3,5-ditrifluoromethylbenzoate 29 as a colorless oil (64.7 mg, 0.076 mmol, 95%). ¹H NMR (CDCl₃, 400 MHz) δ 8.47 (s, 2H), 8.05 (s, 1H), 5.30 (s, 1H), 5.23 (s, 1H), 4.64 – 4.56 (m, 2H), 4.47 (dd, 1H, *J* = 10.1, *J* = 6.2), 4.32 (dd, 1H, *J* = 10.1, *J* = 6.2), 2.69 (st, 1H, *J* = 6.9), 2.59 (st, 1H, *J* = 6.9), 2.45 – 2.35 (m, 1H), 2.35 – 2.24 (m, 2H), 2.05 – 1.51 (m, 14H), 1.42 – 1.13 (m, 14H, including two singlets at 1.40 and 1.37), 0.98 – 0.85 (m, 14H, including two doublets at 0.94 and 0.91 (*J* = 5.5) and a broad doublet at 0.87 (*J* = 7.3)); ¹³C NMR (CDCl₃, 100 MHz) δ 163.79, 133.08, 131.98 (2, d, *J* = 34.0), 129.57 (2, d, *J* = 3.0), 125.94, 122.89 (2, d, *J* = 275.2), 103.12, 102.86, 89.45, 88.77, 81.08, 81.04, 73.31, 70.82, 68.30, 52.24, 52.04, 44.31, 44.03, 37.46 (2), 36.58, 36.47, 34.35 (2), 34.22, 30.57, 30.48, 30.21, 29.98, 25.95, 25.96, 24.92, 24.86, 24.69, 24.53, 20.06, 20.04, 13.04, 12.65; HRMS (EI, *m/z*) for C₄₃H₅₆F₆O₁₀Na requires 869.3670, found 869.3680; [α]_D^{23.4} 46.8 (CHCl₃, *c* = 0.305).

Example 33

Synthesis of bis-trioxane primary alcohol dimethylglycinate 30

To a solution of bis-trioxane primary alcohol 4 (29.2 mg, 0.048 mmol) and *N,N*-dimethylglycine hydrochloride (22.1 mg, 0.158 mmol) in anhydrous

methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (22.8 mg, 0.187 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (36.2 mg, 0.189 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 15 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), saturated sodium bicarbonate solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give an off-white solid. Gradient column chromatography on silica eluting with 60%, 70% and then 80% ethyl acetate/petroleum ether isolated bis-trioxane primary alcohol dimethylglycinate **30** as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white oily solid (23.4 mg, 0.034 mmol, 70%). ¹H NMR (CDCl₃, 400 MHz) δ 5.30 (s, br, 1H), 5.29 (s, br, 1H), 4.41 – 4.22 (m, 4H), 3.28 (s, 2H), 2.68 (st, 1H, *J* = 6.8), 2.57 (st, 1H, *J* = 6.8), 2.44 (s, 6H), 2.37 – 2.15 (m, 3H), 2.06 – 1.96 (m, 2H), 1.95 – 1.85 (m, 2H), 1.83 – 1.54 (m, 7H), 1.53 – 1.19 (m, 17H, including singlet at 1.39), 1.00 – 0.81 (m, 14H, including two doublets at 0.96 and 0.95 (*J* = 5.8) and apparent triplet at 0.85 (*J* = 6.9)); ¹³C NMR (CDCl₃, 100 MHz) δ 169.83, 103.18, 102.87, 89.42, 88.80, 81.13, 81.10, 73.47, 71.11, 66.97, 59.52, 52.35, 52.10, 44.72 (2), 44.41, 44.11, 37.45, 37.42, 34.45, 34.40, 34.01, 30.55, 30.48, 29.97 (2), 26.08, 25.99, 24.82 (2), 24.74, 24.70, 20.20, 20.10, 13.12, 12.73; HRMS (EI, *m/z*) for C₃₈H₆₁NO₁₀Na requires 714.4188, found 714.4236; IR (film, cm⁻¹) 2944, 2873, 1738, 1452, 1373, 1195, 1105, 1048, 1005, 933, 877; [α]_D^{23.8} 71.3 (CHCl₃, *c* = 0.29).

Example 34

Synthesis of dihydroartemisinin **31**

To a solution of artemisinin **1** (0.44 g, 1.55 mmol) in methanol (30 mL) at 0°C was added sodium borohydride (0.47 g, 12.3 mmol) in 4 portions, each at 15 minutes intervals. After addition of the last portion, stirring was continued for 1 hour, at which time TLC analysis confirmed no starting material remained. Concentrated acetic acid was added dropwise until a pH of ~5 was achieved

and the resulting mixture was then concentrated *in vacuo* to give a white solid. Water (30 mL) and methylene chloride (30 mL) were added and organics were extracted with methylene chloride (3 x 30 mL). The combined organic extracts were washed with saturated sodium bicarbonate solution (20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give a white solid (0.46 g). Gradient column chromatography on silica (crude was dry-loaded) eluting with 40% then 50% diethyl ether/petroleum ether isolated dihydroartemisinin **31** as a white solid (0.38 g, 1.33 mmol, 86%, isolated as a 1.0:0.9 mixture of β : α -dihydroartemisinin). ¹H NMR (CDCl₃, 400 MHz) δ 5.61 (s, 1H), 5.40 (s, 0.9H), 5.31 (d, 1H, *J* = 3.4), 4.76 (d, 0.9H, *J* = 9.2) [only characteristic peaks listed].

Example 35

Synthesis of α -dihydroartemisinin dimethylglycinate **32**

To a solution of dihydroartemisinin **31** (25.1 mg, 0.088 mmol) and *N,N*-dimethylglycine hydrochloride (32.9 mg, 0.236 mmol) in anhydrous methylene chloride (2 mL) was added 4-(dimethylamino)-pyridine (36.5 mg, 0.299 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (55.8 mg, 0.291 mmol). A further 0.5 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 3 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), saturated sodium bicarbonate solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo* to give an off-white solid (81.5 mg). Flash column chromatography on silica (crude was dry-loaded) eluting with 100% ethyl acetate isolated α -dihydroartemisinin dimethylglycinate **32** as a colorless oil (31.5 mg, 0.085 mmol, 97%). ¹H NMR (CDCl₃, 400 MHz) δ 5.85 (d, 1H, *J* = 9.9), 5.43 (s, 1H), 3.26 (AB, 2H, *J* = 26.5, *J* = 17.2), 2.61 – 2.50 (m, 1H), 2.36 (s, 6H), 2.41 – 2.32 (m, 1H), 2.02 (d, br, 1H, *J* = 14.4), 1.93 – 1.84 (m, 1H), 1.81 – 1.68 (m, 2H), 1.62 (dt, 1H, *J* = 13.8, *J* = 4.4), 1.50 – 1.39 (m, 4H, including singlet at 1.42), 1.38 – 1.22 (m, 3H), 1.07 – 0.95 (m, 1H), 0.95 (d, 3H, *J* = 5.9), 0.84 (d, 3H, *J* = 7.1); ¹³C NMR (CDCl₃, 100 MHz) δ 169.38, 104.42, 91.86, 91.48, 80.05, 59.95, 51.51, 45.20, 45.17 (2), 37.23, 36.16, 34.03, 31.80, 25.89,

24.53, 21.96, 20.18, 12.08; HRMS (EI, m/z) for $C_{19}H_{31}NO_6Na$ requires 392.2044, found 392.2021; IR (film, cm^{-1}) 2931, 2861, 2766, 1758, 1452, 1376, 1278, 1226, 1190, 1149, 1130, 1098, 1035, 1014, 926, 874, 846.

Example 36

5 **Synthesis of α -dihydroartemisinin isonicotinate N-oxide 33**

To a stirring suspension of dihydroartemisinin **31** (33.6 mg, 0.12 mmol) and isonicotinic acid *N*-oxide (51.7 mg, 0.37 mmol) in anhydrous methylene chloride (1 mL) was added 4-(dimethylamino)-pyridine (52.0 mg, 0.43 mmol) and 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (81.7 mg, 0.43 mmol). A further 1 mL of anhydrous methylene chloride was added to wash down the flask walls and the reaction was stirred at room temperature for 4 hours, at which time TLC analysis showed full consumption of starting material. Water (5 mL), 3M HCl solution (5 mL) and methylene chloride (5 mL) were added and organics were extracted with methylene chloride (3 x 20 mL), dried (Na_2SO_4) and concentrated *in vacuo* to give a yellow tinged colorless oil (76.7 mg). Gradient column chromatography on silica (crude was dry-loaded) eluting with 60%, 80%, 90% and then 95% ethyl acetate/petroleum ether isolated α -dihydroartemisinin isonicotinate N-oxide **33** as a colorless oil. Treatment with methylene chloride (2 x 10 mL) and then hexanes (2 x 10 mL) (followed by drying overnight under high vacuum) gave a white solid (42.2 mg, 0.10 mmol, 88%). 1H NMR ($CDCl_3$, 400 MHz) δ 8.22 (d, 2H, $J = 7.3$), 7.95 (d, 2H, $J = 7.3$), 5.95 (d, 1H, $J = 9.9$), 5.51 (s, 1H), 2.78 – 2.67 (m, 1H), 2.38 (dt, 1H, $J = 13.5$, $J = 3.9$), 2.08 – 2.00 (m, 1H), 1.95 – 1.64 (m, 4H), 1.55 – 1.22 (m, 7H, including singlet at 1.41), 1.09 – 0.84 (m, 7H, including doublet at 0.97 ($J = 5.9$) and doublet at 0.91 ($J = 7.1$); ^{13}C NMR ($CDCl_3$, 100 MHz) δ 162.11, 139.42 (2), 126.77 (2), 125.82, 104.50, 93.33, 91.60, 80.00, 51.47, 45.15, 37.21, 36.11, 33.96, 31.77, 25.83, 24.48, 21.94, 20.15, 12.16; HRMS (EI, m/z) for $C_{21}H_{27}NO_7Na$ requires 428.1680, found 428.1674; IR (film, cm^{-1}) 2925, 2872, 2359, 2343, 1733, 1612, 1486, 1444, 1262, 1165, 1132, 1091, 1053, 1036, 1016, 872, 855, 827, 768; Mp. 118 – 122 °C (morphological change observed beginning at 90 °C); $[\alpha]_D^{24.7}$ 4.67 ($CHCl_3$, $c = 0.465$); Anal ($C_{21}H_{27}NO_7$) C, H, N.

Example 37**Synthesis of bis-trioxane allyl tertiary alcohol 34**

A solution of bis-trioxane ketone **7** (20.0 mg, 0.034 mmol) in anhydrous tetrahydrofuran (1.5 mL) was cooled to -78 °C before addition of allyl
5 magnesium bromide (1.0 M solution in diethyl ether, 0.140 mL, 0.140 mmol). After stirring for one hour, TLC analysis confirmed that no starting material remained. The reaction was quenched with 1.0 mL distilled water. The reaction mixture was poured into a mixture of brine (20 mL) and diethyl ether (30 mL) and the organic layer was separated, dried (Na₂SO₄) and
10 concentrated *in vacuo* to give a yellow oil. Gradient column chromatography on silica eluting with 15 to 20 % ethyl acetate/hexanes isolated bis-trioxane allyl tertiary alcohol **34** as a foamy solid (20.6 mg, 0.033 mmol, 96%). ¹H NMR (CDCl₃, 400 MHz) δ 5.95 - 5.87 (m, 1H), 5.36 (s, 1H), 5.33 (s, 1H), 5.15 (apparent d, 1H, *J* = 17.6), 5.09 (dd, 1H, *J* = 10.4, *J* = 2.4), 4.80 – 4.63 (m, 1H), 4.60 – 4.54 (m, 1H), 2.70 - 2.41 (m, 4H), 2.34 - 2.27 (m, 2H), 1.40 (s, 1H), 1.39 (s, 3H), 0.95 (apparent d, 6H, *J* = 5.6), 0.87 (apparent t, 6H, *J* = 8.4);
15 HRMS (ES, *m/z*) calcd for C₃₆H₅₆O₉Na (M+Na) 655.3817, found 655.3771.

Example 38**Synthesis of bis-trioxane tertiary alcohol carboxylic acid 35**

20 To a solution of bis-trioxane allyl tertiary alcohol **34** (20.6 mg, 0.033 mmol) in ethyl acetate (0.6 mL), acetonitrile (0.6 mL) and H₂O (0.2 mL) was added ruthenium(III) chloride hydrate (1.3 mg, 0.007 mmol) and sodium periodate (27.8 mg, 0.13 mmol) (on addition of ruthenium chloride the solution turned black). After stirring for 30 mins at room temperature (during
25 which time the color of the solution changed to pale orange), the reaction mixture was poured into a mixture of ethyl acetate (30 mL) and saturated aqueous NH₄Cl solution (30 mL). Organics were extracted with ethyl acetate (2 x 30 mL), dried (MgSO₄), filtered through a pad of celite and concentrated *in vacuo*. Flash column chromatography on silica eluting with 1% acetic acid
30 in 35% ethyl acetate/hexane isolated bis-trioxane tertiary alcohol carboxylic acid **35** as a white solid (16.7 mg, 0.026 mmol, 79 %). Mp = 108-110 °C; ¹H NMR (400 MHz, CDCl₃) δ 5.35 (s, 1H), 5.34 (s, 1H), 4.75 – 4.68 (m, 1H), 4.65

– 4.59 (m, 1H), 2.90 (ABq, 2H, $J_{AB} = 15.6$, $\Delta\nu_{AB} = 39.6$), 2.61 - 2.48 (m, 2H), 2.36 - 2.25 (m, 2H), 2.05 - 1.60 (m, 15H), 1.46 - 1.20 (m, 15H, including two siglets at 1.40 and 1.38), 0.98 - 0.82 (m, 2H), 0.96 (apparent doublet, 6H, $J = 5.2$), 0.87 (apparent triplet, 6H, $J = 7.2$); ^{13}C NMR (100 MHz, CDCl_3) δ 103.01, 102.81, 89.87, 89.59, 81.11, 81.04, 74.26, 70.65, 70.03, 51.91, 44.13, 43.57, 38.91, 37.48, 37.43, 36.54, 36.47, 34.28, 30.91, 30.76, 25.92, 24.88, 24.75, 20.06, 12.38, 12.20; HRMS(ES) m/z calcd for $\text{C}_{35}\text{H}_{54}\text{O}_{11}\text{Na}$ (M+Na) 673.3558, found 673.3510.

Example 39

10 **Synthesis of bis-trioxane primary alcohol methyl carbonate 36**

To a solution of bis-trioxane primary alcohol 4 (26.5 mg, 0.044 mmol) in anhydrous pyridine (1.5 mL) at -20°C was added methyl chloroformate (0.017 mL, 0.022 mmol). The reaction was allowed to warm to room temperature and was stirred overnight before being quenched with distilled water (1.0 mL). The reaction mixture was then poured into a mixture of methylene chloride (20 mL) and saturated aqueous NH_4Cl solution (20 mL). Organics were extracted with methylene chloride (3 x 20 mL), dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 20% ethyl acetate/hexanes isolated bis-trioxane primary alcohol methyl carbonate 36 as a sticky solid (6.4 mg, 0.010 mmol, 22 %). $[\alpha]_D^{23.2}$ 64.2 (CHCl_3 , $c = 0.17$); ^1H NMR (400 MHz, CDCl_3) δ 5.30 (apparent s, 2H), 4.41 – 4.36 (m, 1H), 4.33 (apparent doublet, 2H, $J = 4.8$), 4.28 – 4.23 (m, 1H), 2.72 – 2.55 (m, 2H), 2.38 – 2.15 (m, 3H), 2.04 – 1.96 (m, 2H), 1.95 – 1.81 (m, 2H), 1.80 – 1.18 (m, 27H, including three singlets at 1.58, 1.40, 1.39), 0.97 – 0.80 (m, 2H), 0.95 (apparent d, 6H, $J = 6.0$), 0.86 (d, 3H, $J = 7.6$), 0.84 (d, 3H, $J = 7.2$); ^{13}C NMR (100 MHz, CDCl_3) δ 155.79, 103.13, 102.83, 89.47, 88.86, 81.16, 81.15, 77.20, 73.60, 71.02, 70.05, 54.56, 52.38, 52.11, 44.45, 44.14, 37.42, 37.40, 36.67, 36.61, 34.48, 34.43, 34.22, 30.54, 30.49, 29.76, 26.09, 26.00, 24.83, 24.76, 24.70, 20.22, 20.11, 13.13, 12.69; IR (film, cm^{-1}) 2952, 2876, 1748, 1443, 1377, 1270, 1105, 1039, 1008, 928, 732; HRMS(ES) m/z calcd for $\text{C}_{36}\text{H}_{56}\text{O}_{11}\text{Na}$ (M+Na) 687.3715, found 687.3728.

Example 40

Synthesis of bis-trioxane primary alcohol ethyl carbonate 37

To a solution of bis-trioxane primary alcohol 4 (15.5 mg, 0.026 mmol) in anhydrous pyridine (1.5 mL) at 0 °C was added ethyl chloroformate (large excess, ~ 10.0 equiv). The reaction was allowed to warm to room temperature and was stirred overnight, before being quenched with distilled water (1.0 mL). The reaction mixture was then poured into a mixture of methylene chloride (20 mL) and saturated aqueous NH₄Cl solution (20 mL). Organics were extracted with methylene chloride (3 x 20 mL), dried (MgSO₄) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 20% ethyl acetate/hexane isolated bis-trioxane primary alcohol ethyl carbonate 37 as a sticky solid (8.5 mg, 0.013 mmol, 49 %). $[\alpha]_D^{23.7}$ 75.9 (CHCl₃, c = 0.23); ¹H NMR (400 MHz, CDCl₃) δ 5.30 (apparent s, 2H), 4.41 – 4.35 (m, 1H), 4.32 (apparent d, 2H, *J* = 4.8), 4.28 – 4.22 (m, 1H), 4.17 (q, 2H, *J* = 6.8), 2.73 - 2.55 (m, 2H), 2.38 - 2.15 (m, 3H), 2.04 - 1.96 (m, 2H), 1.96 - 1.18 (m, 29H, including a singlet at 1.39 and a triplet at 1.30 (*J* = 7.2)), 0.97 - 0.80 (m, 2H), 0.95 (apparent d, 6H, *J* = 5.6), 0.85 (apparent t, 6H, *J* = 7.2); ¹³C NMR (100 MHz, CDCl₃) δ 155.23, 103.13, 102.84, 89.43, 88.80, 81.15, 81.13, 73.72, 71.13, 69.80, 63.67, 52.39, 52.13, 44.48, 44.16, 37.42, 37.39, 36.67, 36.61, 34.50, 34.44, 34.26, 30.53, 30.42, 29.73, 26.10, 26.09, 24.83, 24.76, 24.70, 20.21, 20.12, 14.31, 13.17, 12.71; IR (film, cm⁻¹) 2939, 2872, 1744, 1453, 1377, 1260, 1105, 1038, 1008, 928, 880, 792; HRMS(ES) *m/z* calcd for C₃₇H₅₈O₁₁Na (M+Na) 701.3873, found 701.3919.

Example 41

Synthesis of bis-trioxane tertiary alcohol methyl sulfonate 38

To a solution of methyl methanesulfonate (0.027 mL, 0.32 mmol) in anhydrous tetrahydrofuran (1.5 mL) at -78 °C was added *n*-butyl lithium (1.3 M sol. in hexanes, 0.26 mL, 0.35 mmol). The reaction was stirred at -78 °C for 30 mins before slow addition of a solution of bis-trioxane ketone 7 (37.6 mg, 0.064 mmol) in anhydrous tetrahydrofuran (0.8 mL). After stirring for 30 min., the reaction was quenched with distilled water (1.0 mL) and the reaction mixture was poured into a mixture of diethyl ether (20 mL) and saturated

aqueous NH_4Cl solution (20 mL). The organic layer was then separated, dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 30% ethyl acetate/hexanes isolated bis-trioxane tertiary alcohol methyl sulfonate **38** as a white solid (42.0 mg, 0.060 mmol, 94 %). Mp = 87 – 90 °C; ^1H NMR (400 MHz, CDCl_3) δ 5.44 (s, 1H), 5.36 (s, 1H), 4.85 – 4.79 (m, 1H), 4.71 – 4.65 (m, 2H), 3.81 (ABq, 2H, $J_{\text{AB}} = 14.4$, $\Delta\nu_{\text{AB}} = 142.4$), 3.96 (s, 3H), 2.76 – 2.67 (m, 1H), 2.59 – 2.91 (m, 1H), 2.38 – 2.15 (m, 3H), 2.06 – 1.82 (m, 6H), 1.81 – 1.19 (m, 21H, including two singlets at 1.40 and 1.37), 0.97 – 0.80 (m, 2H), 0.96 (d, 3H, $J = 5.6$), 0.95 (d, 3H, $J = 6.0$), 0.89 (d, 3H, $J = 7.2$), 0.88 (d, 3H, $J = 7.2$ Hz); ^{13}C NMR (100 MHz, CDCl_3) δ 103.17, 102.81, 89.97, 88.96, 81.12, 72.84, 70.74, 70.28, 56.60, 55.69, 52.25, 51.84, 44.27, 43.37, 39.02, 37.48, 37.39, 36.62, 36.49, 34.27, 30.74, 30.56, 25.96, 25.92, 24.91, 24.77, 24.54, 20.15, 20.12, 13.04, 12.00; IR (film, cm^{-1}) 3469, 2951, 2875, 1451, 1377, 1356, 1175, 1099, 1053, 1007, 911, 879, 731; HRMS(ES) m/z calcd for $\text{C}_{35}\text{H}_{56}\text{O}_{12}\text{SNa}$ ($\text{M}+\text{Na}$) 723.3385, found 723.3397.

Example 42

Synthesis of bis-trioxane tertiary alcohol isopropyl sulfonate **39**

To a solution of isopropyl methanesulfonate (0.009 mL, 0.066 mmol) in anhydrous tetrahydrofuran (1.0 mL) at -78 °C was added *n*-butyl lithium (1.3 M sol. in hexanes, 0.055 mL, 0.073 mmol). The reaction was stirred at -78 °C for 30 mins before slow addition of a solution of bis-trioxane ketone **7** (7.8 mg, 0.013 mmol) in anhydrous tetrahydrofuran (0.4 mL). After stirring for a further 30 mins, the reaction was quenched with distilled water (1.0 mL) and the reaction mixture was poured into a mixture of diethyl ether (20 mL) and saturated aqueous NH_4Cl solution (20 mL). The organic layer was then separated, dried (MgSO_4) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 20% ethyl acetate/hexanes to afford bis-trioxane tertiary alcohol isopropyl sulfonate **39** as a sticky solid (5.8 mg, 0.008 mmol, 60 %). ^1H NMR (400 MHz, CDCl_3) δ 5.46 (s, 1H), 5.36 (s, 1H), 5.11 (st, 1H, $J = 6.4$), 4.88 – 4.84 (m, 1H), 4.67 – 4.63 (m, 2H), 3.78 (ABq, 2H, $J_{\text{AB}}=14.0$, $\Delta\nu_{\text{AB}} = 146.8$), 1.43 (s, 3H), 1.41 (s, 3H), 0.96 (d, 3H, $J = 6.0$), 0.94

(d, 3H, $J = 6.0$), 0.88(d, 3H, $J = 7.2$), 0.87 (d, 3H, $J = 7.2$); HRMS(ES) m/z calcd for $C_{37}H_{60}O_{12}SNa$ ($M+Na$) 751.3698, found 751.3749.

Example 43

Synthesis of bis-trioxane tertiary alcohol N,N-dimethylsulfonamide 40

5 To a solution of *N,N*-dimethyl methanesulfonamide (10.8 mg, 0.088 mmol) in anhydrous tetrahydrofuran (1.5 mL) at $-78^{\circ}C$ was added *n*-butyl lithium (1.3 M sol. in hexanes, 0.072 mL, 0.097 mmol). The reaction was stirred at $-78^{\circ}C$ for 30 mins before slow addition of a solution of bis-trioxane ketone **7** (10.4 mg, 0.018 mmol) in anhydrous tetrahydrofuran (0.5 mL). After
10 stirring for a further 30 mins, the reaction was quenched with distilled water (1.0 mL) and the reaction mixture was poured into a mixture of diethyl ether (20 mL) and saturated aqueous NH_4Cl solution (20 mL). The organic layer was then separated, dried ($MgSO_4$) and concentrated *in vacuo*. Gradient column chromatography on silica eluting with 25% and then 30% ethyl
15 acetate/hexanes isolated bis-trioxane tertiary alcohol N,N-dimethylsulfonamide **40** as a white solid (7.9 mg, 0.011 mmol, 63 %). Mp = $107 - 110^{\circ}C$; $[\alpha]_D^{24.5}$ 74.7 ($CHCl_3$, $c = 0.25$); 1H NMR (400 MHz, $CDCl_3$) δ 5.50 (s, 1H), 5.37 (s, 1H), 5.01 - 4.95 (m, 1H), 4.63 - 4.59 (s, 1H), 4.61 (m, 1H), 3.56 (ABq, 2H, $J_{AB} = 13.6$, $\Delta\nu_{AB} = 207.2$), 2.90 (s, 6H), 2.86 - 2.78 (m, 1H),
20 2.66 - 2.57 (m, 1H), 2.44 - 2.27 (m, 3H), 2.08 - 1.88 (m, 4H), 1.87 - 1.19 (m, 23H, including two singlets at 1.39 and 1.34), 0.97 - 0.80 (m, 2H), 0.95 (d, 3H, $J = 8.0$), 0.94 (d, 3H, $J = 8.4$ Hz), 0.88 (apparent triplet, 6H, $J = 7.2$); ^{13}C NMR (100 MHz, $CDCl_3$) δ 103.38, 102.82, 89.87, 88.33, 81.27, 81.22, 72.80, 71.50, 71.10, 52.51, 51.93, 51.13, 44.86, 43.42, 39.22, 37.85, 37.50, 37.30, 37.00,
25 36.48, 34.51, 34.32, 30.58, 30.51, 26.04, 25.99, 24.98, 24.77, 24.60, 24.36, 20.24, 20.05, 13.61, 11.94; IR (film, cm^{-1}) 3470, 2938, 2875, 1453, 1377, 1330, 1197, 1151, 1099, 1052, 1009, 955, 878, 734, 709; HRMS(ES) m/z calcd for $C_{36}H_{59}NO_{11}SNa$ ($M+Na$) 736.3701, found 736.3660.

Example 44

30 Synthesis of bis-trioxane vicinal diol cyclic carbonate 41

To a solution of bis-trioxane vicinal diol **5** (5.8 mg, 0.009 mmol) in methylene chloride (1.0 mL) was added 1,1'-carbonyl diimidazole (4.5 mg,

0.028 mmol) and 4-(dimethylamino)-pyridine (3.4 mg, 0.028 mmol). The mixture was then stirred at room temperature overnight. The reaction was poured into a mixture of methylene chloride (20 mL) and saturated aqueous NH₄Cl solution (20 mL) and the organic layer was separated. The organics
5 were then washed with brine, dried (MgSO₄) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 25% ethyl acetate/hexanes isolated bis-trioxane vicinal diol cyclic carbonate **41** as a white solid (5.8 mg, 0.009 mmol, 96%). ¹H NMR (400 MHz, CDCl₃) δ 5.31 (s, 1H), 5.30 (s, 1H), 4.84 (d, 1H, *J* = 8.8), 4.67 (dd, 1H, *J* = 6.4, *J* = 1.6), 4.62
10 (dd, 1H, *J* = 6.4, *J* = 1.6), 4.45 (d, 1H, *J* = 8.8), 2.65 - 2.48 (m, 2H), 2.35 - 2.25 (m, 2H), 2.38 - 1.58 (m, 14H), 1.46 - 1.18 (m, 20H, including two singlets at 1.40 and 1.37), 0.95 (d, 6H, *J* = 5.6), 0.88 (d, 3H, *J* = 7.6), 0.86 (d, 3H, *J* = 7.6), 0.98 - 0.86 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 154.9, 102.8, 102.8, 89.9, 89.6, 85.4, 81.0, 80.8, 72.2, 68.9, 68.0, 51.8, 51.8, 43.7, 43.5, 39.0,
15 37.5, 37.4, 36.7, 36.6, 35.6, 34.3, 34.2, 30.8, 30.7, 25.9, 25.7, 24.8, 24.7, 20.0, 12.4, 12.3; HRMS(ES) *m/z* calcd for C₃₅H₅₂O₁₁Na (M+Na) 671.3407, found 671.3412.

Example 45

Synthesis of bis-trioxane vicinal diol cyclic sulfate **42**

20 To a solution of bis-trioxane vicinal diol **5** (18.0 mg, 0.029 mmol) in anhydrous methylene chloride (2.0 mL) at 0 °C was added thionyl chloride (0.021 mL, 0.29 mmol) and triethylamine (0.064 mL, 0.46 mmol) (on addition of triethylamine the solution turned light yellow). The reaction was stirred for one hour, at which time TLC analysis showed complete consumption of
25 starting material. The reaction was quenched with distilled water (1.0 mL) and was poured into a mixture of methylene chloride (20 mL) and saturated aqueous NH₄Cl solution (20 mL). The organic layer was then separated, dried (MgSO₄) and concentrated *in vacuo* to give a yellow oil. A 10 mL round-bottomed flask was charged with the crude material, followed by the addition
30 of ethyl acetate (0.7 mL), and then acetonitrile (0.7 mL) and water (0.3 mL). Ruthenium (III) chloride hydrate (1.2 mg, 0.006 mmol, 0.2 equiv – based on **5**) and sodium periodate (24.8 mg, 0.116 mmol, 4.0 equiv – based on **5**) were

added (on addition of ruthenium chloride the solution turned black). After stirring for 30 mins at room temperature (the color of the solution changed to pale orange), the reaction mixture was poured into a mixture of ethyl acetate (30 mL) and saturated aqueous NH_4Cl solution (30 mL). The aqueous layer
5 was extracted with ethyl acetate (2 x 30 mL), and the combined organics were dried (MgSO_4), filtered through a pad of celite and concentrated *in vacuo*. Gradient column chromatography on silica eluting with 10%, 15% and then 20% ethyl acetate/hexanes isolated bis-trioxane vicinal diol cyclic sulfate **42** as a white solid (17.1 mg, 0.026 mmol, 86 %). $\text{Mp} = 107 - 110^\circ\text{C}$; ^1H NMR
10 (400 MHz, CDCl_3) δ 5.37 (s, 1H), 5.30 (s, 1H), 4.84 (ABq, 2H, $J_{\text{AB}} = 9.6$, $\Delta\nu_{\text{AB}} = 58.4$), 4.61 - 4.55 (m, 2H), 2.75 - 2.66 (m, 1H), 2.60 - 2.48 (m, 2H), 2.36 - 2.20 (m, 4H), 2.07 - 1.85 (m, 6H), 1.82 - 1.56 (m, 8H), 1.42 (s, 3H), 1.37 (s, 3H), 0.97 - 0.80 (m, 2H), 0.96 (apparent triplet, 6H, $J = 6.0$), 0.89 (d, 3H, $J = 7.6$), 0.88 (d, 3H, $J = 7.2$); ^{13}C NMR (100 MHz, CDCl_3) δ 103.20, 102.68,
15 95.49, 89.83, 89.28, 81.16, 80.86, 77.32, 68.95, 68.87, 52.19, 51.76, 44.15, 43.30, 37.53, 37.30, 36.62, 36.54, 34.57, 34.45, 34.19, 30.75, 30.23, 29.69, 25.98, 25.86, 24.83, 24.76, 20.15, 19.99, 12.92, 12.05; IR (film, cm^{-1}) 2941, 2876, 1455, 1380, 1209, 1114, 1093, 1009, 978, 828, 731; HRMS(ES) m/z calcd for $\text{C}_{34}\text{H}_{52}\text{O}_{12}\text{SNa}$ ($\text{M}+\text{Na}$) 707.3072, found 707.3087.

20

Example 46

Synthesis of bis-trioxane tertiary alcohol pyridine sulfide **43**

A 10 mL round-bottomed flask was charged with bis-trioxane epoxide **6** (8.2 mg, 0.013 mmol), 4-mercaptopyridine (15 mg, 0.13 mmol) and neutral aluminum oxide (1.0 g, type W 200 super I, Woelm Pharma, Germany).
25 Anhydrous diethyl ether (0.9 mL) and methylene chloride (0.6 mL) were then added to make a slurry and the reaction was stirred at room temperature overnight. The reaction mixture was then filtered through a pad of celite and the solid was washed with ethyl acetate (2 x 30 mL) before concentration of the organics *in vacuo*. Flash column chromatography on silica eluting with
30 45% ethyl acetate/hexanes isolated bis-trioxane tertiary alcohol pyridine sulfide **43** as a sticky solid (3.0 mg, 0.04 mmol, 31 %). ^1H NMR (400 MHz, CDCl_3) δ 8.36 - 8.32 (m, 2H), 7.31 - 7.25 (m, 2H), 5.36 (s, 1H), 5.35 (s, 1H),

4.75 - 4.71 (m, 1H), 4.57 - 4.53 (m, 1H), 4.37 (s, 1H), 3.51 (ABq, 2H, $J_{AB} = 12.8$, $\Delta\nu_{AB} = 34.0$), 2.61 - 2.45 (m, 2H), 2.35 - 2.35(m, 2H), 2.09 - 1.87 (m, 6H), 1.80 - 1.60 (m, 10H), 1.40 - 1.18 (m, 12H, including two singlets at 1.35 and 1.33), 0.98 - 0.86 (m, 2H), 0.96 (d, 6H, $J = 6.0$), 0.88 (d, 3H, $J = 7.6$),
5 0.85 (d, 3H, $J = 7.6$); HRMS(ES) m/z calcd for $C_{39}H_{57}NO_9SNa$ ($M+Na$) 738.3646, found 738.3664.

Example 47

Synthesis bis-trioxane ketone O-TMS cyanohdrin 44

To a solution of trimethylsilyl cyanide (0.003 mL, 0.023 mmol) in
10 anhydrous tetrahydrofuran (0.8 mL) at room temperature was added lithium tertiary butoxide (1.0 M sol. in tetrahydrofuran, 0.002 mL, 0.002 mmol). The reaction was stirred for 10 mins at room temperature before slow addition of a solution of bis-trioxane ketone 7 (9.0 mg, 0.015 mmol) in anhydrous tetrahydrofuran (0.4 mL). After stirring for 30 mins, the reaction was quenched
15 with distilled water (1.0 mL). The reaction mixture was then poured into a mixture of diethyl ether (20 mL) and 10 % aqueous $NaHCO_3$ solution (20 mL) before the organic layer was separated, dried ($MgSO_4$) and concentrated *in vacuo*. Gradient column chromatography on silica eluting with 10% and then 15% ethyl acetate/hexanes isolated bis-trioxane ketone O-TMS cyanohdrin
20 44 as a sticky solid (9.8 mg, 0.014 mmol, 93 %). 1H NMR (400 MHz, $CDCl_3$) δ 5.41 (s, 1H), 5.37 (s, 1H), 4.65 - 4.61 (m, 1H), 4.54 - 4.50 (m, 1H), 2.85- 2.75 (m, 2H), 2.46 - 2.25 (m, 5H), 1.41 (s, 3H), 1.36 (s, 3H), 0.94 (apparent triplet, 6H, $J = 6.0$), 0.89 (d, 3H, $J = 7.6$), 0.86 (d, 3H, $J = 7.6$), 0.25 (s, 9H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 121.67, 103.40, 103.01, 88.14, 88.07, 80.92, 80.85,
25 72.73, 72.08, 70.25, 52.46, 44.68, 44.63, 38.81, 38.22, 37.23, 37.06, 36.66, 36.60, 34.51, 34.40, 30.58, 30.29, 30.23, 29.69, 26.11, 26.04, 24.56, 24.48, 24.38, 24.30, 20.24, 20.20, 13.59, 13.51, 1.31; HRMS(ES) m/z calcd for $C_{37}H_{59}NO_9Na$ ($M+Na$) 712.3851, found 712.3863.

Example 48

Synthesis of bis-trioxane tertiary alcohol nitrile 45

To a solution of acetonitrile (0.005 mL, 0.096 mmol) and anhydrous tetrahydrofuran (0.8 mL) at $-78^\circ C$ was added LHMDs (1.0 M sol. in THF,

0.096 mL, 0.096 mmol). The reaction was then stirred for 30 mins at -78°C before slow addition of a solution of bis-trioxane ketone **7** (9.5 mg, 0.016 mmol) in anhydrous tetrahydrofuran (0.5 mL). After stirring for 30 mins, the reaction was quenched with distilled water (1.0 mL). The reaction mixture was then poured into a mixture of diethyl ether (20 mL) and saturated aqueous NH₄Cl solution (20 mL). The organic layer was separated, dried (MgSO₄) and concentrated *in vacuo*. Flash column chromatography on silica eluting with 25% ethyl acetate/hexanes isolated bis-trioxane tertiary alcohol nitrile **45** as a white solid (6.0 mg, 0.009 mmol, 59 %). ¹H NMR (400 MHz, CDCl₃) δ 5.38 (s, 1H), 5.34 (s, 1H), 4.74 - 4.70 (m, 1H), 4.60 - 4.55 (m, 1H), 4.44 (s, 1H), 2.94 (ABq, 2H, *J*_{AB} = 16.8, Δ*v*_{AB} = 83.2), 2.05 - 1.86 (m, 8H), 1.82 - 1.62 (m, 6H), 1.45 - 1.19 (m, 18H, including two apparent singlets at 1.39 and 1.38), 0.97 - 0.80 (m, 2H), 0.96 (d, 6H, *J* = 6.0), 0.89 (d, 3H, *J* = 7.2), 0.88 (d, 3H, *J* = 7.2); ¹³C NMR (100 MHz, CDCl₃) δ 118.55, 102.94, 102.69, 90.09, 89.54, 81.13, 81.07, 72.93, 70.07, 69.65, 52.01, 51.81, 43.69, 43.30, 38.87, 37.48, 37.45, 37.06, 36.53, 34.31, 34.24, 30.87, 30.69, 29.14, 25.97, 24.88, 24.81, 24.76, 24.71, 20.06, 20.00, 12.51, 12.13; HRMS(ES) *m/z* calcd for C₃₅H₅₃NO₉Na (M+Na) 654.3613, found 654.3632.

Example 49

Synthesis of bis-trioxane methyl enol ether **46**

To a slurry of (methoxymethyl)triphenylphosphonium chloride (400 mg, 1.2 mmol) in anhydrous tetrahydrofuran (10 mL) at 0°C was added lithium bis(trimethylsilyl)amide (1.0 M solution in tetrahydrofuran, 1.3 mL, 1.3 mmol) resulting in a deep dark red color. The reaction mixture was allowed to warm to room temperature and was stirred for 30 minutes, before being cooled to -78°C. At this time, a solution of bis-trioxane ketone **7** (270 mg, 0.46 mmol) in anhydrous tetrahydrofuran (10 mL) was added dropwise. A color change to pale orange was observed immediately. The reaction was stirred at -78°C for 3 hours, at which time complete consumption of starting material was observed by TLC analysis. The reaction mixture was quenched with distilled water (5 mL) and organics were extracted with methylene chloride (3 x 30 mL), dried (MgSO₄) and concentrated *in vacuo* to give a pale yellow oil. Flash

column chromatography on silica eluting with 20% ethyl acetate/petroleum ether isolated bis-trioxane methyl enol ether **46** as a white solid (220 mg, 0.35 mmol, 76%). ^1H NMR (CDCl_3) δ 5.86 (s, 1H), 5.45 (s, 1H), 5.42 (s, 1H), 4.10 - 4.00 (m, 2H), 3.53 (s, 3H), 2.78 - 2.76 (m, 2H), 2.45 - 2.22 (m, 4H), 2.02 - 1.95 (m, 2H), 1.90 - 1.70 (m, 4H), 1.60 - 1.50 (m, 6H), 1.40 - 1.15 (m, 14H including a singlet at 1.39), 1.00 - 0.85 (m, 14H); HRMS (ES) m/z calcd for $\text{C}_{35}\text{H}_{54}\text{O}_9\text{Na}$ ($\text{M}+\text{Na}$) 641.3660, found 641.3657.

Example 50

Synthesis of bis-trioxane aldehyde **47**

A 50 mL round bottom flask was charged with bis-trioxane methyl enol ether **46** (190 mg, 0.31 mmol), formic acid (10 mL) and diethyl ether (15 mL). The reaction was stirred at room temperature for 8 hours until complete consumption of starting material was observed by TLC analysis. The reaction was then poured into a separatory funnel containing saturated NaHCO_3 (20 mL) and diethyl ether (20 mL). Organics were extracted with diethyl ether (3 x 30 mL), dried (MgSO_4) and concentrated *in vacuo* to give a pale yellow oil. Flash column chromatography on silica eluting with 20% ethyl acetate/petroleum ether isolated bis-trioxane aldehyde **47** as a white solid (160 mg 0.27 mmol, 87%). ^1H NMR (400 MHz, CDCl_3) δ 9.85 (d, 1H, $J = 1.2$ Hz), 5.24 (s, 1H), 5.21 (s, 1H), 4.35 - 4.26 (m, 2H), 2.84 - 2.80 (m, 1H), 2.70 - 2.50 (m, 2H), 2.38 - 2.22 (m, 2H), 2.19 - 2.09 (m, 2H), 2.02 - 1.96 (m, 4H), 1.90 - 1.70 (m, 4H), 1.62 - 1.58 (m, 6H), 1.40 - 1.15 (m, 12H, including two singlets at 1.39 and 1.38), 0.98 - 0.82 (m, 14H); ^{13}C NMR (100 MHz, CDCl_3) δ 205.6, 103.8, 103.1, 89.69, 89.15, 81.47, 81.43, 73.37, 70.94, 52.54, 52.58, 47.79, 44.48, 44.20, 37.72, 37.65, 36.85 (2), 34.70, 34.65, 30.68, 30.51, 29.97, 28.92, 26.30, 26.19, 25.13, 25.11, 25.02, 24.95, 20.43, 20.37, 13.30, 12.74; HRMS (ES) m/z calcd for $\text{C}_{34}\text{H}_{52}\text{O}_9\text{Na}$ ($\text{M}+\text{Na}$) 627.3504, found 627.3445; IR (film, cm^{-1}) 3100, 2800, 2250, 1650, 1400, 1300, 850, 600; Mp. 66-70°C.

Example 51

Synthesis of bis-trioxane aniline **48**

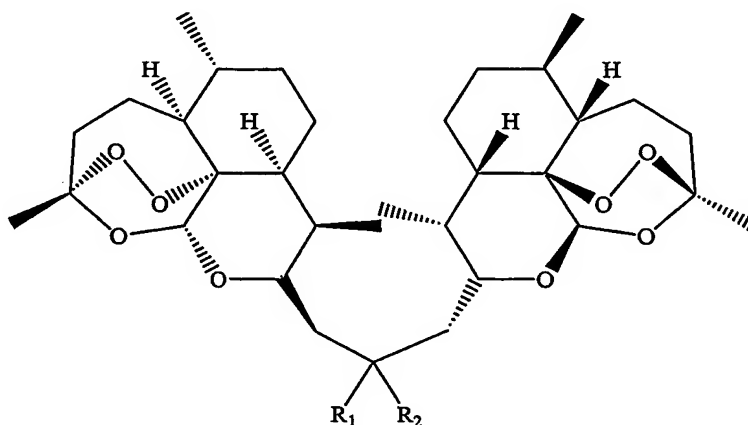
A 10 mL round bottom flask was charged with bis-trioxane aldehyde **47** (25 mg, 0.041 mmol), tetrahydrofuran (1.5 mL), aniline (0.008 mL, 0.088 mmol)

and finally sodium triacetoxo borohydride (18 mg, 0.086 mmol). The reaction was stirred at room temperature for 3 hours until complete consumption of starting material was observed by TLC analysis. The reaction was quenched with saturated NaHCO₃ solution (1.0 mL) and organics were extracted with diethyl ether (3 x 30 mL), dried (MgSO₄) and concentrated *in vacuo* to give a pale yellow oil. Flash column chromatography on silica eluting with 15% ethyl acetate/petroleum ether isolated bis-trioxane aniline **48** as a colorless oil (15 mg, 0.023 mmol, 56%). ¹H NMR (400 MHz, CDCl₃) δ 7.16 - 7.11 (m, 2H), 6.70 - 6.60 (m, 3H), 5.34 (s, 1H), 5.31 (s, 1H), 4.46 - 4.26 (m, 2H), 3.29 - 3.15 (m, 2H), 2.72 - 2.56 (m, 2H), 2.36 - 2.27 (m, 2H), 2.22 - 2.16 (m, 1H), 2.10 - 1.99 (m, 2H), 1.93 - 1.86 (m, 2H), 1.79 - 1.71 (m, 5H), 1.65 - 1.55 (m, 5H), 1.50 - 1.21 (m, 14 H including two singlets at 1.41 and 1.38), 0.96 - 0.84 (m, 14H); ¹³C NMR (100 MHz, CDCl₃) δ 149.1, 129.1, 116.3, 112.7, 103.2, 103.0, 89.37, 88.91, 81.17 (2), 73.90, 71.47, 52.32, 52.16, 47.34, 44.39, 44.21, 37.42, 37.28, 36.61, 36.58, 34.81, 34.42, 31.50, 31.13, 30.70, 30.54, 26.11, 26.02, 24.83, 24.77, 24.73, 24.67, 20.16, 20.13, 13.15, 12.76; HRMS (ES) *m/z* calcd for C₄₀H₅₉NO₈Na (M+Na) 704.4133, found 704.4136.

The foregoing description is considered as illustrative only of the principles of the invention. The words "comprise," "comprising," "include," "including," and "includes" when used in this specification and in the following claims are intended to specify the presence of one or more stated features, integers, components, or steps, but they do not preclude the presence or addition of one or more other features, integers, components, steps, or groups thereof. Furthermore, since a number of modifications and changes will readily occur to those skilled in the art, it is not desired to limit the invention to the exact construction and process shown described above. Accordingly, all suitable modifications and equivalents may be resorted to falling within the scope of the invention as defined by the claims, which follow.

What is claimed is:

1. A compound including resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, said compound having the formula:



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wherein if R_1 is hydrogen or -OH then R_2 is AX, and if R_2 is hydrogen or -OH then R_1 is AX, and A may be absent or A may be any alkyl or aryl group where X is hydrogen, a phosphate group, a phosphonic acid derivative group, an alcohol group, a carboxylic acid group, an ether group, an ester group, a nitrile group, a sulfone group, a sulfide group, an amino acid derivative group, an amine group, and amide group, an aldehyde group, or an aromatic group.

2. The compound of claim 1, wherein said alcohol group is represented by $-R^3OH$, wherein R^3 is a straight chained or branched alkyl group having 1 to 5 carbon atoms.

3. The compound of claim 1, wherein said carboxylic acid group comprises $-R^4COOH$ wherein R^4 is at least one saturated or unsaturated alkyl group, an aryl group an ester group, an ether group or a combination thereof.

4. The compound of claim 3, wherein R^4 is an ester group represented by $-R^5COO-$, wherein R^5 is bonded to the carboxylic acid group and has 0 to 5 carbon atoms.

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5. The compound of claim 3, wherein R^4 is an ether group represented by R^6-O-R^7 wherein R^6 and R^7 are, independently, an alkyl or allyl group having 0 to 5 carbon atoms.

6. The compound of claim 1, wherein said aromatic group
5 comprises $Ar-(R^8)_m$, wherein Ar represents a benzene ring, and m is 1 or 2.

7. The compound of claim 6, wherein R^8 is $-CH=CH_2$, or $-COOH$.

8. The compound of claim 1, wherein the ester group is represented by $-CR^9$, where R^9 is an ester of nicotinic acid, an ester of isonicotinic acid, or the ester group is represented by $-CO(C=O)R^{9a}$, where
10 R^{9a} is $Ph(CY_3)_o$, where o is 1 or 2, and Y may be, independently, H, F, Cl, Br, or I, or where R^{9a} is a substituted heterocyclohexane compound.

9. The compound of claim 1, wherein the phosphonic acid derivative group is represented by $-CO-P(R^{10})(O)OH$, where R^{10} is an alkyl group having 0 to 5 carbon atoms.

10. The compound of claim 1, wherein the phosphate group is –
15 $COP(O)(OR^{11})_2$, where R^{11} is an alkyl group having 0 to 5 carbon atoms, or a phenyl group.

11. The compound of claim 1, wherein the nitrile group is $R^{12}CN$, where R^{12} is an alkyl group having 0 to 5 carbon atoms.

12. The compound of claim 1, wherein the sulfone group is
20 $-CS(=O)_2R^{13}$, wherein R^{13} is $-N(CH_3)_2$, $-OR^{14}$, or $-Ph-COOR^{14}$, where R^{14} is H, CH_3 , or $-CH(CH_3)_2$.

13. The compound of claim 1, wherein the sulfide group is $-CSR^{15}$, where R^{15} is pyridine or $-Ph-COOR^{16}$, where R^{16} is H or CH_3 .

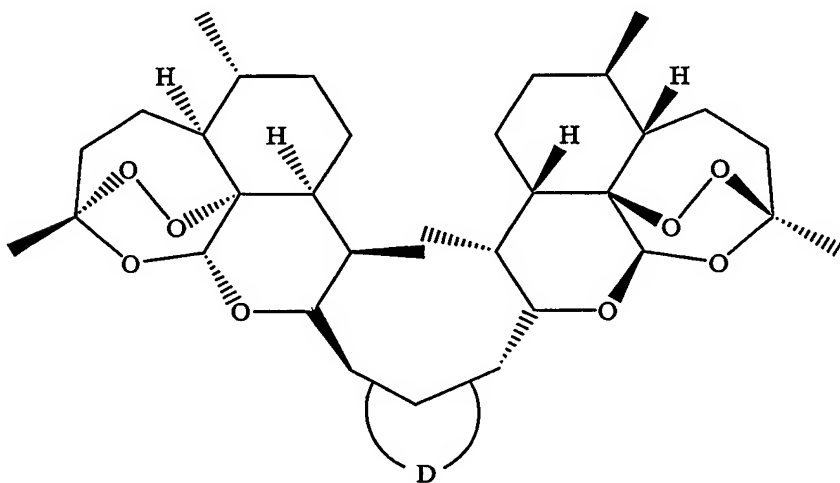
14. The compound of claim 1, wherein the amino acid derivative
25 group is $-COC(=O)CHR^{21}N(R^{17})_2$, where each R^{17} group is, independently, H or CH_3 and R^{21} is hydrogen or any other substituent.

15. The compound of claim 1, wherein the amine group is $\text{CN}(\text{R}^{18})_2$, where each R^{18} group is, independently, H, an alkyl group, or a phenyl group.

16. The compound of claim 1, wherein the ether group is —C—O—CR^{19} , where R^{19} is a substituted pyridine.

17. The compound of claim 1, wherein the amide group is $\text{—(C=O)N(R}^{20})_2$, or $\text{—CH}_2(\text{C=O)N(R}^{20})_2$ where each R^{20} is, independently, H or $\text{—CH}_2\text{CH}_2\text{N(CH}_3)_2$.

18. A compound including resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, said compound having the formula:



where D forms a heterocyclic ring having 3 to 5 atoms.

19. The compound of claim 18, wherein the heterocyclic ring is a 3-membered ring and one of the atoms in the ring is oxygen.

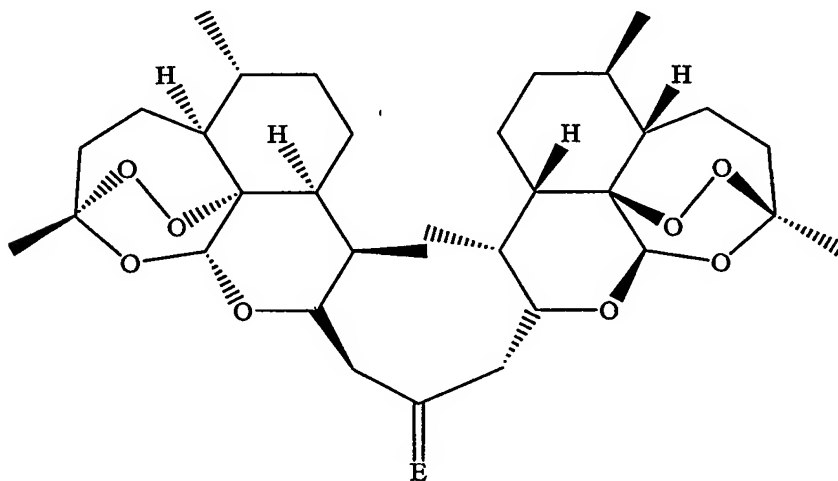
20. The compound of claim 18, wherein the heterocyclic ring is a 5-membered ring and two of the atoms in the ring are oxygen.

21. The compound of claim 20, wherein the heterocyclic ring is substituted with an oxygen atom.

22. The compound of claim 21, wherein another atom in the 5-membered ring is a sulfur or a phosphorous atom.

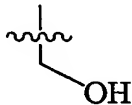
23. The compound of claim 22, wherein the 5-membered ring is substituted with 1 or 2 oxygen atoms bonded to the sulfur atom.

5 24. A compound including resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, said compound having the formula:

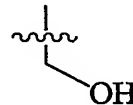


10 where E is H, O, NR, CH₂ or S wherein R may be hydrogen, alkyl, aryl or any other substituent.

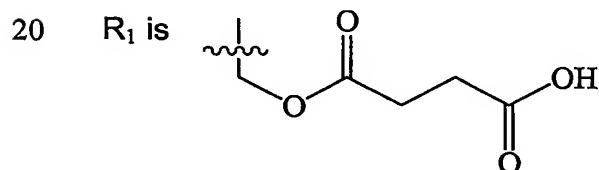
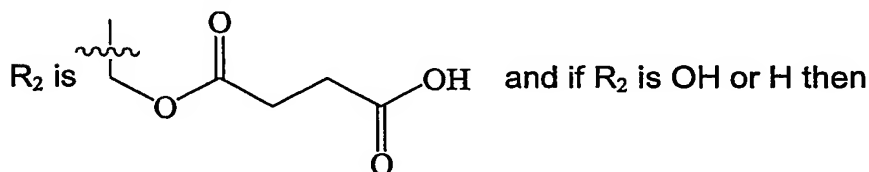
25. The compound of claim 1 wherein if R₁ is H or -OH then R₂ is



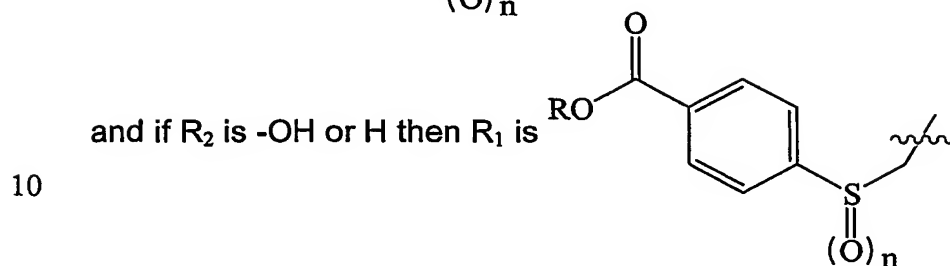
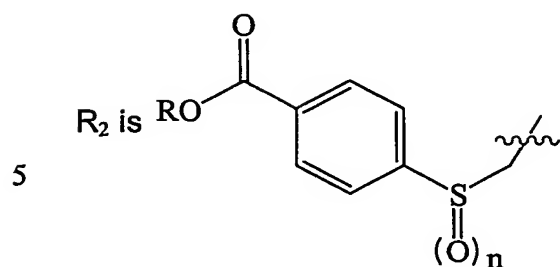
and if R₂ is OH or H then R₁ is



15 26. The compound of claim 1, wherein if R₁ is H or -OH then

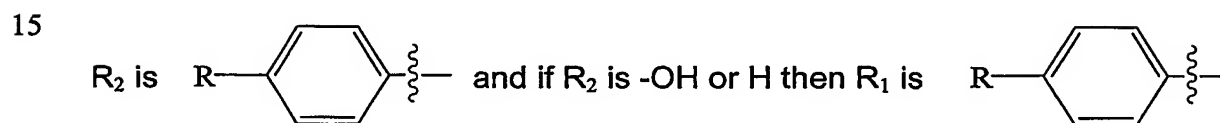


27. The compound of claim 1, wherein if R_1 is H or -OH then



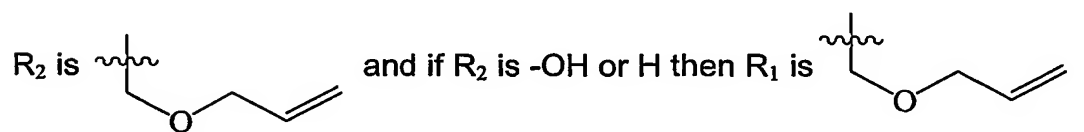
wherein R is hydrogen or a methyl group when n is 0 or 2.

28. The compound of claim 1, wherein if R_1 is H or -OH then

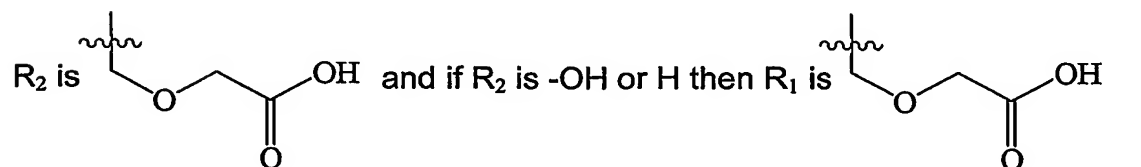


wherein R may be $\text{CH}_2=\text{CH}$ or COOH .

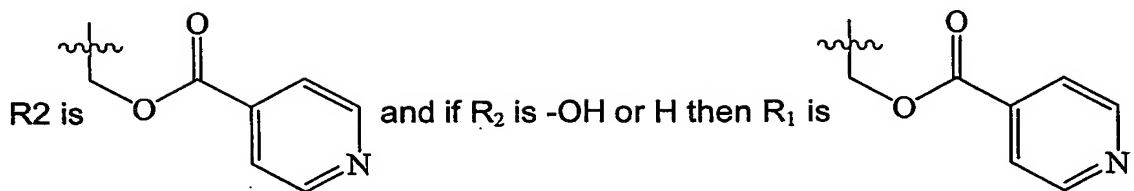
29. The compound of claim 1, wherein if R_1 is H or -OH then



30. The compound of claim 1, wherein if R_1 is H or -OH then

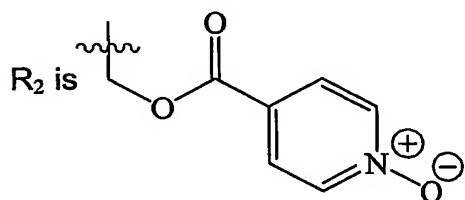


31. The compound of claim 1, wherein if R_1 is H or -OH then



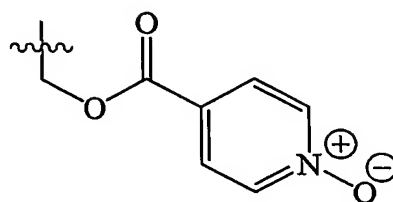
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32. The compound of claim 1, wherein if R₁ is H or -OH then



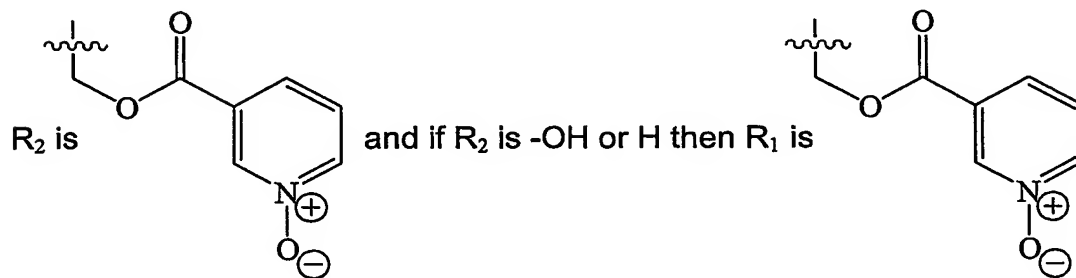
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and if R₂ is -OH or H then R₁ is



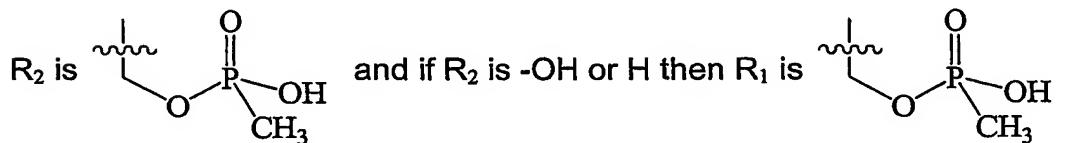
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33. The compound of claim 1, wherein if R₁ is H or -OH then



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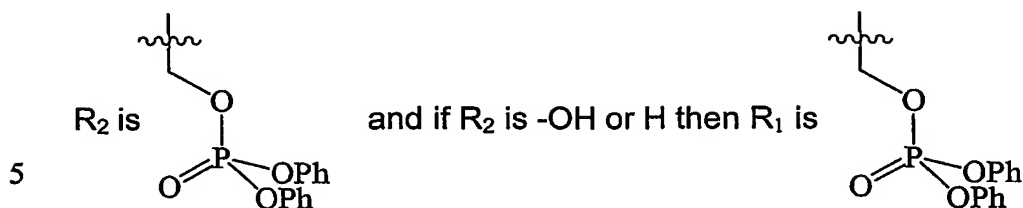
34. The compound of claim 1, wherein if R₁ is H or -OH then



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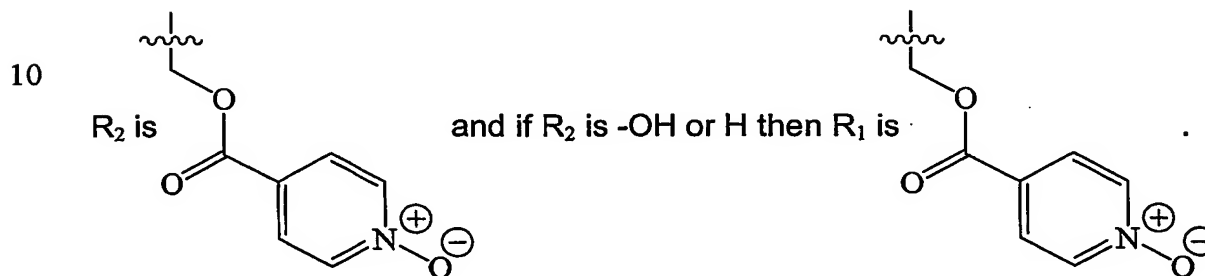
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35. The compound of claim 1, wherein if R_1 is H or -OH then



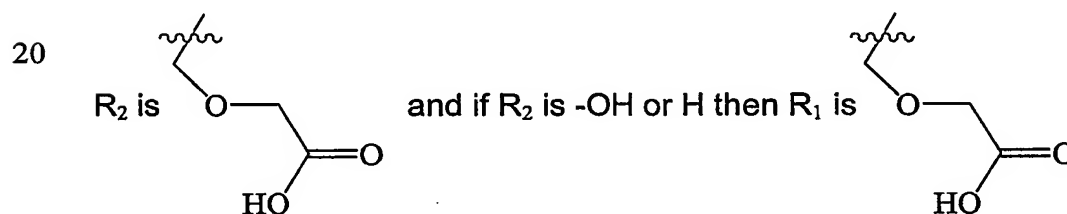
36. The compound of claim 1, wherein if R_1 is H then R_2 -OH.

37. The compound of claim 1, wherein if R_1 is H or -OH then

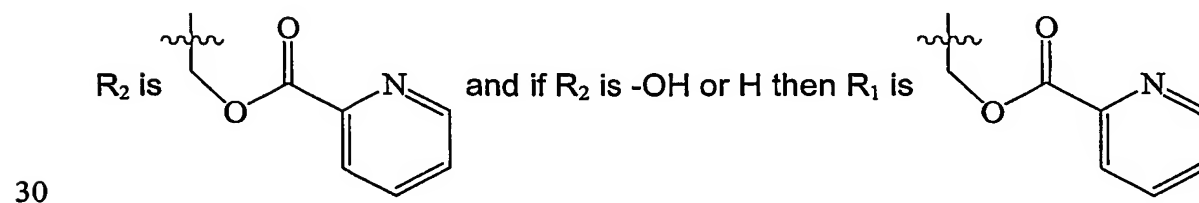


15 38. The compound of claim 1, wherein if R_1 is H then R_2 is carboxylic acid.

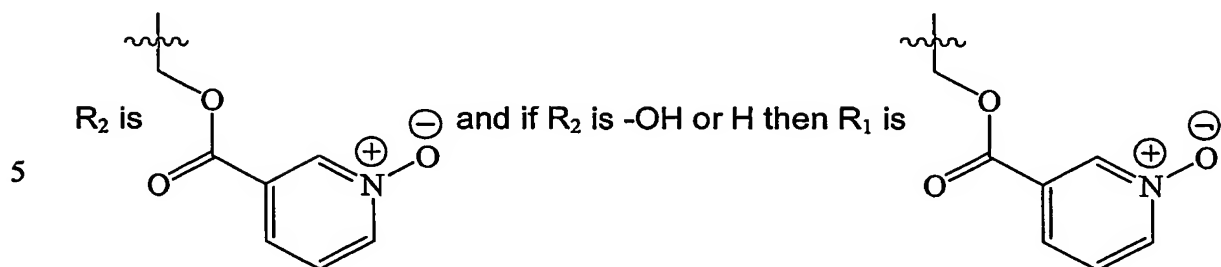
39. The compound of claim 1, wherein if R_1 is H or -OH then



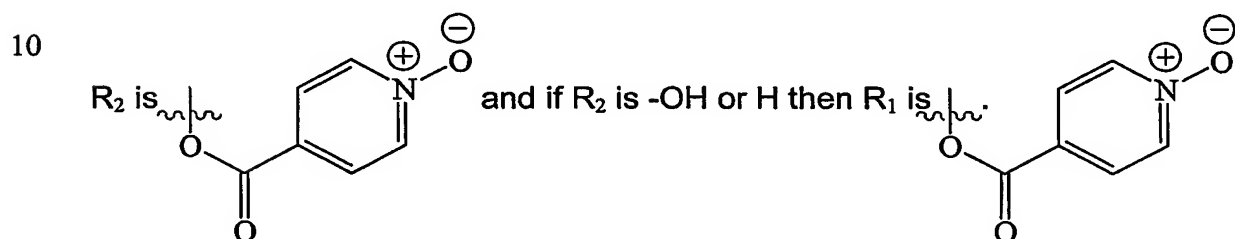
25 40. The compound of claim 1, wherein if R_1 is H or -OH then



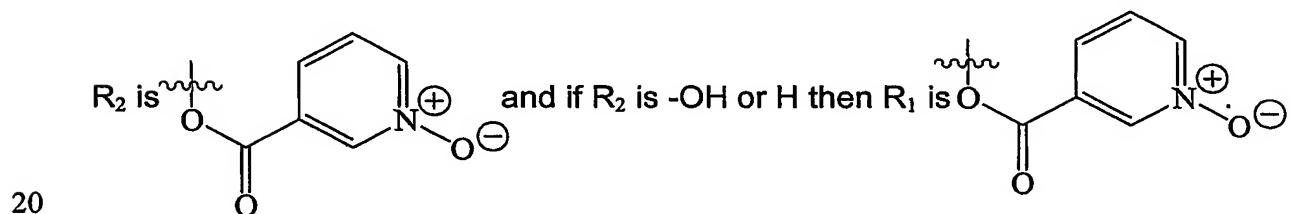
41. The compound of claim 1, wherein if R_1 is H or -OH then



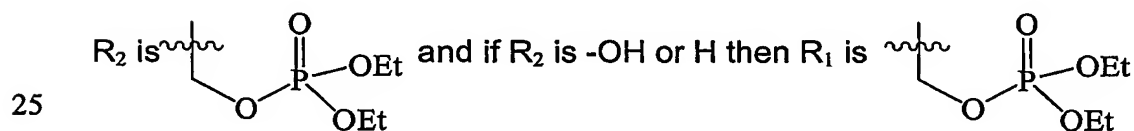
42. The compound of claim 1, wherein if R_1 is H or -OH then



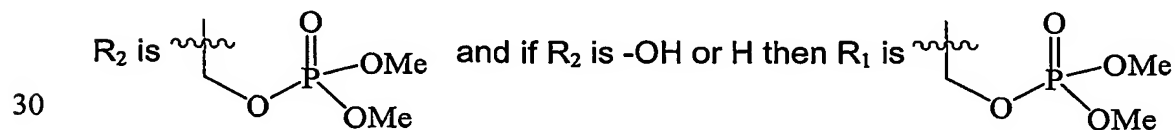
15 43. The compound of claim 1, wherein if R_1 is H or -OH then



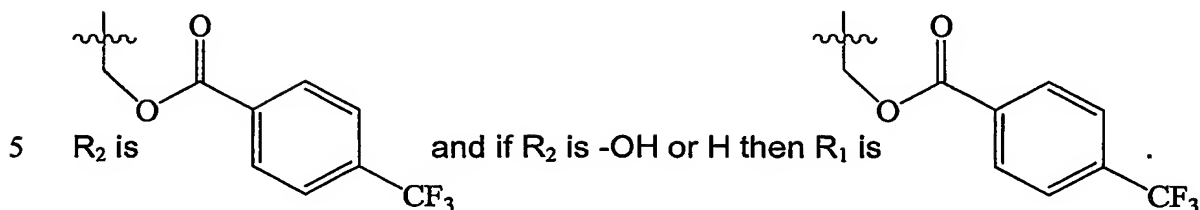
44. The compound of claim 1, wherein if R_1 is H or -OH then



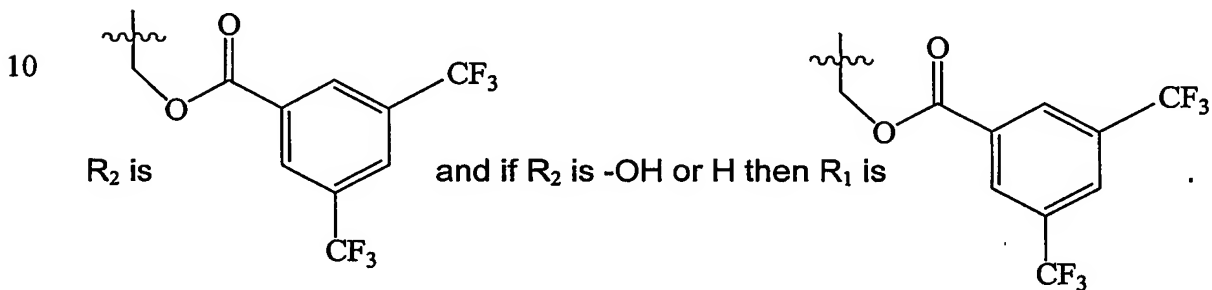
45. The compound of claim 1, wherein if R_1 is H or -OH then



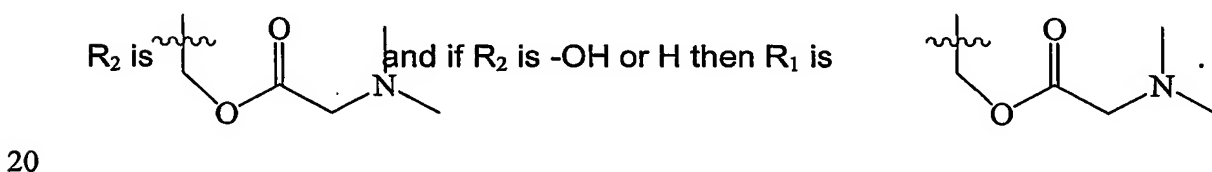
46. The compound of claim 1, wherein if R_1 is H or -OH then



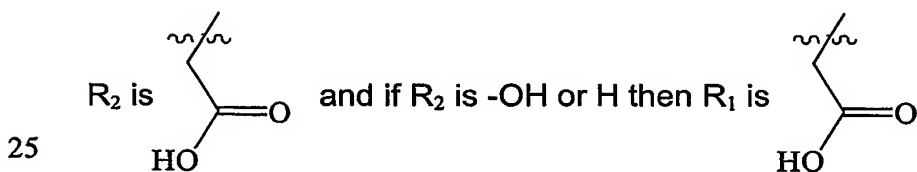
47. The compound of claim 1, wherein if R_1 is H or -OH then



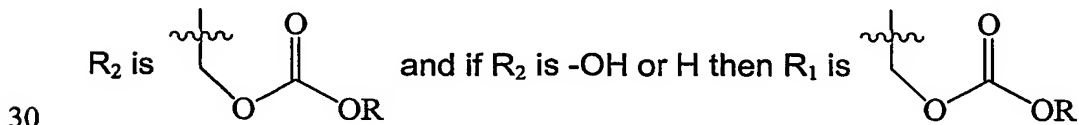
48. The compound of claim 1, wherein if R_1 is H or -OH then



49. The compound of claim 1, wherein if R_1 is H or -OH then

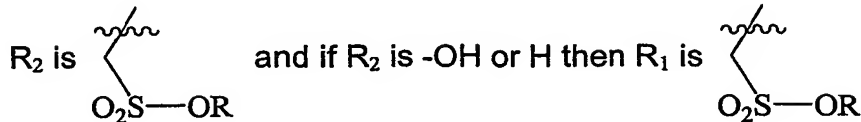


50. The compound of claim 1, wherein if R_1 is H or -OH then



51. The compound of claim 50 wherein R is a methyl or ethyl group.

52. The compound of claim 1, wherein if R_1 is H or -OH then

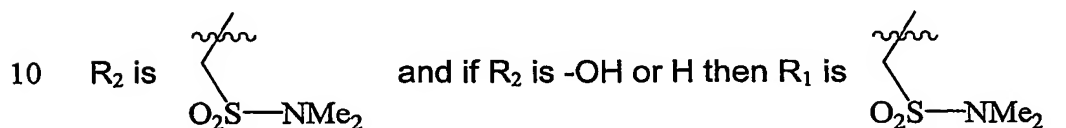


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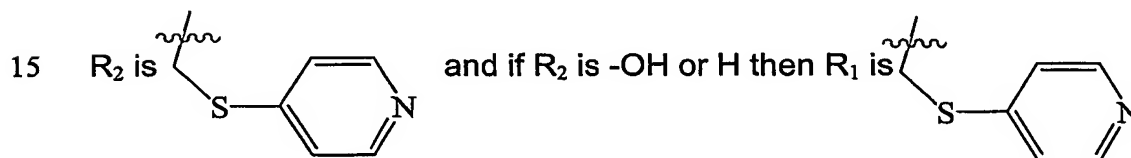
53. The compound of claim 52 wherein R is a methy group.

54. The compound of claim 52 wherein R is an iso-propyl group.

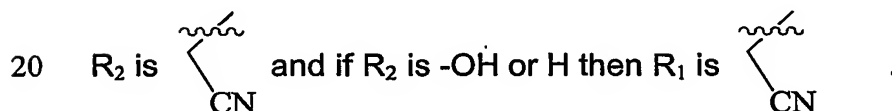
55. The compound of claim 1, wherein if R_1 is H or -OH then



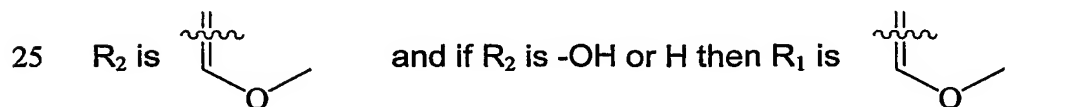
56. The compound of claim 1, wherein if R_1 is H or -OH then



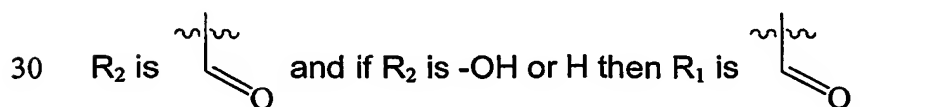
57. The compound of claim 1, wherein if R_1 is H or -OH then



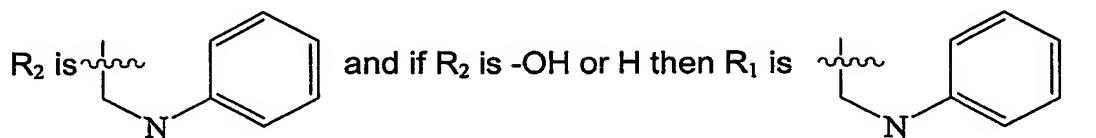
58. The compound of claim 1, wherein if R_1 is H or -OH then



59. The compound of claim 1, wherein if R_1 is H or -OH then

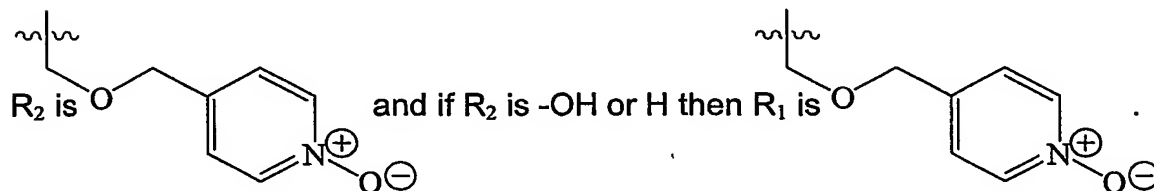


60. The compound of claim 1, wherein if R_1 is H or -OH then



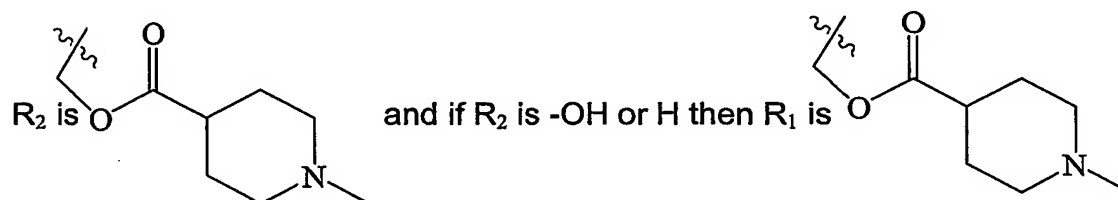
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61. The compound of claim 1, wherein if R_1 is H or -OH then



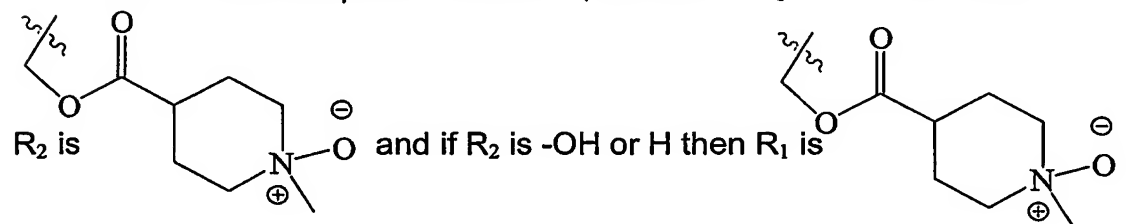
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62. The compound of claim 1, wherein if R_1 is H or -OH then



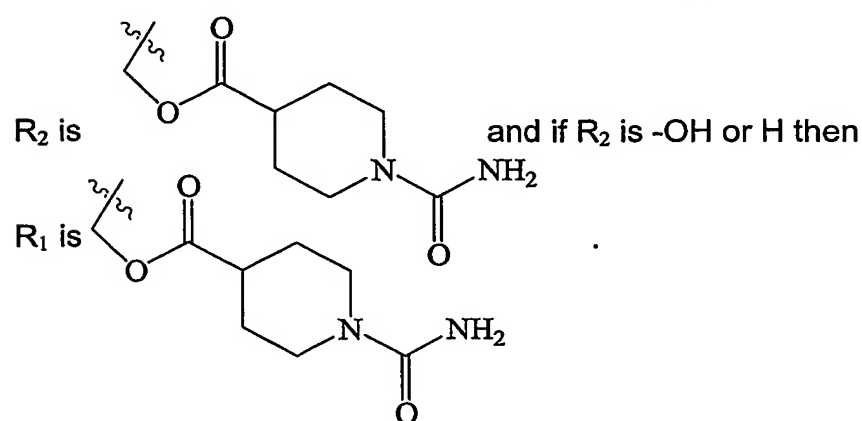
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63. The compound of claim 1, wherein if R_1 is H or -OH then



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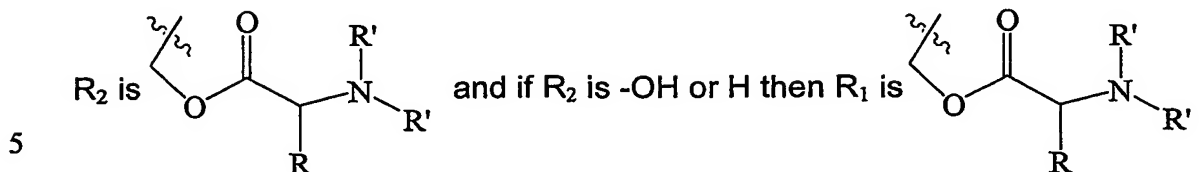
64. The compound of claim 1, wherein if R_1 is H or -OH then



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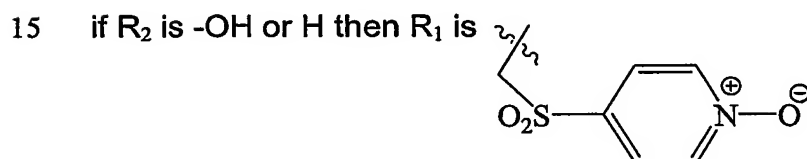
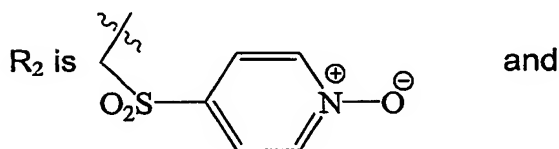
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65. The compound of claim 1, wherein if R_1 is H or -OH then

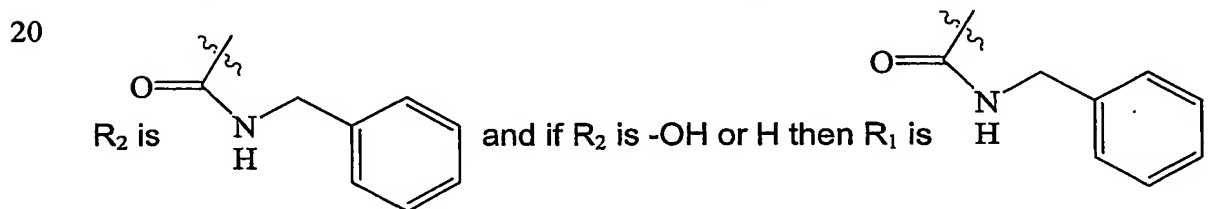


66. The compound of claim 66 wherein each R' and R independently can be any amino acid of all possible stereochemistries and with any degree and choice of protecting group.

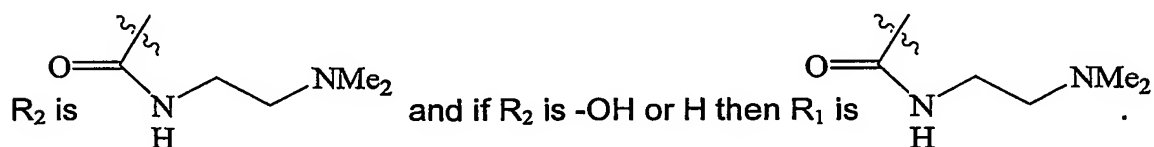
10 67. The compound of claim 1, wherein if R_1 is H or -OH then



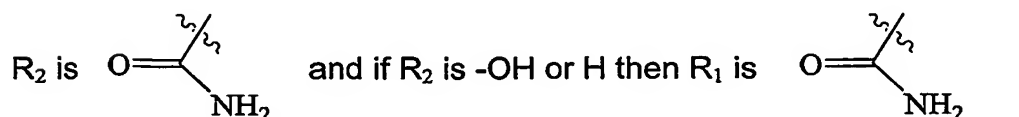
68. The compound of claim 1, wherein if R_1 is H or -OH then



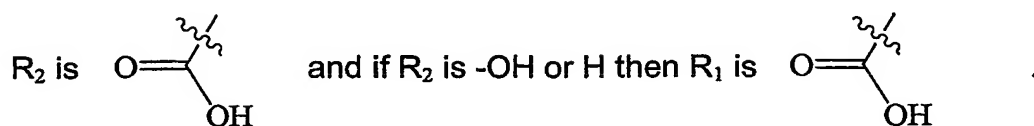
25 69. The compound of claim 1, wherein if R_1 is H or -OH then



30 70. The compound of claim 1, wherein if R_1 is H or -OH then

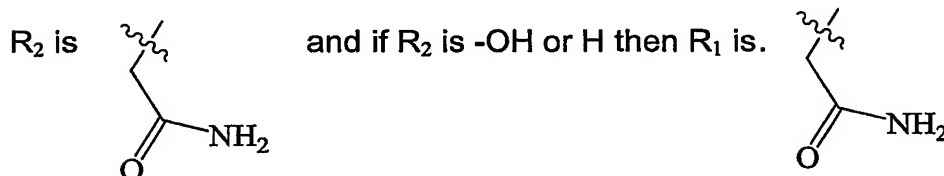


71. The compound of claim 1, wherein if R_1 is H or -OH then



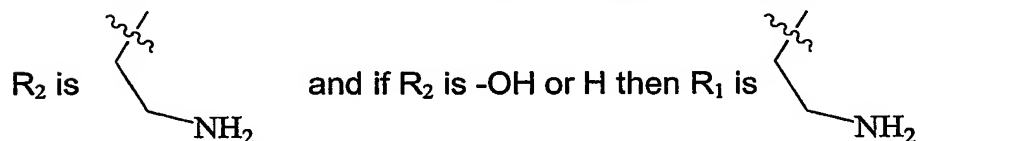
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72. The compound of claim 1, wherein if R_1 is H or -OH then



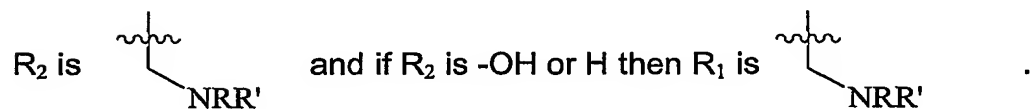
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73. The compound of claim 1, wherein if R_1 is H or -OH then




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74. The compound of claim 1, wherein if R_1 is H or -OH then

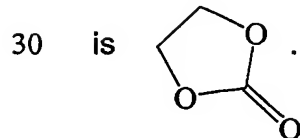


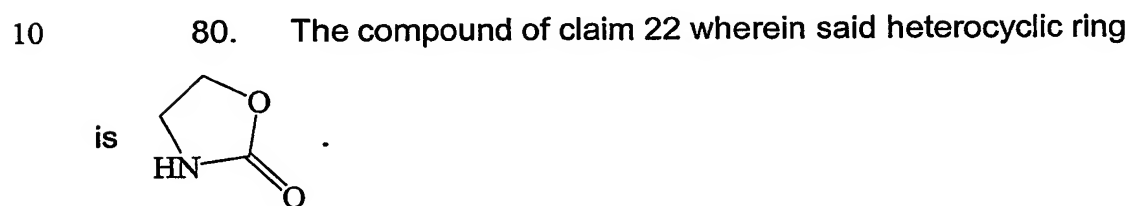
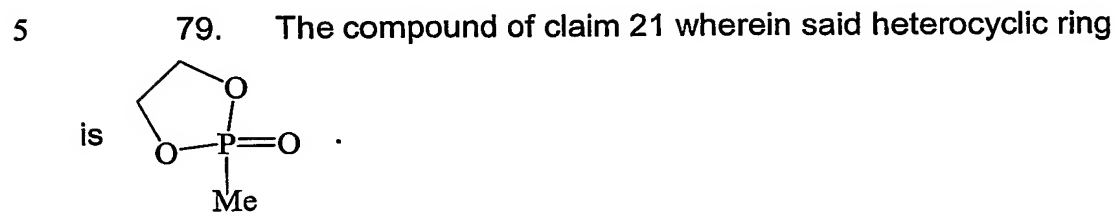
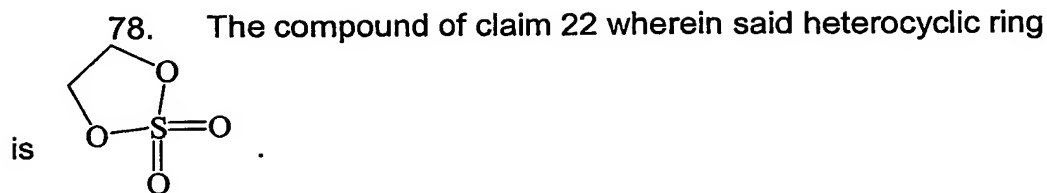
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75. The compound of claim 74, wherein R and R' are independently of each other hydrogen, alkyl, aryl, or allyl.

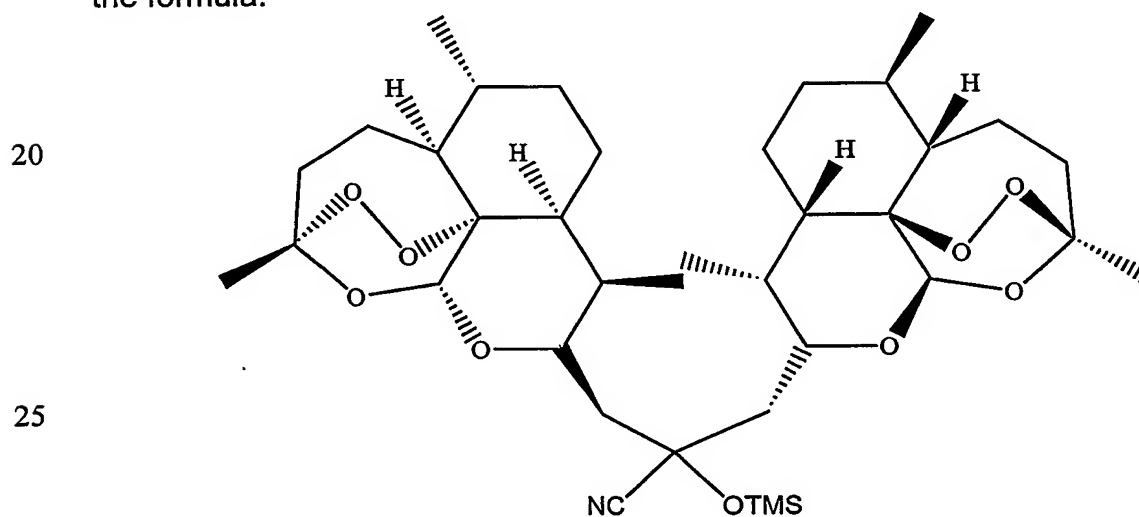
25 76. The compound of claim 19 wherein said heterocyclic ring is .

77. The compound of claim 21 wherein said heterocyclic ring

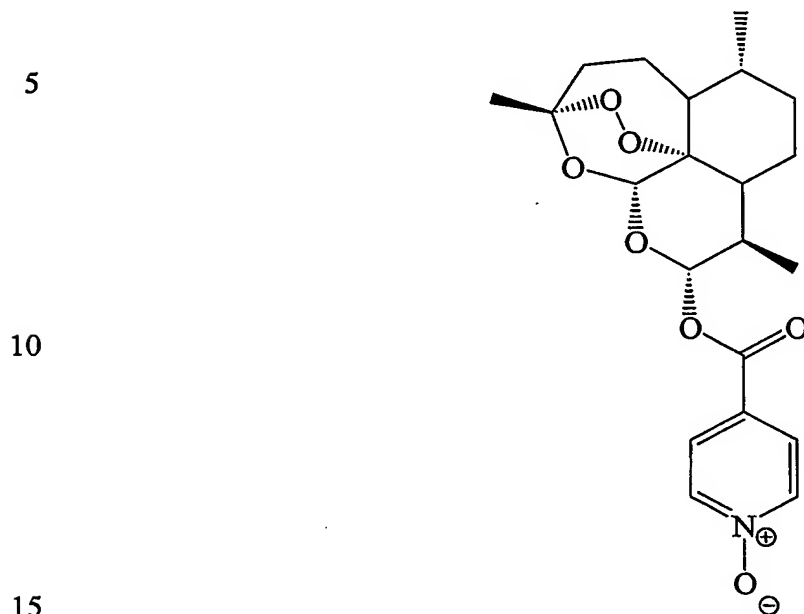




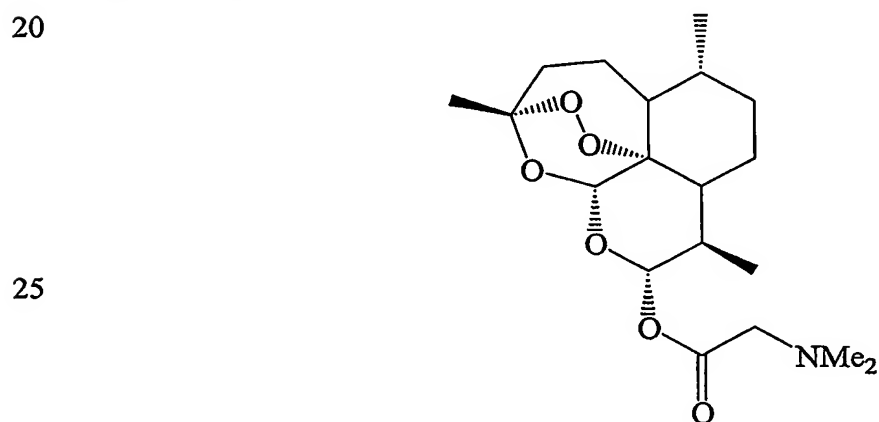
15 81. A compound including resolved enantiomers, diastereomers,
solvates and pharmaceutical acceptable salts thereof, said compound having
the formula:



82. A compound including resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, said compound having the formula:



83. A compound including resolved enantiomers, diastereomers, solvates and pharmaceutical acceptable salts thereof, said compound having the formula:



84. A method of treating cancer, which comprises administering to a patient suffering from said cancer a compound or combination of compounds of claims 1 – 83.

85. A method according to claim 84 wherein said cancer is selected from the group of cancers consisting of leukemia, non-small cell lung cancer, colon cancer, central nervous system cancer, melanoma cancer, ovarian cancer, renal cancer, prostate cancer, and breast cancer.

5 86. A method for treating malaria comprising administering an effective amount of a compound or combination of compounds of claims 1-83.

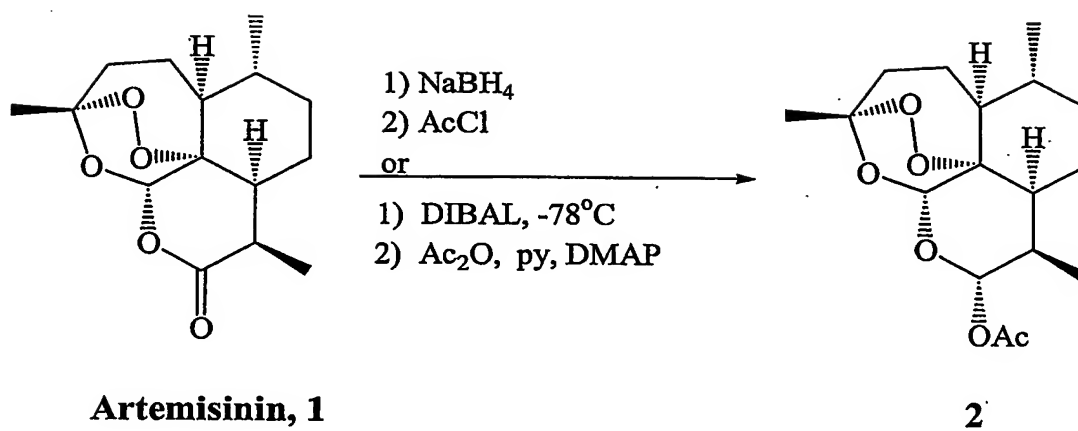


FIGURE 1

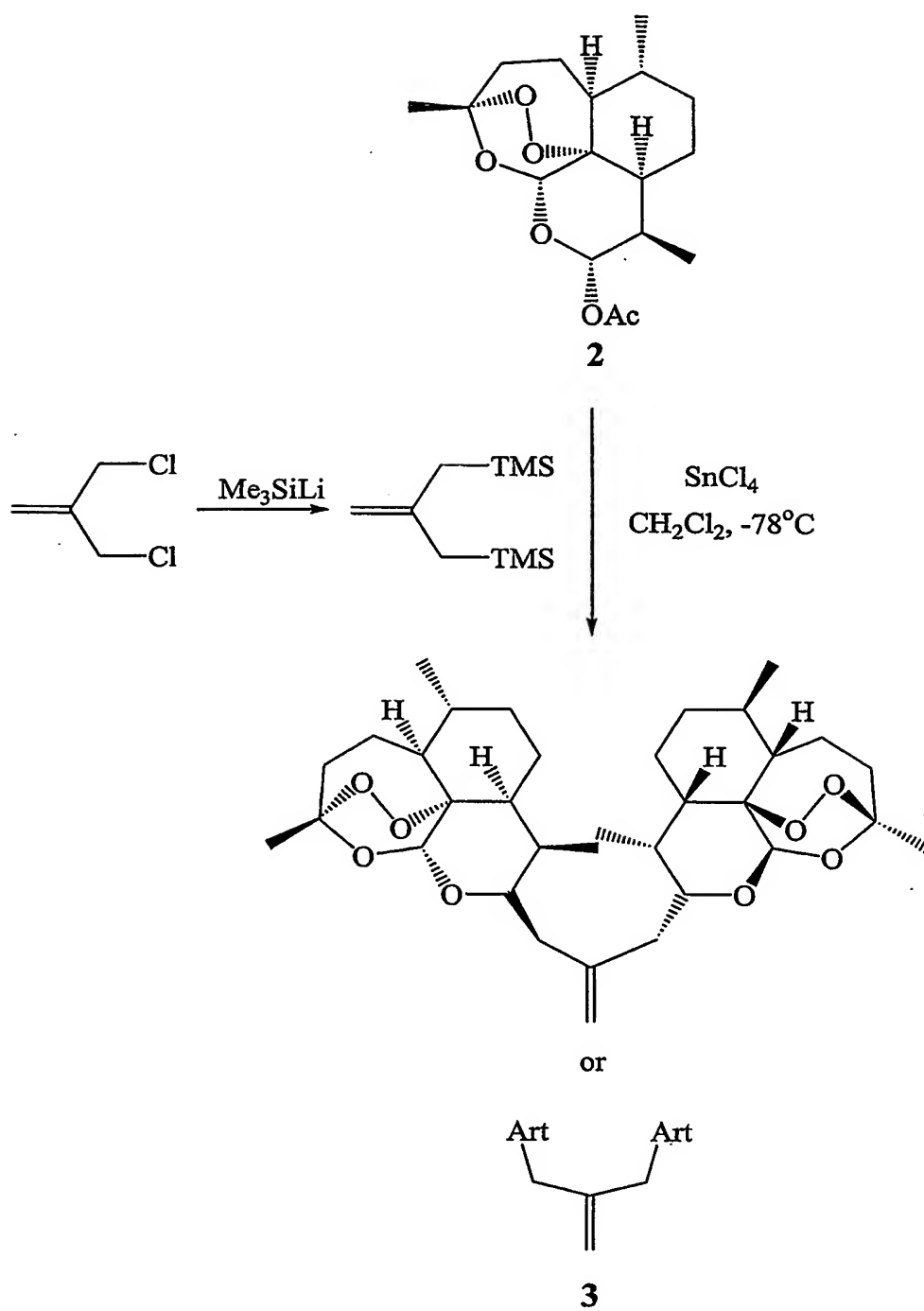


FIGURE 2

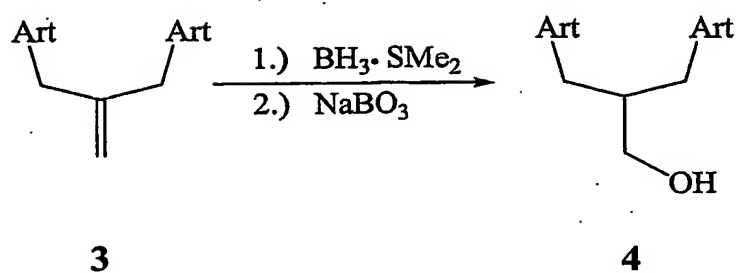


FIGURE 3

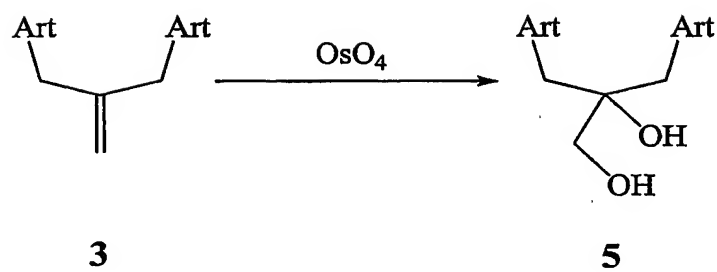


FIGURE 4

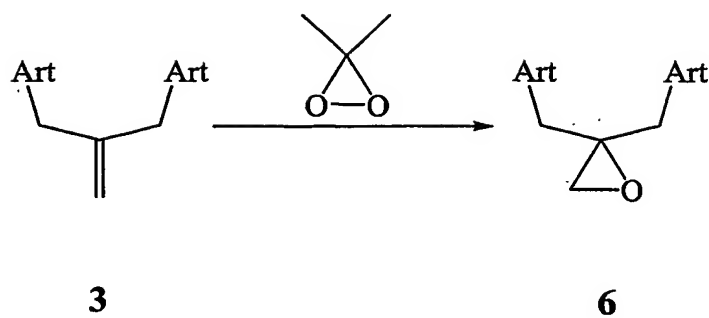


FIGURE 5

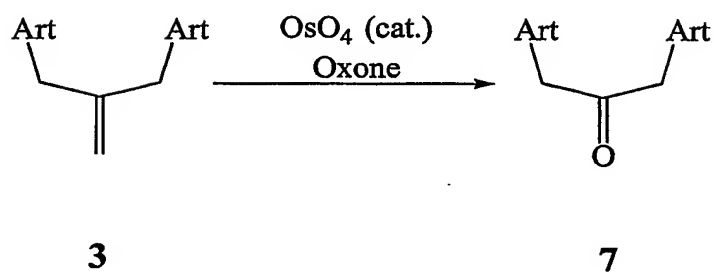


FIGURE 6

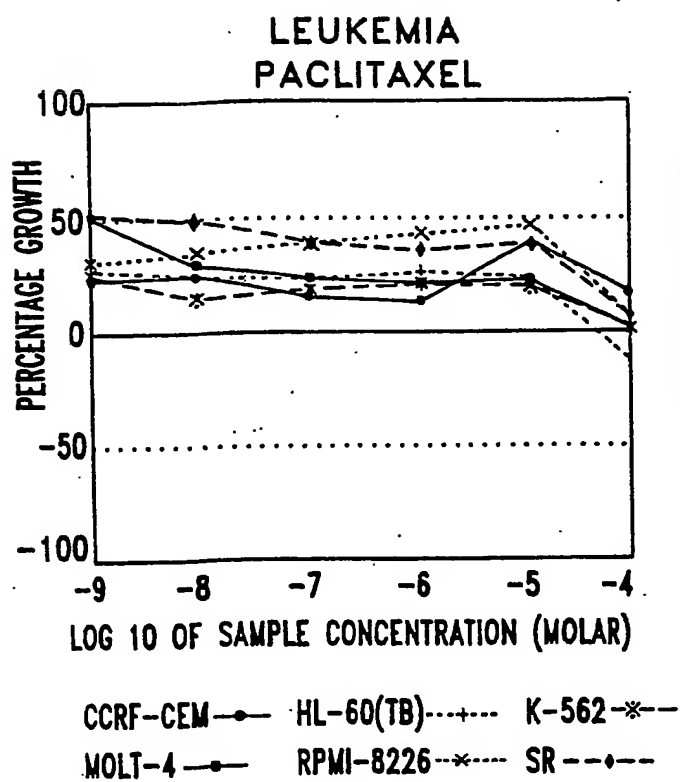


Fig. 7a

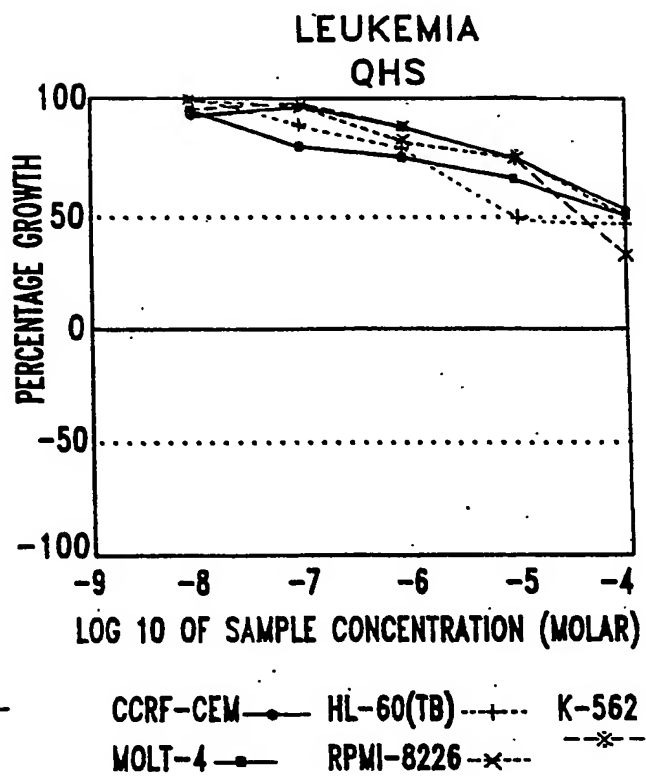


Fig. 7b

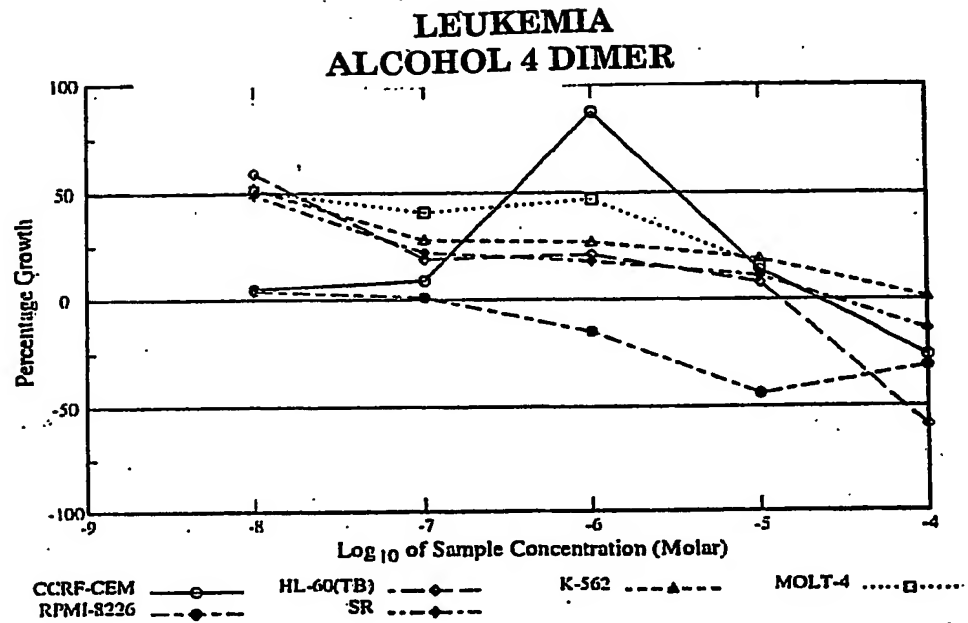


Fig. 7c

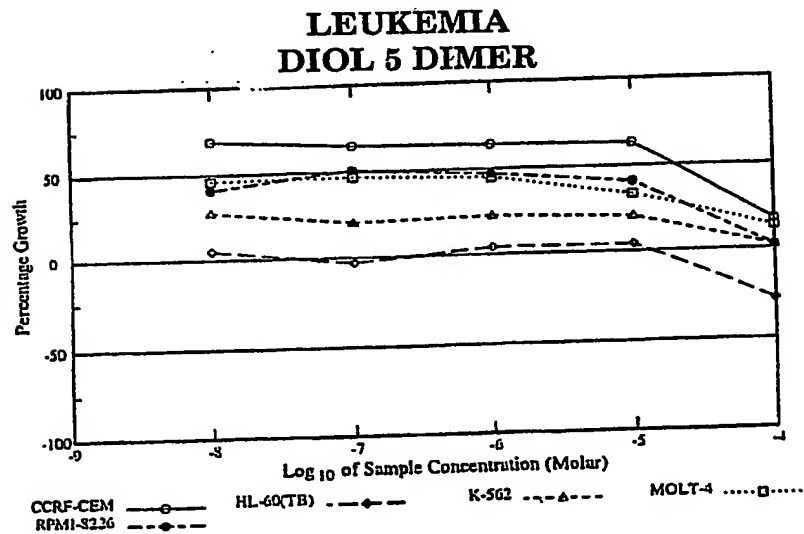


Fig. 7d

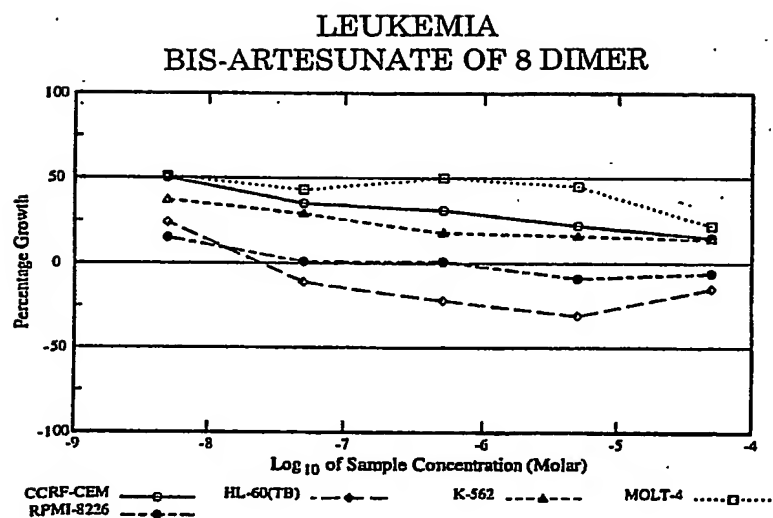


Fig. 7e

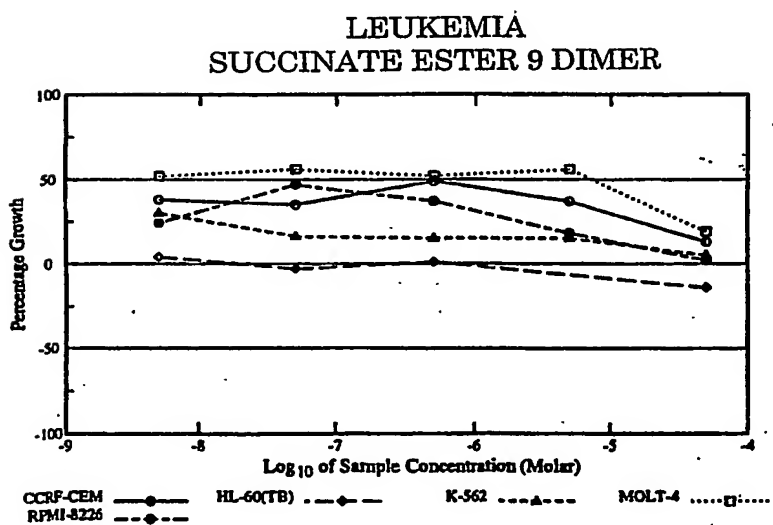


Fig. 7f

LEUKEMIA
BENZOIC ACID 10C DIMER

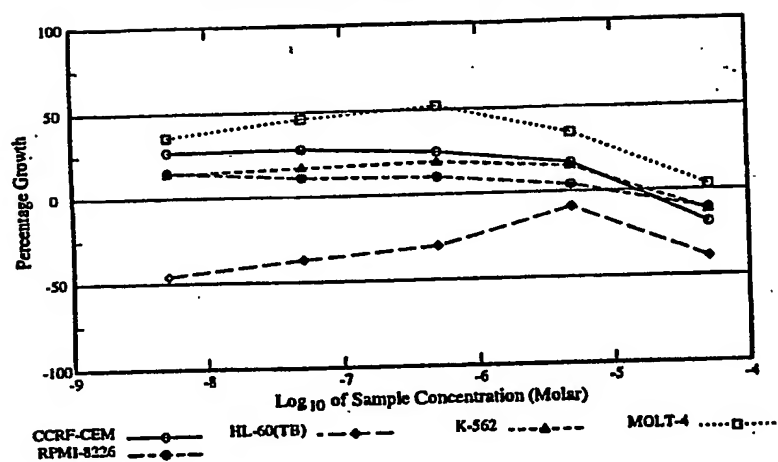


Fig. 7g

LEUKEMIA
DIMER 11B

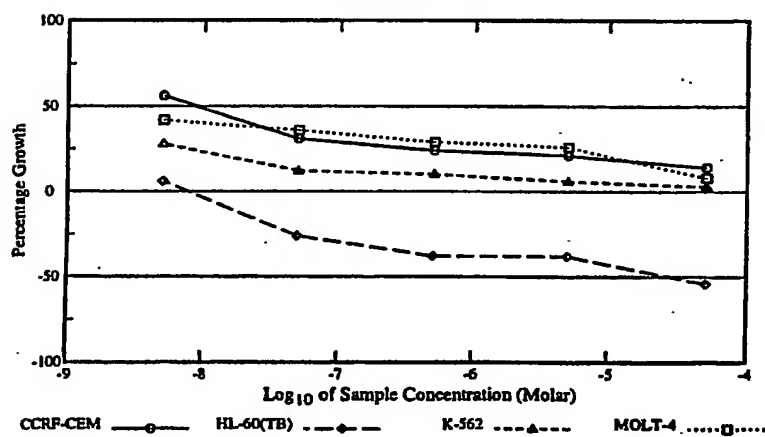
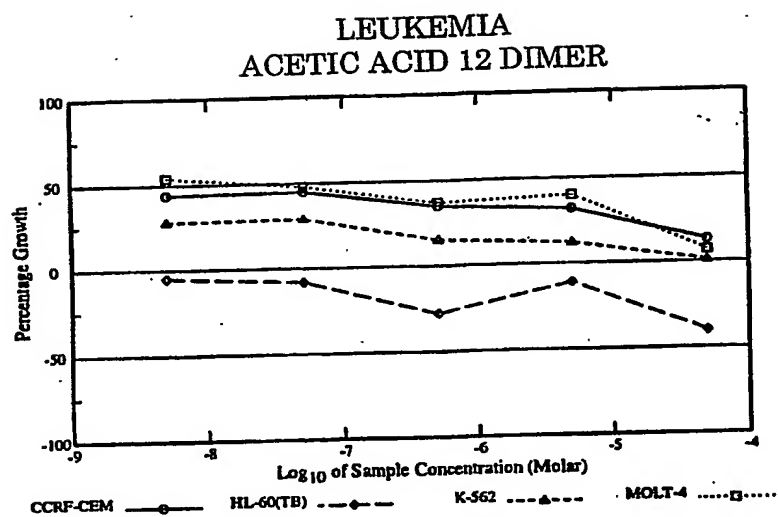
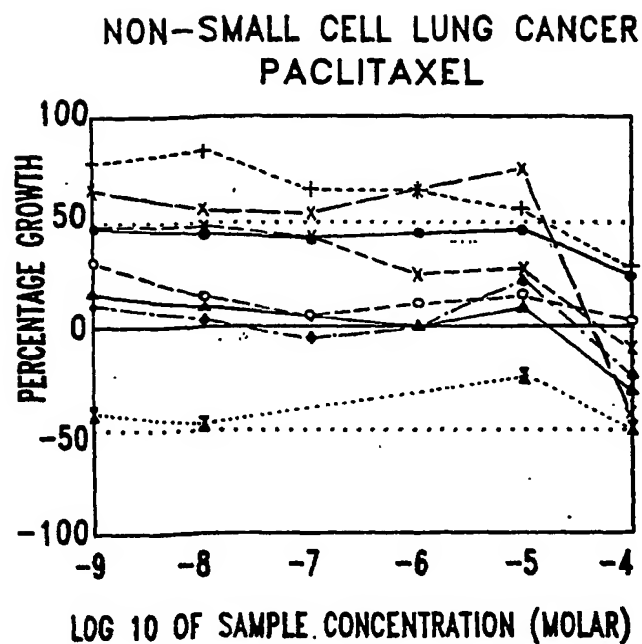


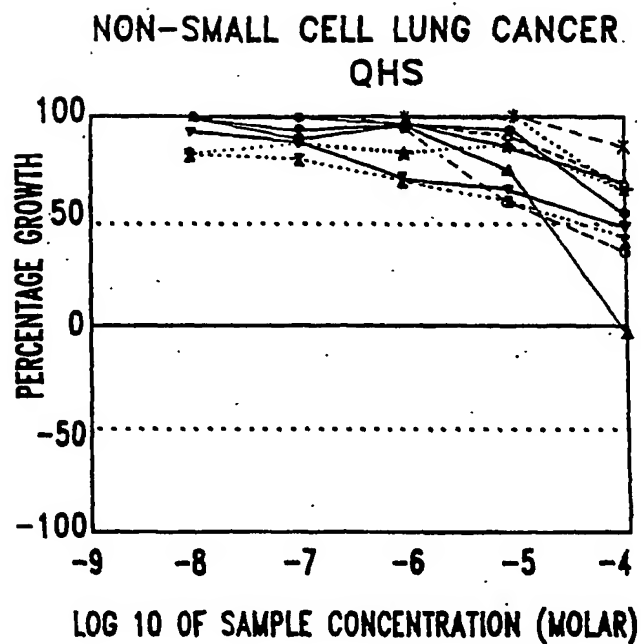
Fig. 7h

**Fig. 7i**



A549/ATCC—●— EKVX—+— HOP-62—*—
 NCI-H226—x— NCI-H23—♦— NCI-H460—▲—
 NCI-H522—x— NCI-H322M—○—

Fig 8a



A549/ATCC—●— EKVX—+— HOP-62—*—
 NCI-H23—■— NCI-H460—○— NCI-H522—▲—
 HOP-18—x— NCI-H322M—○— LXFL-S29—▼—
 NCI-H226—★—

Fig. 8b

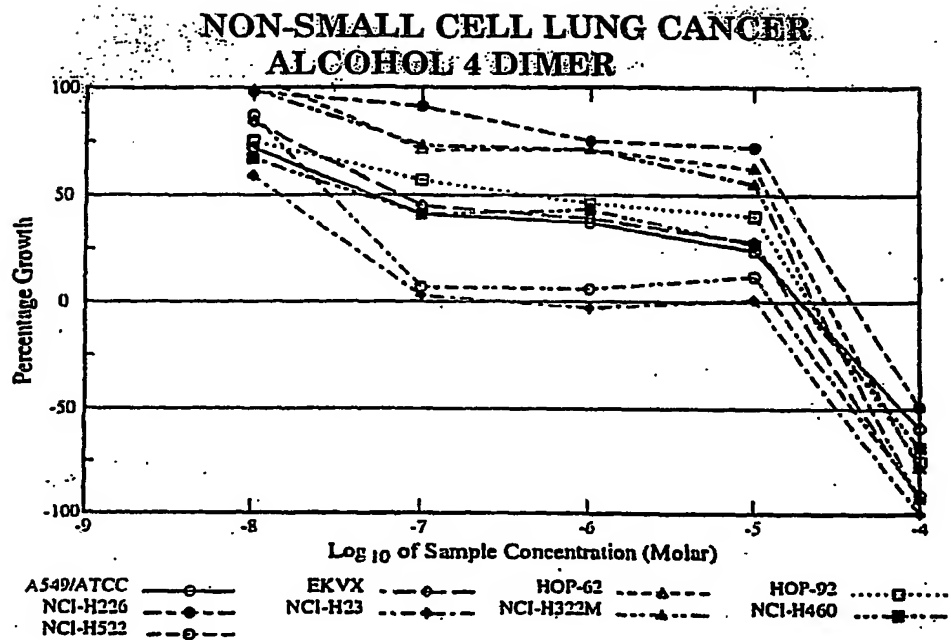


Fig. 8c

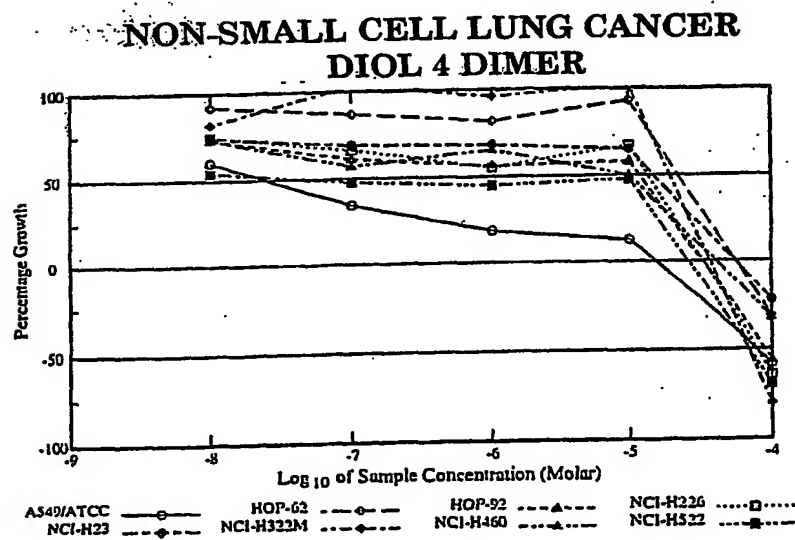


Fig. 8d

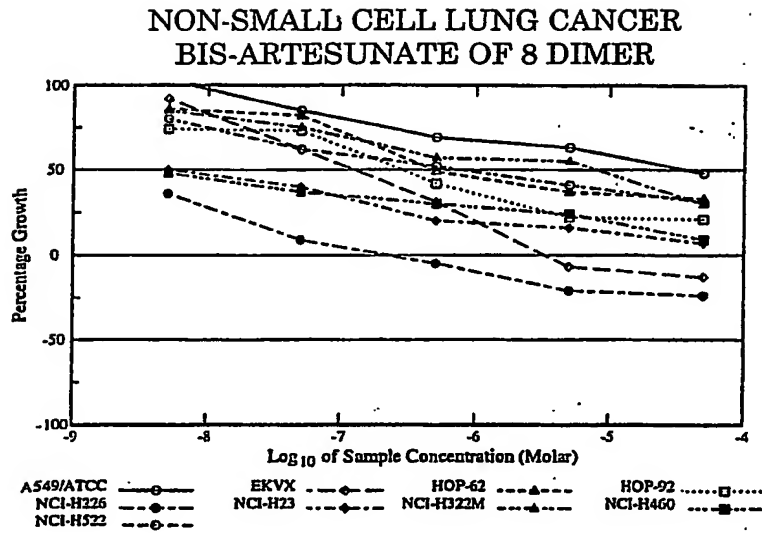


Fig. 8e

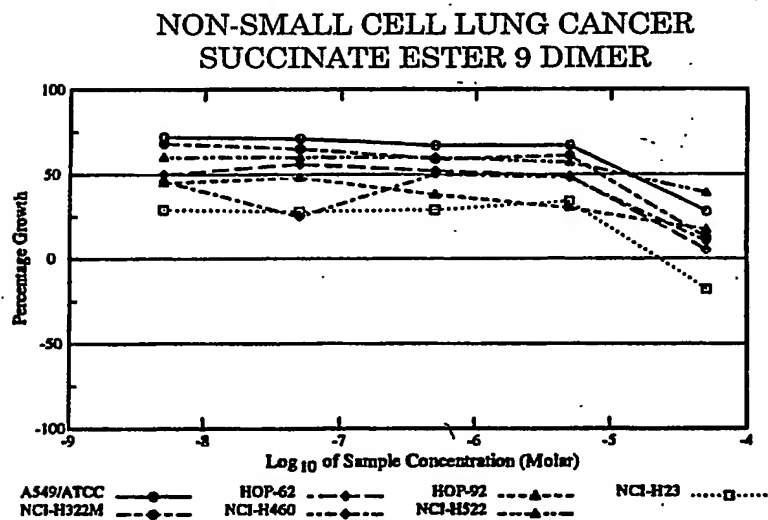


Fig. 8f

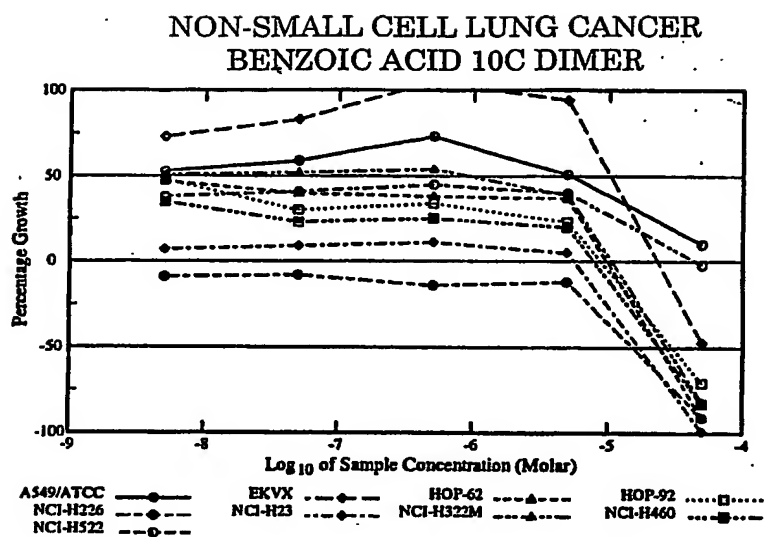


Fig. 8g

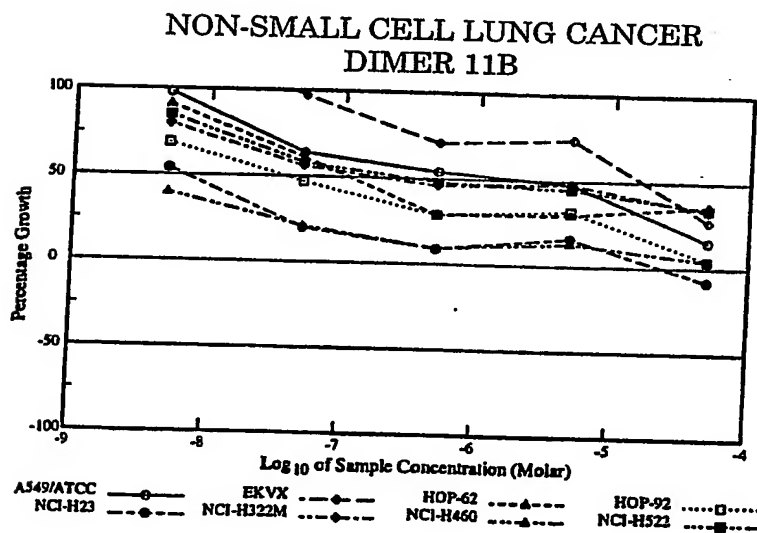
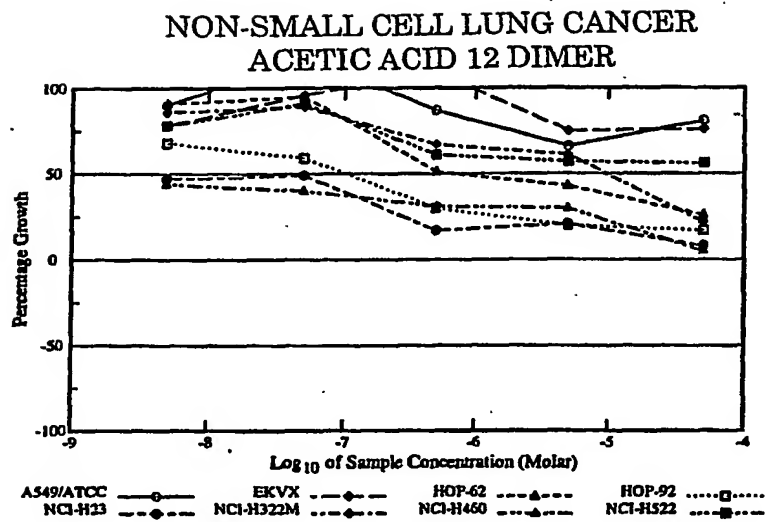


Fig. 8h

**Fig. 8i**

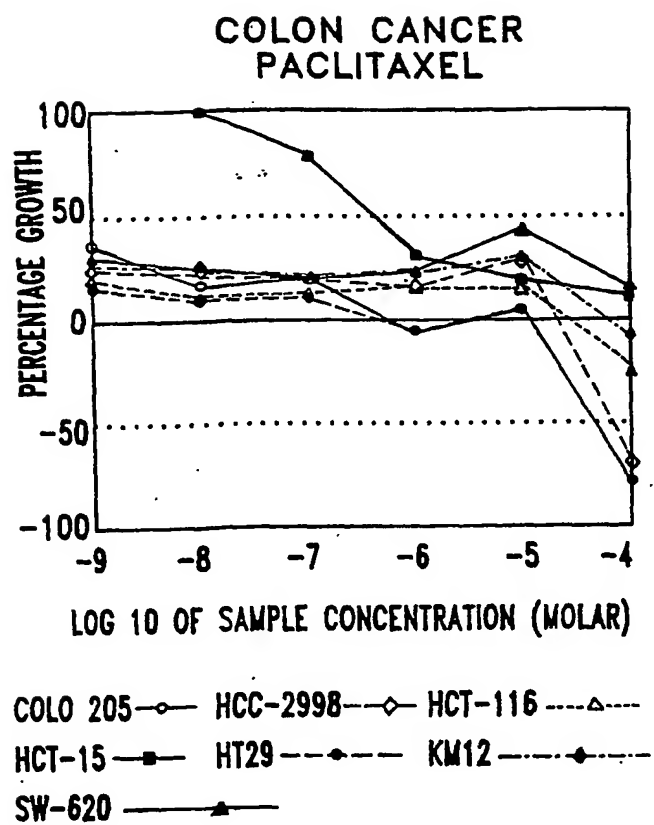


Fig. 9a

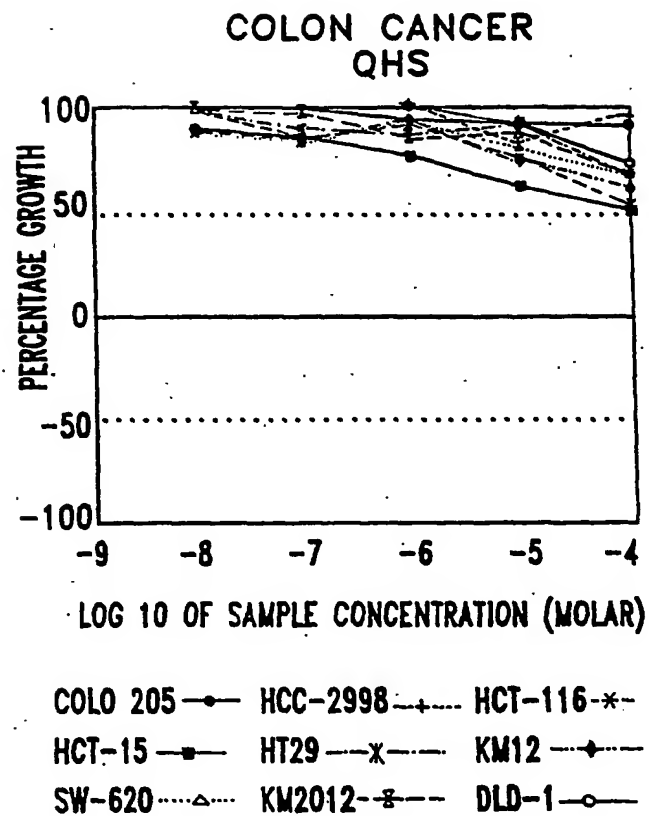


Fig. 9b

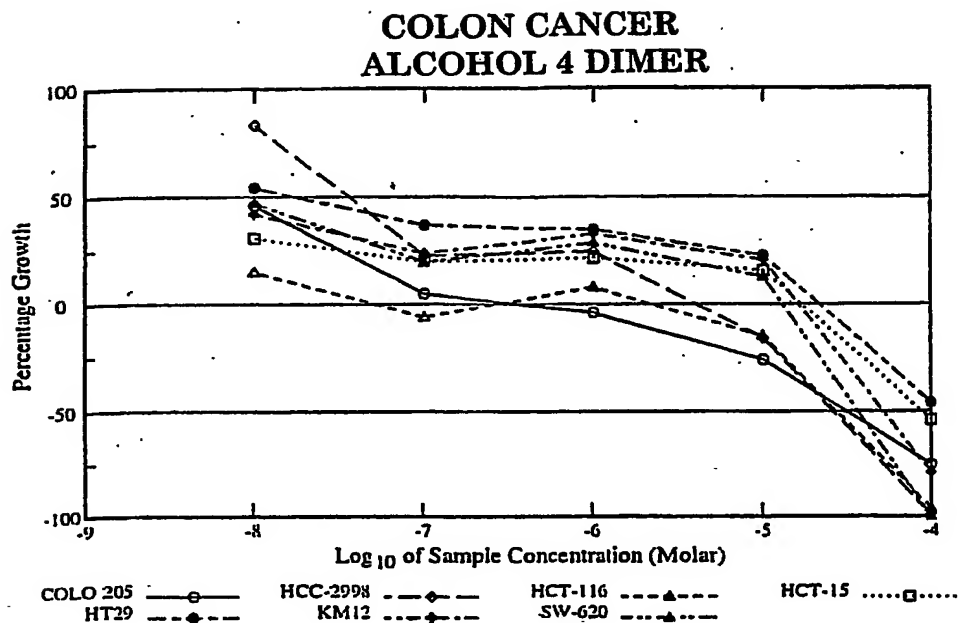


Fig. 9c

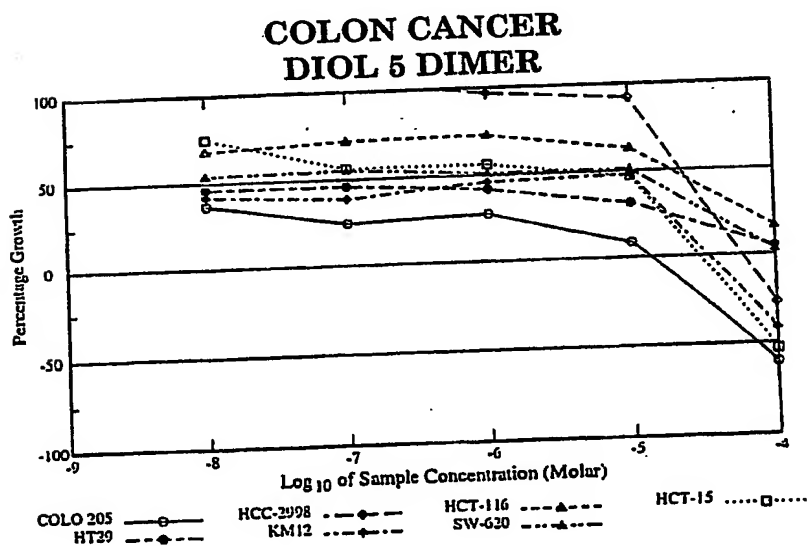
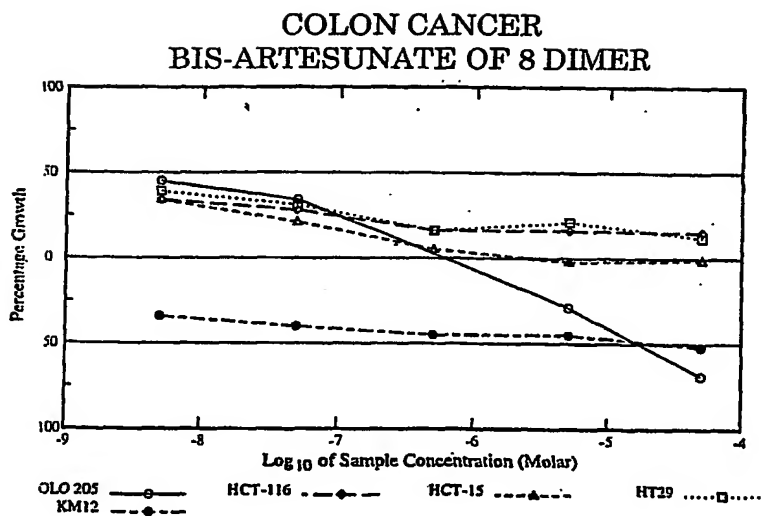
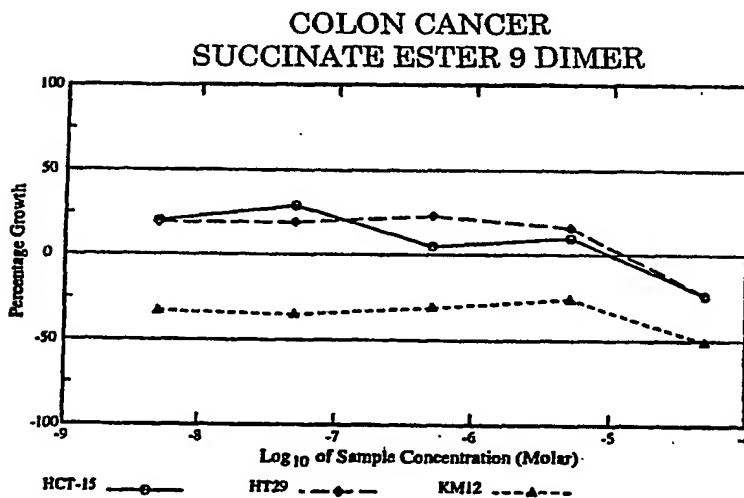


Fig. 9d

**Fig. 9e****Fig. 9f**

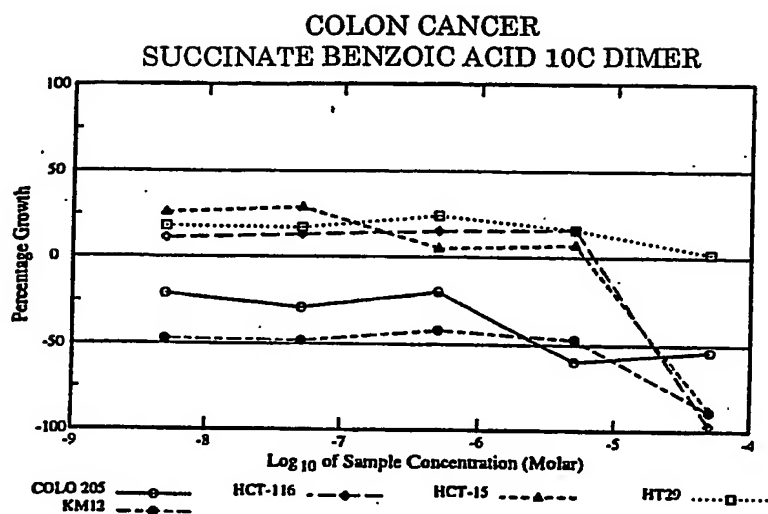


Fig. 9g

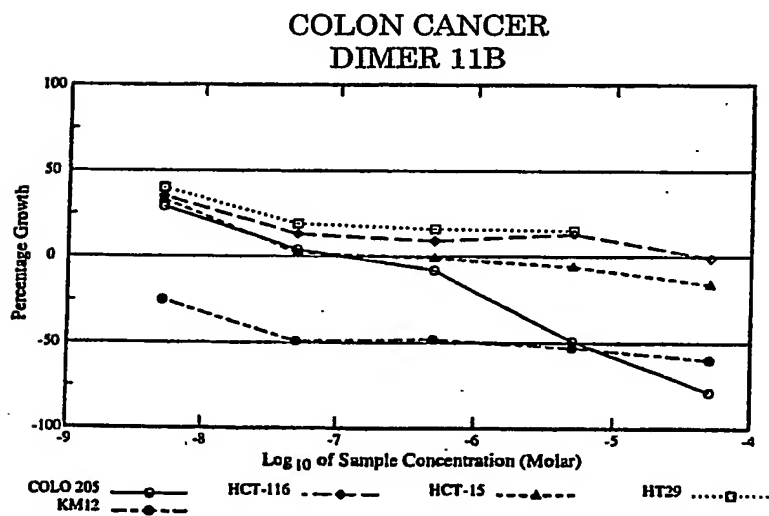
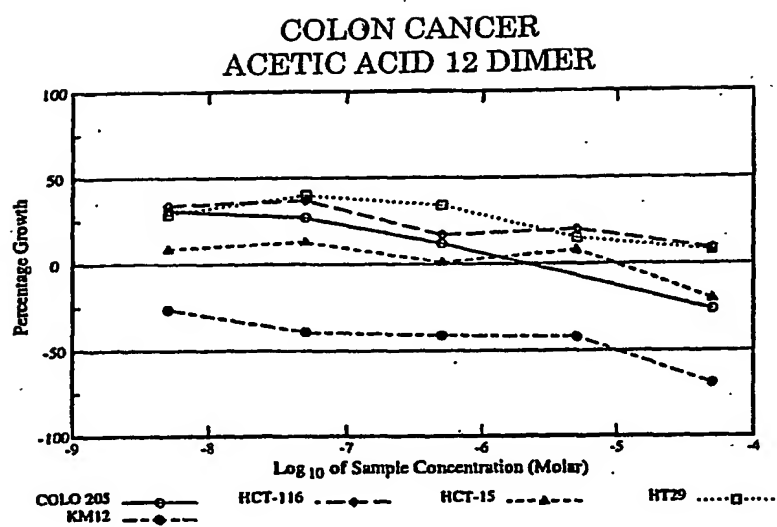


Fig. 9h

**Fig. 9i**

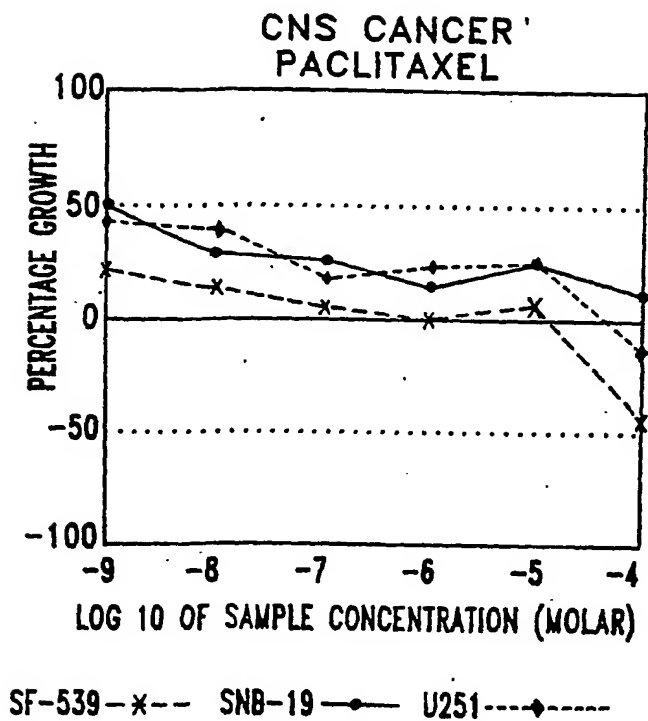


Fig. 10a

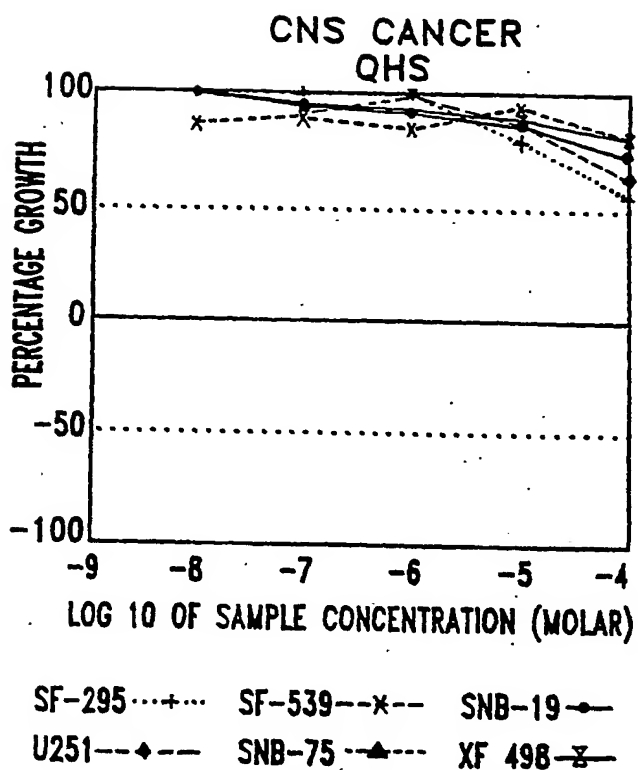


Fig. 10b

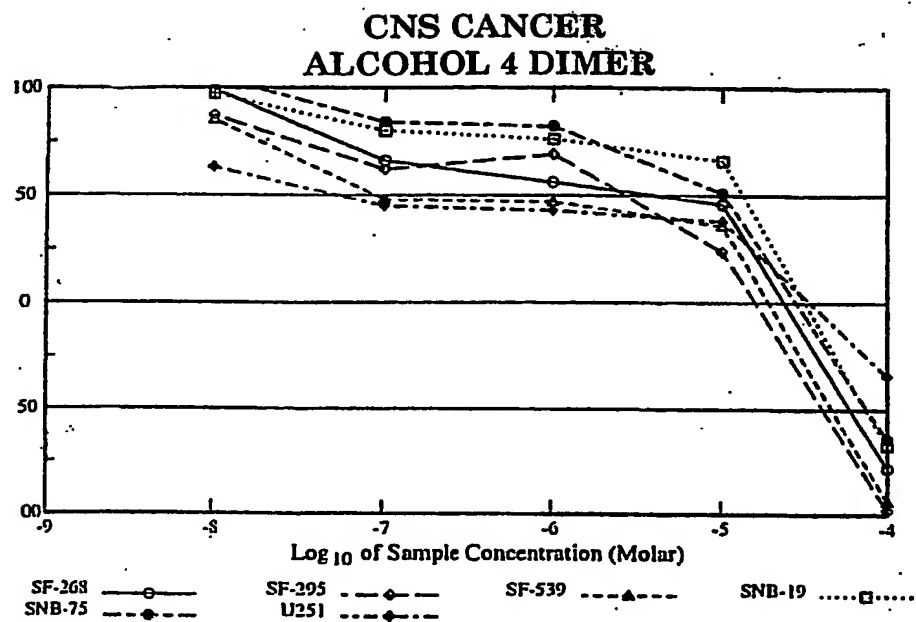


Fig. 10c

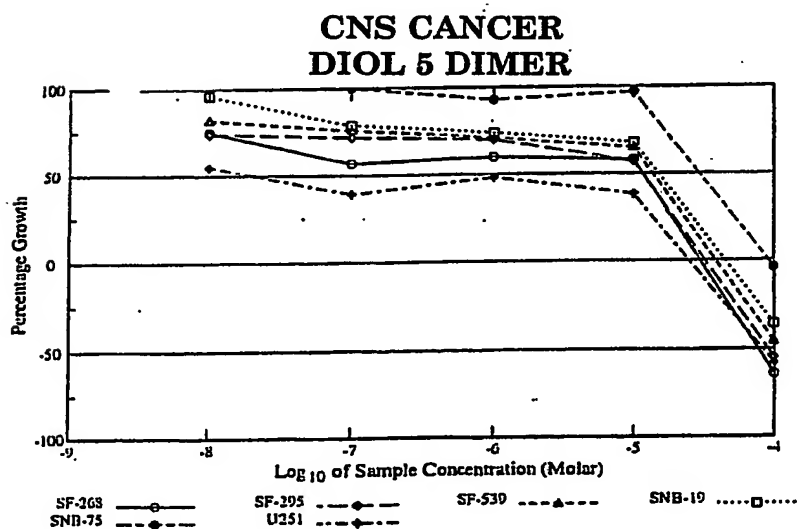


Fig. 10d

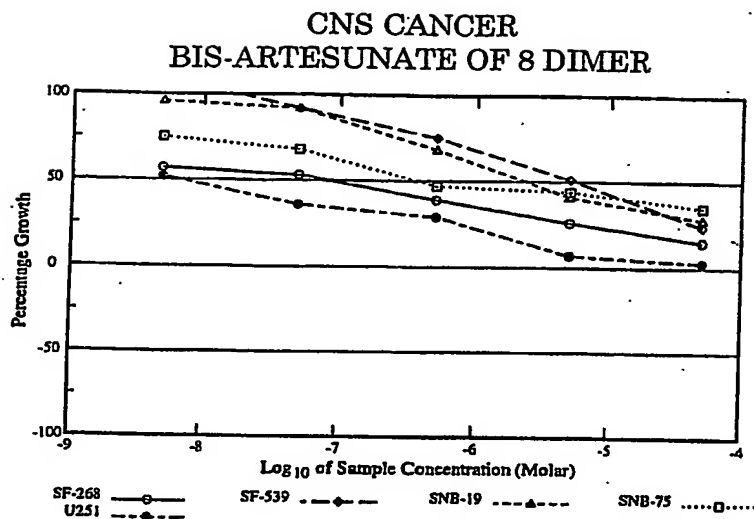


Fig. 10e

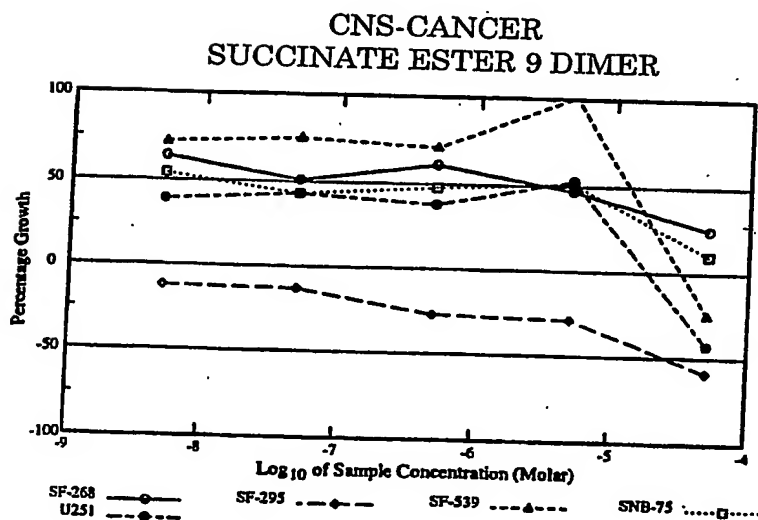


Fig. 10f

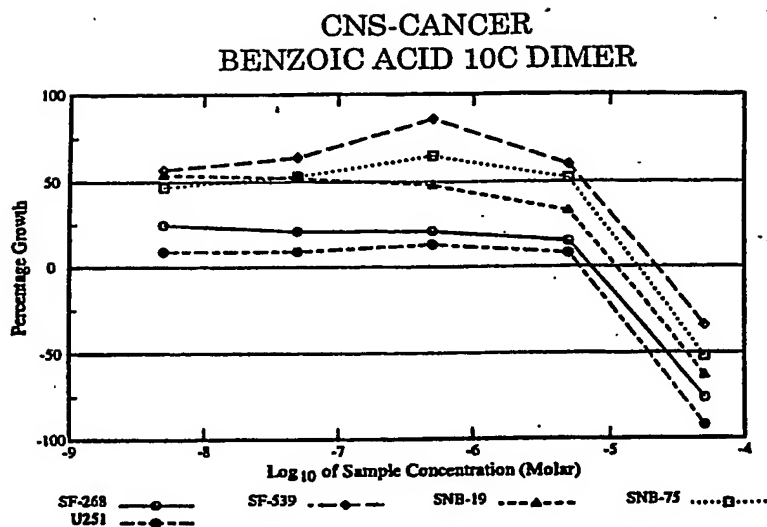


Fig. 10g

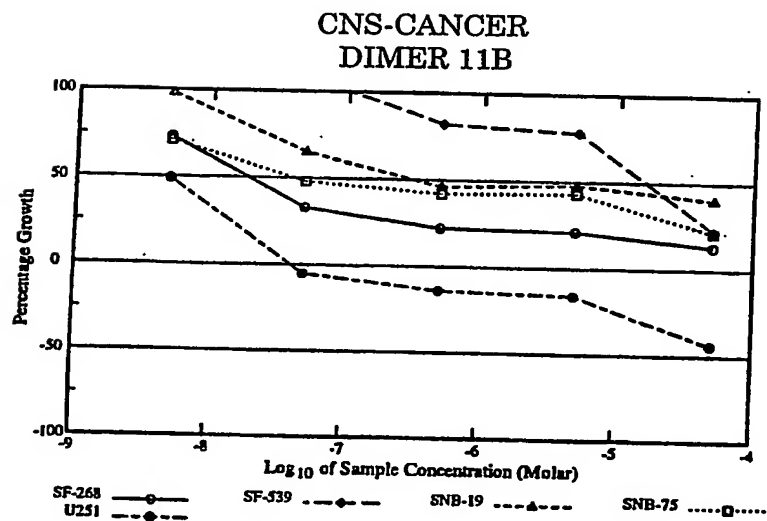
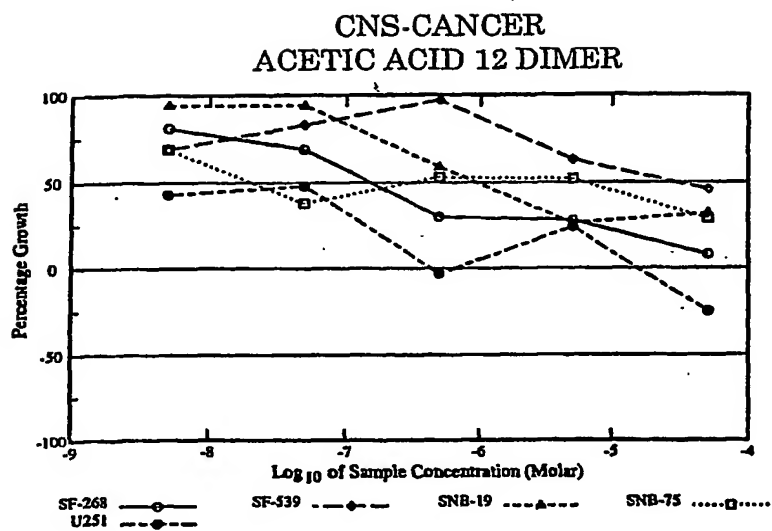
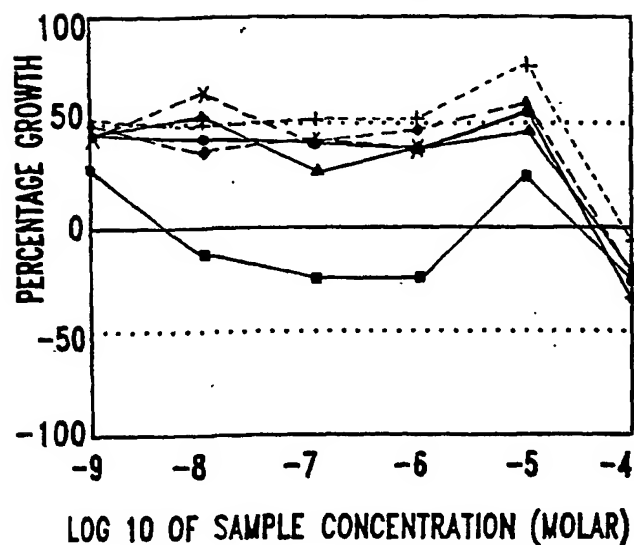


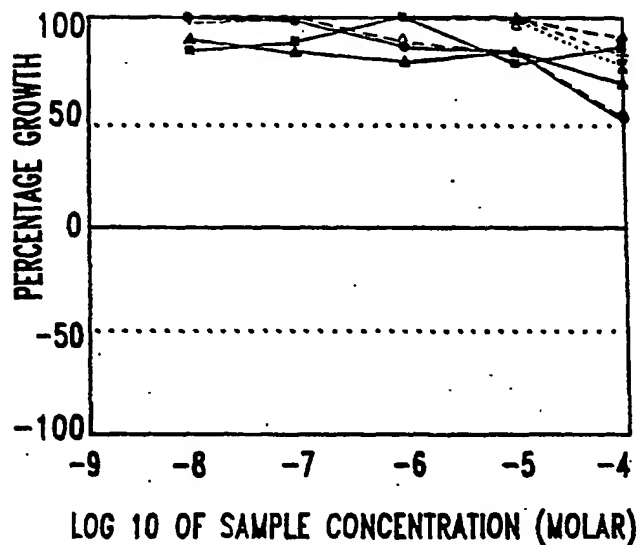
Fig. 10h

**Fig. 10i**

MELANOMA
PACLITAXEL

LOX IMVI —●— MALME-3M —+— M14 —*—
SK-MEL-2 —■— UACC-257 —◆— UACC-62 —▲—

Fig. 11a

MELANOMA
QHS

LOX IMVI —●— MALME-3M —+— SK-MEL-2 —◆—
SK-MEL-5 —○— UACC-257 —▲— UACC-62 —x—
M19-MEL —◆—

Fig. 11b

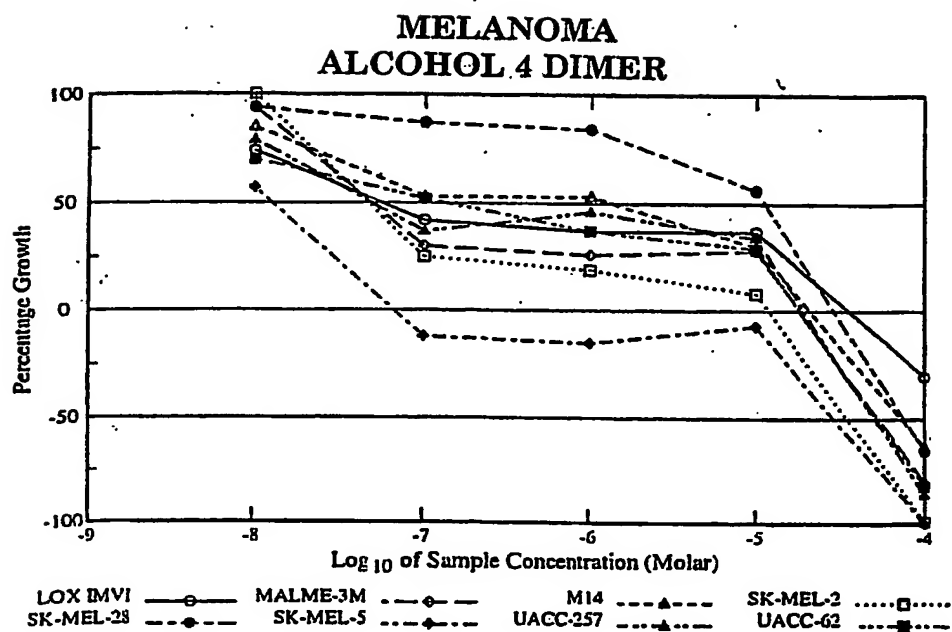


Fig. 11c

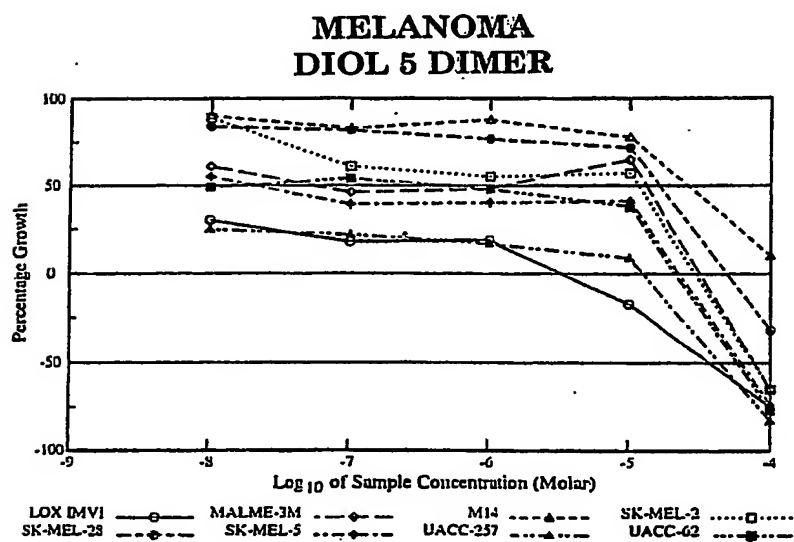


Fig. 11d

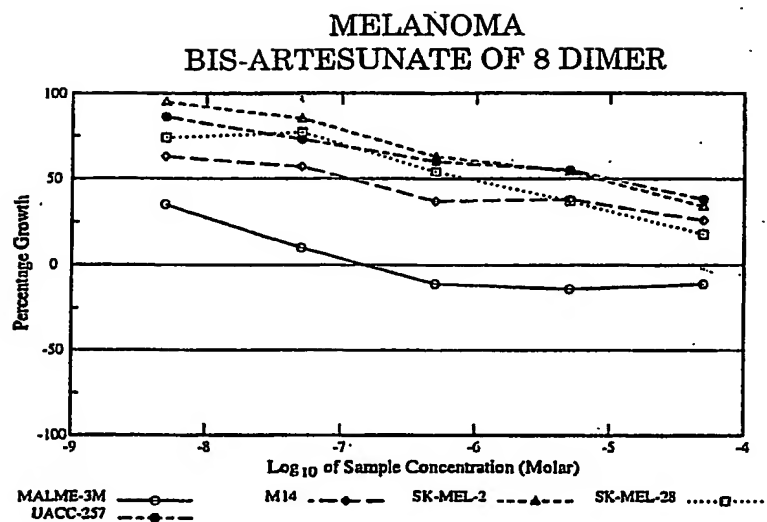


Fig. 11e

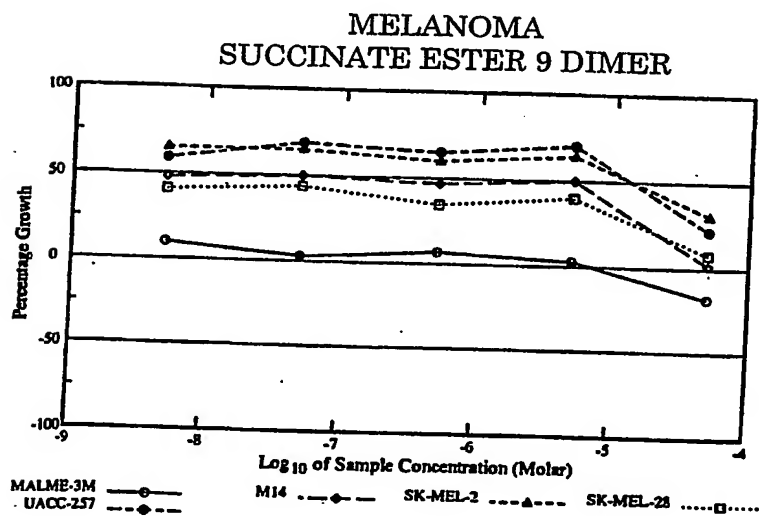


Fig. 11f

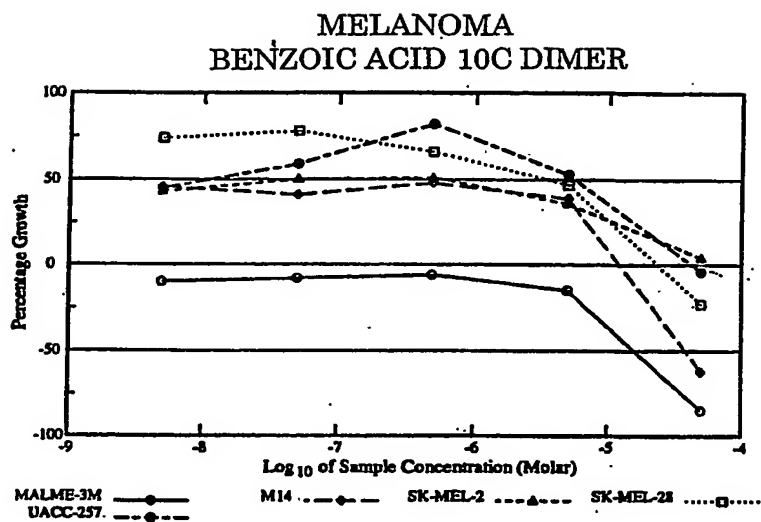


Fig. 11g

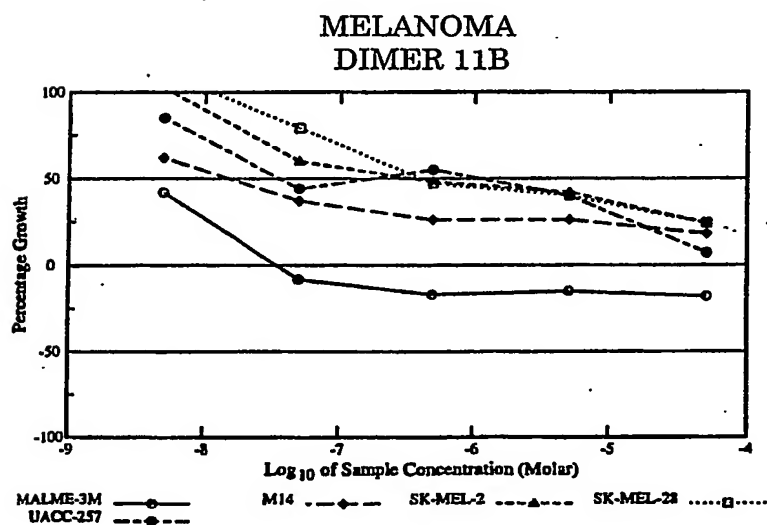


Fig. 11h

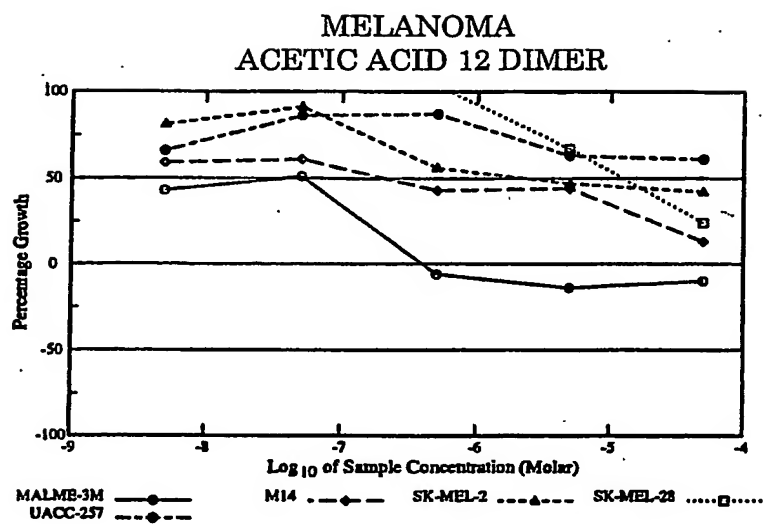


Fig. 11i

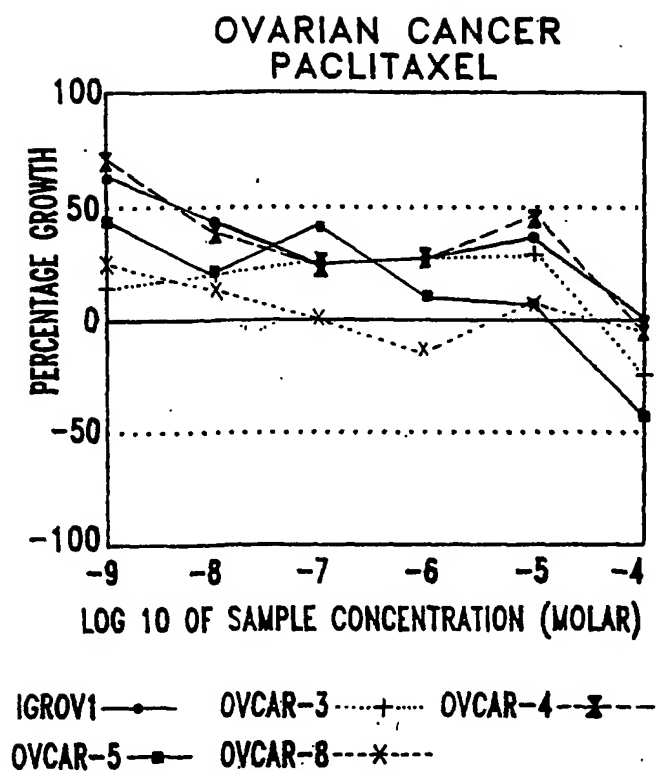


Fig. 12a

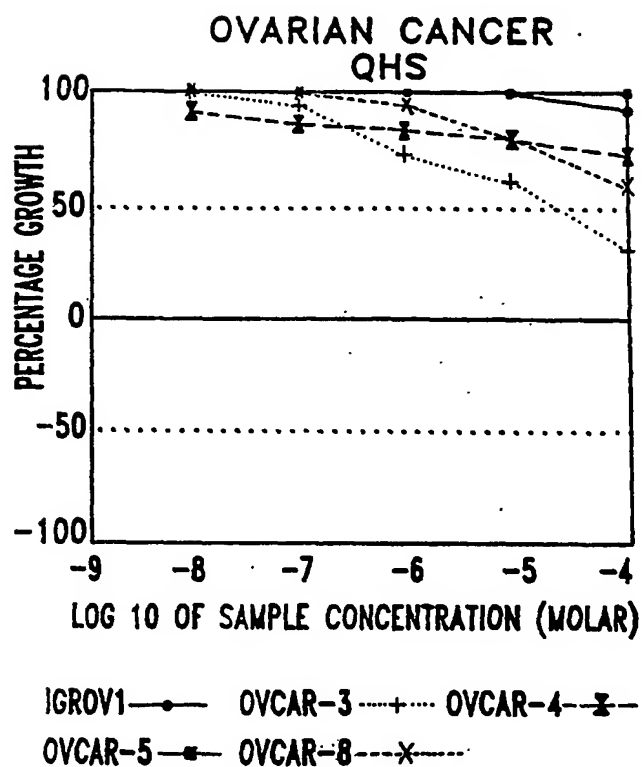


Fig. 12b

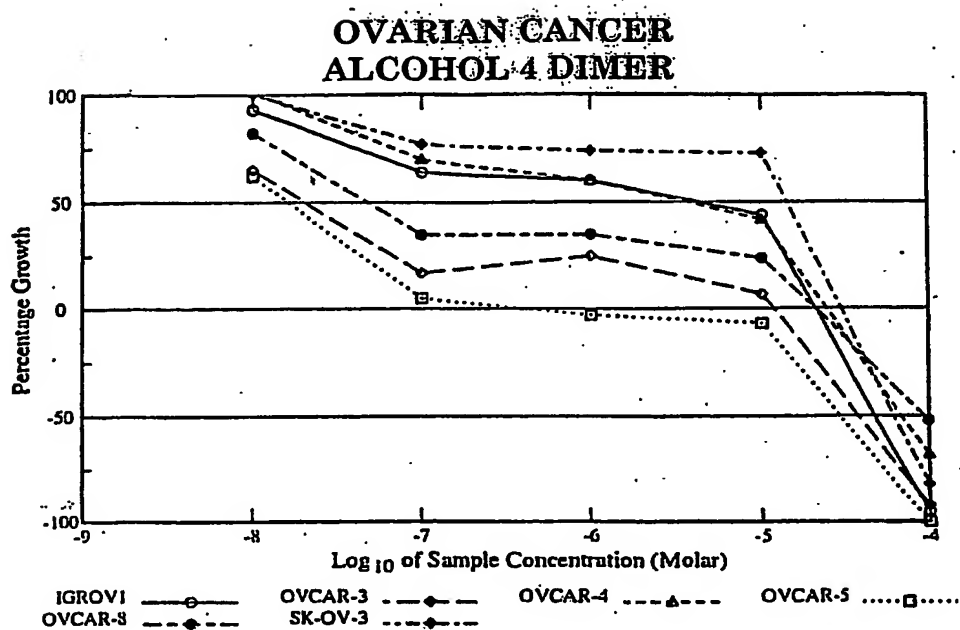


Fig. 12c

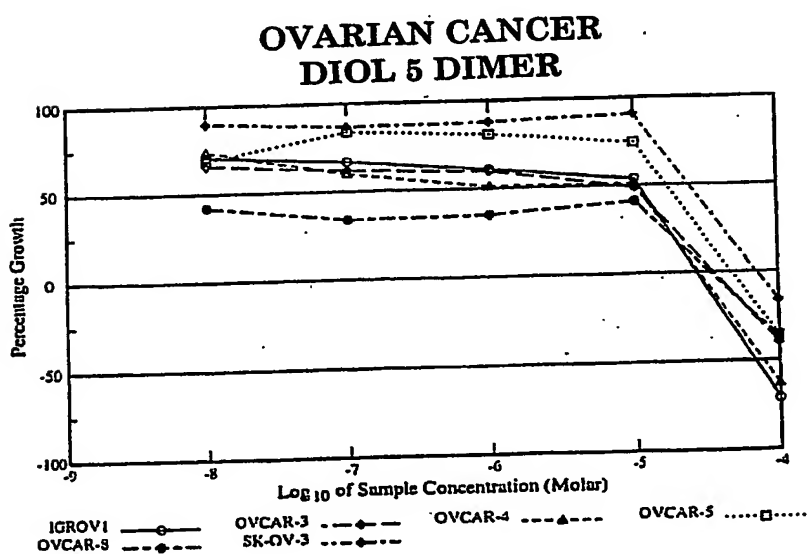


Fig. 12d

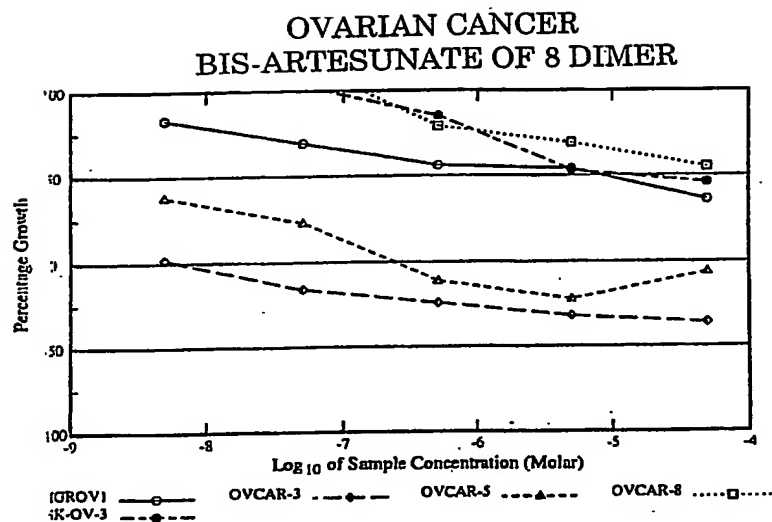


Fig. 12e

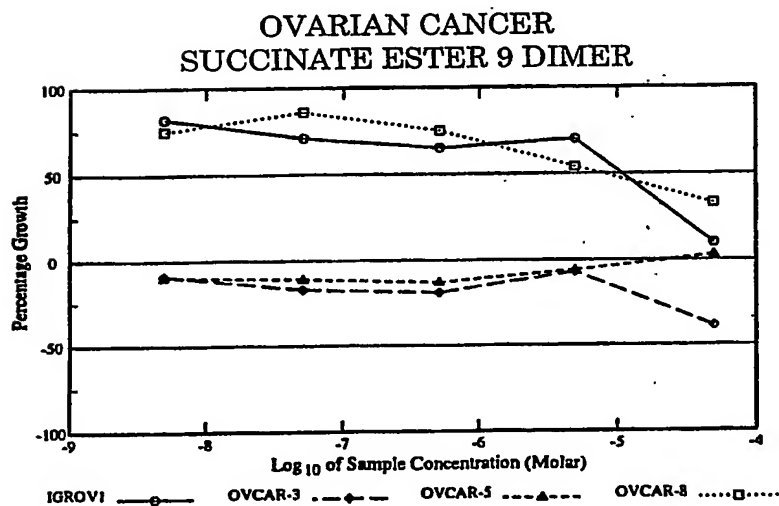


Fig. 12f

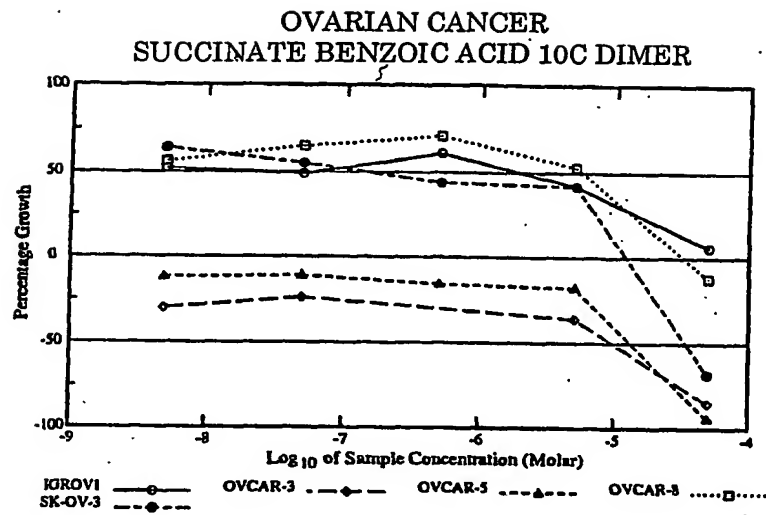


Fig. 12g

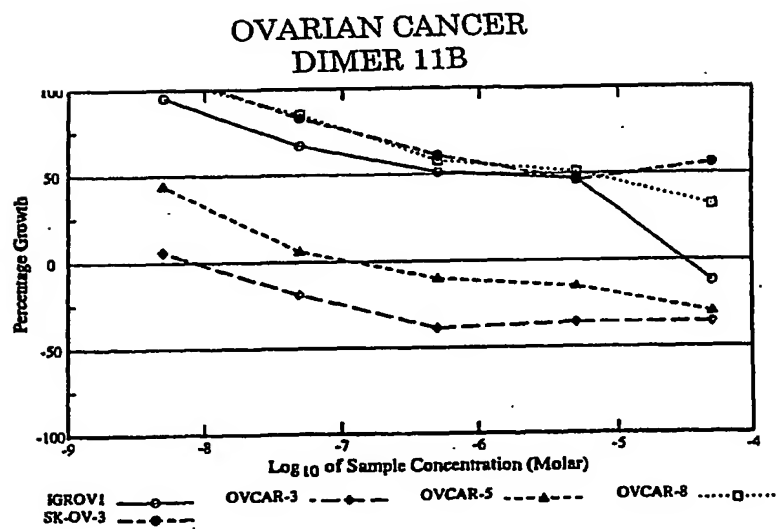


Fig. 12h

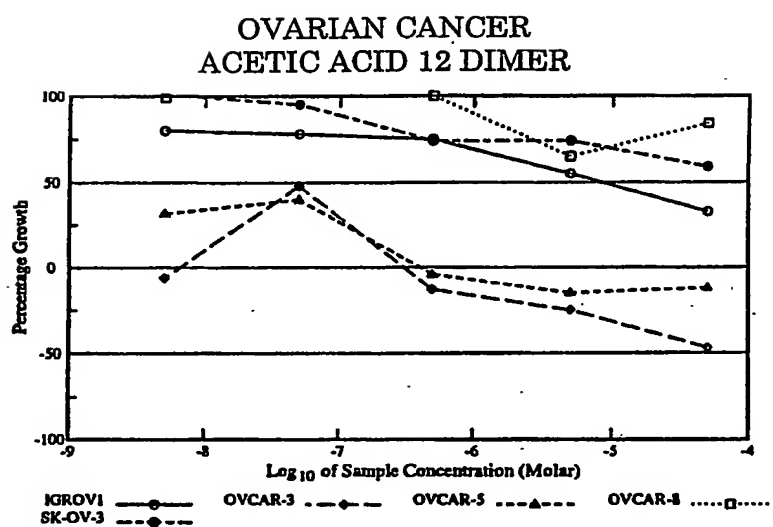
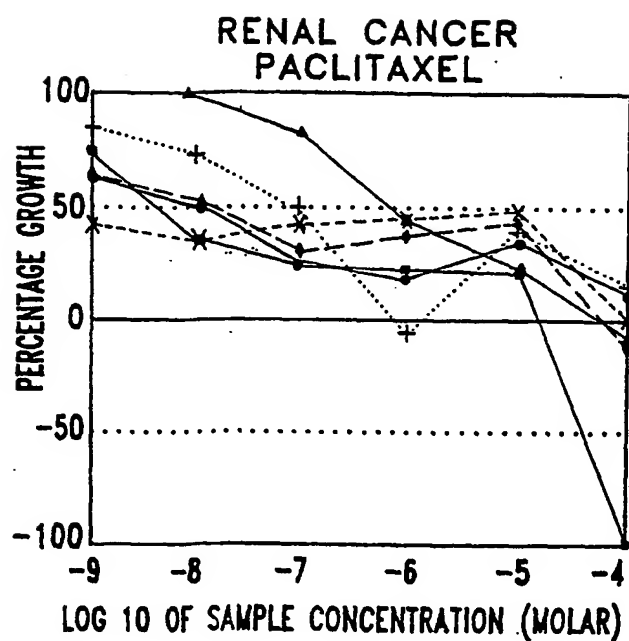
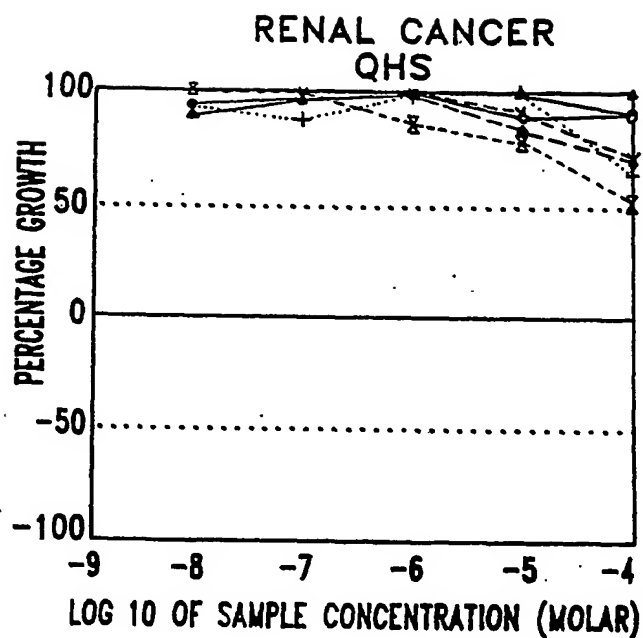


Fig. 12i



786-0 —●— A498 —+— RXF-393 —■—
SN12C —*— TK-10 —◆— UO-31 —▲—

Fig. 13a



786-0 —●— A498 —+— ACHN —x—
SN12C —◆— TK-10 —▲— UO-31 —&—
RXF-631 —○—

Fig. 13b

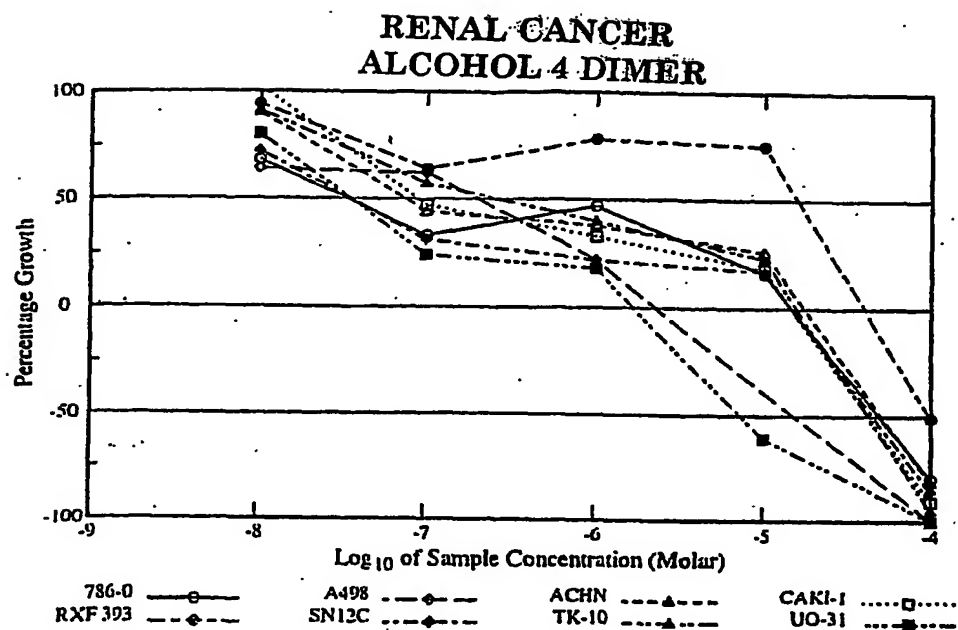


Fig. 13c

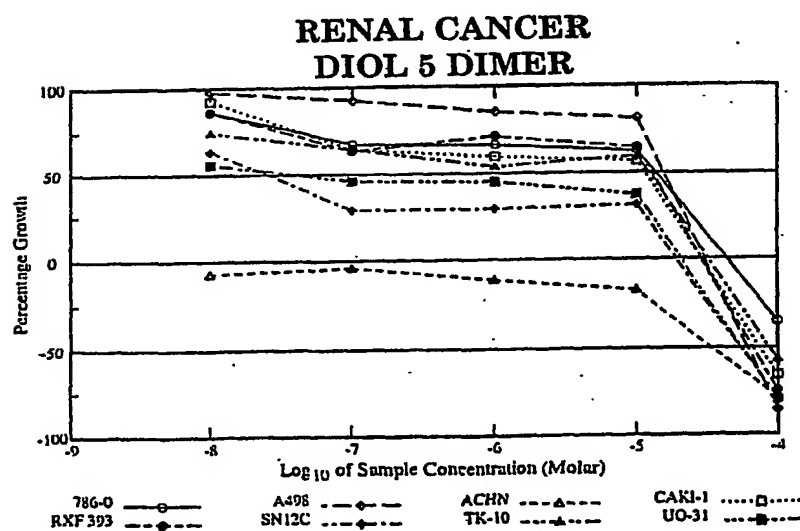


Fig. 13d

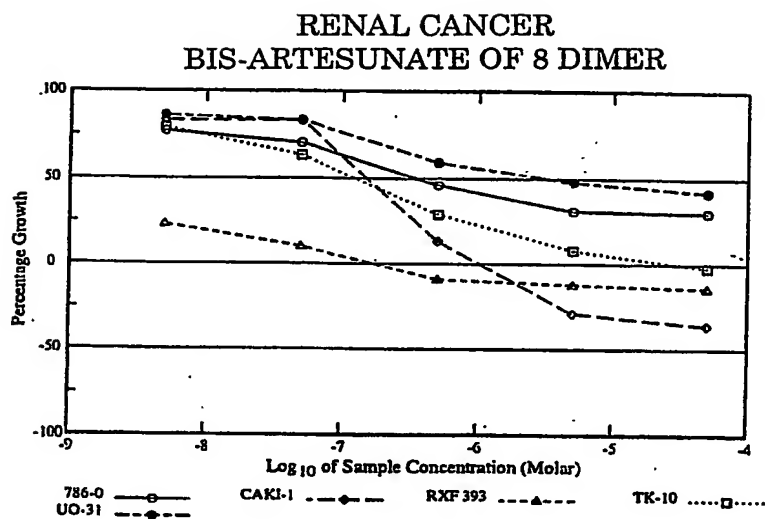


Fig. 13e

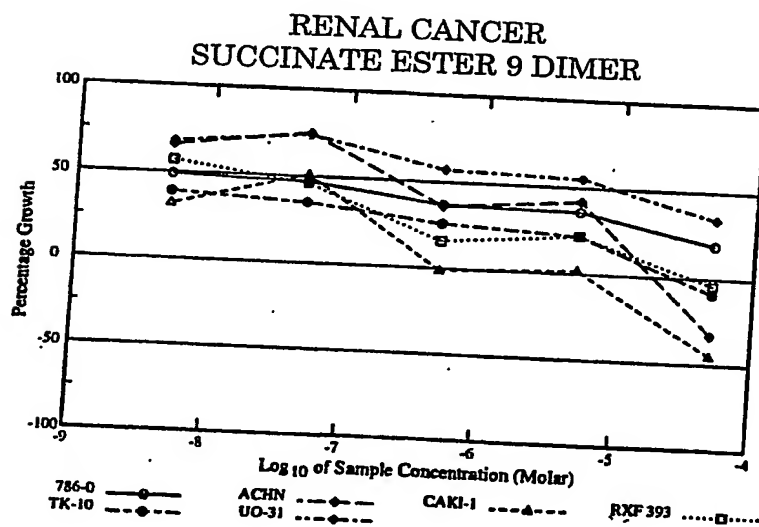


Fig. 13f

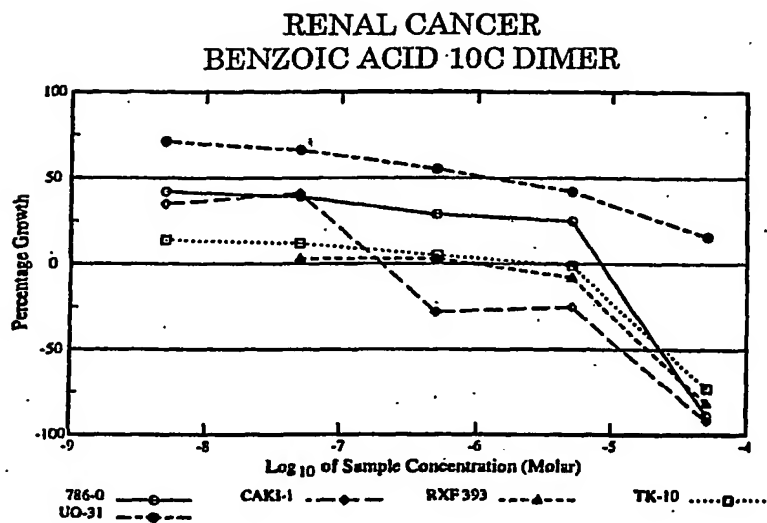


Fig. 13g

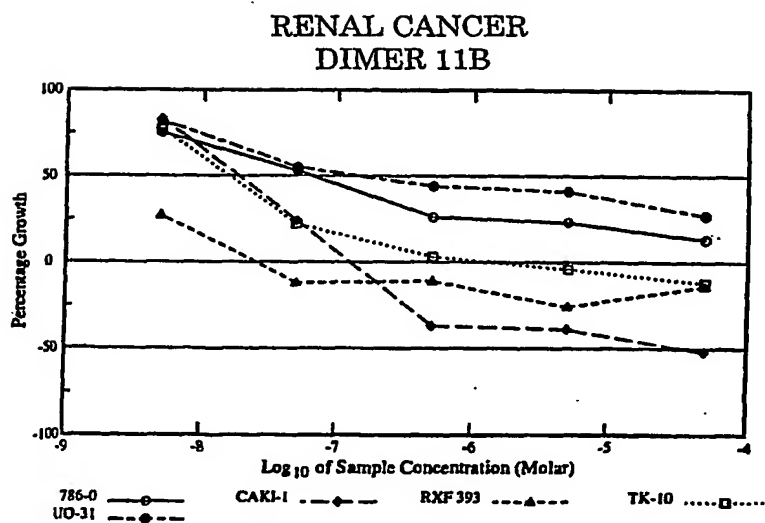


Fig. 13h

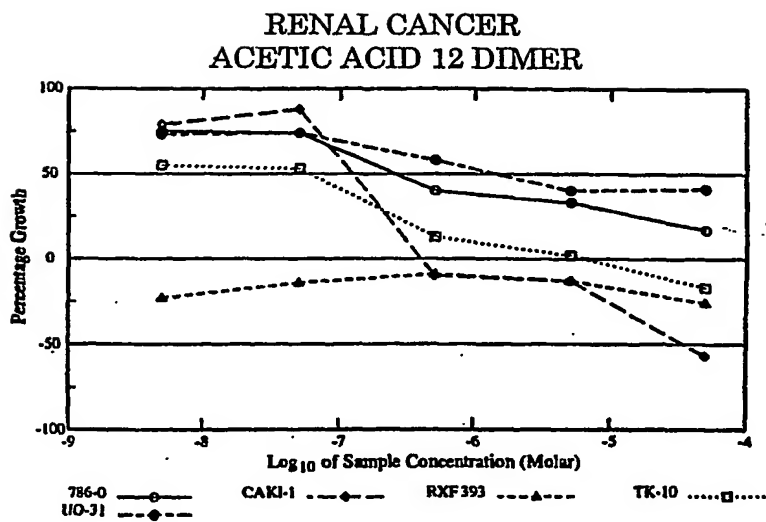
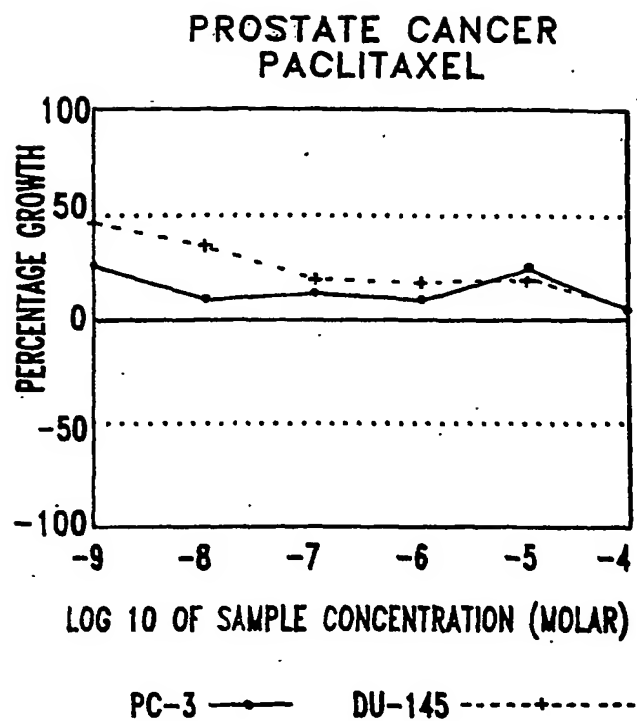


Fig. 13i

**Fig. 14a**

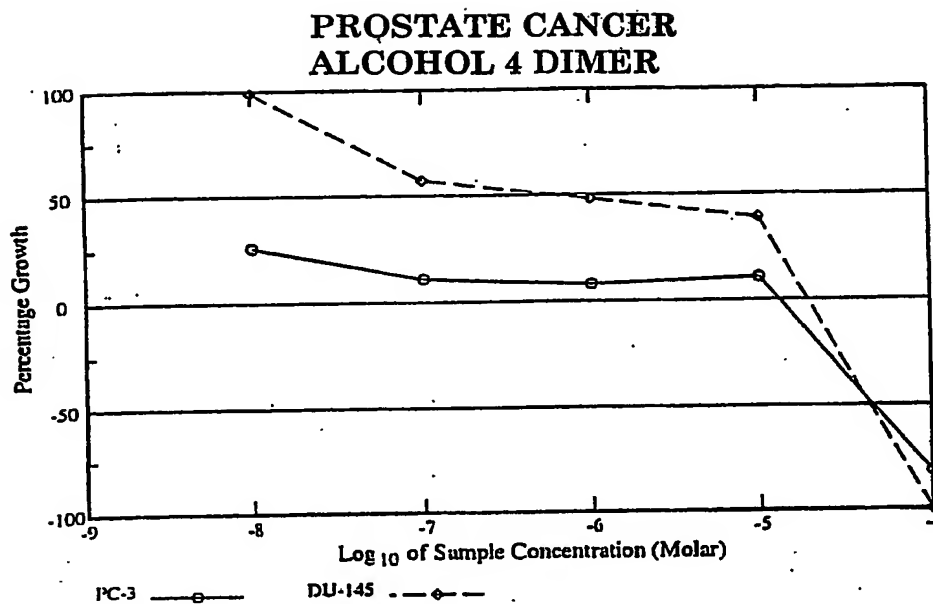


Fig. 14b

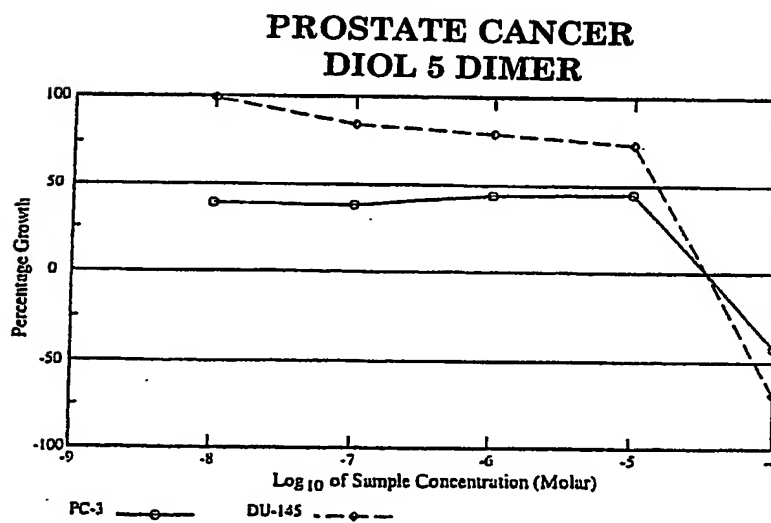
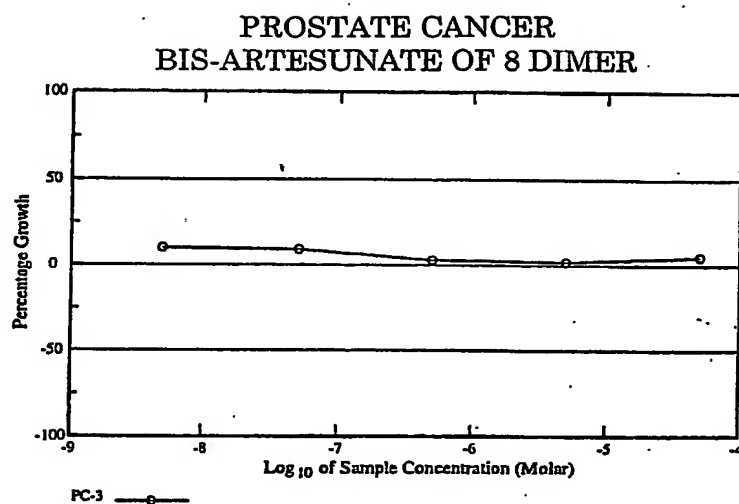
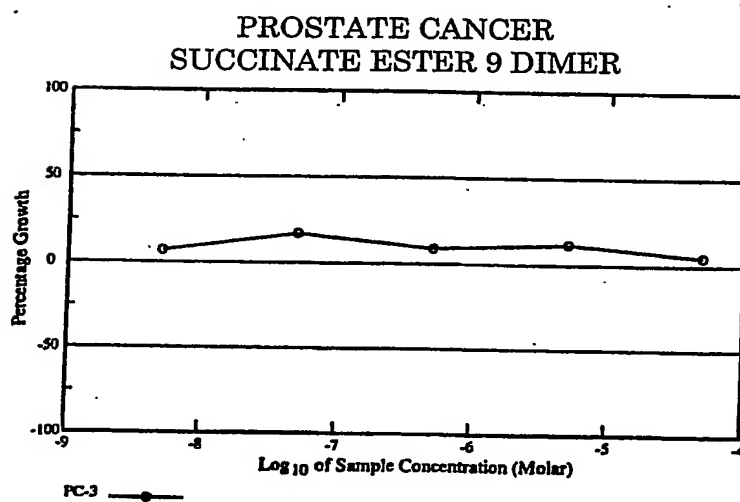


Fig. 14c

**Fig. 14d****Fig. 14e**

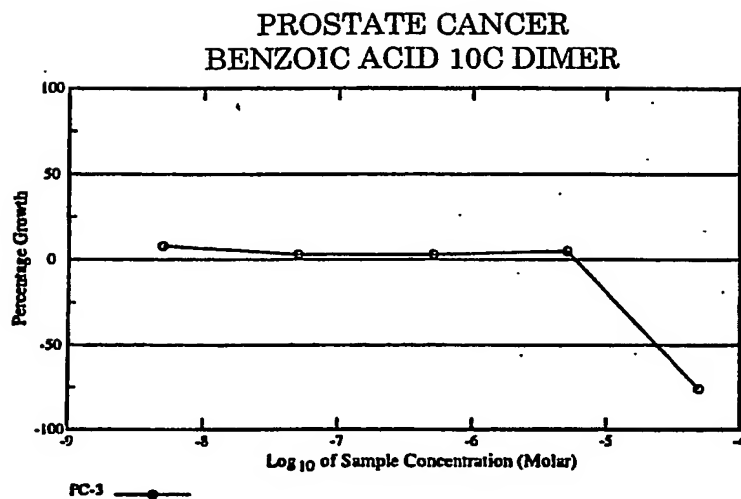


Fig. 14f

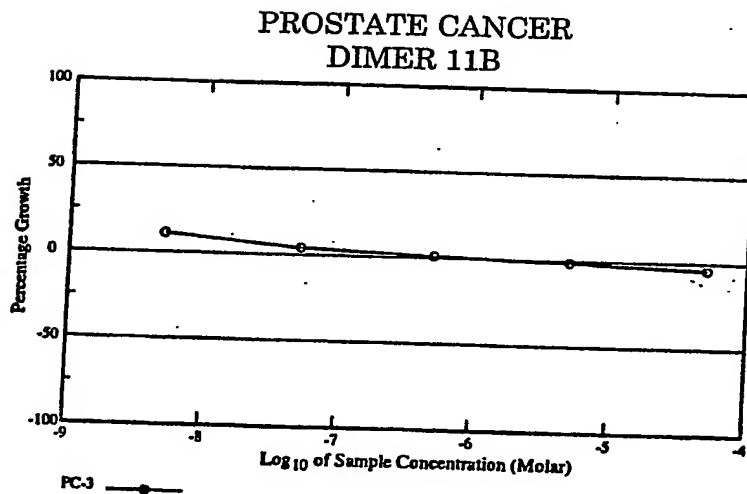


Fig. 14g

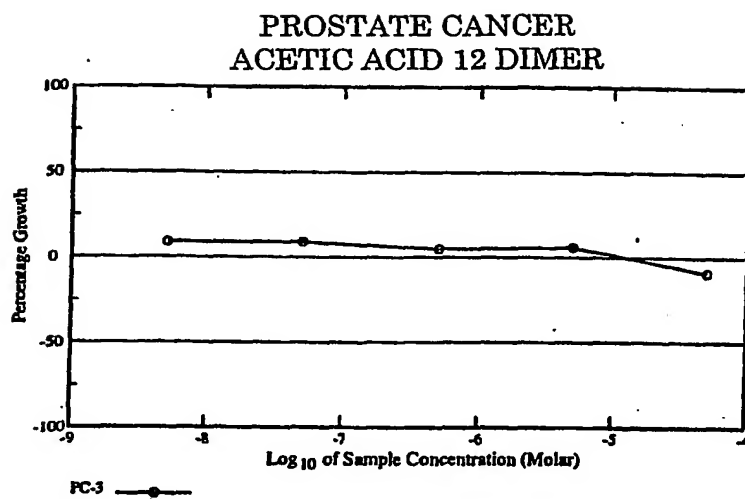


Fig. 14h

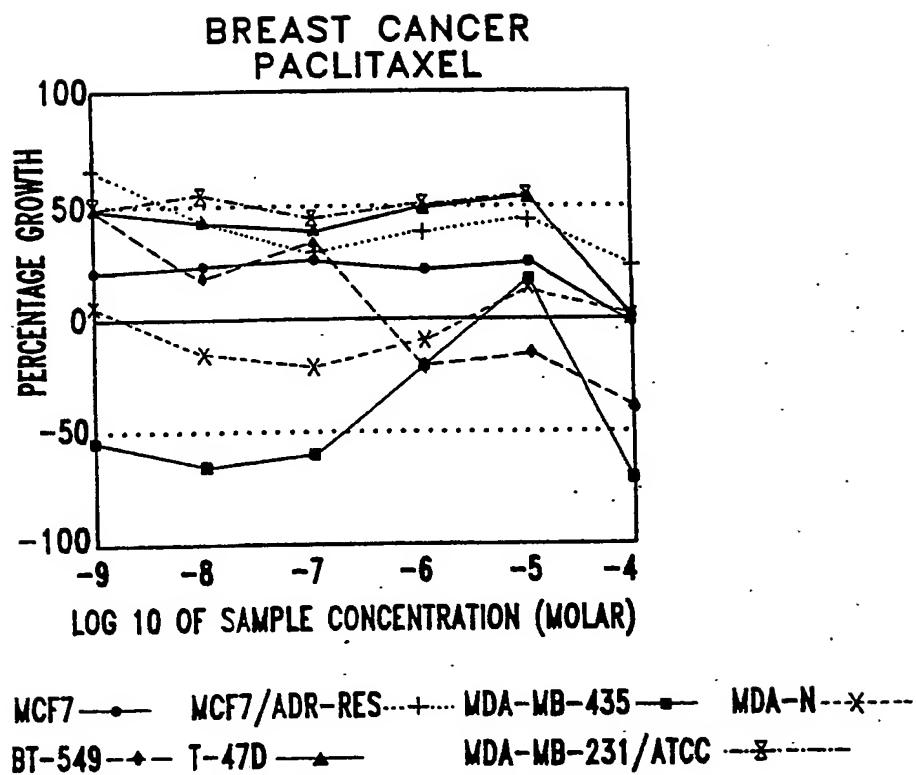


Fig. 15a

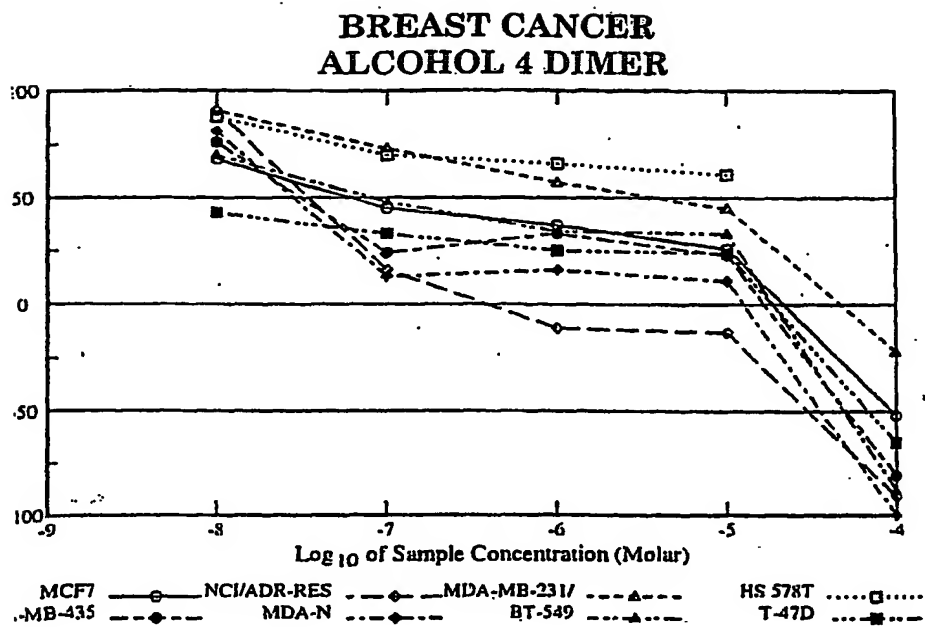


Fig. 15b

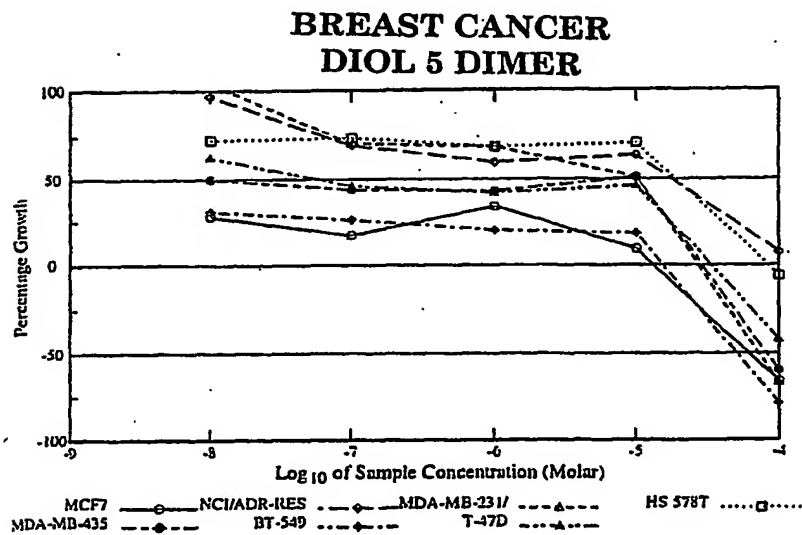


Fig. 15c

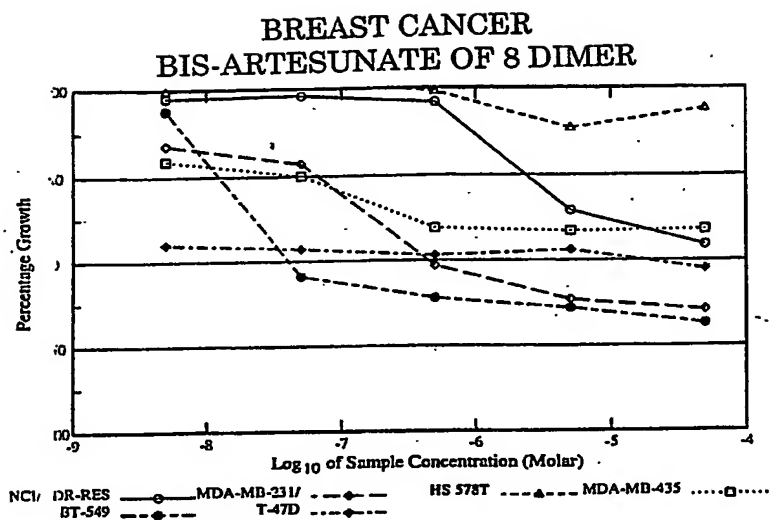


Fig. 15d

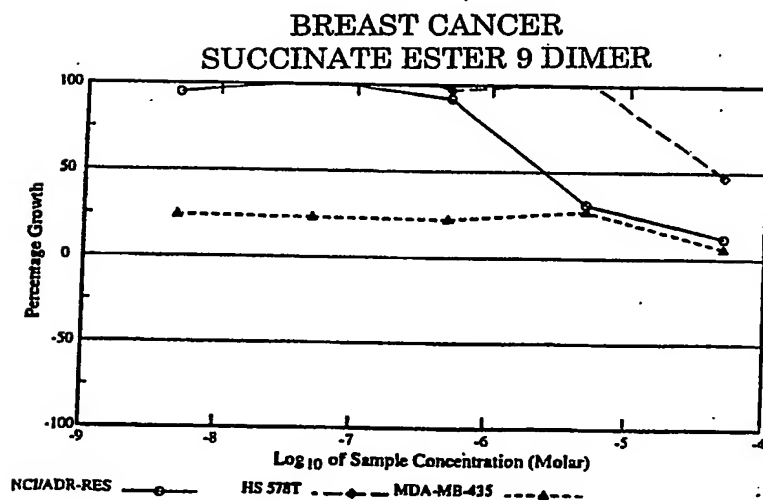


Fig. 15e

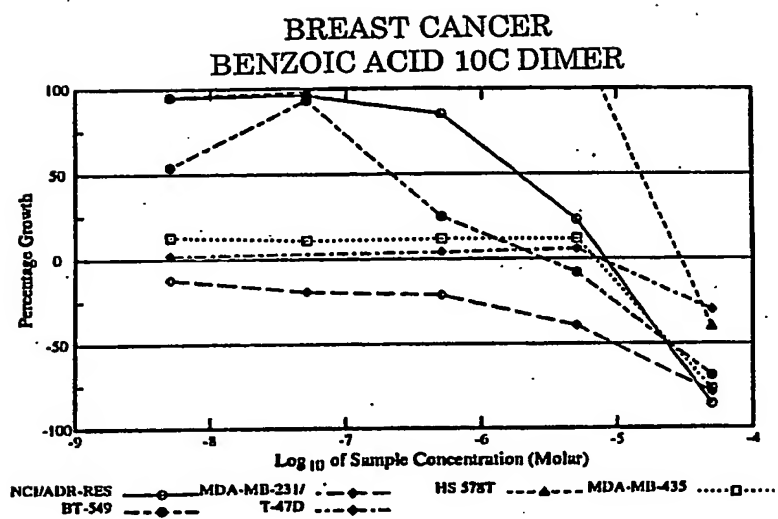


Fig. 15f

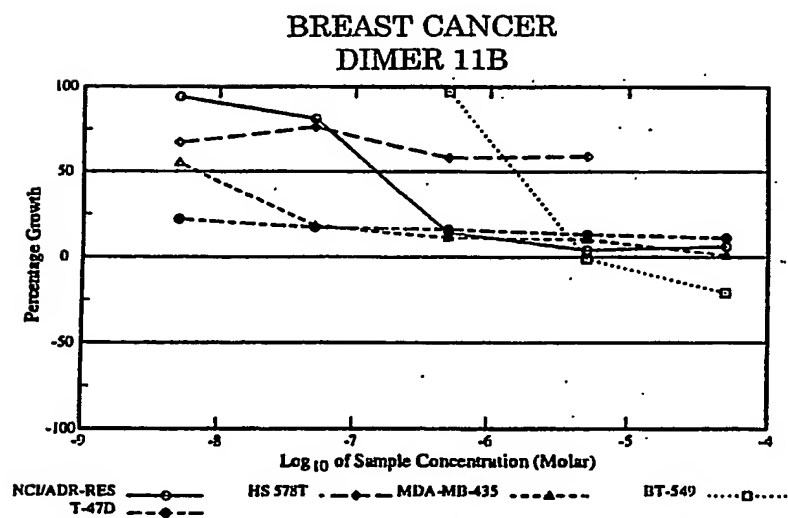
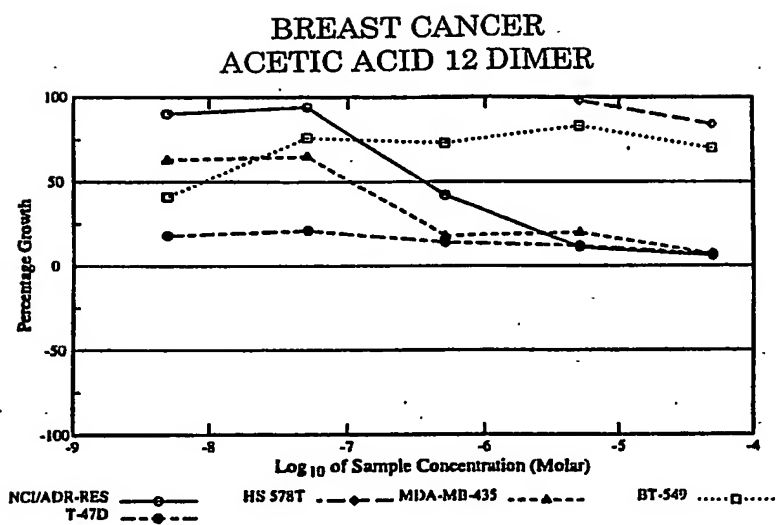


Fig. 15g

**Fig. 15h**

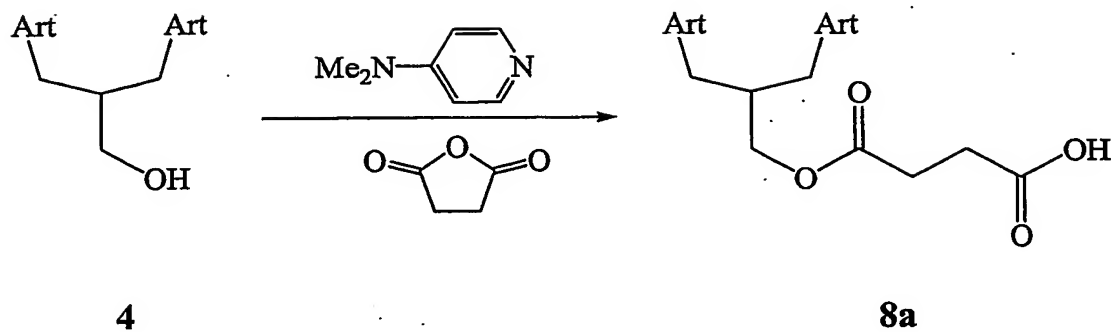


FIGURE 16

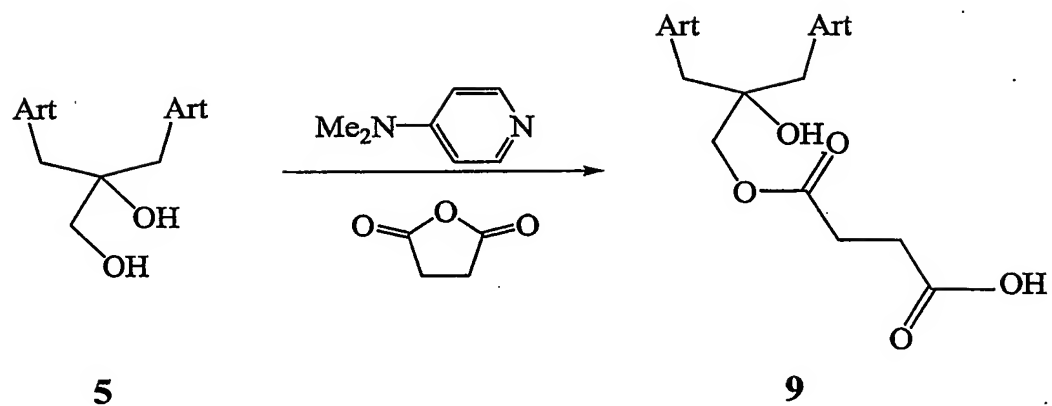


FIGURE 17

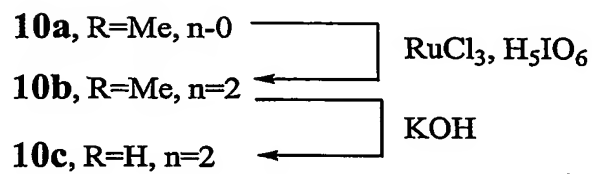
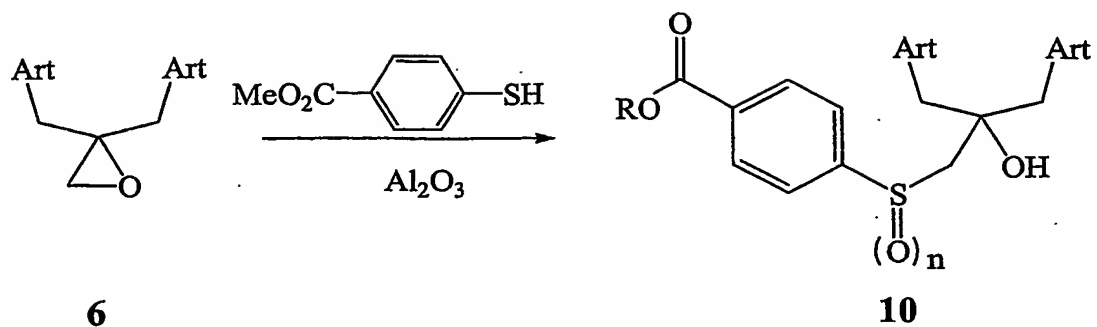


FIGURE 18

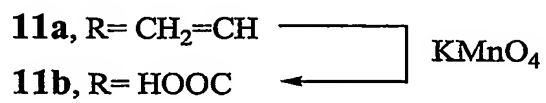
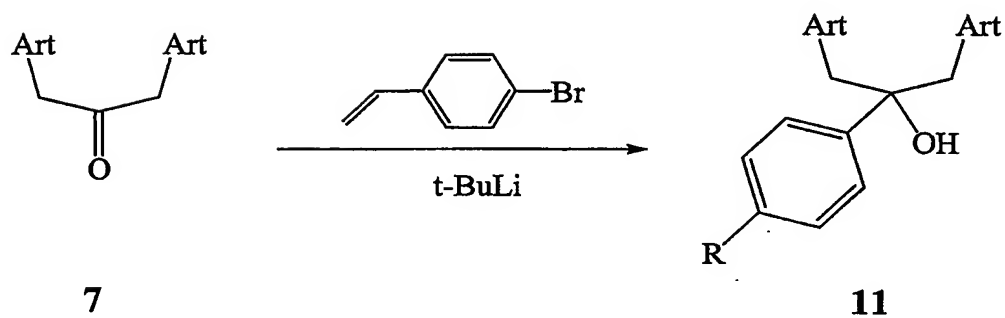


FIGURE 19

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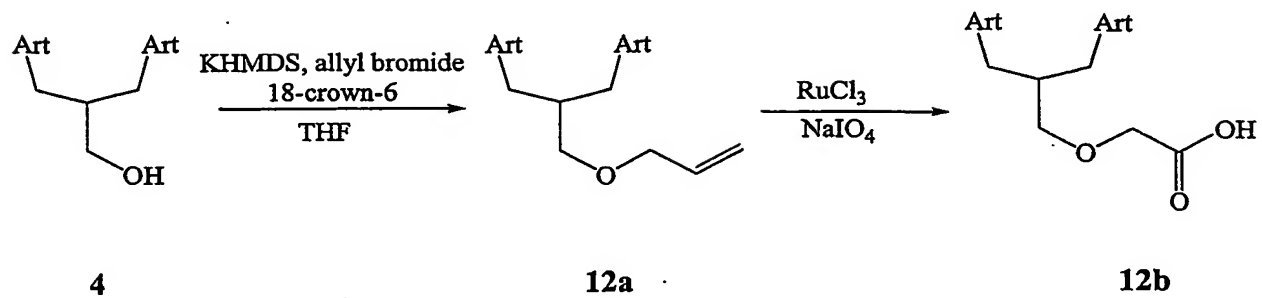


FIGURE 20

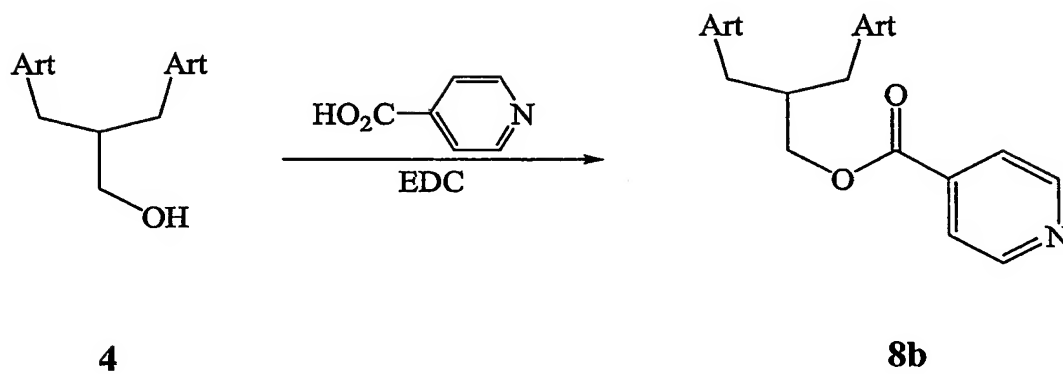


FIGURE 21

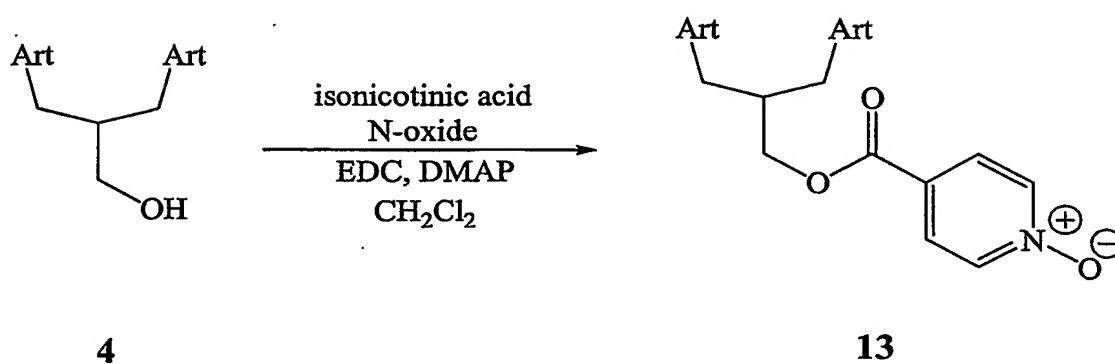


FIGURE 22

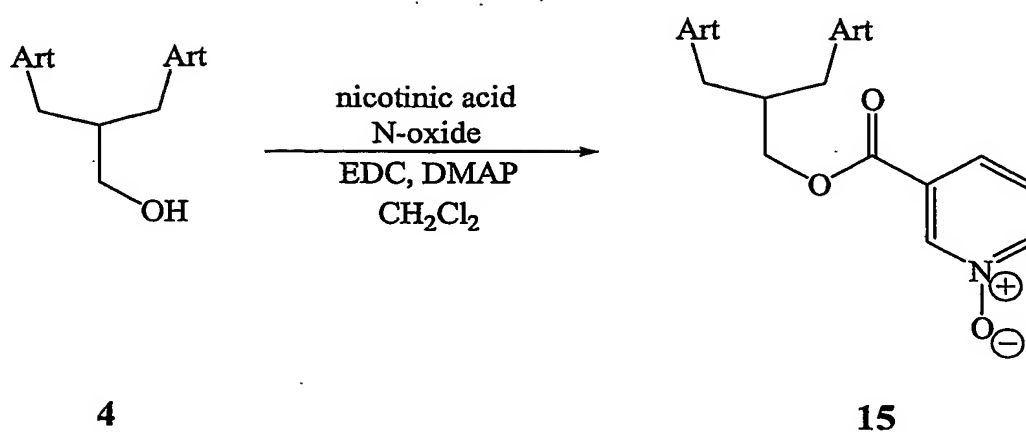


FIGURE 23

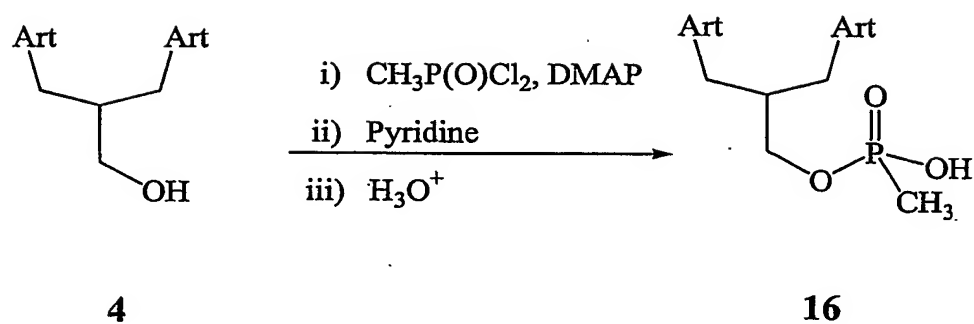


FIGURE 24

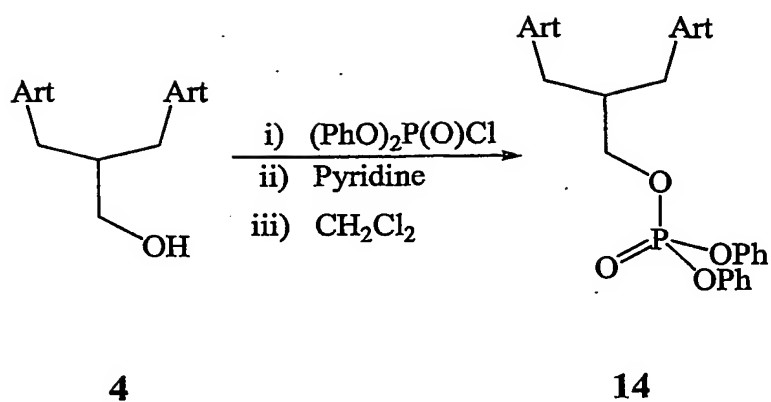


FIGURE 25

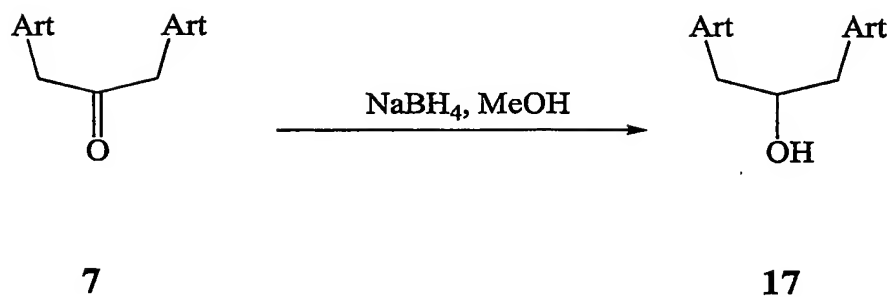


FIGURE 26

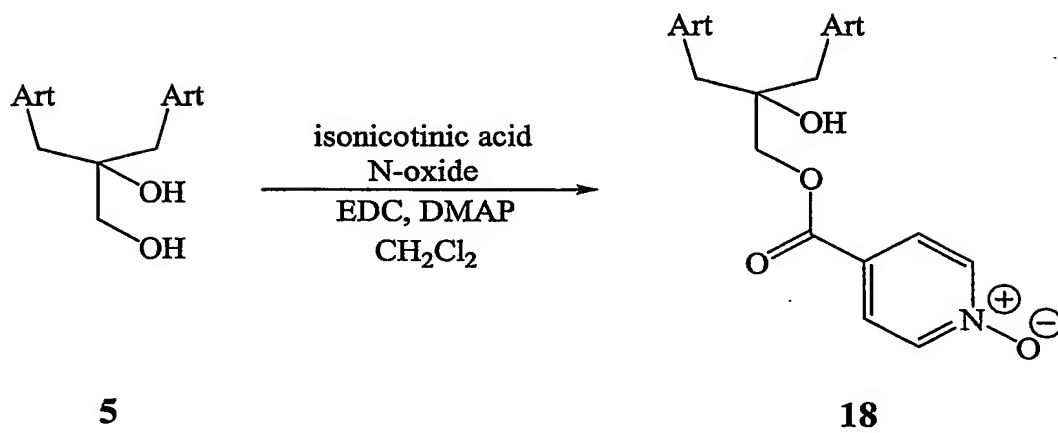


FIGURE 27

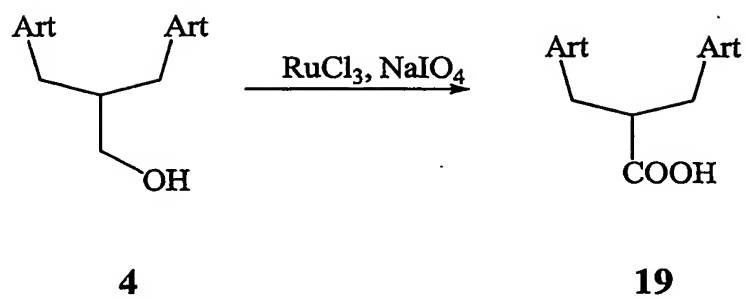


FIGURE 28

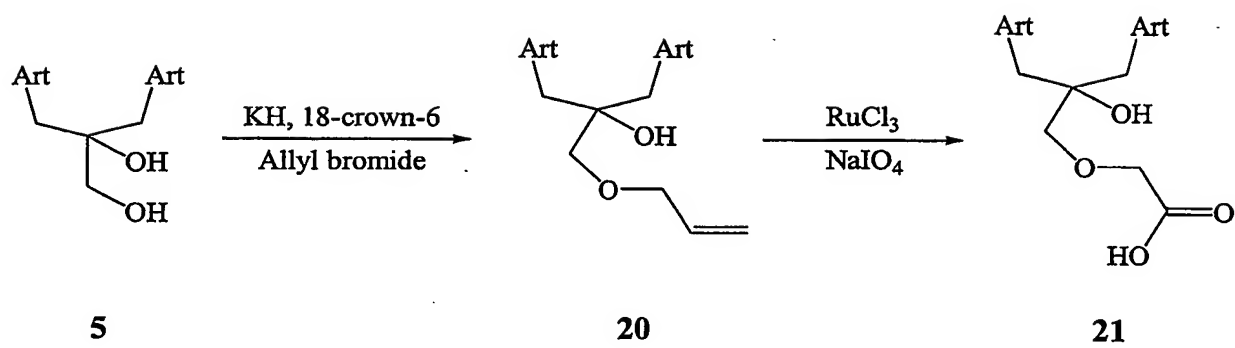


FIGURE 29

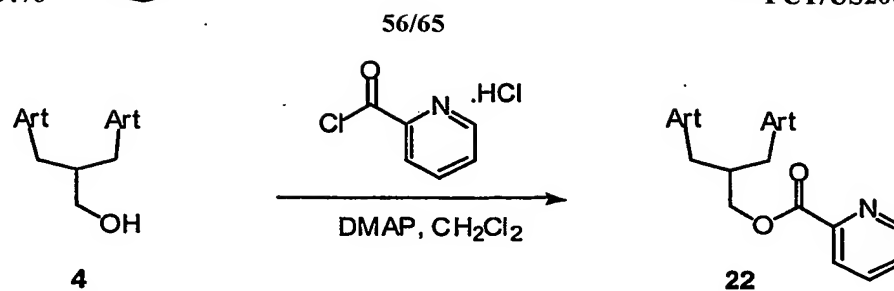


FIGURE 30

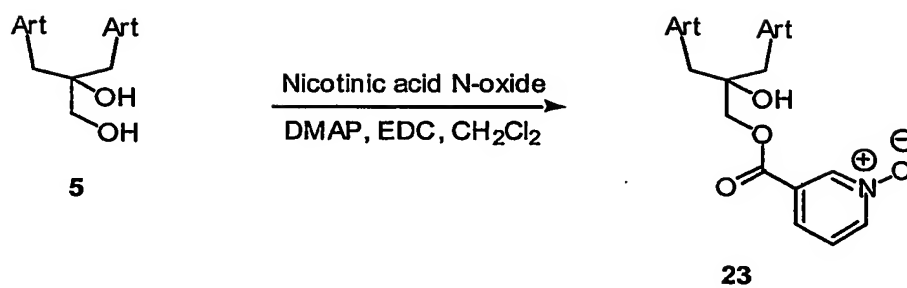


FIGURE 31

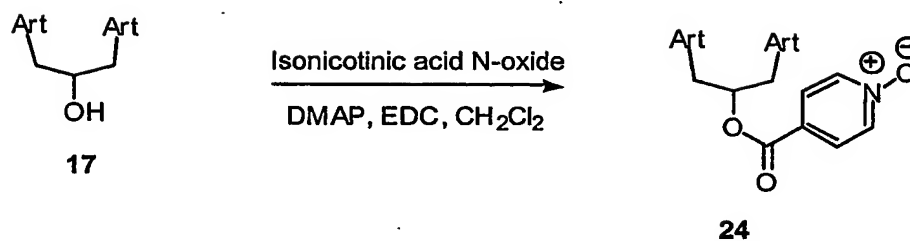


FIGURE 32

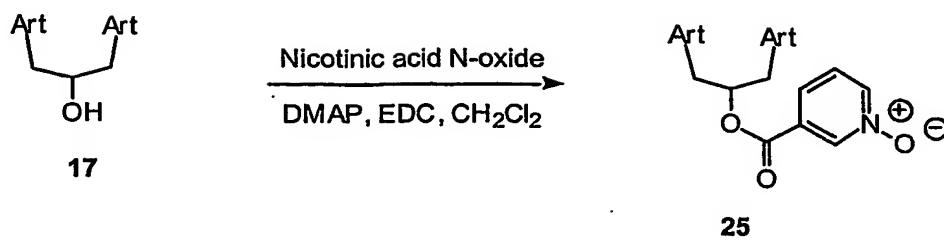


FIGURE 33

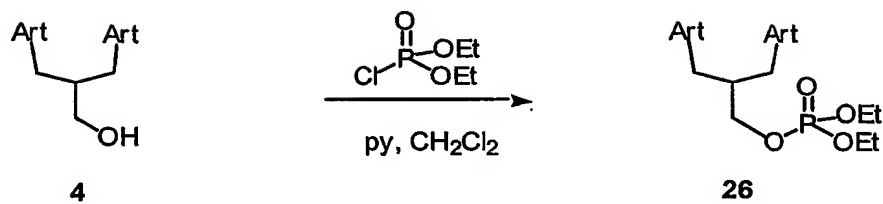


FIGURE 34

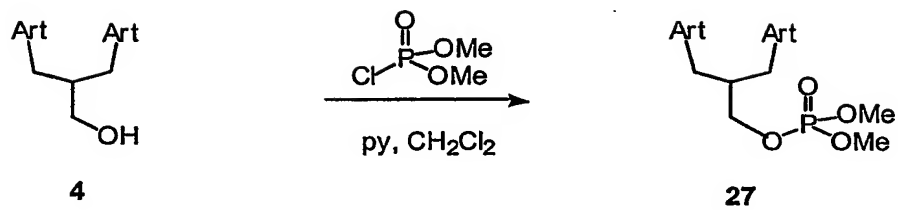


FIGURE 35

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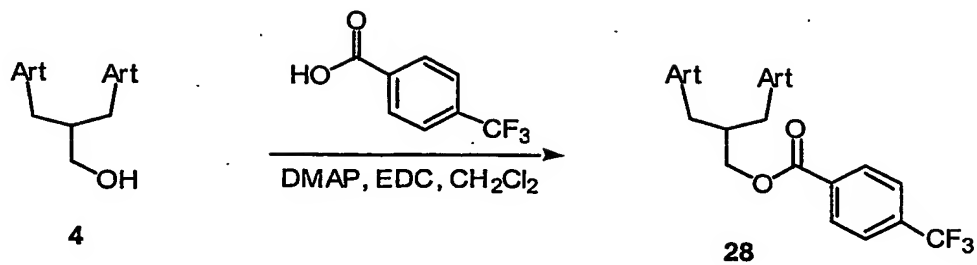


FIGURE 36

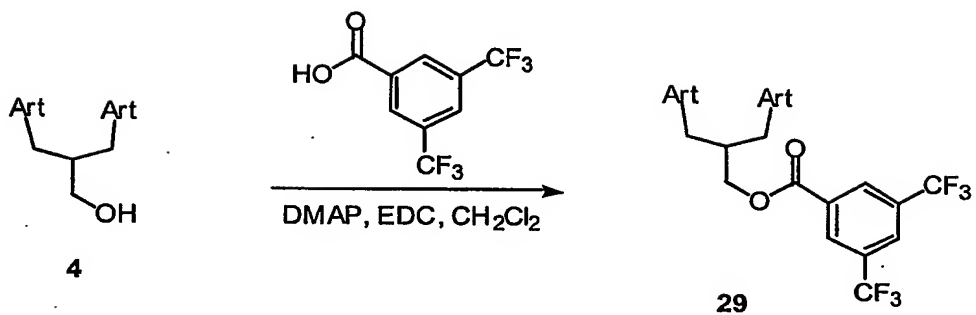


FIGURE 37

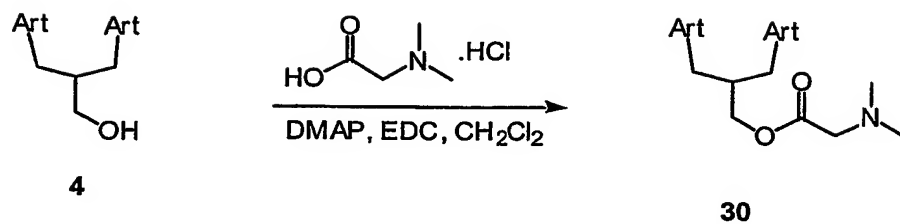


FIGURE 38

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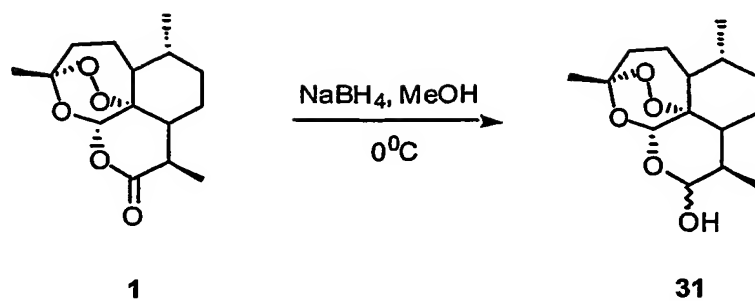


FIGURE 39

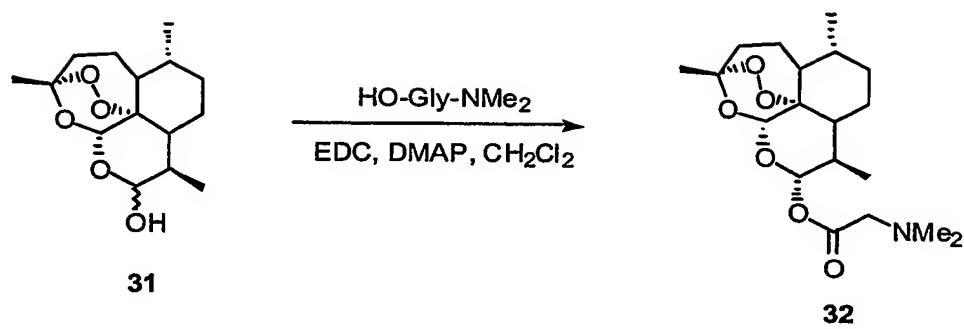


FIGURE 40

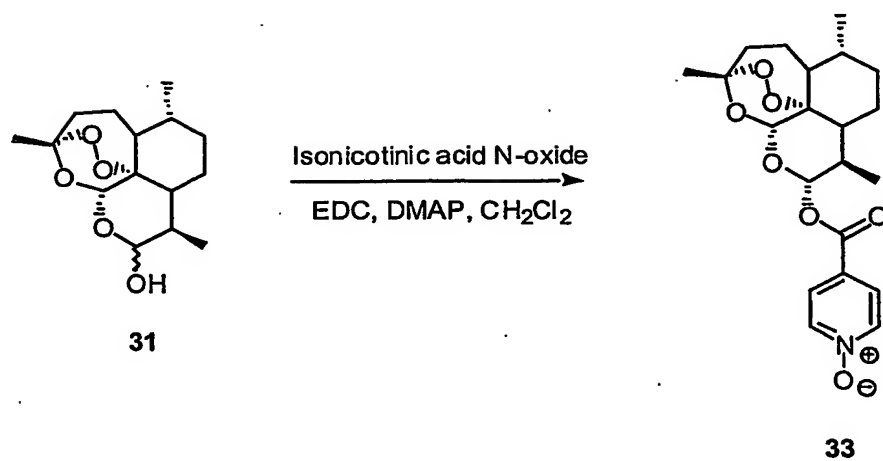


FIGURE 41

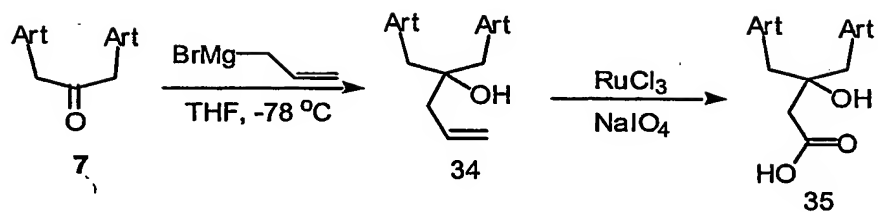


FIGURE 42

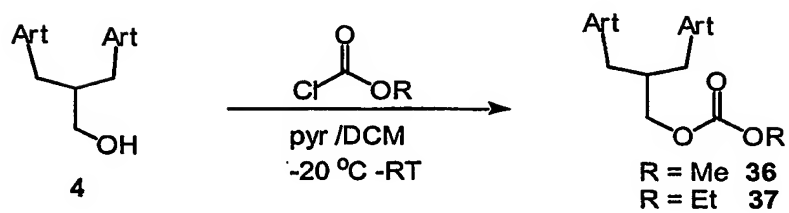


FIGURE 43

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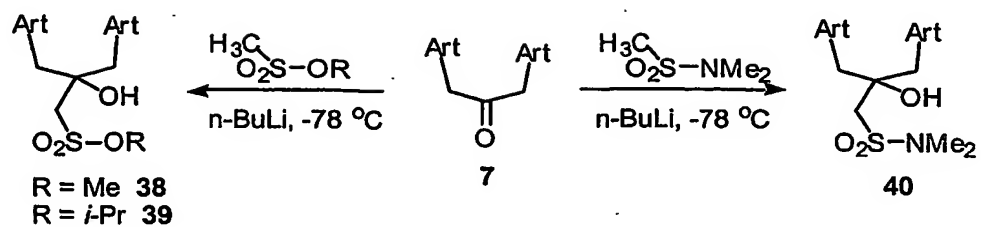


FIGURE 44

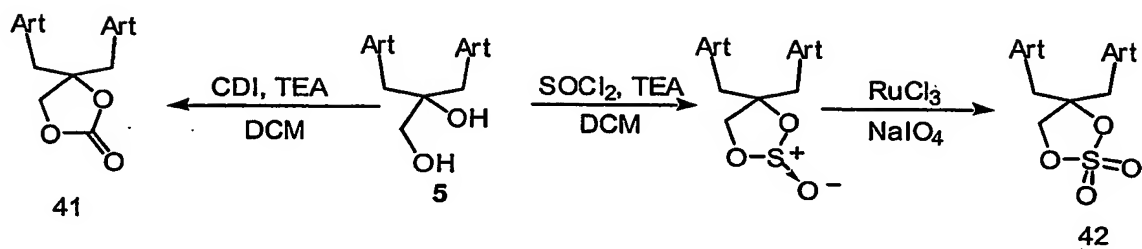


FIGURE 45

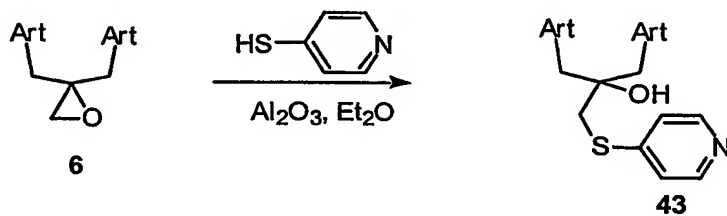


FIGURE 46

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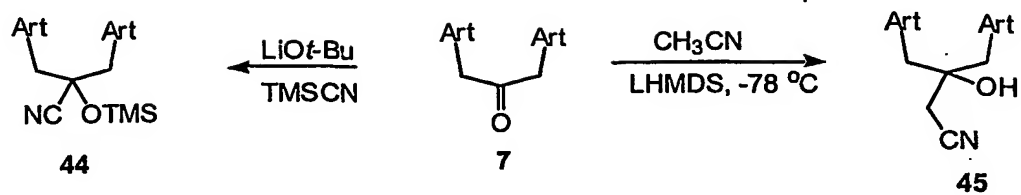


FIGURE 47

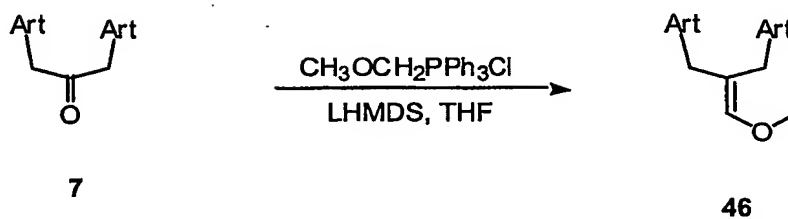


FIGURE 48

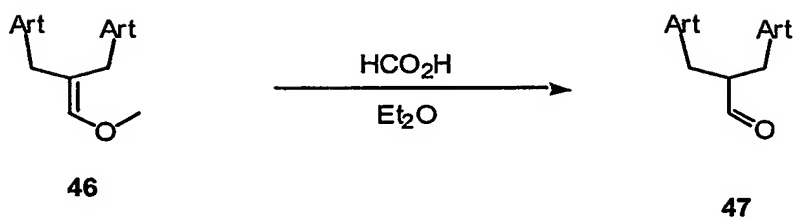


FIGURE 49

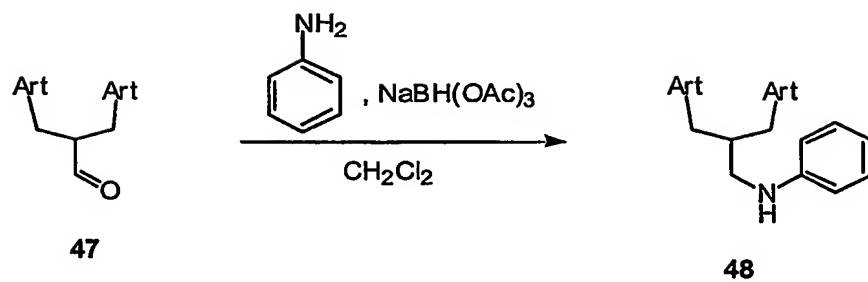


FIGURE 50

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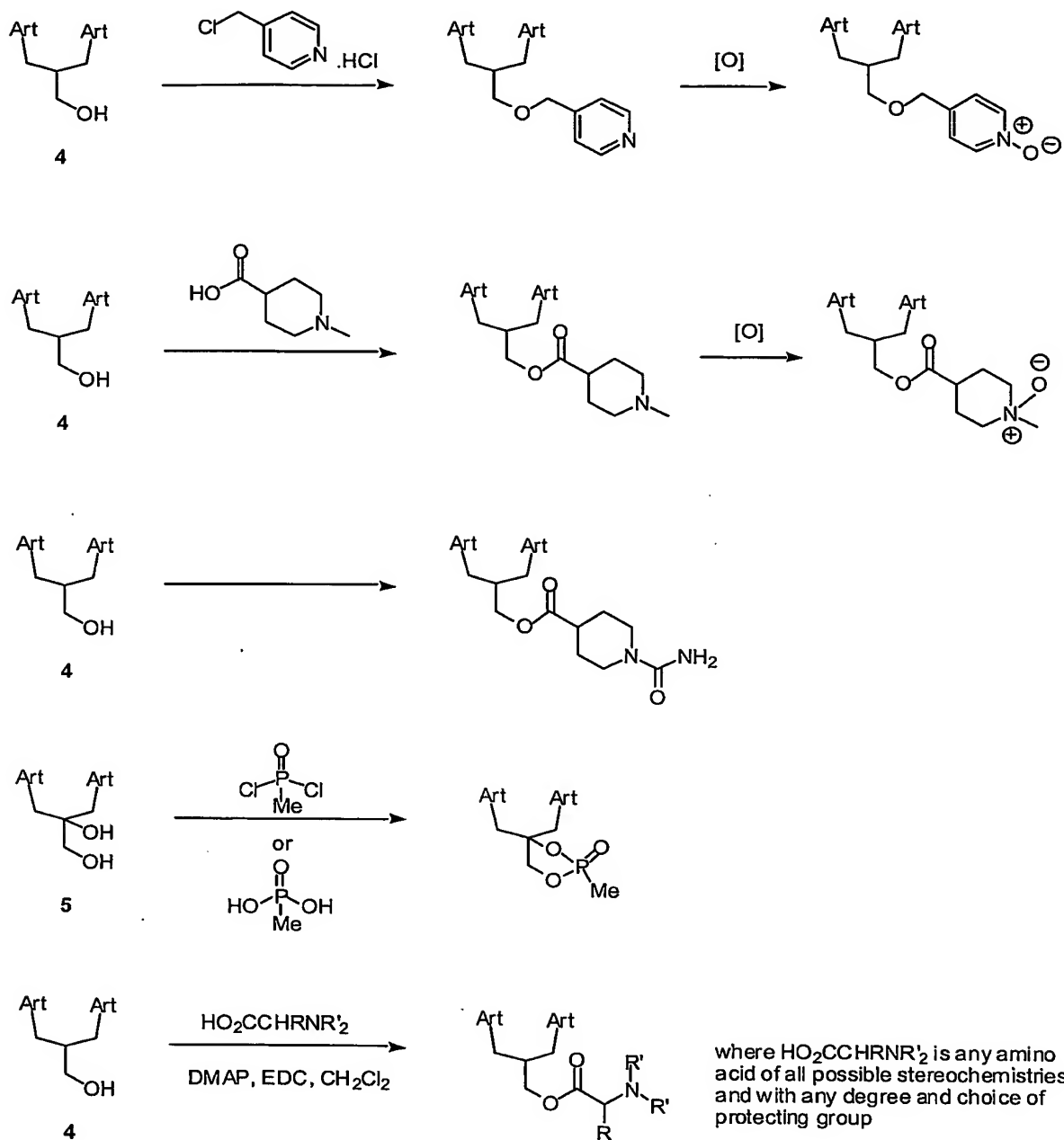


FIGURE 51

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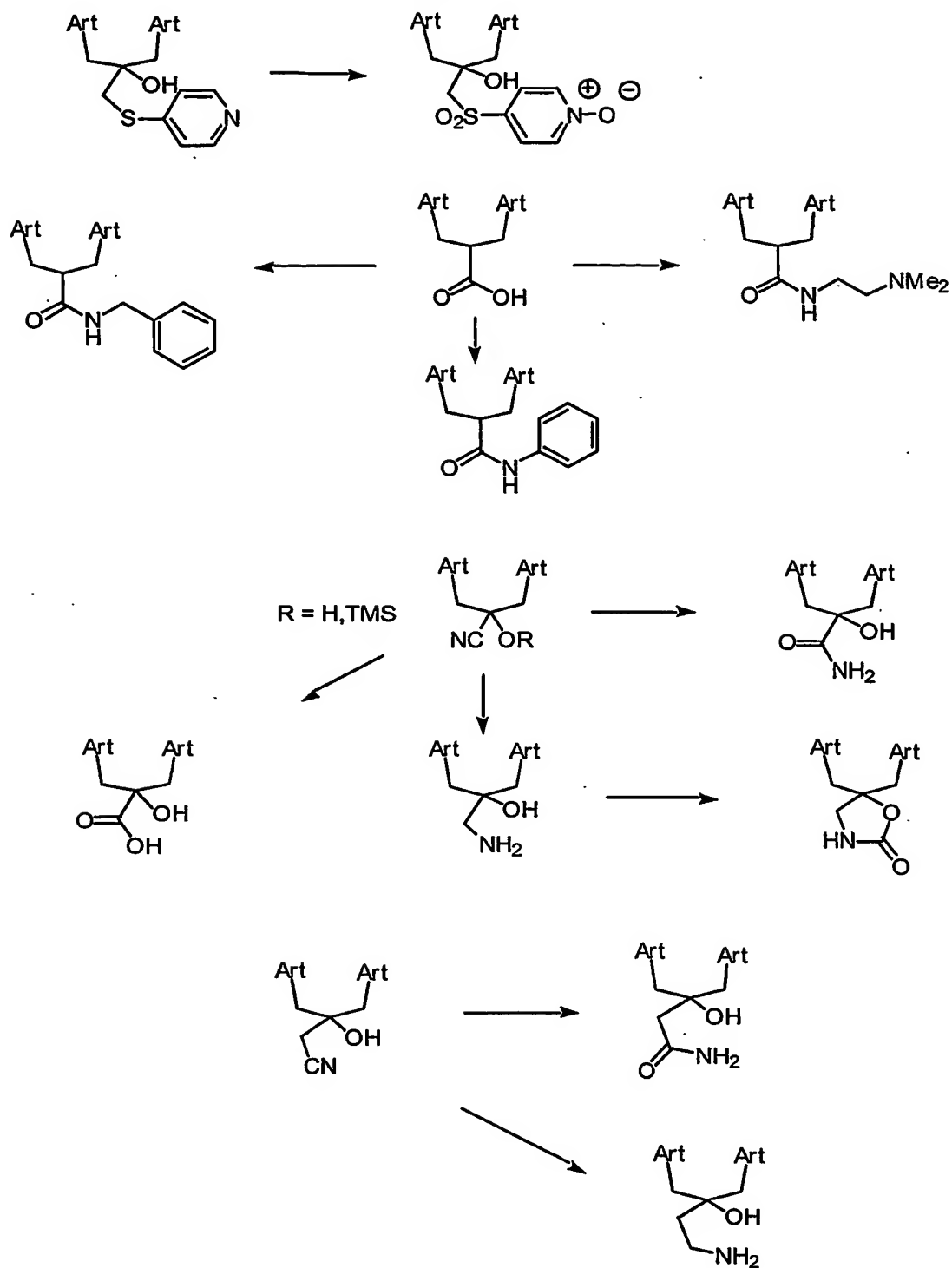
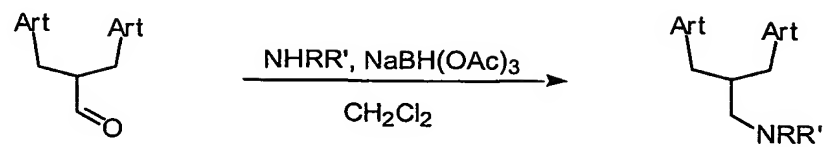


FIGURE 52



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R = H, alkyl, aryl, etc.
R' = H, alkyl, aryl, etc.

FIGURE 53

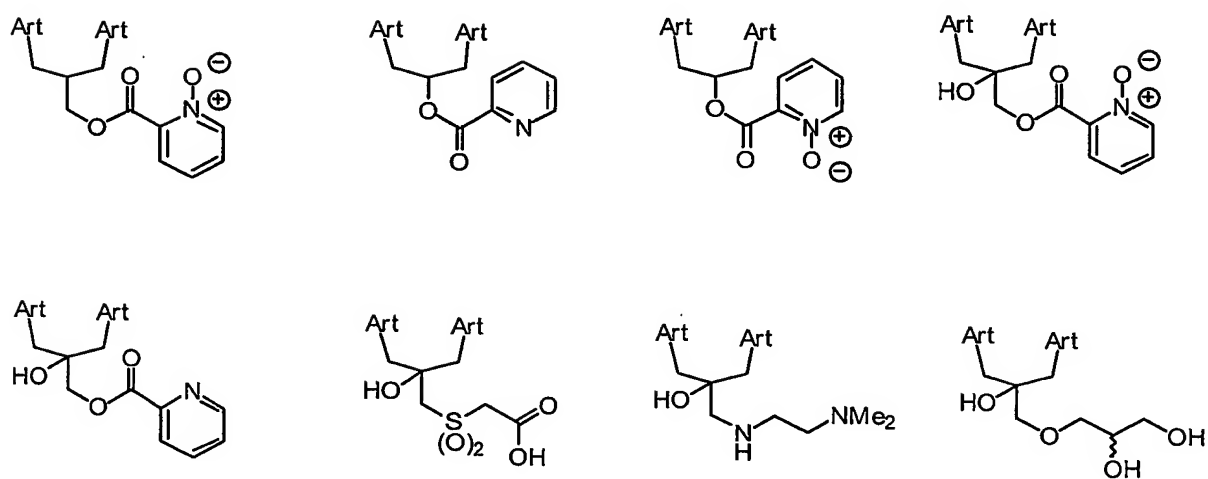


FIGURE 54

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